A Reactive Global Optimization Approach to the Catalytic Reduction of CO₂ by Ni supported nanoclusters

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We discuss a predictive computational methodology that allows one to study the catalytic activity of small metal nanoclusters supported on oxide surfaces¹. This methodology is based on two steps. In the first step, a reactive global minimization (RGO) search is conducted, in which first-principles (i.e., density-functional) methods are used to explore the potential energy surface of (both bare metal clusters and) combined supported-metalcluster+reactant-molecules systems. In a second step, the results of the RGO search are combined with statistical mechanics techniques to simulate of the catalytic behavior of such systems. The computational models necessary to achieve the required compromise between accuracy and efficiency for supported particles are also discussed. The approach is first applied to exploring the structure of bare supported Ni clusters supported on MgO(100), where the MgO(100) surface is taken as an effective model of a simple ionic oxide substrate, along the lines of preliminary explorations². By introducing hydrogen species into the RGO approach, a thermodynamic phase diagram for the hydrogenation of supported Ni particles is further derived. As the core of the work, the real catalytic process is finally tackled: both hydrogen and CO₂ species are simultaneously introduced as ligands of the Ni clusters, and CO₂ reduction by Ni nanoparticles is studied as a function of temperature and chemical potential of the involved species, at the same time illustrating the use of the basic concepts and the power of the approach. It is found that subnanometer Ni supported clusters in proper reaction conditions can act as affective catalysts for CO₂ reduction by H₂. Furthermore, the relevance of these developments to the general field of heterogeneous nanocatalysis (including photo- and electro-catalysis) is also discussed.

References:

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