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An analysis of chemical particles related to the sulfur cycle and the meteorological effects of sulfate aerosols over Europe and the Mediterranean region

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An Analysis of Chemical Particles related to the Sulfur Cycle and the Meteorological Effects of Sulfate Aerosols over Europe and the **Mediterranean Region**

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Outline

- **◎ PRECIS & Physical Parameters**
- Why the Mediterranean Basin?
- The Sulfur Cycle & Aerosols
- Aerosol Impacts
- Sulfur Cycle Validation
- Data Analysis
- Atmospheric Sulfur Chemistry
- Conclusions

PRECIS & Physical Parameters

- ▶ The PRECIS (Providing REgional Climates for Impact Studies)
	- RCM is driven by the GCM HadAM3P. (Gordon et al, 2000)
- ▶ Clouds & precipitation
- Radiation includes the seasonal and diurnal cycles of solar radiation
- ▶ Land Surface characteristics are prescribed according to climatological surface types
- ▶ Surface exchanges
	- Land surface scheme is MOSES (Met Office Surface Exchange Scheme). (Cox et al, 1999)

Why the Mediterranean Basin?

- ▶ Enclosed by 3 major continents.
- Surrounded almost entirely by mountains.
- ▶ Very unique and sensitive to climate changes.
- Simulation details
	- GCM-HadAM3P-150km
	- PRECIS (v 1.7.2) used
	- 1960-1990 (1-year spin up)
	- Resolution: 0.44° x 0.44°
	- 100 cells x (50 km x 50 km)
	- 57° N-18° N, 16° W-46° E

The Sulfur Cycle & Aerosols

- The Sulfur Cycle is based 5 variables:
	- SO₂, DMS and 3 modes of SO_4^{2-} (1 dissolved & 2 free modes)
- Model simulates transport of above, via horizontal and vertical advection, convection and turbulent mixing
- Aerosols act as CCN

Increase in aerosol causes climate impacts

Aerosol Impacts

- With increasing aerosol activity:
- ▶ Direct: Scattering effect
	- Decreased: SW, Skin T
- ▶ Indirect: Cloud albedo (Twomey) effect
	- Decreased: Diurnal Temperature Range (DTR)
- ▶ Indirect: Cloud lifetime effect
	- Decreased: Precipitation, DTR

Aerosol Impacts within the PRECIS output:

- ▶ Studied through correlation between aerosolsensitive parameters.
- Direct effect:
	- Correlations between 0.60-0.82
- Indirect effects:
	- Correlations between 0.18-0.82
	- Complex interactions
	- Cannot isolate Cloud albedo effect from the Twomey

Net Downward SW Flux (ESRL)

Net Downward SW Flux (PRECIS)

Statistics of Parameters

- Bias of Conv PPN, DTR, Net LW, Net SW, RH:
	- Suggest SO_4^2 over-prediction
- Bias of Surface Temperature:
	- Possible result of Net LW over-prediction

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Atmospheric Sulfur Chemistry

- **RECIS**
	- STOCHEM Model
		- UK Meteorological Office Global Three-Dimensional Lagrangian
- ▶ Bacterial Reduction of Sulfate: SO_4^{2-} + H⁺ + 2CH₂O \rightarrow HS⁻ + 2H₂O + 2CO₂ ▶ Gaseous Oxidation of Sulfide: (Andreae et al. 2008) $2H_2S + 3O_2 \xrightarrow{H_0} 2SO_2 + 2H_2O$ ▶ Oxidation of DMS: (Norris, 2003)

 $CH_3SCH_3 + HO^{\bullet} \rightarrow SO_2$ (unbalanced)

Atmospheric Sulfur Chemistry

- Ammonic Oxidation of Sulfur Dioxide: $NH_3 + SO_2 + H_2O \rightarrow NH_4^+ + HSO_3^-$ Gaseous Oxidation of Sulfur Dioxide: (Harrison, 1999) $HO\cdot + SO_2 \rightarrow HSO_3\cdot \stackrel{02}{\longrightarrow} SO_3 + HO_2\cdot \stackrel{H20}{\longrightarrow} H_2SO_4$
- Aqueous oxidation Sulfur (IV) compounds:

 ${O} + {S} \rightarrow 2H^+ + SO₄²$ (unbalanced)

$$
\{O\} = H_2O_2, \text{ HO-}, O_3, O_2
$$

• $\{S\} = SO_2$, HSO_3^- , SO_3^{2-}

Aqueous Oxidation of SO_2 via H_2O_2 : $SO_2 + H_2O_2 \rightarrow SO_3 + H_2O \rightarrow H_2SO_4$

Chemistry Module

- PRECIS yearly average concentrations:
	- HO• constant.
		- \cdot HO \cdot causes no variation in DMS and SO₂ oxidation
- ▶ Using Pearson Correlation:

WWW

- \circ SO₂ is correlated to dissolved SO₄²⁻
- $\, \circ \,$ DMS is negatively correlated to dissolved SO $_4{}^{2-}$
- H2O2 is negatively correlated to Aitken & accumulation SO_4^2 ⁻
- $\, \circ \,$ Aitken is correlated to accumulation SO $_4{}^{2-}$
- $\, \circ \,$ Accumulation is correlated to dissolved SO $_4{}^{2-}$

[All correlations are statistically significant (p<0.01), except the first, which is $p < 0.05$]

Conclusions

- Sulfur Cycle Validation:
	- Direct: High evidence
	- Indirect: Varying evidence
- Results suggest problems with the Sulfur cycle:
	- Possible over-estimation of SO_4^2 concentrations
- Correlation of SO_4^2 ⁻ and H_2O_2 shows the SO_2 to H_2O_2 reaction is the most important within PRECIS
- Correlations show evidence of conversion between:
	- Aitken and accumulation SO $_4$ ²⁻;
	- Accumulation and dissolved SO_4^{2-} ;
	- DMS and dissolved SO_4^{2-} ;

WWW

- Negative correlation of DMS not yet understood
- SO₂ and dissolved SO₄²⁻
- More detailed chemical analysis is needed within PRECIS.

Further Work

References

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Thank you for your attention

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What is PRECIS?

- It is an Atmospheric and LSM of limited area.
- The model is described in 3 units:
	- The dynamics
	- The physical parameterizations
	- The Sulfur cycle

Boundary Conditions

- Model requires surface boundary conditions from surface temperatures and ice extents.
- Dynamical atmospheric information at the boundary of the model's domain are obtained from the LBCs which include
	- standard atmospheric variables of surface pressure
	- horizontal wind components

MARINES

- atmospheric temperature and humidity measures
- ▶ LBCs are updated every 6 hours of the model simulation time.
- Surface boundary conditions are updated every day.

Sulfate Abundance:

- 1. Aitken
- 2. Accumulation
- 3. Dissolved

Aerosol-sensitive parameters

▶ Direct Effect

- Strong correlations
- Expected relationships
- Strong evidence of effect

Aerosol-sensitive parameters

Indirect Effects:

- Low detectability of effect
- Expected due to complex interactions
- ▶ "Incorrect" correlations:
	- Model errors?

Consider a cloud with dissolved aerosol Influenced by the indirect effect (↑ CLW/thickness/albedo)

SO4 aerosol: dissolved mode - 1965

Cloud Liquid Water - 1965

Atmospheric Sulfur Chemistry

Atmospheric Sulfur Compounds and Sources:

- $\,\circ\,$ Sulfates (SO $_4$ ²⁻): sea spray (salts), dust (rocks)
- \circ Sulfur dioxide (SO₂): volcanoes, combustion
- Hydrogen Sulfide (H2S): bacterial reduction
- Dimethyl Sulfide (DMS): marine phytoplankton
- \triangleright SO₂ is the most abundant.

Atmospheric Sulfur Chemistry

 Aqueous oxidation Sulfur (IV) compounds: ${O} + {S} \rightarrow 2H^+ + SO_4^2$ (unbalanced) \circ {O} = H₂O₂, HO•, O₃, O₂ $\text{S} = SO_2$, HSO_3^- , SO_3^{2-}

$$
SO_{2} + H_{2}O_{2} \rightarrow SO_{3} + H_{2}O \rightarrow H_{2}SO_{4}
$$

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$$
SO_{2} + O_{3} \rightarrow SO_{3} + O_{2} \rightarrow H_{2}SO_{4} \text{ (unbalanced)}
$$

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$$
2SO_{2} + O_{2} + 2H_{2}O \rightarrow 2H_{2}SO_{4}
$$

\n
$$
HSO_{3}^{-} + H_{2}O_{2} \rightarrow HSO_{4}^{-} \text{ (unbalanced)}
$$

\n
$$
SO_{3}^{2-} + O_{3} \rightarrow SO_{4}^{2-} + O_{2}
$$

Chemical Correlation

Correlations

*. Correlation is significant at the 0.05 level (2-tailed).

**. Correlation is significant at the 0.01 level (2-tailed).

Future Scenarios

SO4 Aitken: 1981

SO4 Aitken: 2081 B2

Future Scenarios

2070-2100 B2 Aitken Annual Anomaly Field

Future Scenarios

2070-2100 B2 SW Annual Anomaly Field

