



**The Abdus Salam
International Centre for Theoretical Physics**



2332

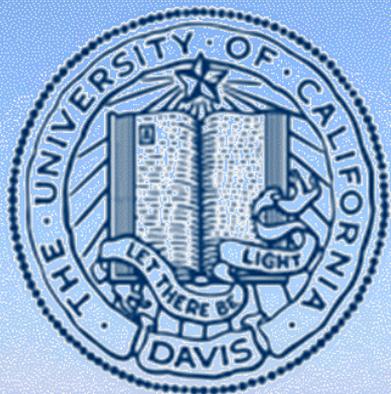
**School on Synchrotron and FEL Based Methods and their Multi-Disciplinary
Applications**

19 - 30 March 2012

Additional information lecture 1

Charles S. Fadley
*Univ. of California Davis & LBNL
USA*

Characterization of Surfaces, Interfaces, and Complex Materials with Core and Valence Photoemission



Chuck Fadley
Dept. of Physics, UC Davis
and



Materials Sciences Division

Lawrence Berkeley National Laboratory

Supported by: U.S. Dept. of Energy
Humboldt Foundation & Helmholtz Association
Jülich Research Center

School on Synchrotron Radiation and FEL Based
Methods and their Multi-Disciplinary Applications
March 26, 2012

Photoelectron spectroscopy = Photoemission: The Basic Elements

Friday, 23 March 2012

Room: Adriatico Guest House Giambiasi Lecture Hall

09:00	Surfaces, Interfaces & Photoelectron Spectroscopy 01h30'	A. Goldoni Sincrotrone Trieste
10:30	BREAK 30'	
11:00	ARPES 01h30'	A. Goldoni Sincrotrone Trieste
12:30	LUNCH BREAK 01h30'	
14:00	Inelastic X-ray Scattering: principles and application 01h30'	F. Bencivenga Sincrotrone Trieste
15:30	BREAK 30'	
16:00	Electronic structure and atomistic computations for the interpretation of synchrotron experiments 01h00'	N. Binggeli ICTP
17:00		

Monday, 26 March 2012

Room: Adriatico Guest House Giambiasi Lecture Hall

09:00	Photoelectron Diffraction and holography 01h30'	Charles S. Fadley Univ. of California Davis & LBNL (USA)
10:30	BREAK 30'	
11:00	Core Level shifts, splitting and dichroism 01h30'	Charles S. Fadley Univ. of California Davis & LBNL (USA)
12:30	LUNCH BREAK 01h30'	
14:00	Low density matter: spectroscopy and scattering 01h30'	M. Simon University Pierre et Marie Curie, France
15:30	BREAK 30'	
16:00	Low-density matter: ultrafast dynamics 01h00'	M. Simon University Pierre et Marie Curie, France
17:00	Theoretical spectroscopy of gas-phase systems 01h30'	P. Decleva University of Trieste
18:30		

Link to archive of longer set of lectures in 2010, plus guides to use of SESSA and EDAC simulation programs:

<http://www.yousendit.com/download/M3Btb2VDeFU3N0RFdzhUQw>

General references on various aspects of photoelectron spectroscopy, diffraction, holography (available at School website and/or Fadley group website: <http://www.physics.ucdavis.edu/fadleygroup/> :

Paper [1] "Basic Concepts of X-ray Photoelectron Spectroscopy", C.S.F, in Electron Spectroscopy, Theory, Techniques, and Applications, Brundle and Baker, Eds. (Pergamon Press, 1978) Vol. II, Ch. 1.

Paper [2] "Atomic-Level Characterization of Materials with Core- and Valence-Level Photoemission: Basic Phenomena and Future Directions", C.S. Fadley, Surf. Interface Anal. 2008, 40, 1579–1605.

**Paper [3] "X-ray photoelectron spectroscopy: Progress and perspectives"
C.S. Fadley, Journal of Electron Spectroscopy and Related Phenomena 178–179 (2010) 2–32**

Key Reference: "X-ray Data Booklet", Center for X-Ray Optics and the Advanced Light Source, LBNL, January, 2001, available online at: <http://xdb.lbl.gov/>

Additional very useful websites:

**X-ray optical calculations: reflectivities, penetration depths for a variety of mirror/surface geometries—
http://www-cxro.lbl.gov/optical_constants/**

General properties of the elements and their compounds: <http://www.webelements.com>

Simulation of photoelectron and Auger spectra with program SESSA: See "SESSA: A Brief Manual", with download instructions, etc.

Simulation of x-ray photoelectron diffraction (XPD) with program EDAC: See guide example calculation for CO on Fe.

X-RAY DATA BOOKLET

Center for X-ray Optics and Advanced Light Source
Lawrence Berkeley National Laboratory

<http://xdb.lbl.gov/>

- [Introduction](#)
- [X-Ray Properties of Elements](#)
- [Electron Binding Energies](#)
- [X-Ray Energy Emission Energies](#)
- [Fluorescence Yields for K and L Shells](#)
- [Principal Auger Electron Energies](#)
- [Subshell Photoionization Cross-Sections](#)
- [Mass Absorption Coefficients](#)
- [Atomic Scattering Factors](#)
- [Energy Levels of Few Electron Ions](#)
- [Periodic Table of X-Ray Properties](#)
- [Synchrotron Radiation](#)
- [Characteristics of Synchrotron Radiation](#)
- [History of X-rays and Synchrotron Radiation](#)
- [Synchrotron Facilities](#)
- [Scattering Processes](#)
- [Scattering of X-rays from Electrons and Atoms](#)
- [Low-Energy Electron Ranges in Matter](#)
- [Optics and Detectors](#)
- [Crystal and Multilayer Elements](#)
- [Specular Reflectivities for Grazing-Incidence Mirrors](#)
- [Gratings and Monochromators](#)
- [Zone Plates](#)
- [X-Ray Detectors](#)
- [Miscellaneous](#)
- [Physical Constants](#)
- [Physical Properties of the Elements](#)
- [Electromagnetic Relations](#)
- [Radioactivity and Radiation Protection](#)
- [Useful Formulas](#)

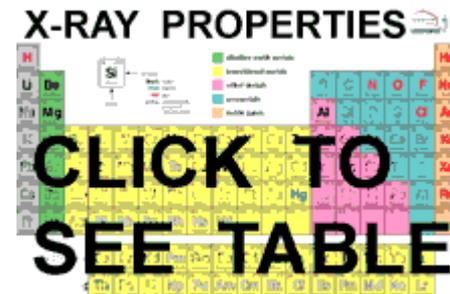
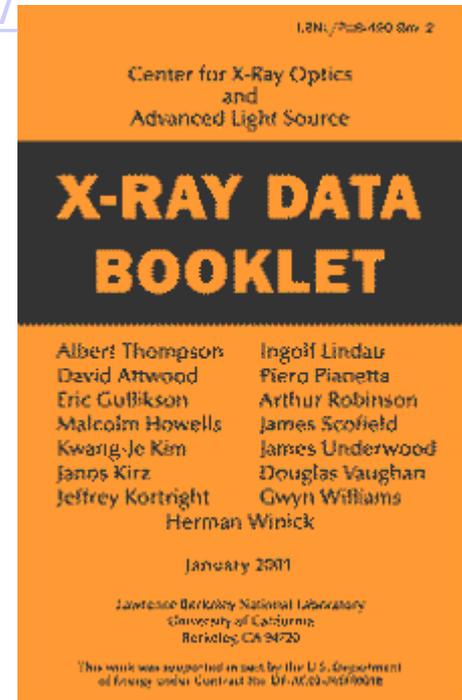


Table 3 Crystal structures of the elements

The data given are at room temperature for the most common form, or at the stated temperature in deg K. For further descriptions of the elements see Wyckoff, Vol. 1, Chap. 2. Structures labeled complex are described there.

H¹ 4K hcp 3.75 6.12																He⁴ 2K hcp 3.57 5.83													
Li 78K bcc 3.491																Be hcp 2.27 3.59		B rhomb. 3.567		C diamond 5.66 (N ₂)		N 20K cubic (O ₂)		O complex (Cl ₂)		F fcc 4.46		Ne 4K fcc 5.31	
Na 5K bcc 4.225																Mg hcp 3.21 5.21		Al fcc 4.05		Si diamond 5.430		P complex		S complex		Cl complex (Br ₂)		Ar 4K fcc 5.64	
←————— Crystal structure —————→ ←————— a lattice parameter, in Å —————→ ←————— c lattice parameter, in Å —————→																													
K 5K bcc 5.225		Ca fcc 5.58	Sc hcp 3.31 5.27	Ti hcp 2.95 4.68	V bcc 3.03	Cr bcc 2.88	Mn cubic complex	Fe bcc 2.87	Co hcp 2.51 4.07	Ni fcc 3.52	Cu fcc 3.61	Zn hcp 2.66 4.95	Ga complex	Ge diamond 5.658	As rhomb.	Se hex. chains	Br complex (Br ₂)	Kr 4K fcc 5.64											
Rb 5K bcc 5.585		Sr fcc 6.08	Y hcp 3.65 5.73	Zr hcp 3.23 5.15	Nb bcc 3.30	Mo bcc 3.15	Tc hcp 2.74 4.40	Ru hcp 2.71 4.28	Rh fcc 3.80	Pd fcc 3.89	Ag fcc 4.09	Cd hcp 2.98 5.62	In tetr. 3.25 4.95	Sn (α) diamond 6.49	Sb rhomb.	Te hex. chains	I complex (I ₂)	Xe 4K fcc 6.13											
Cs 5K bcc 6.045		Ba bcc 5.02	La hex. 3.77 ABAC	Hf hcp 3.19 5.05	Ta bcc 3.30	W bcc 3.16	Re hcp 2.76 4.46	Os hcp 2.74 4.32	Ir fcc 3.84	Pt fcc 3.92	Au fcc 4.08	Hg rhomb.	Tl hcp 3.46 5.52	Pb fcc 4.95	Bi rhomb.	Po sc 3.34	At —	Rn —											
Fr —	Ra —	Ac fcc 5.31																											
			Ce fcc 5.16	Pr hex. 3.67 ABAC	Nd hex. 3.66	Pm —	Sm complex	Eu bcc 4.58	Gd hcp 3.63 5.78	Tb hcp 3.60 5.70	Dy hcp 3.59 5.65	Ho hcp 3.58 5.62	Er hcp 3.56 5.59	Tm hcp 3.54 5.56	Yb fcc 5.48	Lu hcp 3.50 5.55													
			Th fcc 5.08	Pa tetr. 3.92 3.24	U complex	Np complex	Pu complex	Am hex. 3.64 ABAC	Cm —	Bk —	Cf —	Es —	Fm —	Md —	No —	Lr —													

Table 4 Density and atomic concentration

The data are given at atmospheric pressure and room temperature, or at the stated temperature in deg K. (Crystal modifications as for Table 3.)

H 4K												He 2K																																																																																																																													
0.088												0.205 (at 37 atm)																																																																																																																													
Li 78K	Be	<div style="display: flex; justify-content: space-around; align-items: center;"> <div style="background-color: yellow; padding: 5px; border: 1px solid black;"> Atomic radius $= r_{MT}$ $= 0.5 \text{ n-n dist.}$ </div> <div style="background-color: yellow; padding: 5px; border: 1px solid black;"> Average surface density $= \rho_S \approx (\rho_V)^{2/3}$ </div> </div>										B	C	N 20K	O	F	Ne 4K																																																																																																																								
0.542	1.82											2.47	3.516	1.03			1.51																																																																																																																								
4.700	12.1	13.0	17.6				4.36																																																																																																																																		
3.023	2.22		1.54			1.44	3.16																																																																																																																																		
Na 5K	Mg	<div style="display: flex; justify-content: space-between; margin-top: 10px;"> <div style="width: 30%; text-align: center;">←</div> <div style="width: 40%; text-align: center;">Density in g cm^{-3} (10^3kg m^{-3})</div> <div style="width: 30%; text-align: center;">→</div> </div> <div style="display: flex; justify-content: space-between; margin-top: 5px;"> <div style="width: 30%; text-align: center;">←</div> <div style="width: 40%; text-align: center;">Concentration in 10^{22}cm^{-3} (10^{28}m^{-3})</div> <div style="width: 30%; text-align: center;">→</div> </div> <div style="display: flex; justify-content: space-between; margin-top: 5px;"> <div style="width: 30%; text-align: center;">←</div> <div style="width: 40%; text-align: center;">Nearest-neighbor distance, in Å (10^{-10}m)</div> <div style="width: 30%; text-align: center;">→</div> </div>										Al	Si	P	S	Cl 93K	Ar 4K																																																																																																																								
1.013	1.74											2.70	2.33			2.03	1.77																																																																																																																								
2.652	4.30											6.02	5.00				2.66																																																																																																																								
3.659	3.20	2.86	2.35			2.02	3.76																																																																																																																																		
K 5K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br 123K	Kr 4K																																																																																																																								
0.910	1.53	2.99	4.51	6.09	7.19	7.47	7.87	8.9	8.91	8.93	7.13	5.91	5.32	5.77	4.81	4.05	3.09																																																																																																																								
1.402	2.30	4.27	5.66	7.22	8.33	8.18	8.50	8.97	9.14	8.45	6.55	5.10	4.42	4.65	3.67	2.36	2.17																																																																																																																								
4.525	3.95	3.25	2.89	2.62	2.50	2.24	2.48	2.50	2.49	2.56	2.66	2.44	2.45	3.16	2.32		4.00																																																																																																																								
Rb 5K	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe 4K																																																																																																																								
1.629	2.58	4.48	6.51	8.58	10.22	11.50	12.36	12.42	12.00	10.50	8.65	7.29	5.76	6.69	6.25	4.95	3.78																																																																																																																								
1.148	1.78	3.02	4.29	5.56	6.42	7.04	7.36	7.26	6.80	5.85	4.64	3.83	2.91	3.31	2.94	2.36	1.64																																																																																																																								
4.837	4.30	3.55	3.17	2.86	2.72	2.71	2.65	2.69	2.75	2.89	2.98	3.25	2.81	2.91	2.86	3.54	4.34																																																																																																																								
Cs 5K	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg 227	Tl	Pb	Bi	Po	At	Rn																																																																																																																								
1.997	3.59	6.17	13.20	16.66	19.25	21.03	22.58	22.55	21.47	19.28	14.26	11.87	11.34	9.80	9.31																																																																																																																										
0.905	1.60	2.70	4.52	5.55	6.30	6.80	7.14	7.06	6.62	5.90	4.26	3.50	3.30	2.82	2.67	—	—																																																																																																																								
5.235	4.35	3.73	3.13	2.86	2.74	2.74	2.68	2.71	2.77	2.88	3.01	3.46	3.50	3.07	3.34																																																																																																																										
Fr	Ra	Ac	<table border="1" style="width: 100%; border-collapse: collapse;"> <tr> <td></td> <td>Ce</td> <td>Pr</td> <td>Nd</td> <td>Pm</td> <td>Sm</td> <td>Eu</td> <td>Gd</td> <td>Tb</td> <td>Dy</td> <td>Ho</td> <td>Er</td> <td>Tm</td> <td>Yb</td> <td>Lu</td> </tr> <tr> <td></td> <td>6.77</td> <td>6.78</td> <td>7.00</td> <td></td> <td>7.54</td> <td>5.25</td> <td>7.89</td> <td>8.27</td> <td>8.53</td> <td>8.80</td> <td>9.04</td> <td>9.32</td> <td>6.97</td> <td>9.84</td> </tr> <tr> <td></td> <td>2.91</td> <td>2.92</td> <td>2.93</td> <td>—</td> <td>3.03</td> <td>2.04</td> <td>3.02</td> <td>3.22</td> <td>3.17</td> <td>3.22</td> <td>3.26</td> <td>3.32</td> <td>3.02</td> <td>3.39</td> </tr> <tr> <td></td> <td>3.65</td> <td>3.63</td> <td>3.66</td> <td></td> <td>3.59</td> <td>3.96</td> <td>3.58</td> <td>3.52</td> <td>3.51</td> <td>3.49</td> <td>3.47</td> <td>3.54</td> <td>3.88</td> <td>3.43</td> </tr> <tr> <td></td> <td>Th</td> <td>Pa</td> <td>U</td> <td>Np</td> <td>Pu</td> <td>Am</td> <td>Cm</td> <td>Bk</td> <td>Cf</td> <td>Es</td> <td>Fm</td> <td>Md</td> <td>No</td> <td>Lr</td> </tr> <tr> <td></td> <td>11.72</td> <td>15.37</td> <td>19.05</td> <td>20.45</td> <td>19.81</td> <td>11.87</td> <td></td> <td></td> <td></td> <td></td> <td></td> <td></td> <td></td> <td></td> </tr> <tr> <td></td> <td>3.04</td> <td>4.01</td> <td>4.80</td> <td>5.20</td> <td>4.26</td> <td>2.96</td> <td>—</td> <td>—</td> <td>—</td> <td>—</td> <td>—</td> <td>—</td> <td>—</td> <td>—</td> </tr> <tr> <td></td> <td>3.60</td> <td>3.21</td> <td>2.75</td> <td>2.62</td> <td>3.1</td> <td>3.61</td> <td></td> <td></td> <td></td> <td></td> <td></td> <td></td> <td></td> <td></td> </tr> </table>																Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu		6.77	6.78	7.00		7.54	5.25	7.89	8.27	8.53	8.80	9.04	9.32	6.97	9.84		2.91	2.92	2.93	—	3.03	2.04	3.02	3.22	3.17	3.22	3.26	3.32	3.02	3.39		3.65	3.63	3.66		3.59	3.96	3.58	3.52	3.51	3.49	3.47	3.54	3.88	3.43		Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr		11.72	15.37	19.05	20.45	19.81	11.87										3.04	4.01	4.80	5.20	4.26	2.96	—	—	—	—	—	—	—	—		3.60	3.21	2.75	2.62	3.1	3.61								
	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu																																																																																																																											
	6.77	6.78	7.00		7.54	5.25	7.89	8.27	8.53	8.80	9.04	9.32	6.97	9.84																																																																																																																											
	2.91	2.92	2.93	—	3.03	2.04	3.02	3.22	3.17	3.22	3.26	3.32	3.02	3.39																																																																																																																											
	3.65	3.63	3.66		3.59	3.96	3.58	3.52	3.51	3.49	3.47	3.54	3.88	3.43																																																																																																																											
	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr																																																																																																																											
	11.72	15.37	19.05	20.45	19.81	11.87																																																																																																																																			
	3.04	4.01	4.80	5.20	4.26	2.96	—	—	—	—	—	—	—	—																																																																																																																											
	3.60	3.21	2.75	2.62	3.1	3.61																																																																																																																																			
		10.07																																																																																																																																							
		2.66																																																																																																																																							
		3.76																																																																																																																																							

Table 1 Debye temperature and thermal conductivity^a

Li	Be											B	C	N	O	F	Ne
344	1440												2230				75
0.85	2.00											0.27	1.29				
Na	Mg											Al	Si	P	S	Cl	Ar
158	400	Low temperature limit of θ , in Kelvin										428	645				92
1.41	1.56	Thermal conductivity at 300 K, in $\text{W cm}^{-1}\text{K}^{-1}$										2.37	1.48				
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
91	230	360	420	380	630	410	470	445	450	343	327	320	374	282	90		72
1.02		0.16	0.22	0.31	0.94	0.08	0.80	1.00	0.91	4.01	1.16	0.41	0.60	0.50	0.02		
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn _w	Sb	Te	I	Xe
56	147	280	291	275	450		600	480	274	225	209	108	200	211	153		64
0.58		0.17	0.23	0.54	1.38	0.51	1.17	1.50	0.72	4.29	0.97	0.82	0.67	0.24	0.02		
Cs	Ba	La β	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
38	110	142	252	240	400	430	500	420	240	165	71.9	78.5	105	119			
0.36		0.14	0.23	0.58	1.74	0.48	0.88	1.47	0.72	3.17		0.46	0.35	0.08			
Fr	Ra	Ac															
			Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu	
									200		210				120	210	
			0.11	0.12	0.16		0.13		0.11	0.11	0.11	0.16	0.14	0.17	0.35	0.16	
			Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr	
			163		207												
			0.54		0.28	0.06	0.07										

^aMost of the θ values were supplied by N. Pearlman; references are given the *A.I.P. Handbook*, 3rd ed; the thermal conductivity values are from R. W. Powell and Y. S. Touloukian, *Science* **181**, 999 (1973).

X-Ray Data Booklet--Section 1.1 ELECTRON BINDING ENERGIES

The energies are given in eV relative to the vacuum level for the rare gases and for H₂, N₂, O₂, F₂, and Cl₂; relative to the Fermi level for the metals; and relative to the top of the valence bands for semiconductors (and insulators).

	Electronic Element configuration	K 1s	L ₁ 2s	L ₂ 2p _{1/2}	L ₃ 2p _{3/2}	M ₁ 3s	M ₂ 3p _{1/2}	M ₃ 3p _{3/2}
1s	1 H	13.6						
1s ²	2 He	24.6*						
1s ² 2s	3 Li	54.7*						
1s ² 2s ²	4 Be	111.5*						
1s ² 2s ² 2p	5 B	188*						
1s ² 2s ² 2p ²	6 C	284.2*						
1s ² 2s ² 2p ³	7 N	409.9*	37.3*	~ 9	~ 9			
1s ² 2s ² 2p ⁴	8 O	543.1*	41.6*	~ 13	~ 13			
1s ² 2s ² 2p ⁵	9 F	696.7*	~ 45	~ 17	~ 17			
1s ² 2s ² 2p ⁶	10 Ne	870.2*	48.5*	21.7*	21.6*			
[Ne] 3s	11 Na	1070.8†	63.5†	30.65	30.81			
[Ne] 3s ²	12 Mg	1303.0†	88.7	49.78	49.50			
[Ne] 3s ² 3p	13 Al	1559.6	117.8	72.95	72.55			
[Ne] 3s ² 3p ²	14 Si	1839	149.7*b	99.82	99.42			
[Ne] 3s ² 3p ³	15 P	2145.5	189*	136*	135*			
[Ne] 3s ² 3p ⁴	16 S	2472	230.9	163.6*	162.5*			
[Ne] 3s ² 3p ⁵	17 Cl	2822.4	270*	202*	200*			
[Ne] 3s ² 3p ⁶	18 Ar	3205.9*	326.3*	250.6†	248.4*	29.3*	15.9*	15.7*
[Ar] 4s	19 K	3608.4*	378.6*	297.3*	294.6*	34.8*	18.3*	18.3*
[Ar] 4s ²	20 Ca	4038.5*	438.4†	349.7†	346.2†	44.3 †	25.4†	25.4†
	21 Sc	4492	498.0*	403.6*	398.7*	51.1*	28.3*	28.3*
	22 Ti	4966	560.9†	460.2†	453.8†	58.7†	32.6†	32.6†

Valence levels

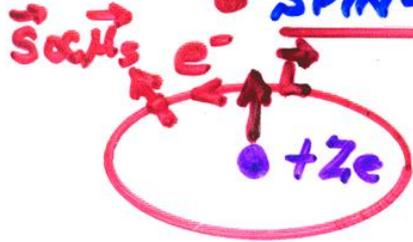
Interpolated,
extrapolated

Missing
valence
B.E.s

Valence levels

Adding relativity to the Hamiltonian—spin-orbit coupling

• SPIN-ORBIT SPLITTING OF LEVELS:



⇒ EFFECTIVE \vec{B} (NUCLEUS AROUND e^-) $\propto \vec{L}$

$$\hat{H}_{s.o.} = \xi(r) \vec{L} \cdot \vec{S}$$

- SPLITS ALL nl LEVELS $2(2l+1)$
 - $nl_j = l + 1/2 \rightarrow 2l+2$
 - $nl_j = l - 1/2 \rightarrow 2l$

• MIXES SPIN + ORBITAL ANGULAR MOM.::

$$\psi_{nljm_j} = C_1 \psi_{nl, m_j - 1/2} \begin{pmatrix} 1 \\ 0 \end{pmatrix} + C_2 \psi_{nl, m_j + 1/2} \begin{pmatrix} 0 \\ 1 \end{pmatrix}$$

\parallel
 $m_s = +1/2$
 \parallel
 \uparrow

\parallel
 $m_s = -1/2$
 \parallel
 \downarrow

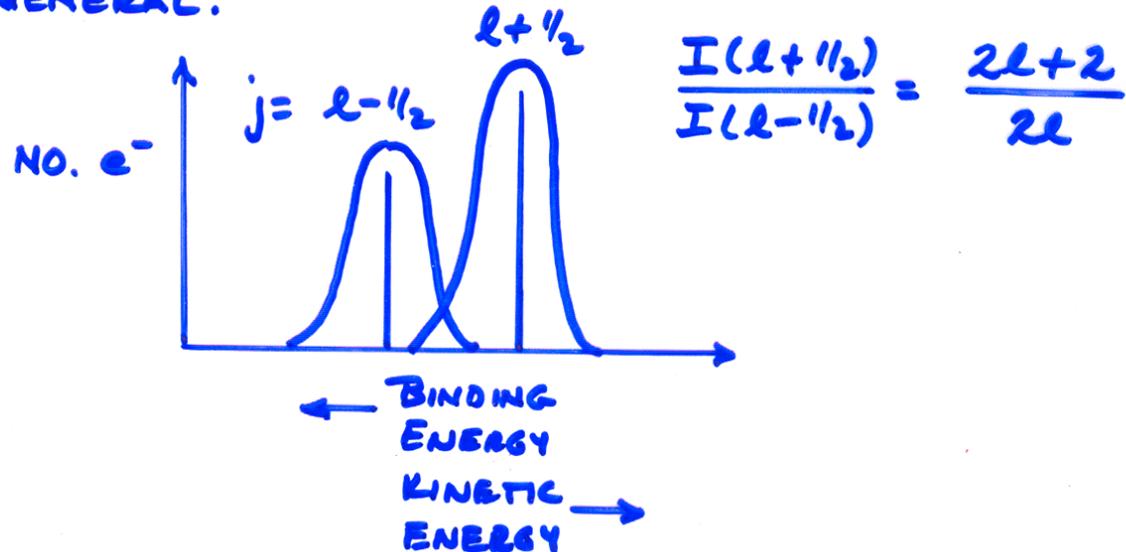
WITH C_1 AND C_2 TABULATED CLEBSCH-GORDAN OR WIGNER $3j$ SYMBOLS

SOME SPIN-ORBIT SPLITTINGS: (IN eV)

$2p^6$ \swarrow \searrow $2p_{1/2}^2$ $2p_{3/2}^4$	$Z = 13$ (Al) 0.4	28 (Ni) 17.8	46 (Pd) 157.0
$3d^{10}$ \swarrow \searrow $3d_{3/2}^4$ $3d_{5/2}^6$	$Z = 30$ (Zn) 0.1	48 (Cd) 6.7	64 (Gd) 32.3
$4f^{14}$ \swarrow \searrow $4f_{5/2}^6$ $4f_{7/2}^8$	$Z = 74$ (W) 2.2	84 (Pb) 7.0	92 (U) 64

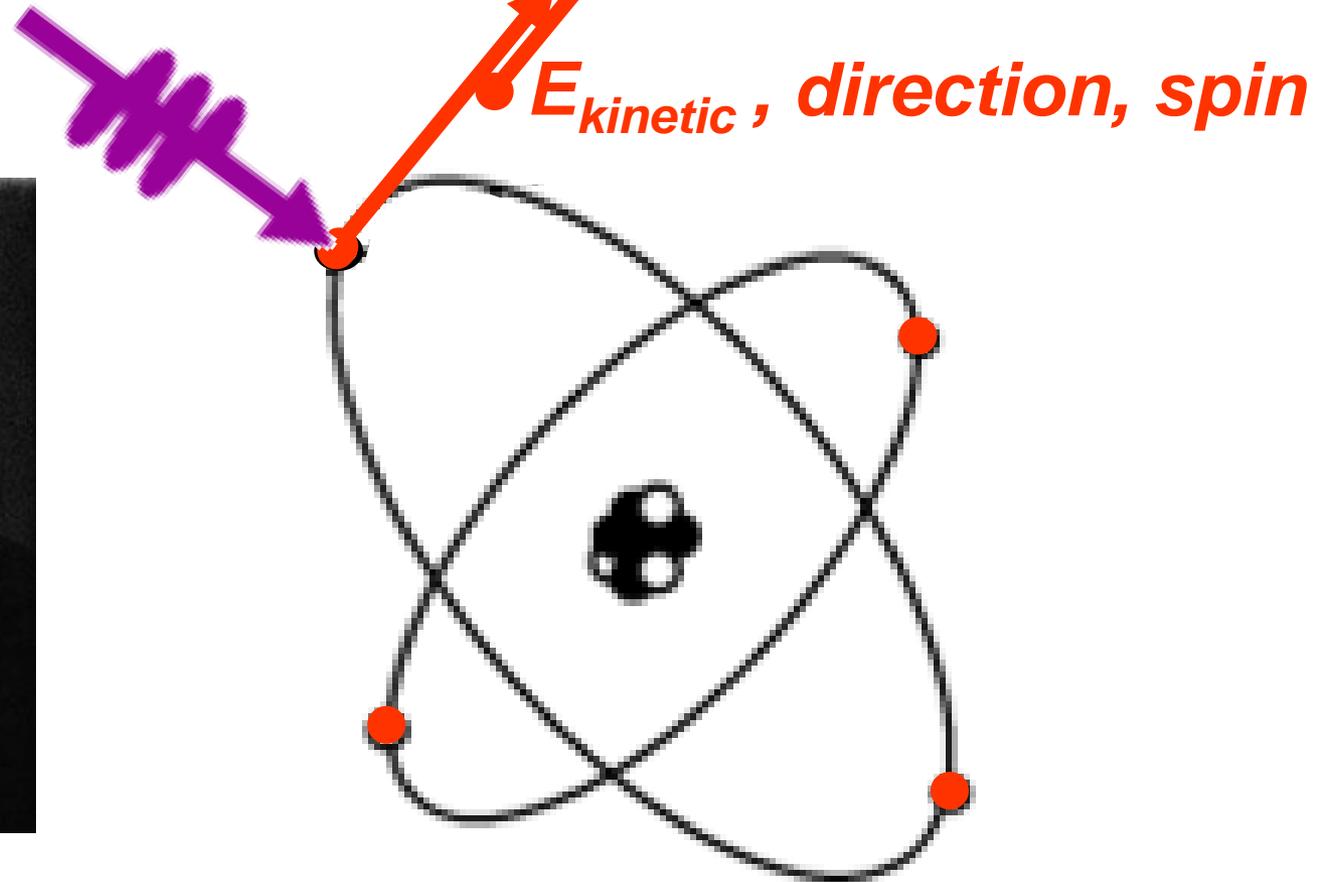
INCREASE WITH Z FOR A GIVEN LEVEL.

IN GENERAL:

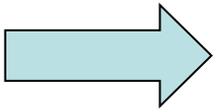


The Photoelectric Effect, Einstein, 1905

Light can behave like a Particle!



$$h\nu = E_{initial} - E_{final} = E_{binding} + E_{kinetic}$$



Basic Concepts and Experiments

Core-Level Photoemission

Intensities and Quantitative Analysis, the 3-Step Model
Varying Surface and Bulk Sensitivity
Chemical shifts
Multiplet Splittings
Electron Screening and Satellite Structure
Magnetic and Non-Magnetic Dichroism
Resonant Photoemission
Photoelectron Diffraction and Holography

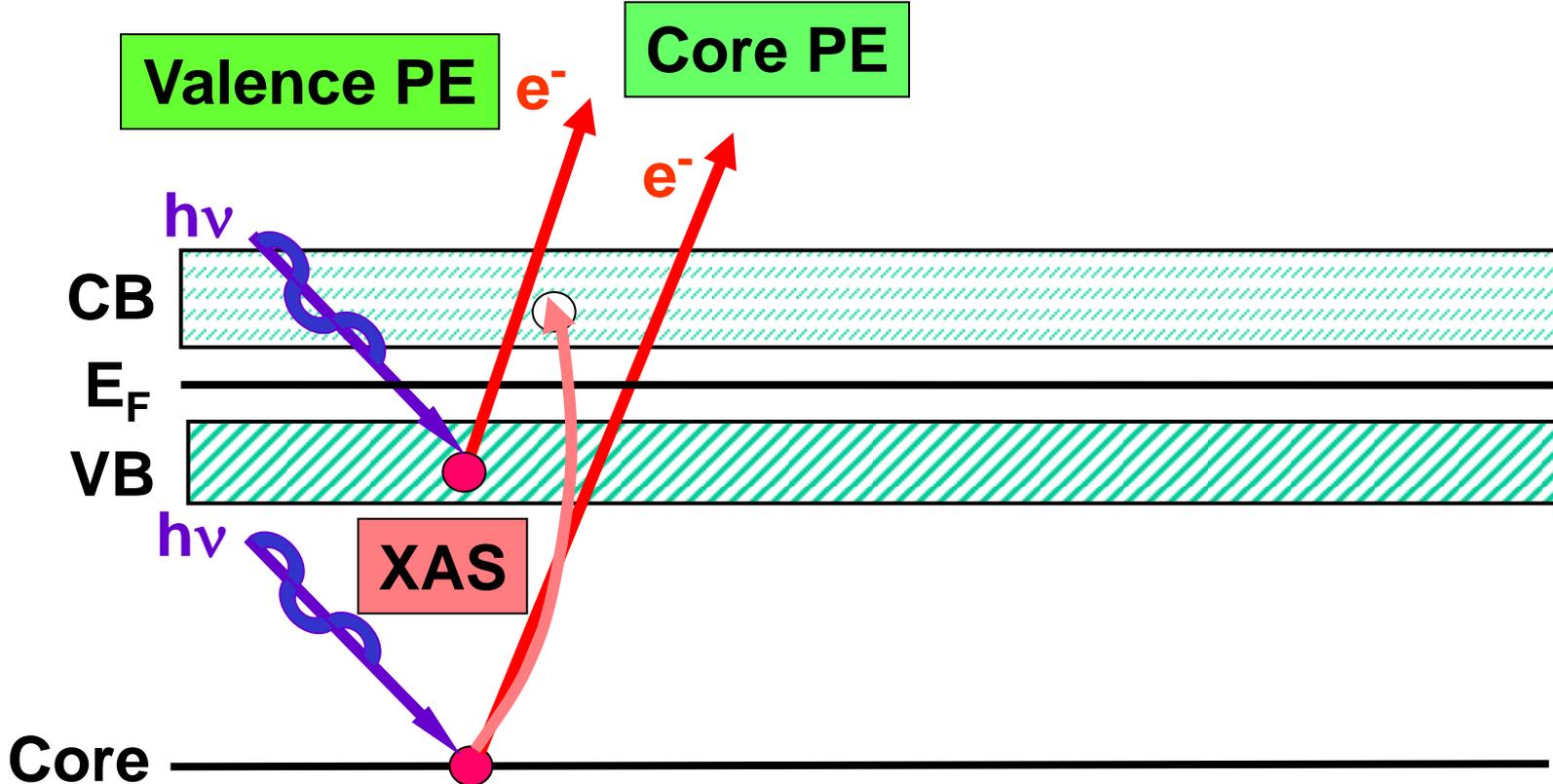
Valence-Level Photoemission

Band-Mapping in the Ultraviolet Photoemission Limit
Densities of States in the X-Ray Photoemission Limit

Some New Directions

Photoemission with Hard X-Rays (throughout lectures)
Photoemission with Standing Wave Excitation
Photoemission with Spatial Resolution/Photoelectron Microscopy
Temporal Resolution
@ Higher Pressures

The Soft X-Ray Spectroscopies



PE = photoemission = photoelectron spectroscopy

XAS = x-ray absorption spectroscopy

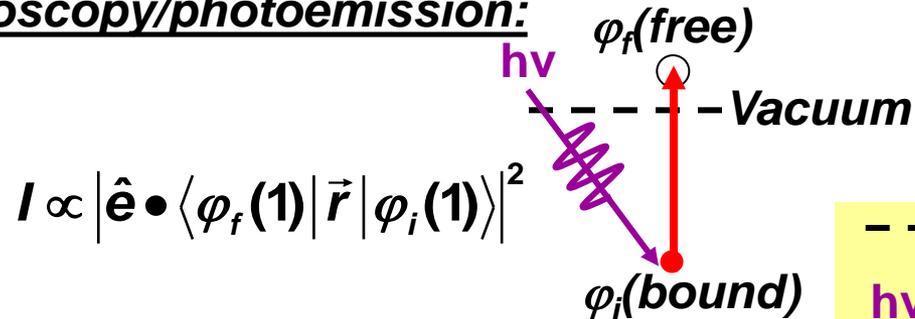
AES = Auger electron spectroscopy

XES = x-ray emission spectroscopy

RIXS = resonant inelastic x-ray scattering / x-ray Raman scatt.

MATRIX ELEMENTS IN THE SOFT X-RAY SPECTROSCOPIES: DIPOLE LIMIT

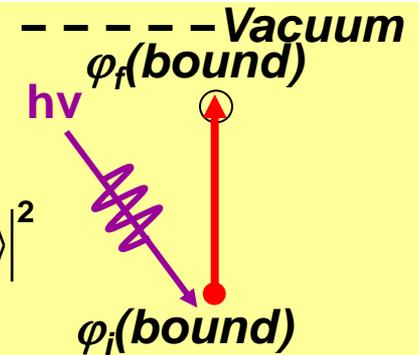
- Photoelectron spectroscopy/photoemission:



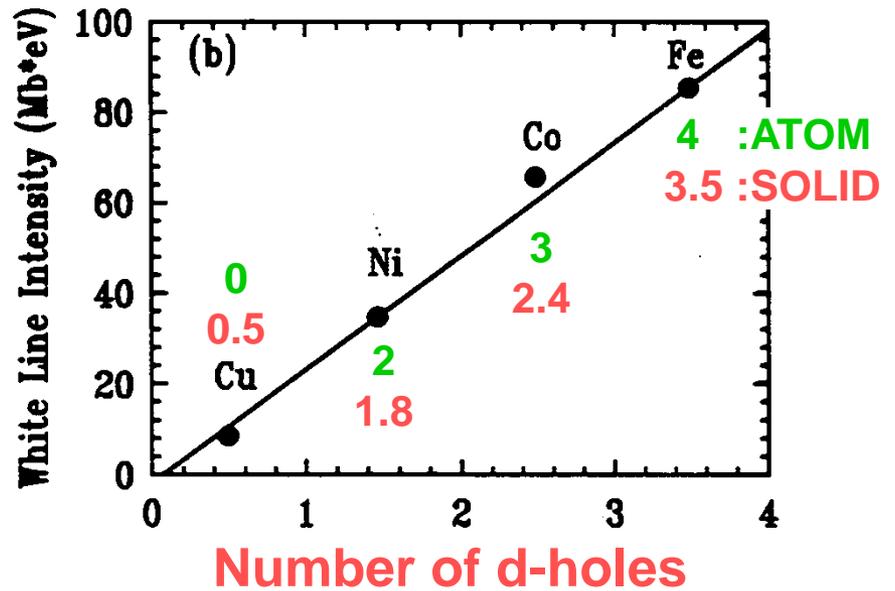
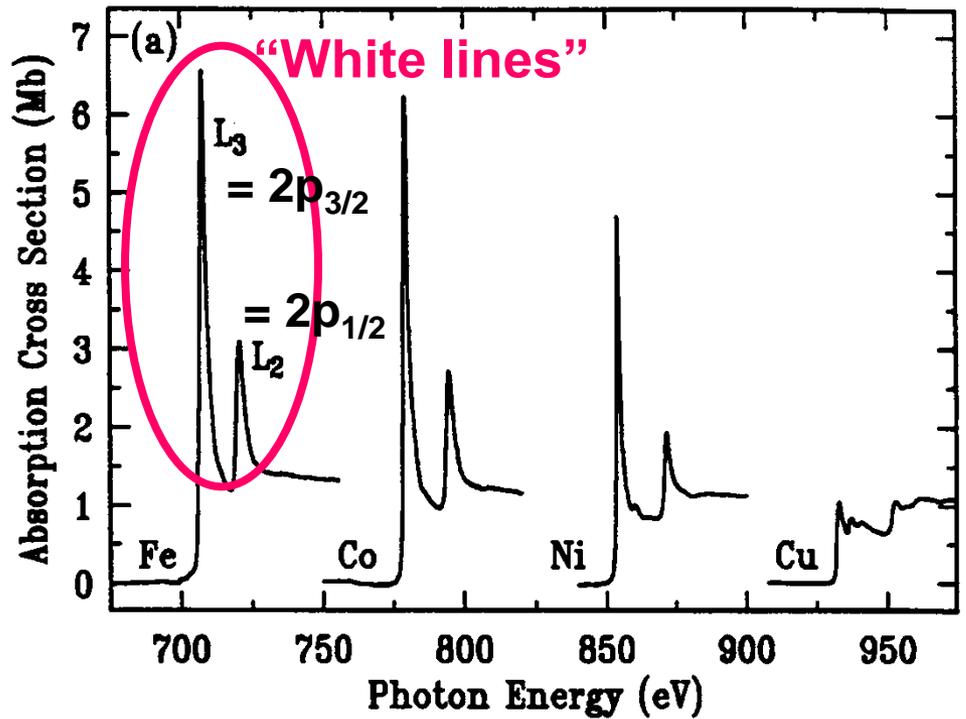
$$I \propto |\hat{\mathbf{e}} \cdot \langle \varphi_f(\mathbf{1}) | \vec{r} | \varphi_i(\mathbf{1}) \rangle|^2$$

- Near-edge x-ray absorption:

$$I \propto |\hat{\mathbf{e}} \cdot \langle \varphi_f(\mathbf{1}) | \vec{r} | \varphi_i(\mathbf{1}) \rangle|^2$$



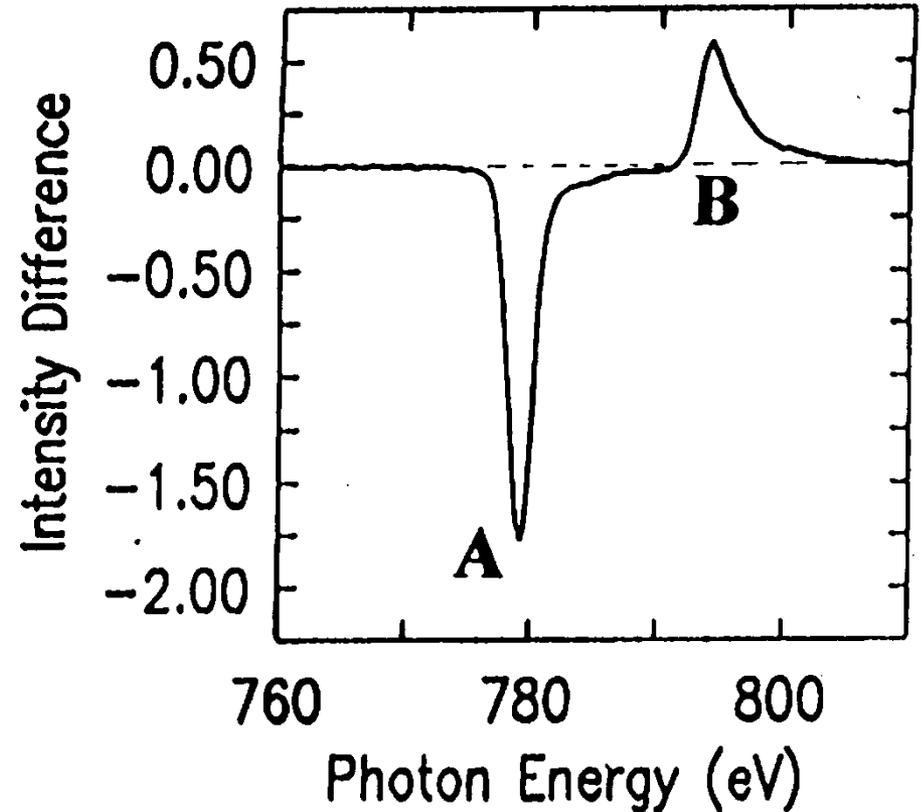
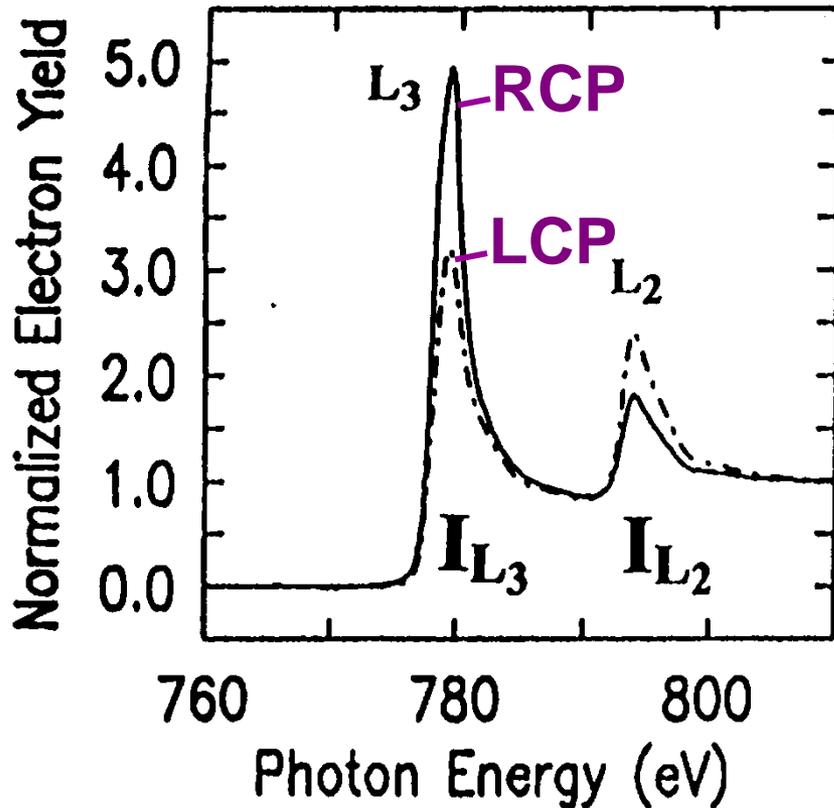
Variation of
Near-Edge X-Ray
Absorption Fine
Structure
(NEXAFS) with Atomic
No. for Some 3d
Transition Metals



J. Stohr, "NEXAFS
Spectroscopy"

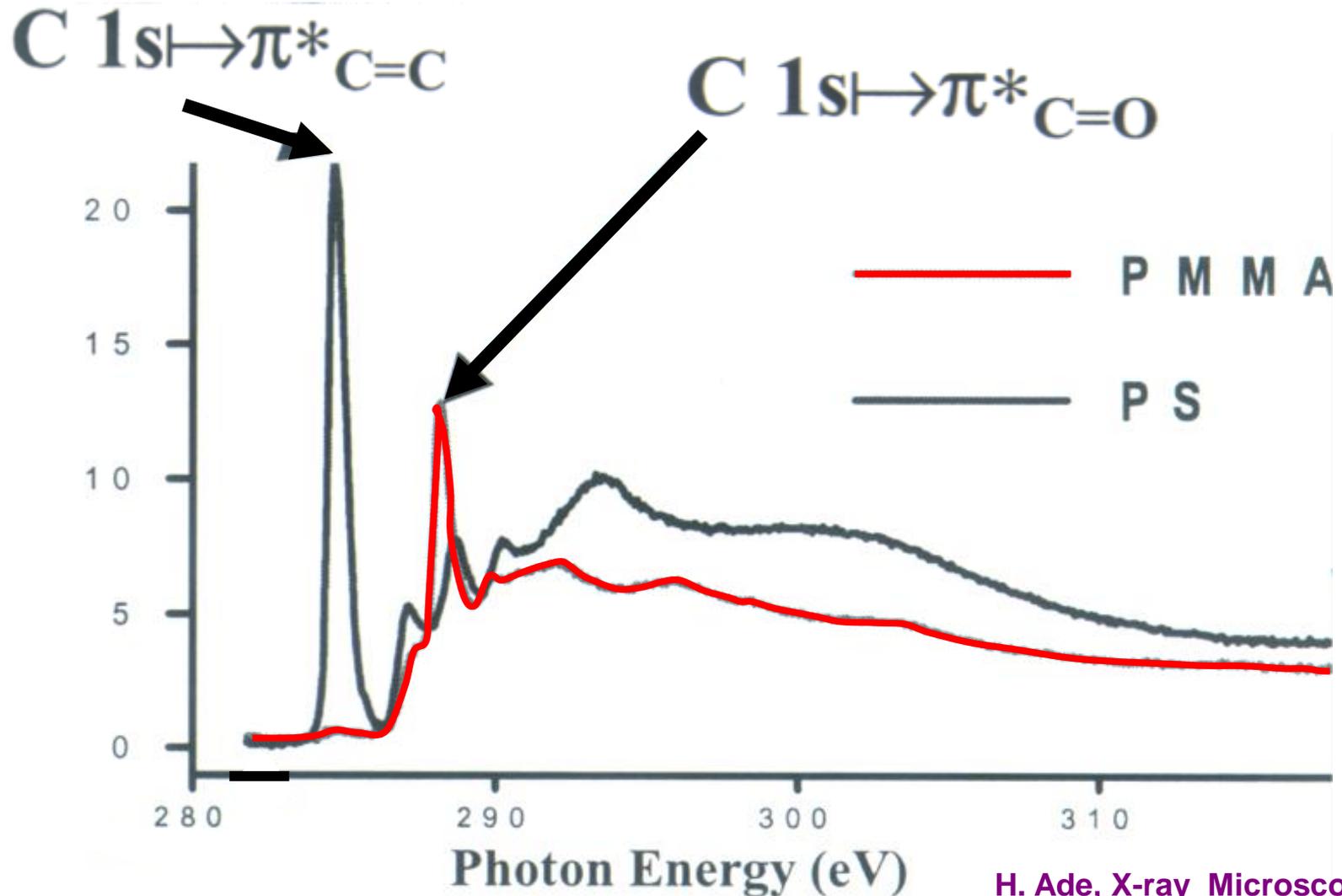
Magnetic Circular Dichroism in X-Ray Absorption (XMCD)

Ferromagnetic cobalt with magnetization along incident light direction



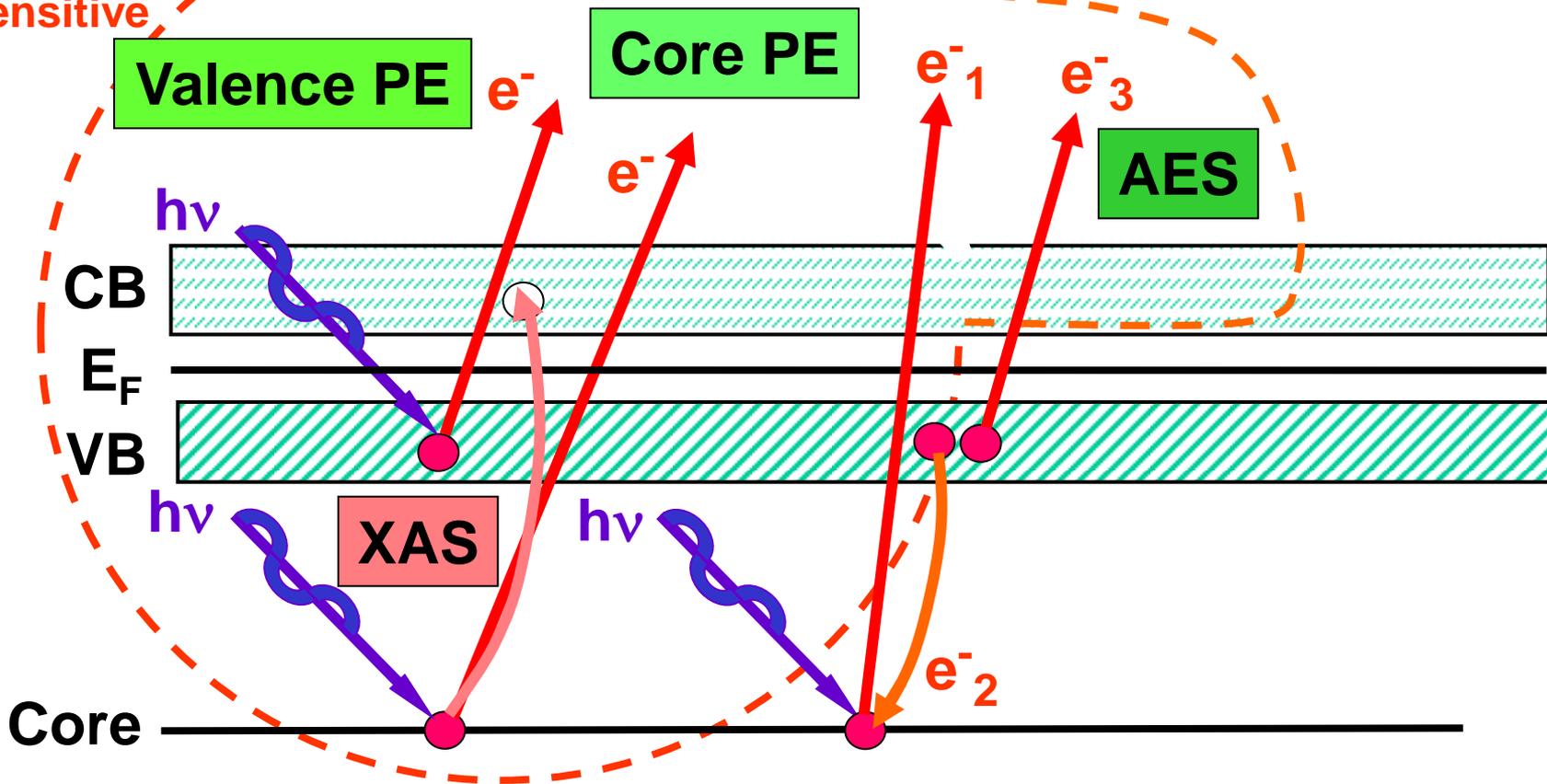
→ Sum rules yield spin and orbital magnetic moments on a given atom

Variation of Near-Edge X-Ray Absorption Fine Structure (NEXAFS) for Different Polymers



The Soft X-Ray Spectroscopies

Electron-out:
surface
sensitive



PE = photoemission = photoelectron spectroscopy

XAS = x-ray absorption spectroscopy

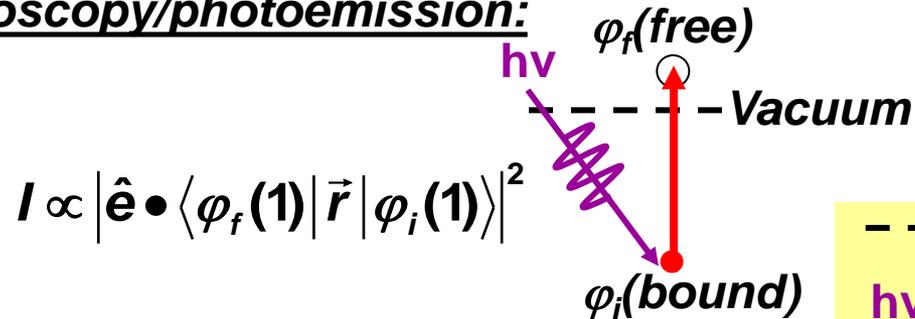
AES = Auger electron spectroscopy

XES = x-ray emission spectroscopy

REXS/RIXS = resonant elastic/inelastic x-ray scattering

MATRIX ELEMENTS IN THE SOFT X-RAY SPECTROSCOPIES: DIPOLE LIMIT

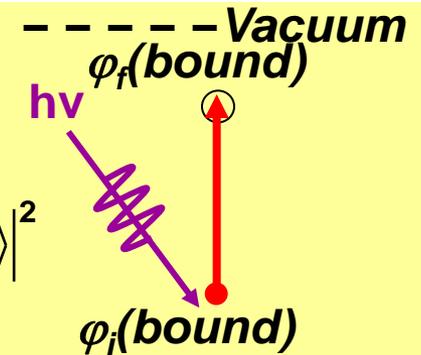
- Photoelectron spectroscopy/photoemission:



$$I \propto |\hat{\mathbf{e}} \cdot \langle \varphi_f(\mathbf{1}) | \vec{r} | \varphi_i(\mathbf{1}) \rangle|^2$$

- Near-edge x-ray absorption:

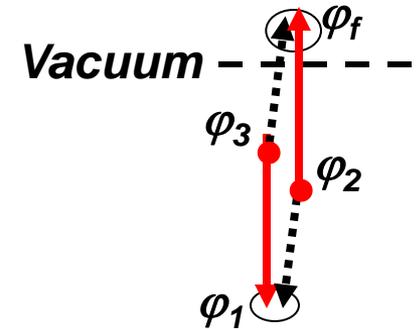
$$I \propto |\hat{\mathbf{e}} \cdot \langle \varphi_f(\mathbf{1}) | \vec{r} | \varphi_i(\mathbf{1}) \rangle|^2$$



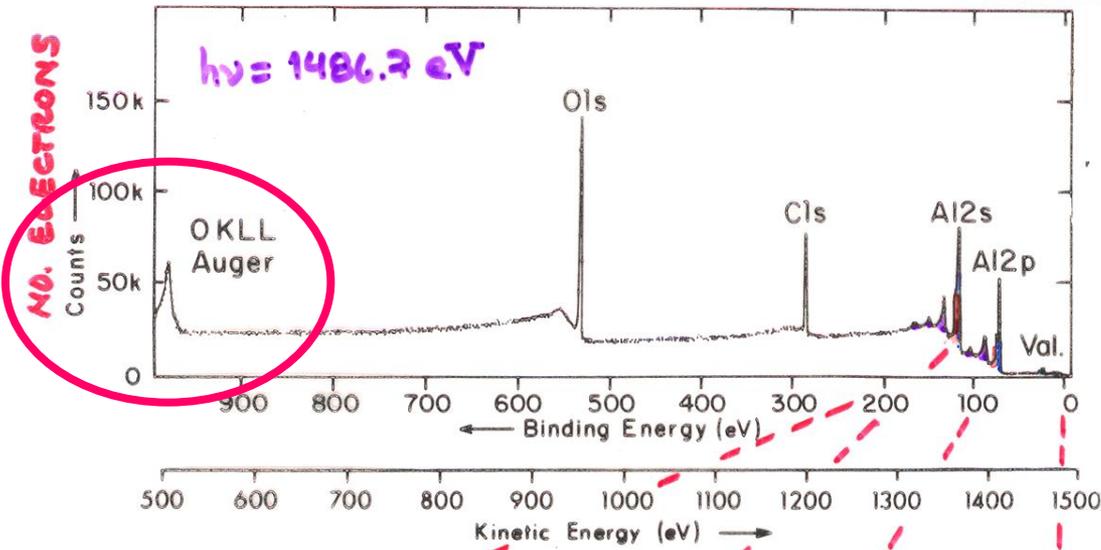
- Auger electron emission:

$$I \propto \left| \langle \varphi_f(\mathbf{1})\varphi_1(\mathbf{2}) | \frac{e^2}{r_{12}} | \varphi_3(\mathbf{1})\varphi_2(\mathbf{2}) \rangle - \langle \varphi_1(\mathbf{1})\varphi_f(\mathbf{2}) | \frac{e^2}{r_{12}} | \varphi_3(\mathbf{1})\varphi_2(\mathbf{2}) \rangle \right|^2$$

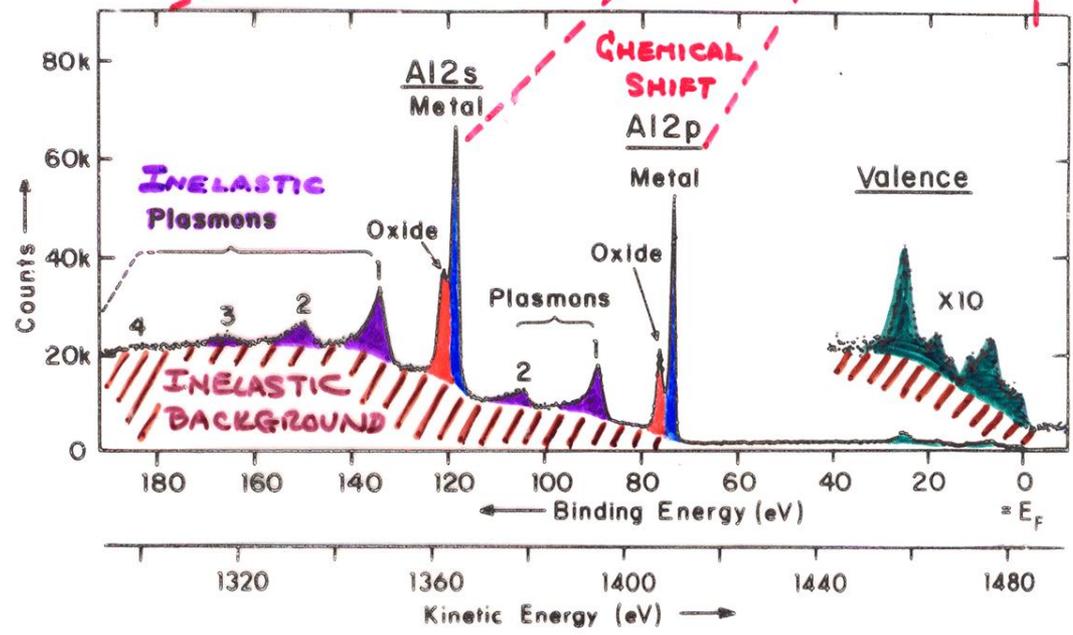
Direct
Exchange



TYPICAL PHOTOELECTRON SPECTRA: OXIDIZED ALUMINUM



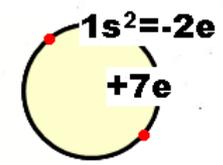
A very useful diagnostic observation:
 Auger kinetic energies do not change with photon energy
 Photoelectron kinetic energies shift linearly with photon energy



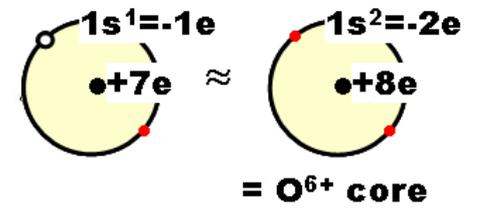
"Basic Concepts of XPS"
 Figure 1

THE AUGER PROCESS

N core = N 1s² = N⁵⁺

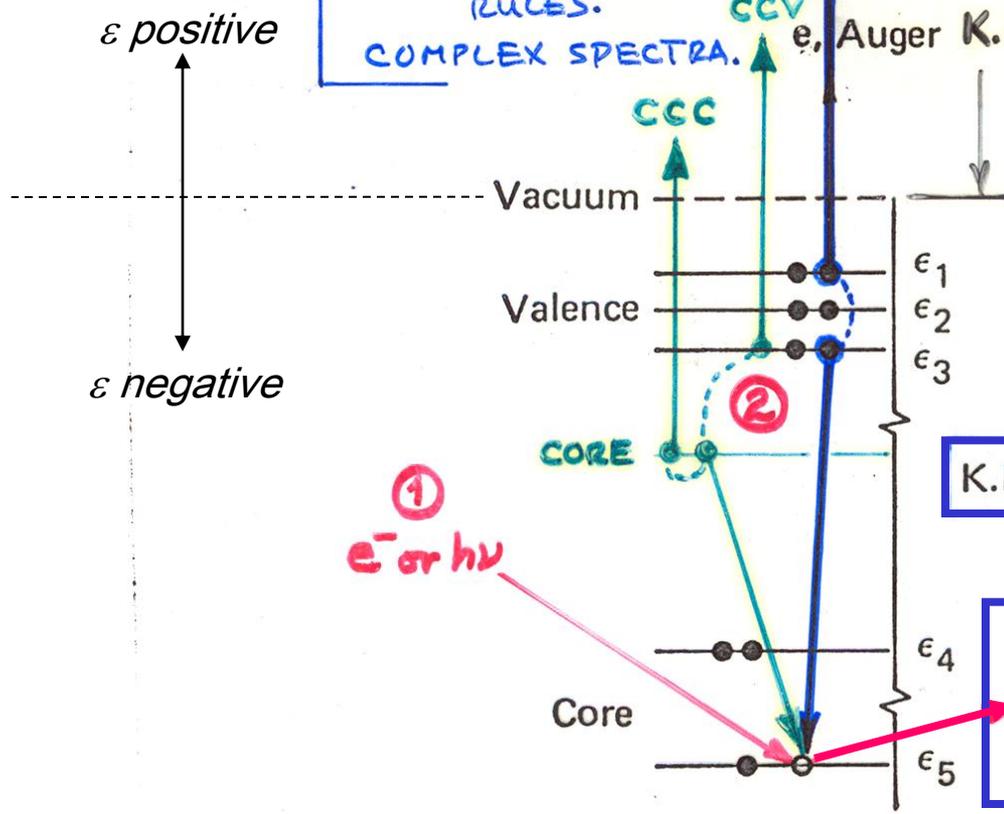


Assume:
N⁶⁺ core with
1s hole = N⁶⁺ =



The equivalent core or Z+1 approximation

TWO-STEP PROCESS.
TWO-HOLE FINAL STATE.
NO STRONG SELEC. RULES.
COMPLEX SPECTRA.

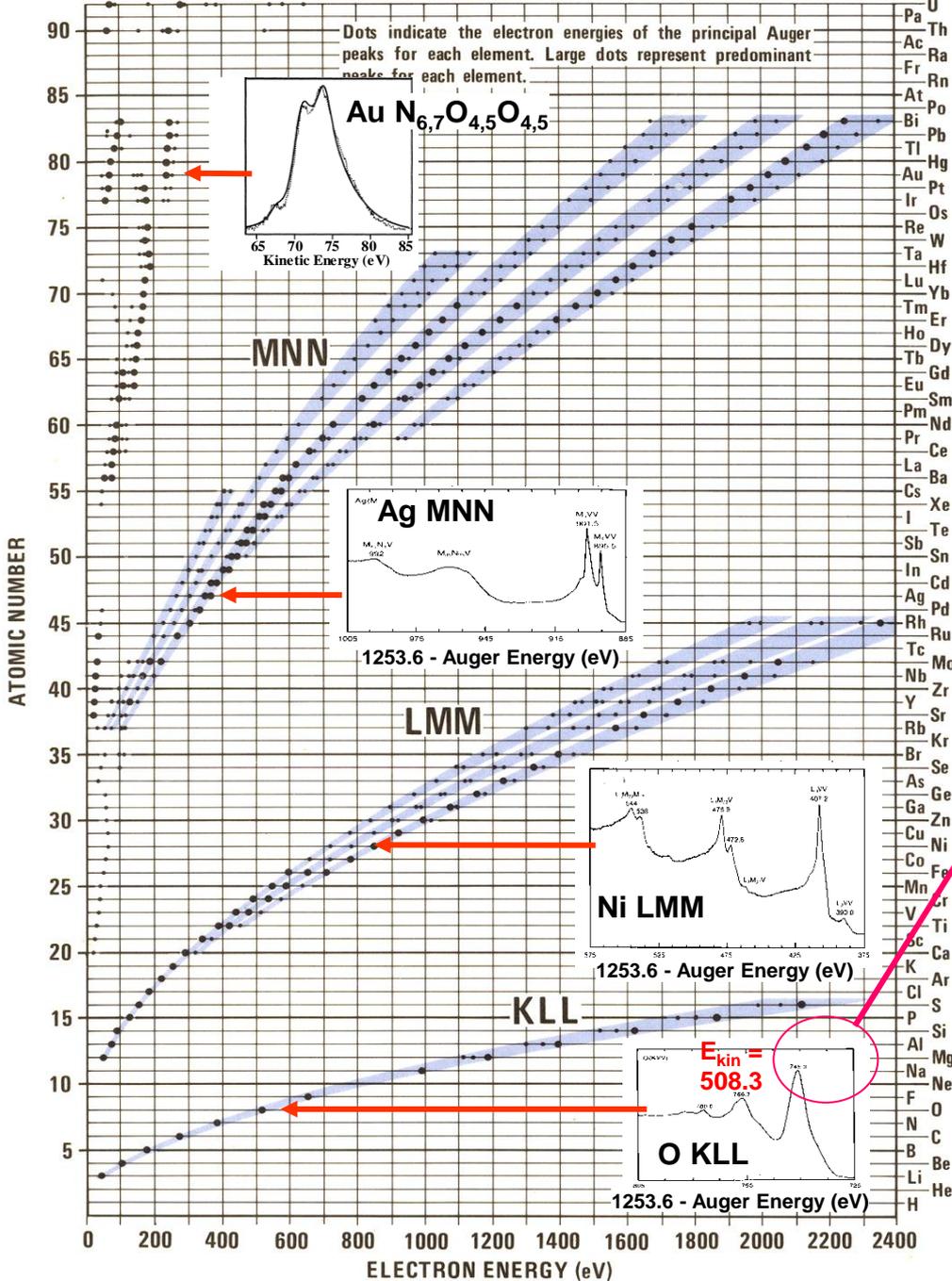


K.E. $\sim -\epsilon_5 + \epsilon_3 + \epsilon_1$ DOES NOT DEPEND ON $h\nu$!

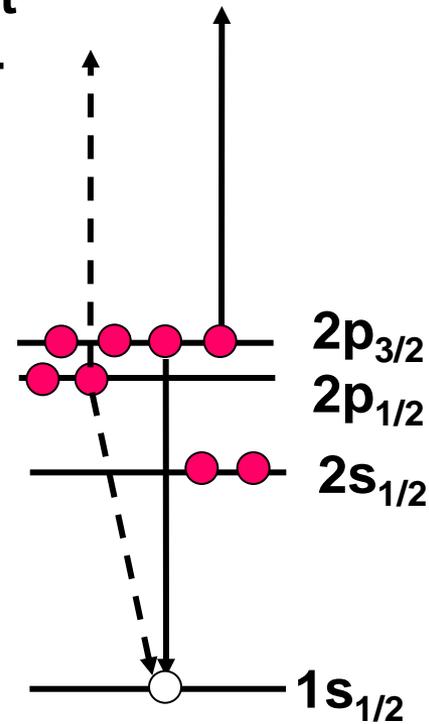
Or more accurately:
K.E. $\approx B.E._5^Z - B.E._3^{Z+1} - B.E._1^Z$
 $e^- \approx B.E._5^Z - B.E._3^Z - B.E._1^{Z+1}$
 \approx (average of two above)

Figure 2. Scheme of the Auger process. A valence-level involved Auger emission is illustrated here, but the two electrons involved also could have come from core level, ϵ_4 , provided $\epsilon_5 - 2\epsilon_4 > 0$.

PRINCIPAL AUGER ELECTRON ENERGIES



X-Ray Data Booklet Fig. 1.4



$$\begin{aligned}
 \text{K.E.} &\approx \text{B.E.}_{1s}^{Z=8} - \text{B.E.}_{2p}^9 - \text{B.E.}_{2p}^8 \\
 &\approx \text{B.E.}_{1s}^8 + \text{B.E.}_{2p}^8 - \text{B.E.}_{2p}^9 \\
 &\approx 543.1 - 17 - 13 \approx \mathbf{513 \text{ eV}}
 \end{aligned}$$

X-Ray Data Booklet--Section 1.1 ELECTRON BINDING ENERGIES

The energies are given in eV relative to the vacuum level for the rare gases and for H₂, N₂, O₂, F₂, and Cl₂; relative to the Fermi level for the metals; and relative to the top of the valence bands for semiconductors (and insulators).

Electronic configuration	Element	K 1s	L ₁ 2s	L ₂ 2p _{1/2}	L ₃ 2p _{3/2}	M ₁ 3s	M ₂ 3p _{1/2}	M ₃ 3p _{3/2}
1s	1 H	13.6						
1s²	2 He	24.6*						
1s² 2s	3 Li	54.7*						
1s² 2s²	4 Be	111.5*						
1s² 2s² 2p	5 B	188*						
1s² 2s² 2p²	6 C	284.2*						
1s² 2s² 2p³	7 N	409.9*	37.3*	~ 9	~ 9			
1s² 2s² 2p⁴	8 O	543.1*	41.6*	~ 13	~ 13			
1s² 2s² 2p⁵	9 F	696.7*	~ 45	~ 17	~ 17			
1s² 2s² 2p⁶	10 Ne	870.2*	48.5*	21.7*	21.6*			
[Ne] 3s	11 Na	1070.8†	63.5†	30.65	30.81			
[Ne] 3s²	12 Mg	1303.0†	88.7	49.78	49.50			
[Ne] 3s² 3p	13 Al	1559.6	117.8	72.95	72.55			
[Ne] 3s² 3p²	14 Si	1839	149.7*b	99.82	99.42			
[Ne] 3s² 3p³	15 P	2145.5	189*	136*	135*			
[Ne] 3s² 3p⁴	16 S	2472	230.9	163.6*	162.5*			
[Ne] 3s² 3p⁵	17 Cl	2822.4	270*	202*	200*			
[Ne] 3s² 3p⁶	18 Ar	3205.9*	326.3*	250.6†	248.4*	29.3*	15.9*	15.7*
[Ar] 4s	19 K	3608.4*	378.6*	297.3*	294.6*	34.8*	18.3*	18.3*
[Ar] 4s²	20 Ca	4038.5*	438.4†	349.7†	346.2†	44.3 †	25.4†	25.4†
	21 Sc	4492	498.0*	403.6*	398.7*	51.1*	28.3*	28.3*
	22 Ti	4966	560.9†	460.2†	453.8†	58.7†	32.6†	32.6†

Valence levels

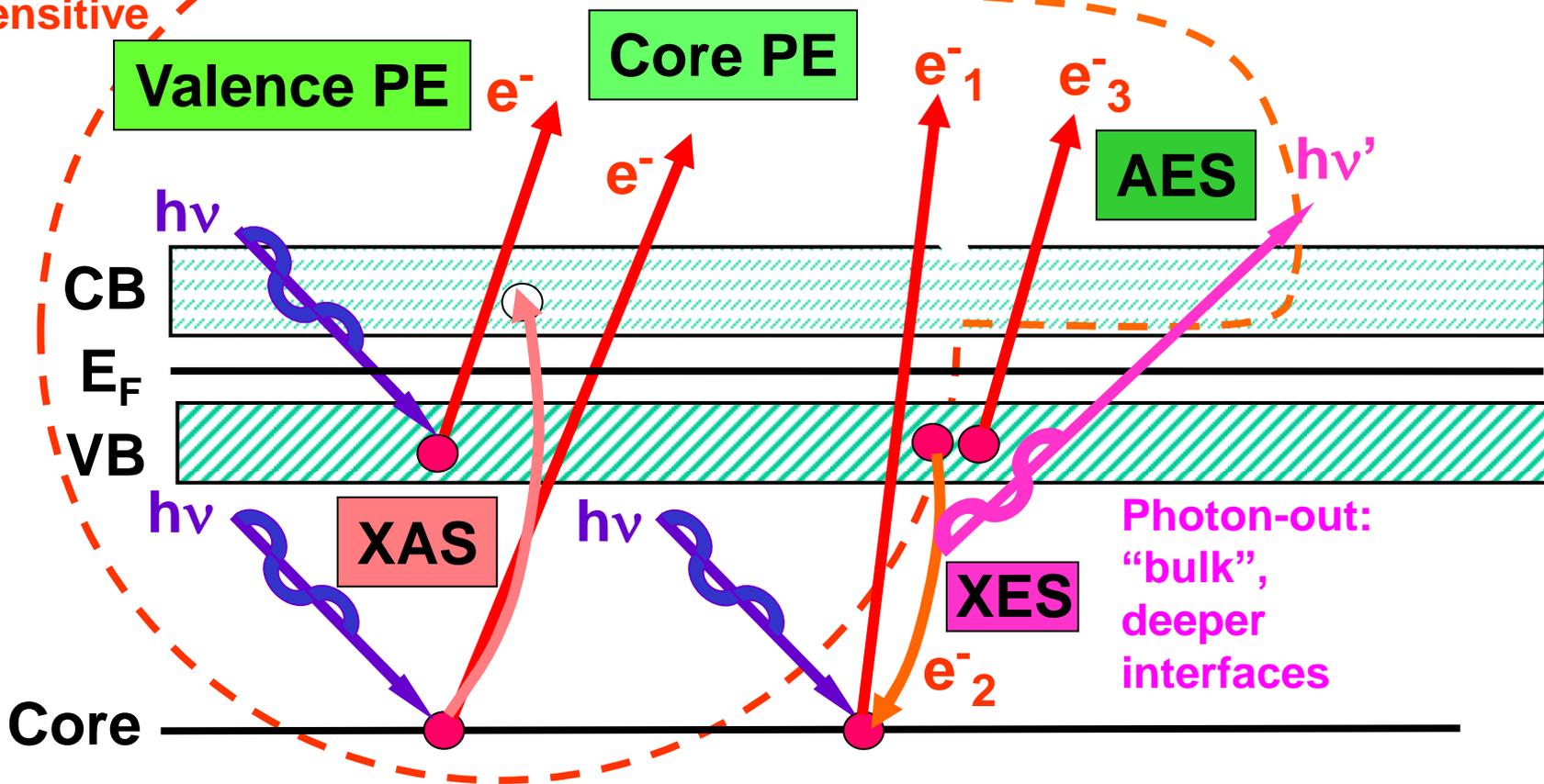
Interpolated, extrapolated

Missing valence B.E.s

Valence levels

The Soft X-Ray Spectroscopies

Electron-out:
surface
sensitive



PE = photoemission = photoelectron spectroscopy

XAS = x-ray absorption spectroscopy

AES = Auger electron spectroscopy

XES = x-ray emission spectroscopy

REXS/RIXS = resonant elastic/inelastic x-ray scattering

THE AUGER PROCESS

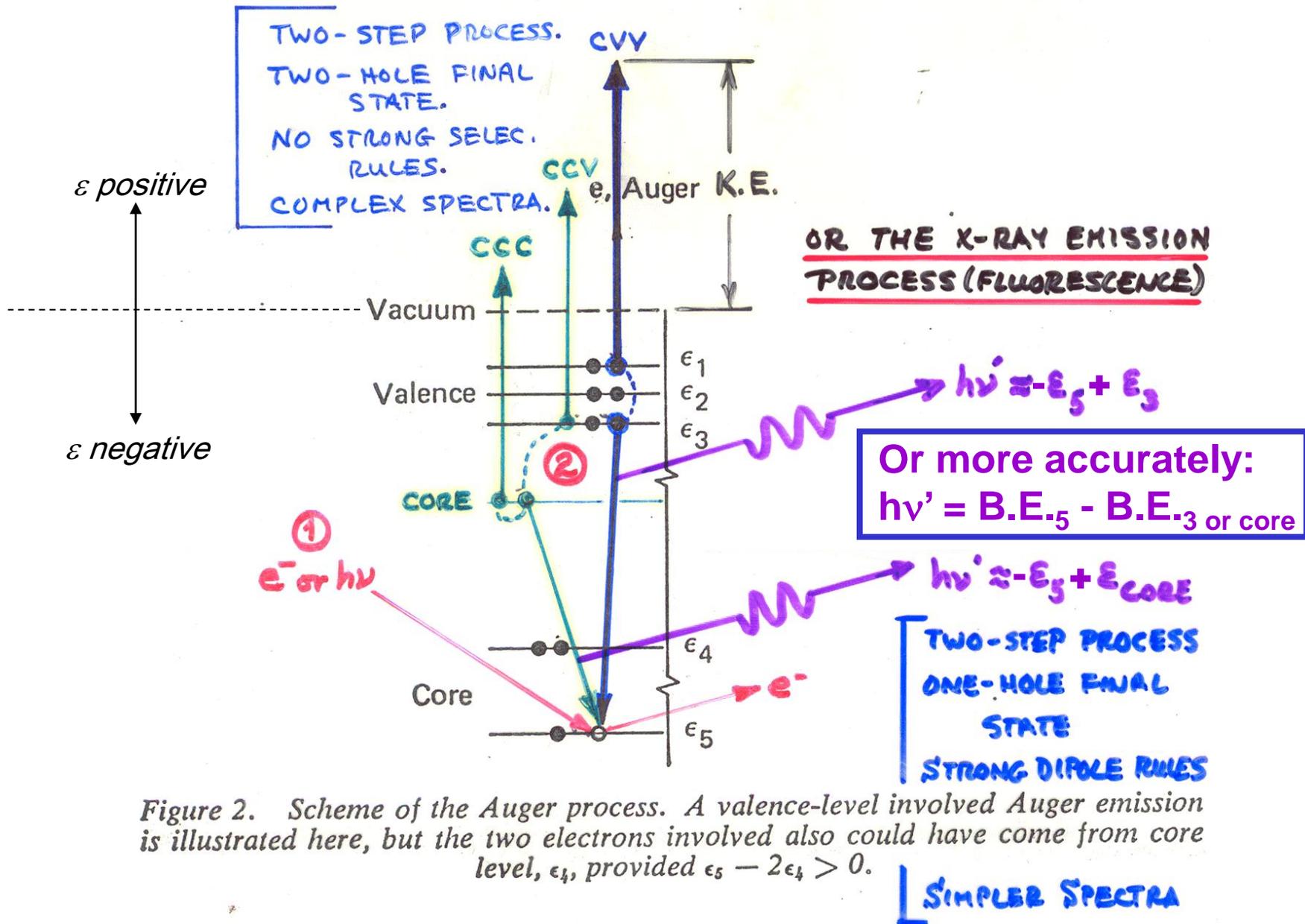
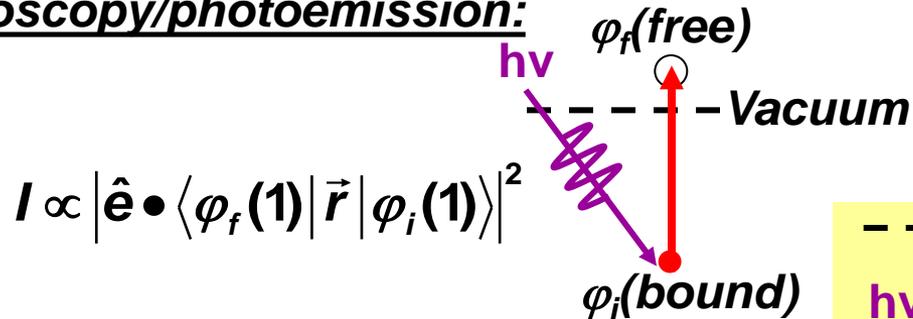


Figure 2. Scheme of the Auger process. A valence-level involved Auger emission is illustrated here, but the two electrons involved also could have come from core level, ϵ_4 , provided $\epsilon_5 - 2\epsilon_4 > 0$.

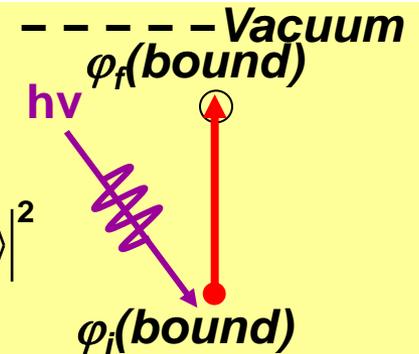
MATRIX ELEMENTS IN THE SOFT X-RAY SPECTROSCOPIES: DIPOLE LIMIT

- Photoelectron spectroscopy/photoemission:



- Near-edge x-ray absorption:

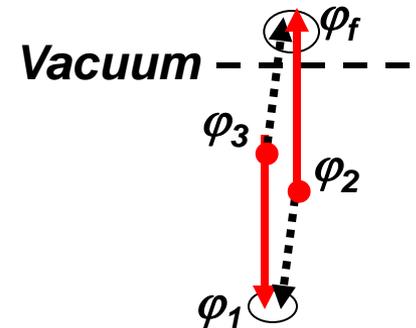
$$I \propto |\hat{\mathbf{e}} \cdot \langle \varphi_f(\mathbf{1}) | \vec{r} | \varphi_i(\mathbf{1}) \rangle|^2$$



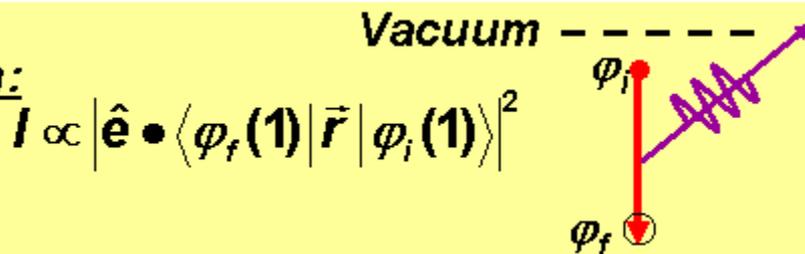
- Auger electron emission:

$$I \propto \left| \langle \varphi_f(\mathbf{1})\varphi_1(\mathbf{2}) | \frac{e^2}{r_{12}} | \varphi_3(\mathbf{1})\varphi_2(\mathbf{2}) \rangle - \langle \varphi_1(\mathbf{1})\varphi_f(\mathbf{2}) | \frac{e^2}{r_{12}} | \varphi_3(\mathbf{1})\varphi_2(\mathbf{2}) \rangle \right|^2$$

Direct Exchange



- X-ray emission:



1.3 FLUORESCENCE YIELDS FOR K AND L SHELLS

Jeffrey B. Kortright

Fluorescence yields for the *K* and *L* shells for the elements $5 \leq Z \leq 110$ are plotted in Fig. 1-2; the data are based on Ref. 1. These yields represent the probability of a core hole in the *K* or *L* shells being filled by a radiative process, in competition with nonradiative processes. Auger processes are the only nonradiative processes competing with fluorescence for the *K* shell and

If fluorescence yield \equiv FY

FY = probability of radiative decay \rightarrow x-ray emission)

$1 - \text{FY} =$ probability of non-radiative decay \rightarrow Auger electron emission

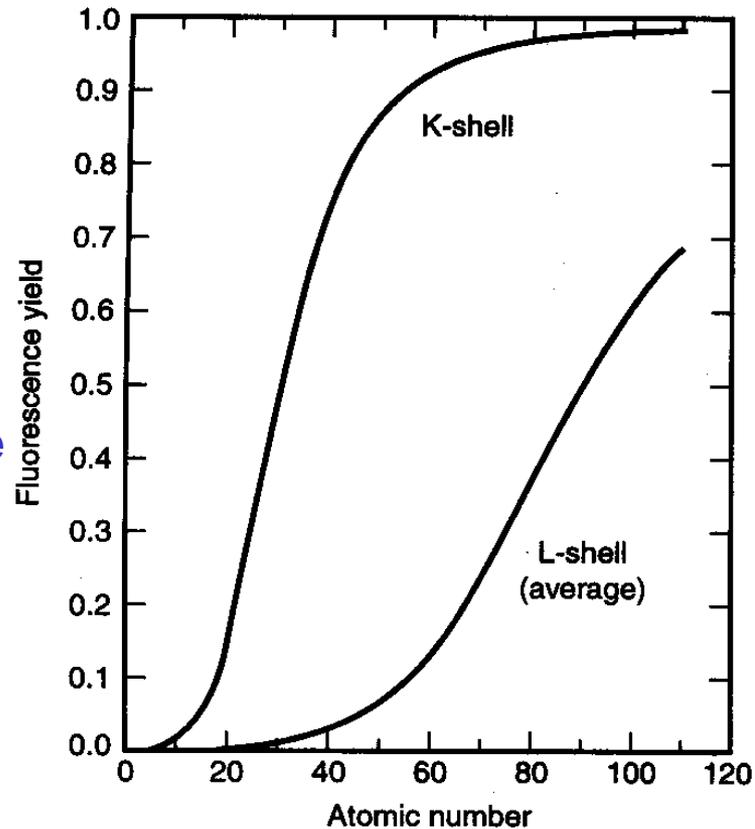
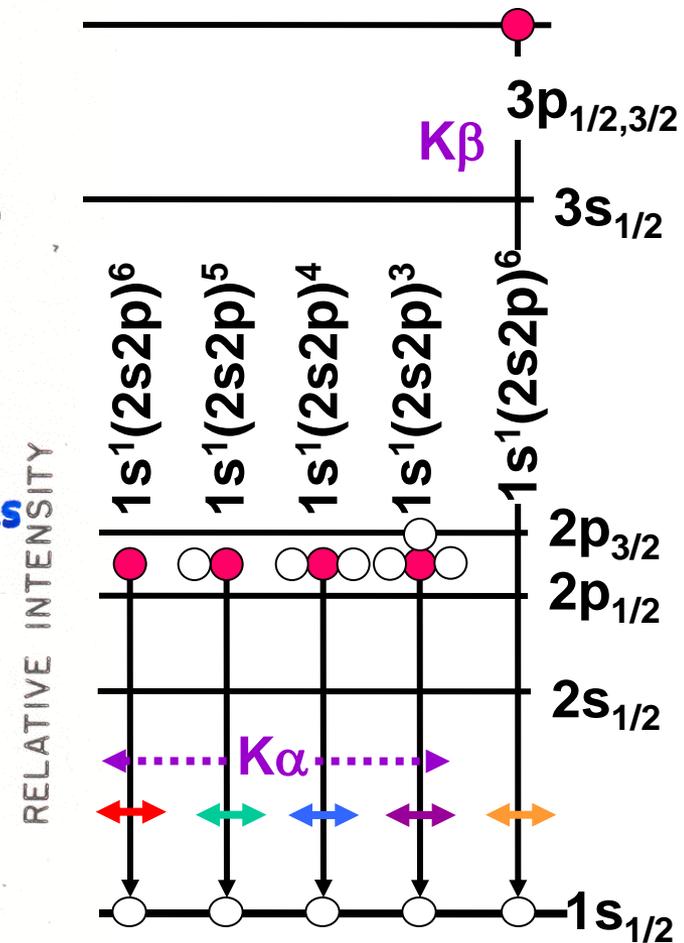
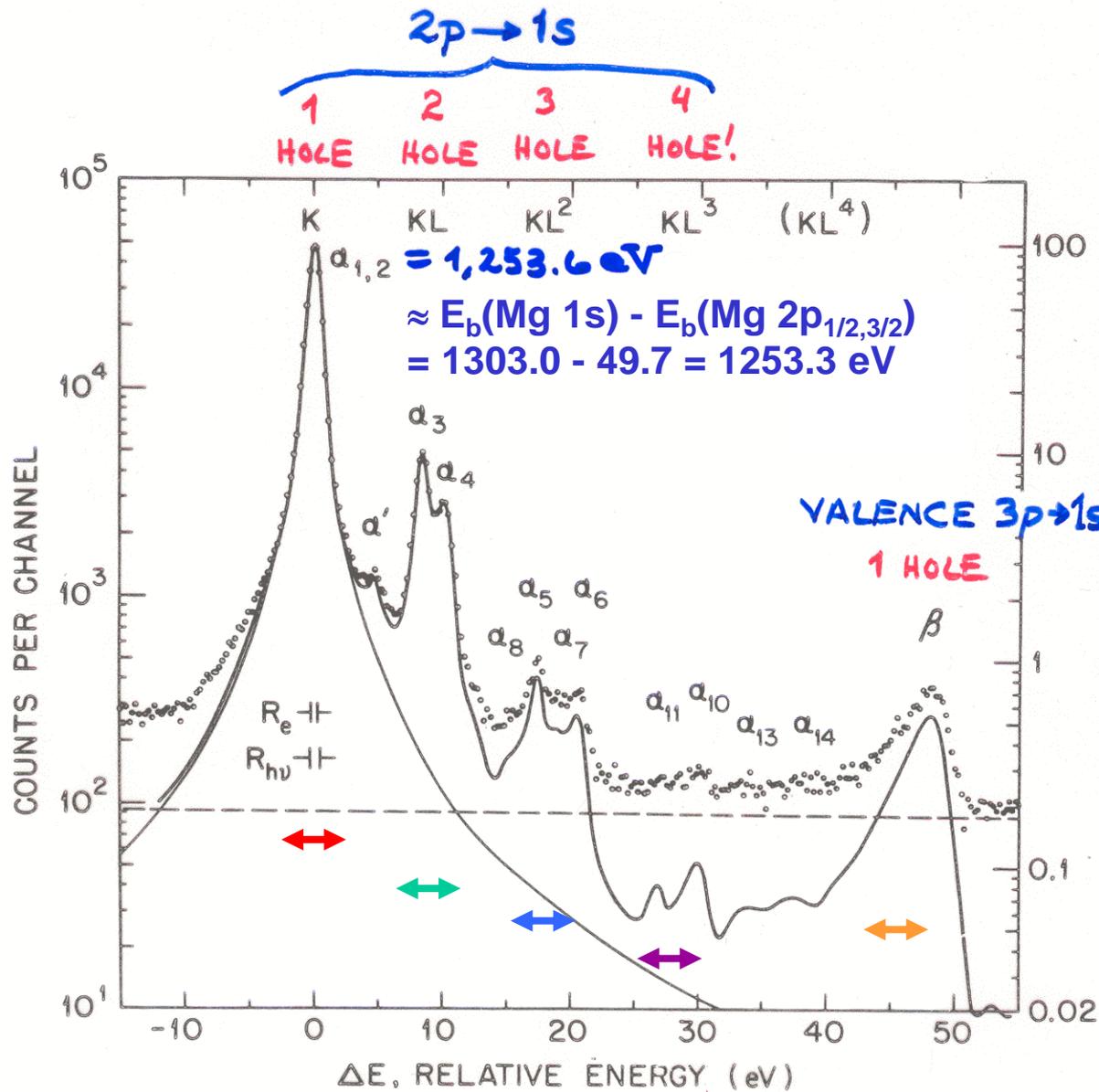


Fig. 1-2. Fluorescence yields for K and L shells for 110. The plotted curve for the L shell represents the average of L_1 , L_2 , and L_3 effective yields.



Mg K series of x-rays:
 atomic no. = 12
 Fluorescence Yield ≈ 0.03

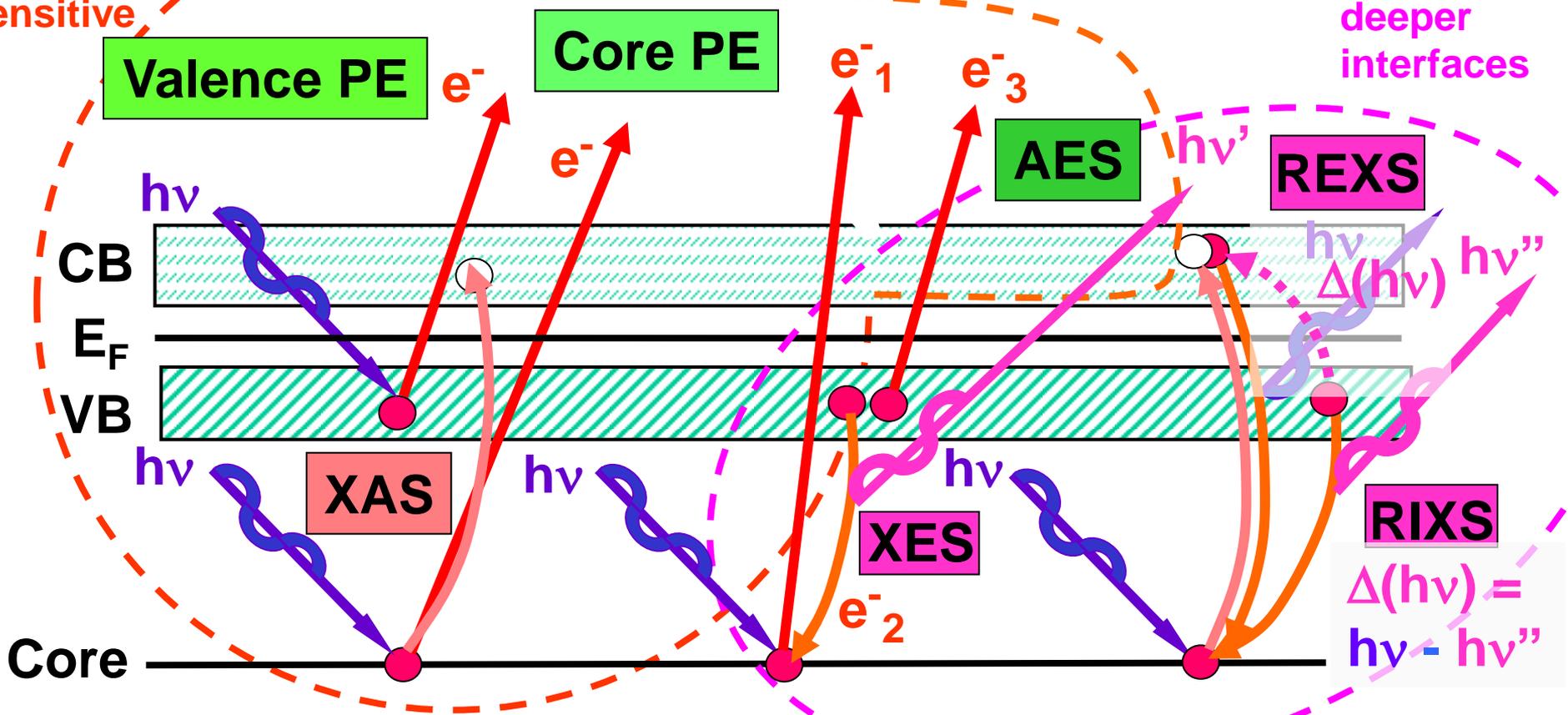
A STANDARD LABORATORY X-RAY SOURCE

“Basic Concepts of XPS”
 Figure 2

The Soft X-Ray Spectroscopies

Electron-out:
surface
sensitive

Photon-out:
"bulk",
deeper
interfaces



PE = photoemission = photoelectron spectroscopy

XAS = x-ray absorption spectroscopy

AES = Auger electron spectroscopy

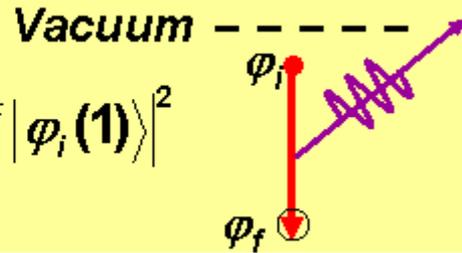
XES = x-ray emission spectroscopy

REXS/RIXS = resonant elastic/inelastic x-ray scattering

MATRIX ELEMENTS IN THE SOFT X-RAY SPECTROSCOPIES: RESONANT EFFECTS

- X-ray emission:

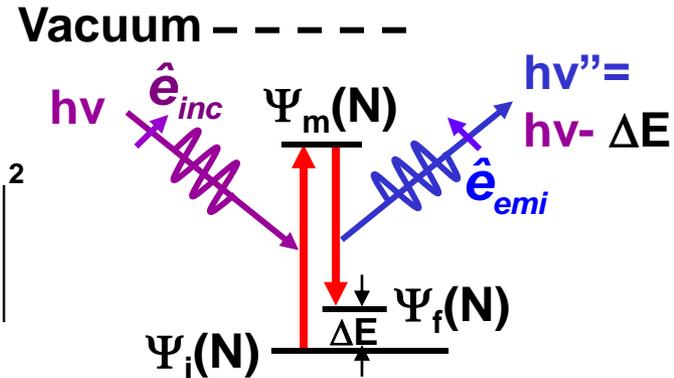
$$I \propto \left| \hat{\mathbf{e}} \cdot \langle \varphi_f(\mathbf{1}) | \vec{r} | \varphi_i(\mathbf{1}) \rangle \right|^2$$



- Resonant inelastic x-ray scattering:

$$I \propto \sum_f \left| \sum_m \frac{\langle \Psi_f(N) | \hat{\mathbf{e}}_{emi} \cdot \vec{r} | \Psi_m(N) \rangle \langle \Psi_m(N) | \hat{\mathbf{e}}_{inc} \cdot \vec{r} | \Psi_i(N) \rangle}{h\nu + E_i(N) - E_m(N) - i\Gamma_m} \right|^2$$

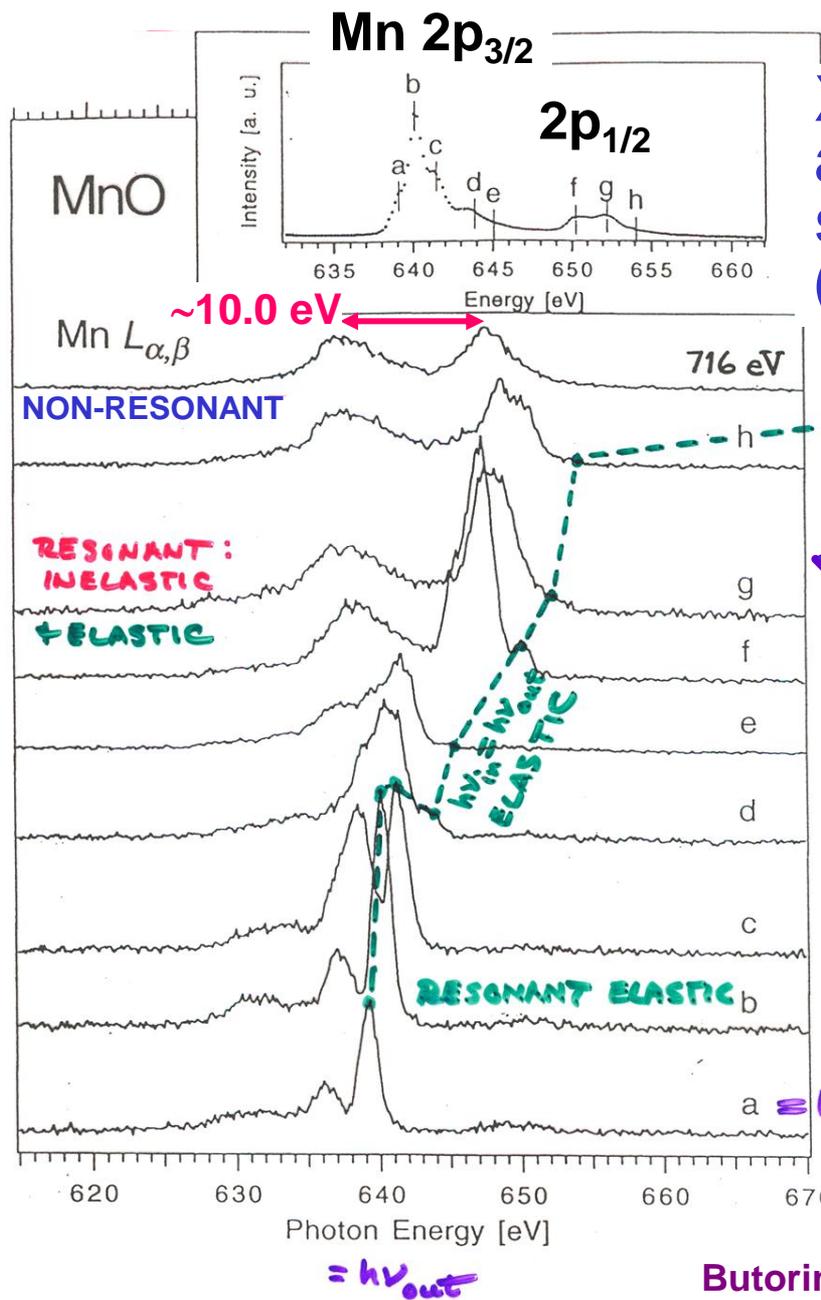
$$\times \delta(h\nu - (E_m(N) - E_i(N)))$$



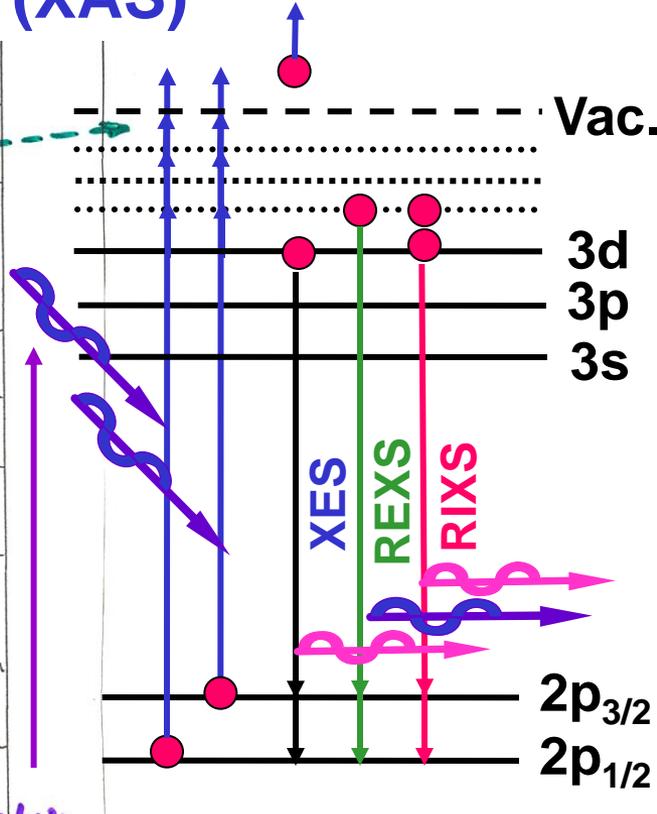
$$N_m(t) = N_m(0) e^{-\frac{2\Gamma_m t}{\hbar}} = N_m(0) e^{-\frac{t}{T_{lifetime}}}$$

X-ray
fluorescence
spectroscopy
=X-ray
emission
spectroscopy
(XES)

Resonant
inelastic
x-ray
scattering
(RIXS)
and
Resonant
elastic
x-ray
scattering
(REXS)



X-ray
absorption
spectroscopy
(XAS)

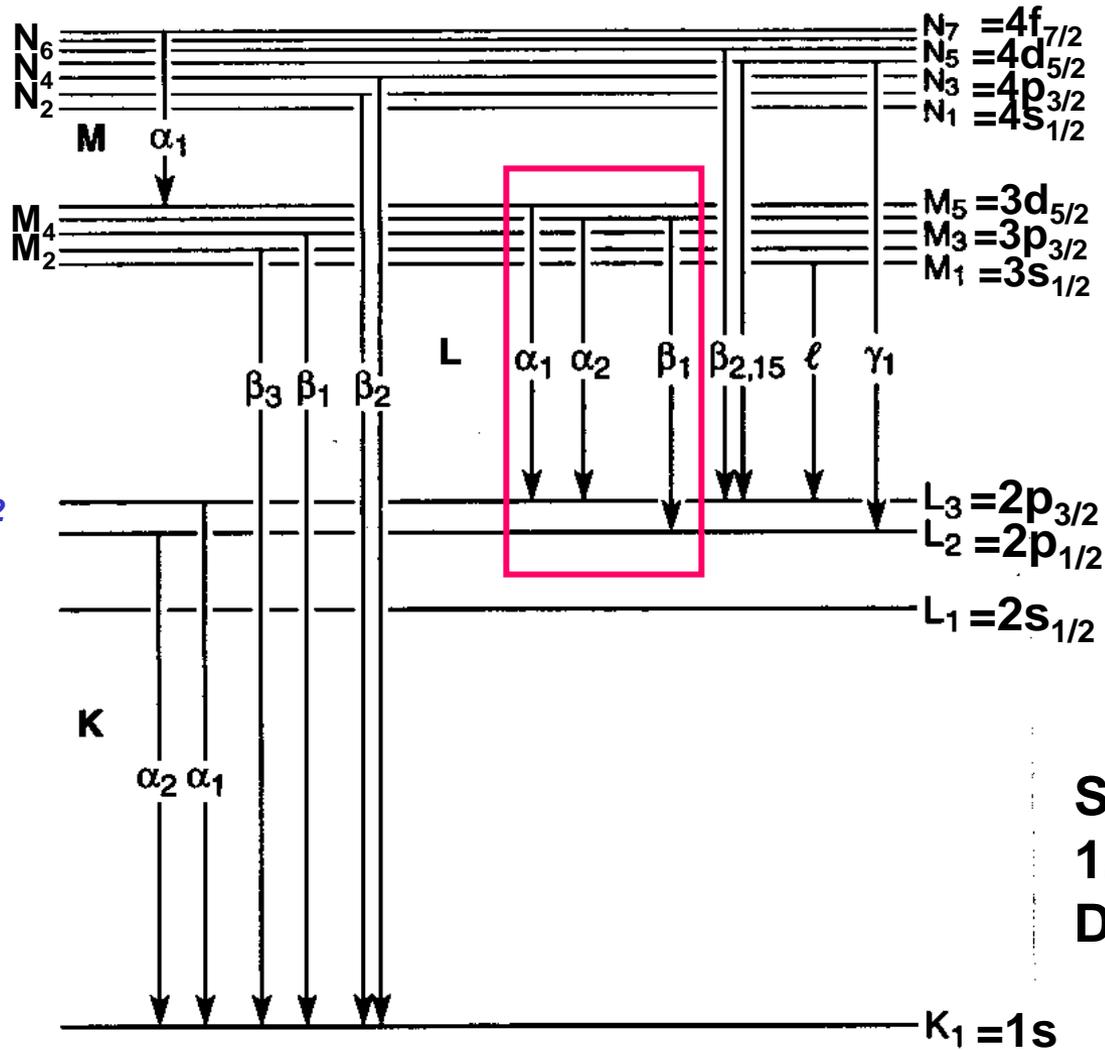


Butorin et al., Phys. Rev.
B 54, 4405 ('96)

**X-Ray
Nomenclature
(from "X-Ray
Data Booklet")**

In general:

$$nl \rightarrow \begin{cases} \text{Spin-}nl_{j=l+1/2} \\ \text{orbit } nl_{j=l-1/2} \end{cases}$$



$\Delta j = 0, \pm 1$

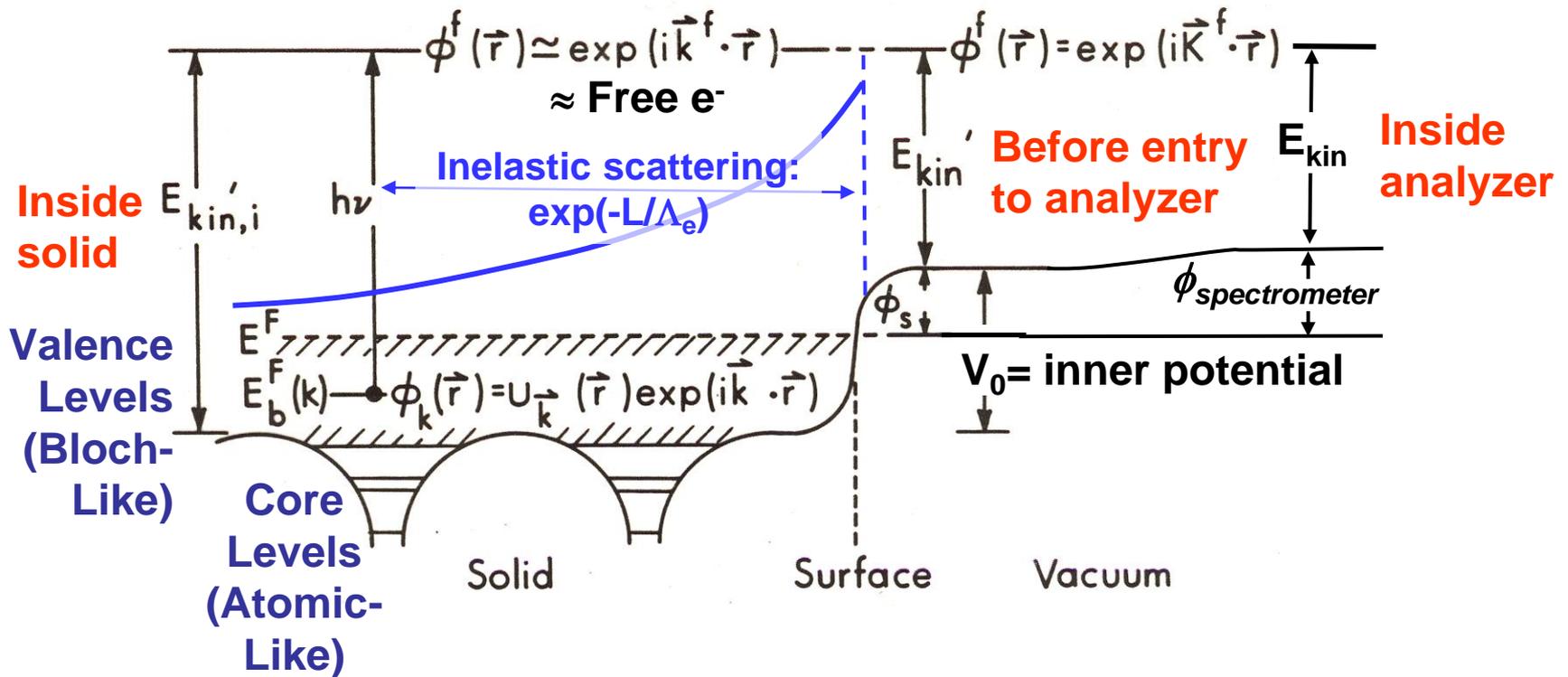
**See Section
1.2 in "X-Ray
Data Booklet"**

Fig. 1-1. Transitions that give rise to the emission lines in Table 1-3.

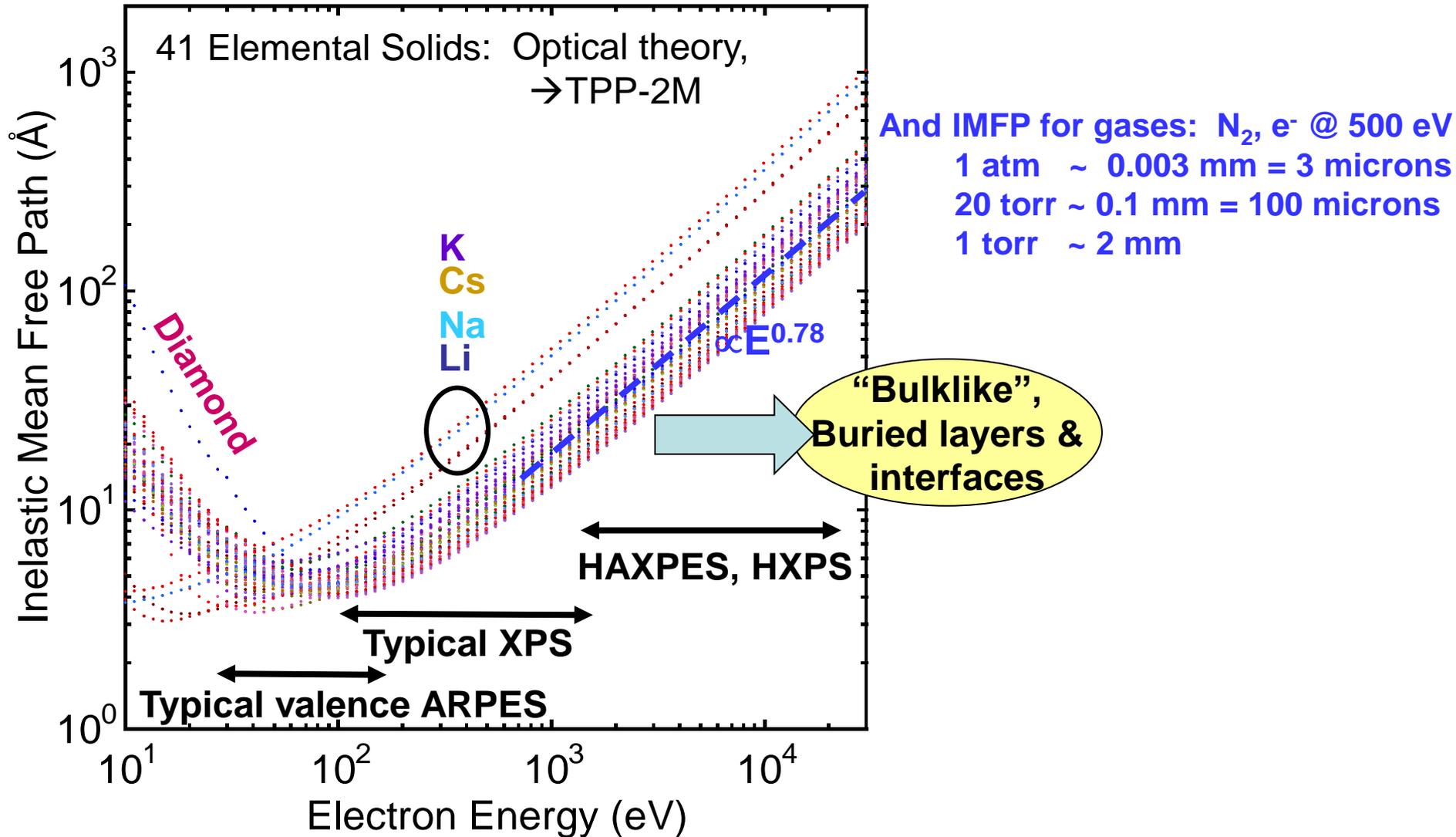
Basic energetics

$$h\nu = E_{\text{binding}}^{\text{Vacuum}} + E_{\text{kinetic}} = E_{\text{binding}}^{\text{Fermi}} + \phi_{\text{spectrometer}} + E_{\text{kinetic}}$$

One-Electron Picture of Photoemission from a Surface



Surface sensitivity and why we may want to go to 5-10 keV in XPS



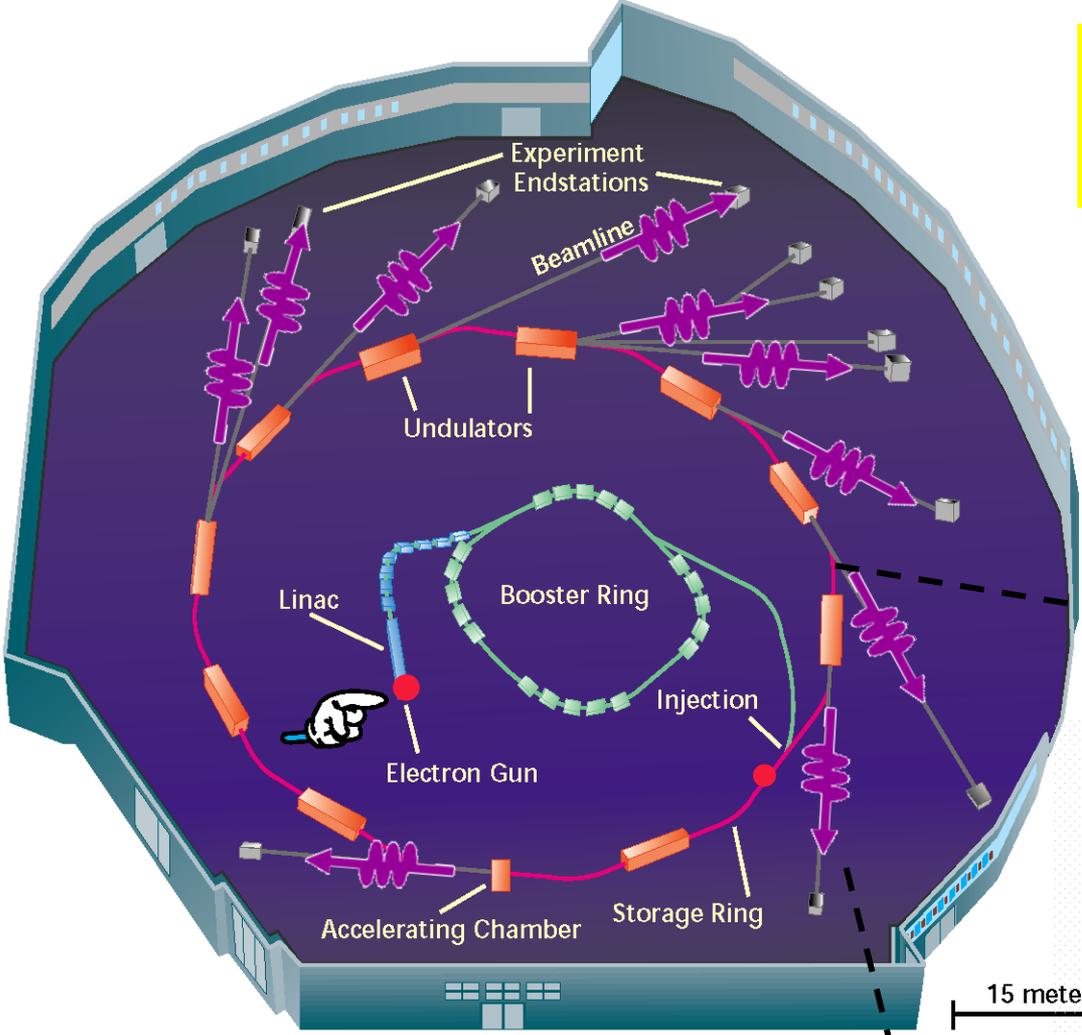
The Advanced Light Source



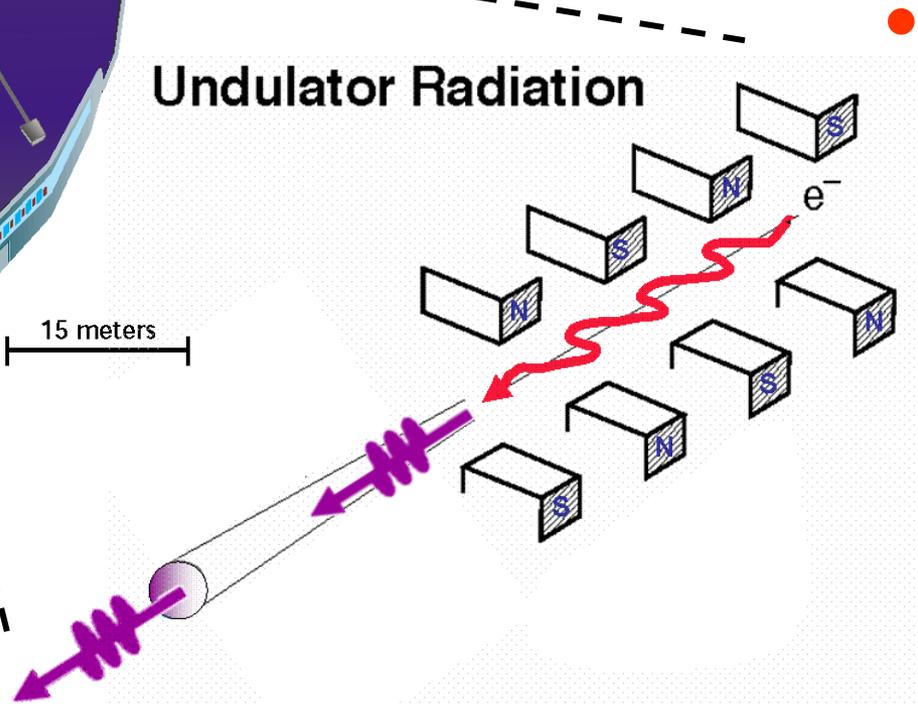
One of 40→50 synchrotron radiation facilities in the world

Inside a synchrotron radiation source

Electron speed near c :
ALS-0.99999997 c ,
very relativistic



Undulator Radiation



+FELs: FLASH, LCLS, Fermi (seeded),...

**EXPERIMENTAL STATION:
MULTI-TECHNIQUE
SPECTROMETER/
DIFFRACTOMETER (MTSD)**

**5-axis
sample
manipulator**

**Scienta
electron
spectrometer
(hidden)**

**Sample prep.
chamber: LEED,
Knudsen cells,
electromagnet,...**

**ALS
BL 9.3.1
 $h\nu = 2-5 \text{ keV}$**

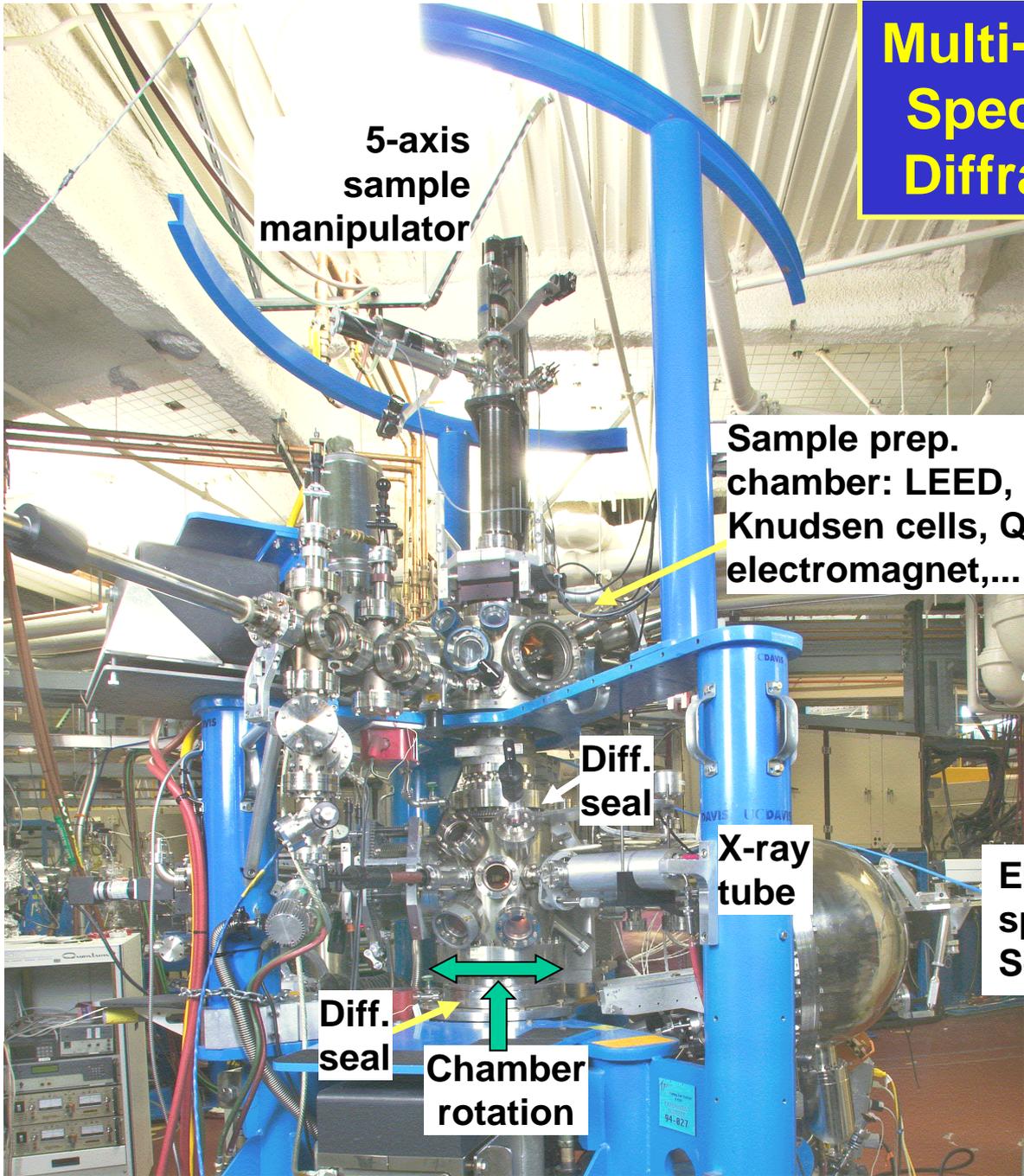


**Chamber
rotation**

**Scienta
soft x-ray
spectrometer**

**Permits using all relevant soft and hard x-ray spectroscopies on a single sample:
PS, PD, PH; XAS (e^- or photon detection), XES/RIXS, with MCD, MLD**

Multi-Technique Spectrometer/ Diffractometer



5-axis sample manipulator

Sample prep. chamber: LEED, Knudsen cells, QCM, electromagnet,...

Loadlock for sample introduction

Diff. seal

X-ray tube

Electron spectrometer: Scienta SES 200

Soft x-ray spectrometer: Scienta XES 300

Diff. seal

Chamber rotation

Hard X-Ray Photoemission (HXPS, HAXPES, HX-PES, HIKE...) in the World

Past workshops:

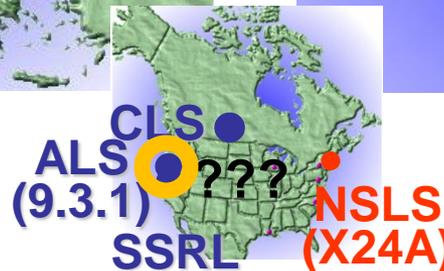
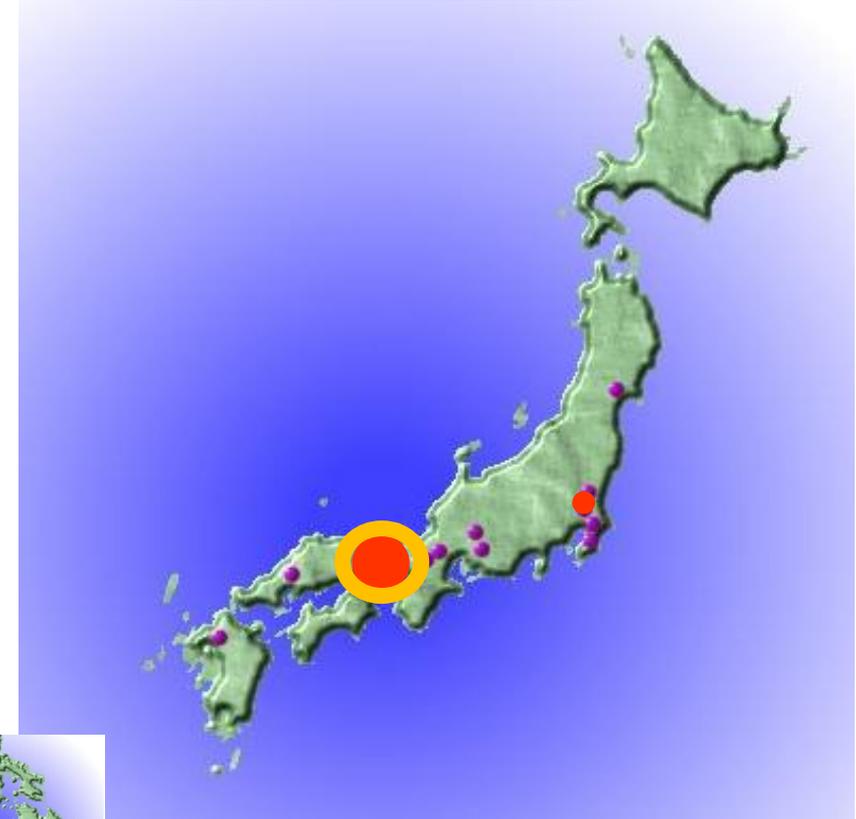
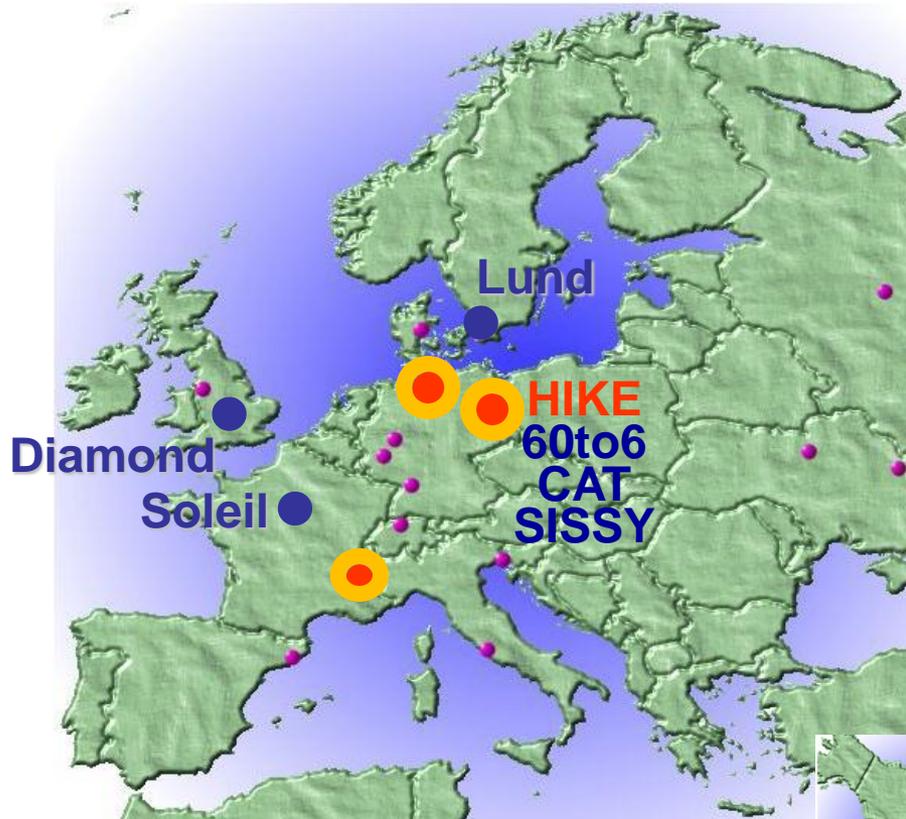
HAXPES03, ESRF--Nucl. Inst. and Meth. A, Volume 547, Issue 1, Pages 1-238 (2005)

HAXPES06, SPring8-- <http://haxpes2006.spring8.or.jp/program.html>

HAXPES-ALS User Meeting-- <http://ssg.als.lbl.gov/ssgdirectory/fedorov/workshops/index.html>

HAXPES09-NSLS-- <http://www.nsls.bnl.gov/newsroom/events/workshops/2009/haxpes/>

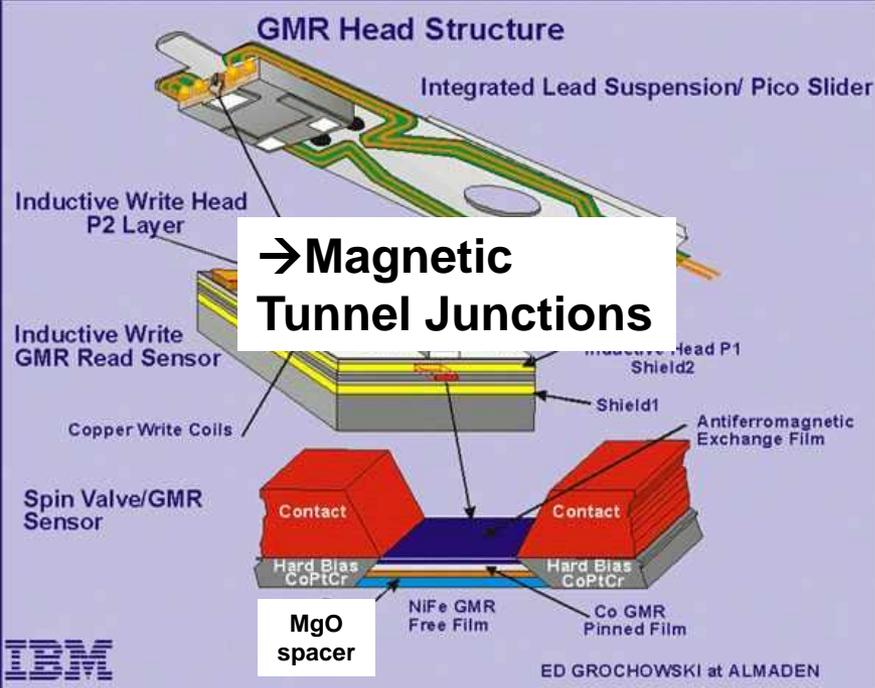
HAXPES11-Hasylab-- <http://indico.desy.de/conferenceDisplay.py?confId=3713>



● Existing
● Under construction

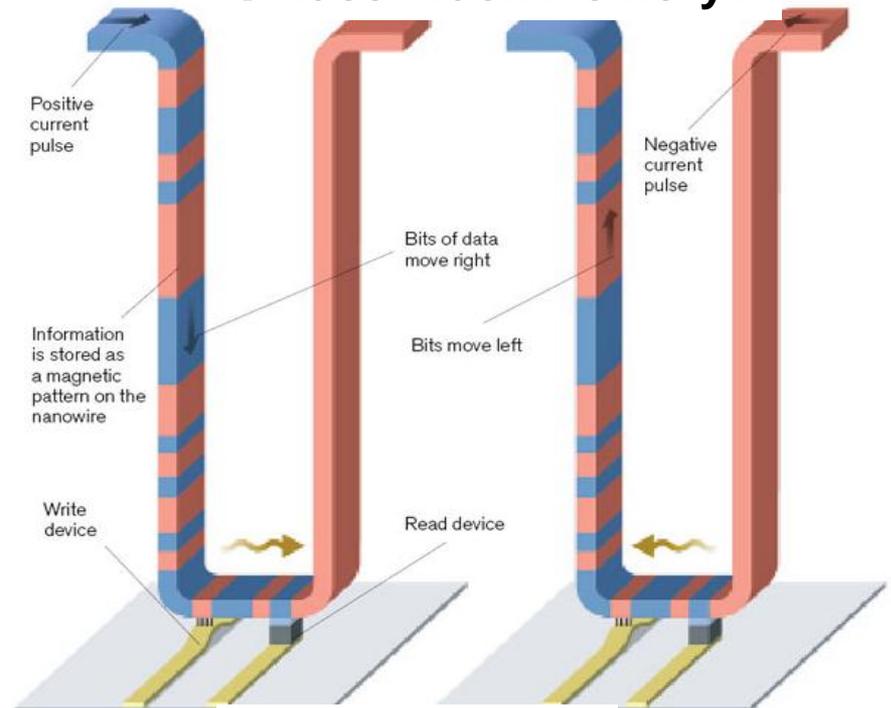
○ Our group

Some key elements in Spintronics/Semiconductors/Sensors—multilayer nanostructures



Magnetic Random Access Memory (MRAM-Non Volatile)

→Race Track Memory?



S. Parkin, IBM Almaden

Crucial buried functional layers & interfaces everywhere-

What do we want to know:

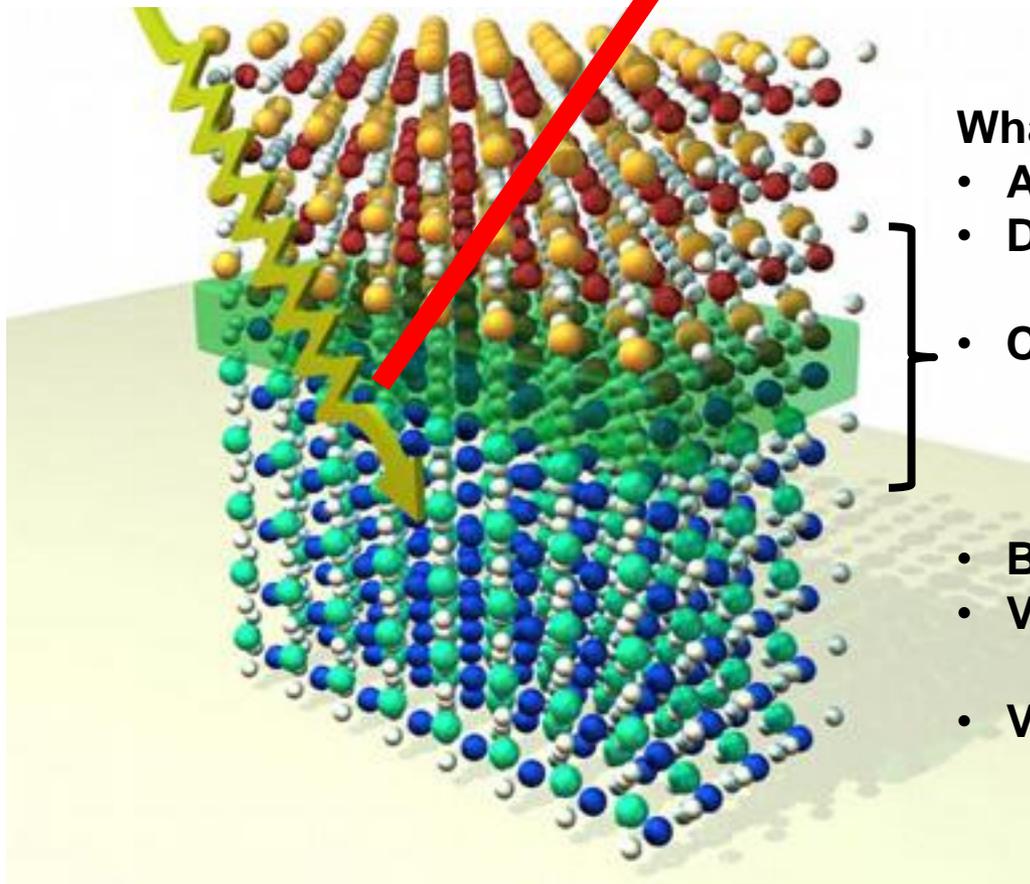
- Interface mixing/roughness/ concentration profiles
 - Depth-dependent magnetization profiles
- Depth-dependent densities of states
 - Depth-dependent band structure?

Photoemission from surfaces, interfaces, bulk materials

Photoelectron

$$E_{\text{kin}}, \vec{p} = \hbar\vec{k}, \vec{s}$$

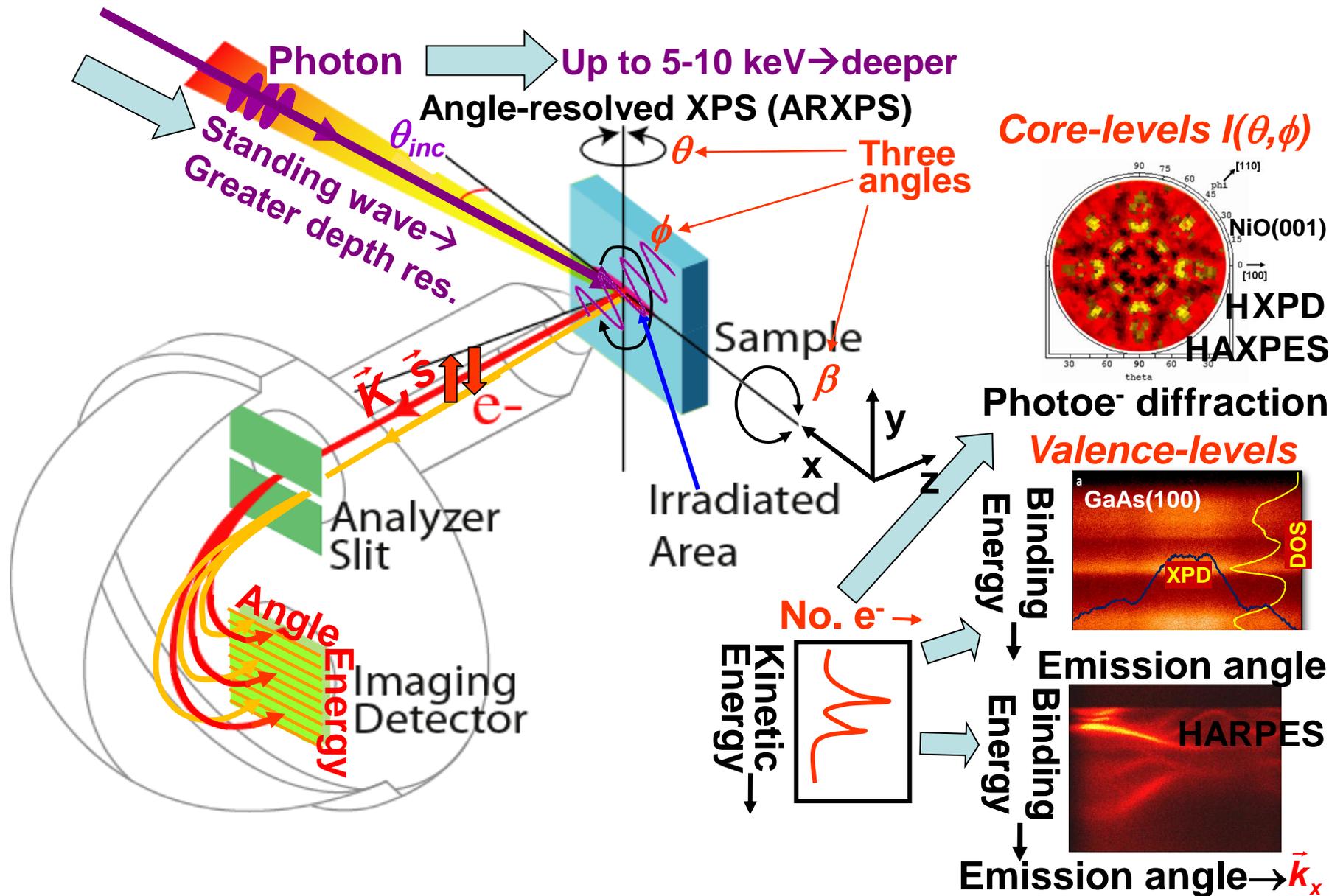
Photon
 $h\nu$



What do we want to know?

- Atomic structure, lattice distortions
- Depth profiles of composition and optical properties
- Core-levels \rightarrow element specific binding energies, charge states magnetic moments, electronic configurations
- Band offsets
- Valence-band densities of states bandgaps, behavior near E_F
- Valence-band dispersions, via depth-resolved ARPES

X-ray photoemission: some key elements



Photoemission with time-of-flight analysis

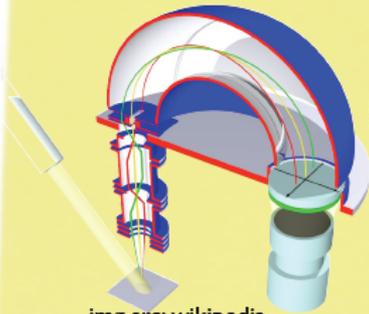


UPPSALA

Andreas,
Svensson,...

High resolution & High transmission

Hemisphere

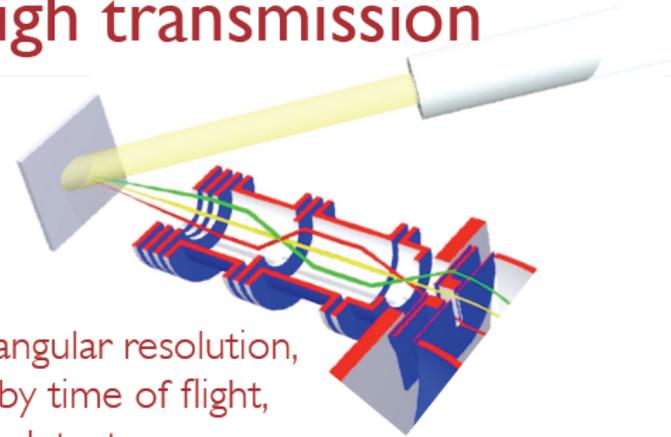


img src: wikipedia

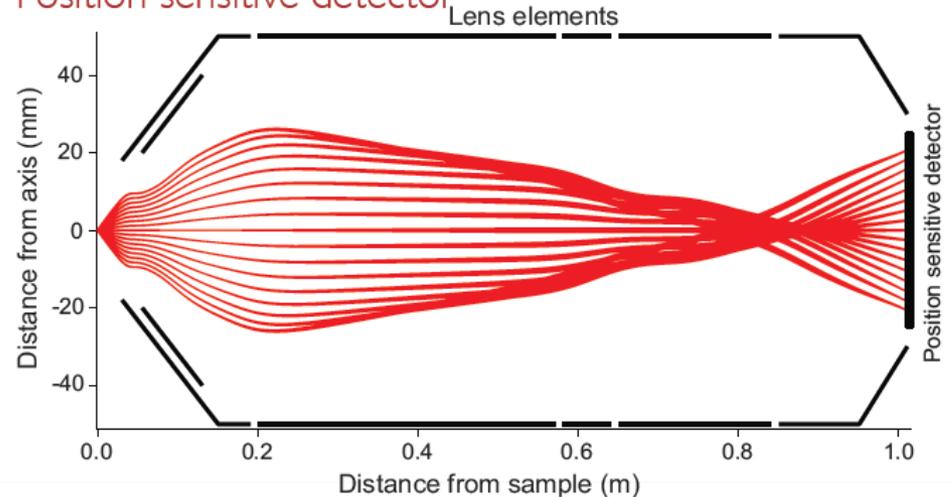
$$\frac{\Delta E}{E} \propto E_{\text{pass}} \left[\frac{W_{\text{slit}}}{2 \cdot r} + \frac{\alpha^2}{2} \right]$$

- **Resolution** set by: slitwidth, hemisphere's mean radius and angular acceptance.
- **Transmission** limited by slitwidth and angular acceptance.

- Measures energy and only one momentum component (k_x)



- Get rid of slit,
- Lens system for angular resolution,
- Energy analyzed by time of flight,
- Position sensitive detector



- Measures energy and two momentum components (k_x and k_y)

ARTOF (+ Themis, LBNL,...)

King, ... et al. *Phys. Rev. Lett.* **107**, 096802 (2011)

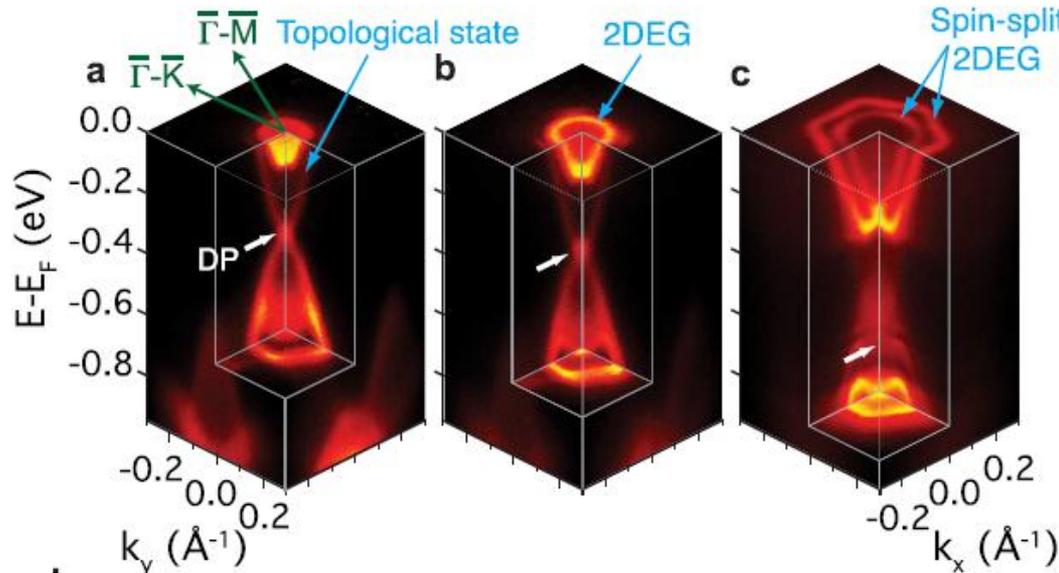


UPPSALA
UNIVERSITET

ARToF spectroscopy: Bi_2Se_3

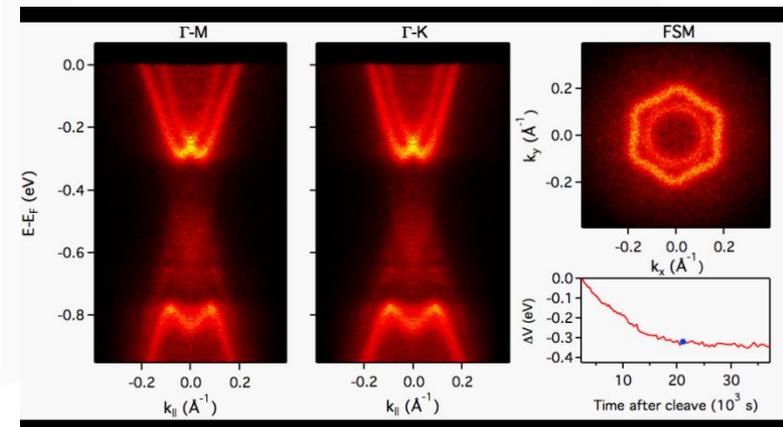
High resolution & High transmission

- Full 3d-bandstructure in ~ 15 min



$$E - E_F = -810 \text{ meV}$$

Time-resolved video in UHV

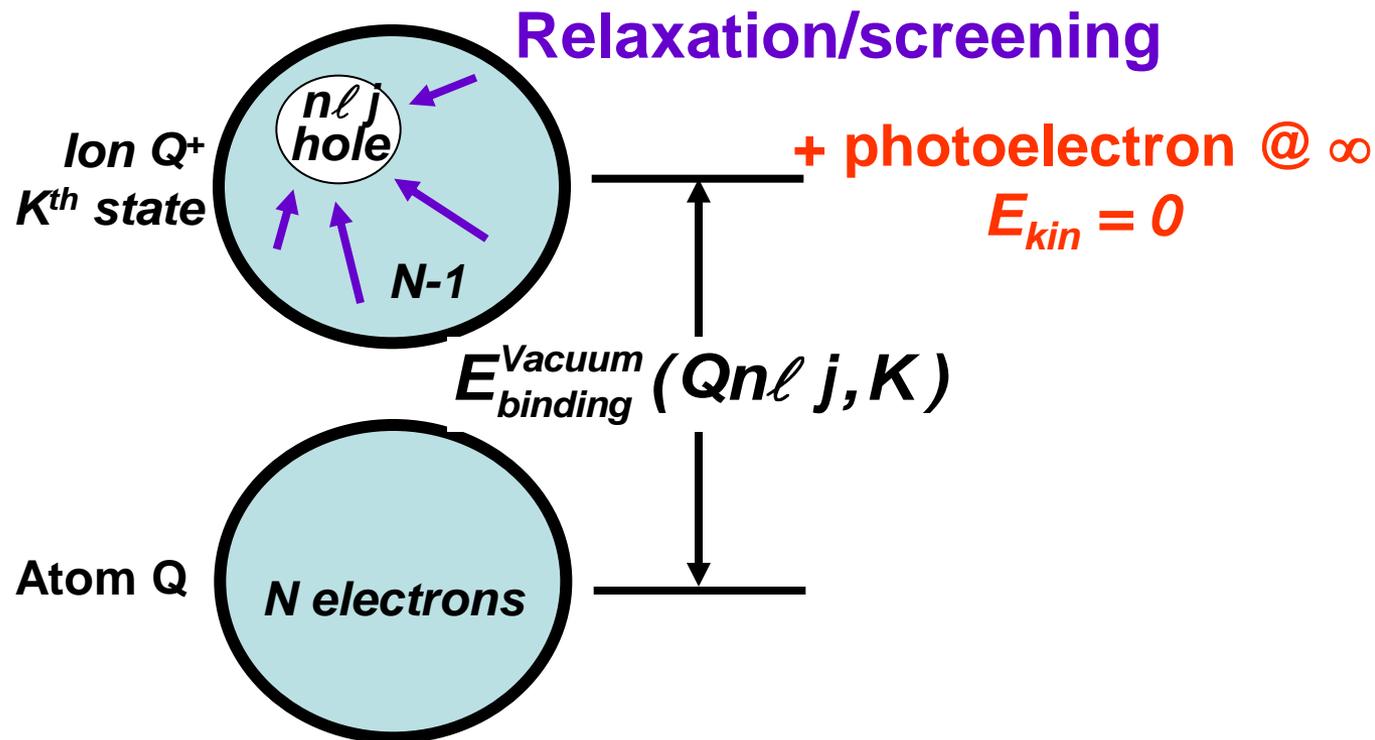


Basic energetics—Many e⁻ picture

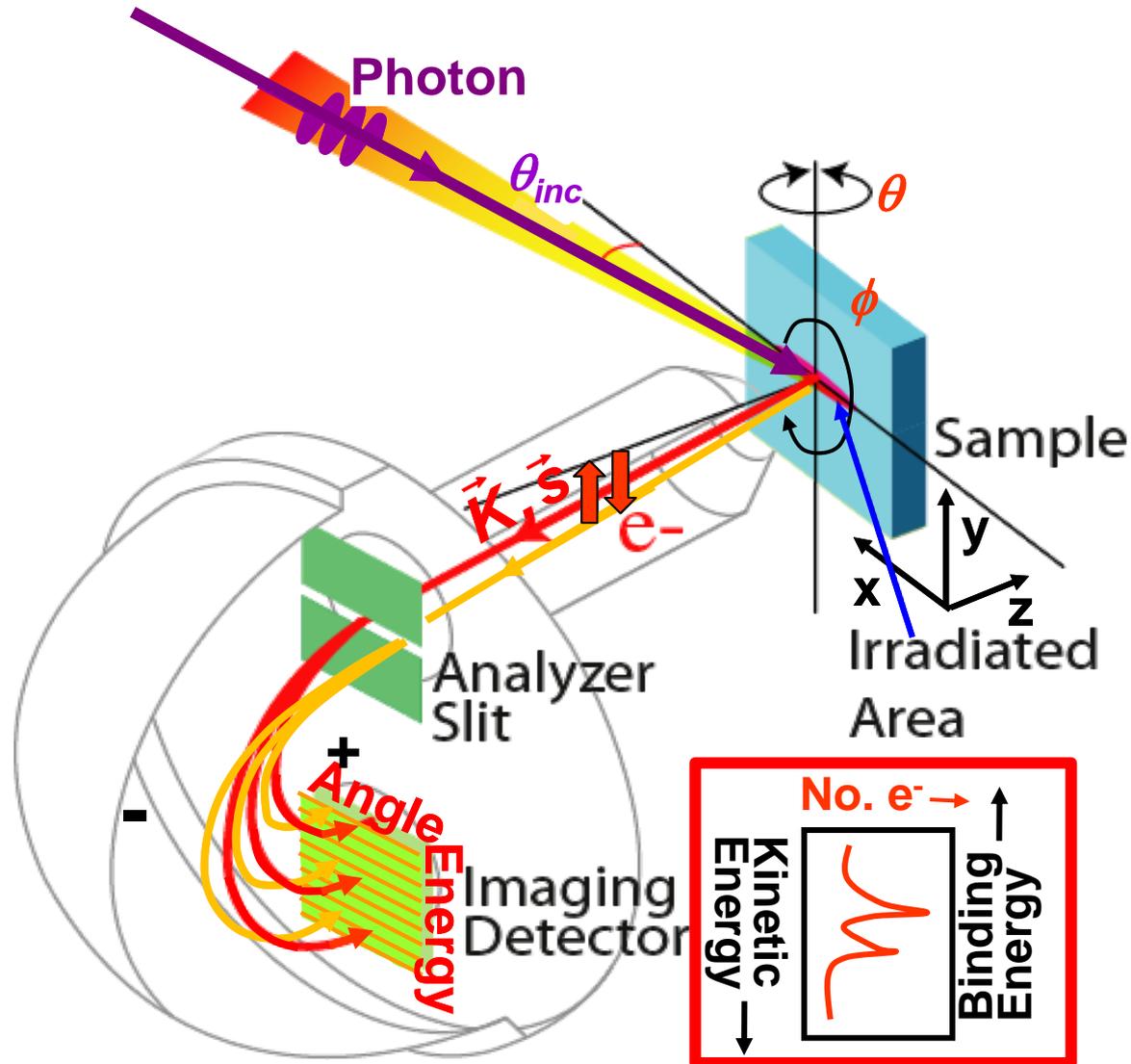
$$h\nu = E_{\text{binding}}^{\text{Vacuum}} + E_{\text{kinetic}} = E_{\text{binding}}^{\text{Fermi}} + \varphi_{\text{spectrometer}} + E_{\text{kinetic}}$$

↓

$$E_{\text{binding}}^{\text{Vacuum}}(Qn\ell j, K) = E_{\text{final}}(N-1, Qn\ell j \text{ hole}, K) - E_{\text{initial}}(N)$$

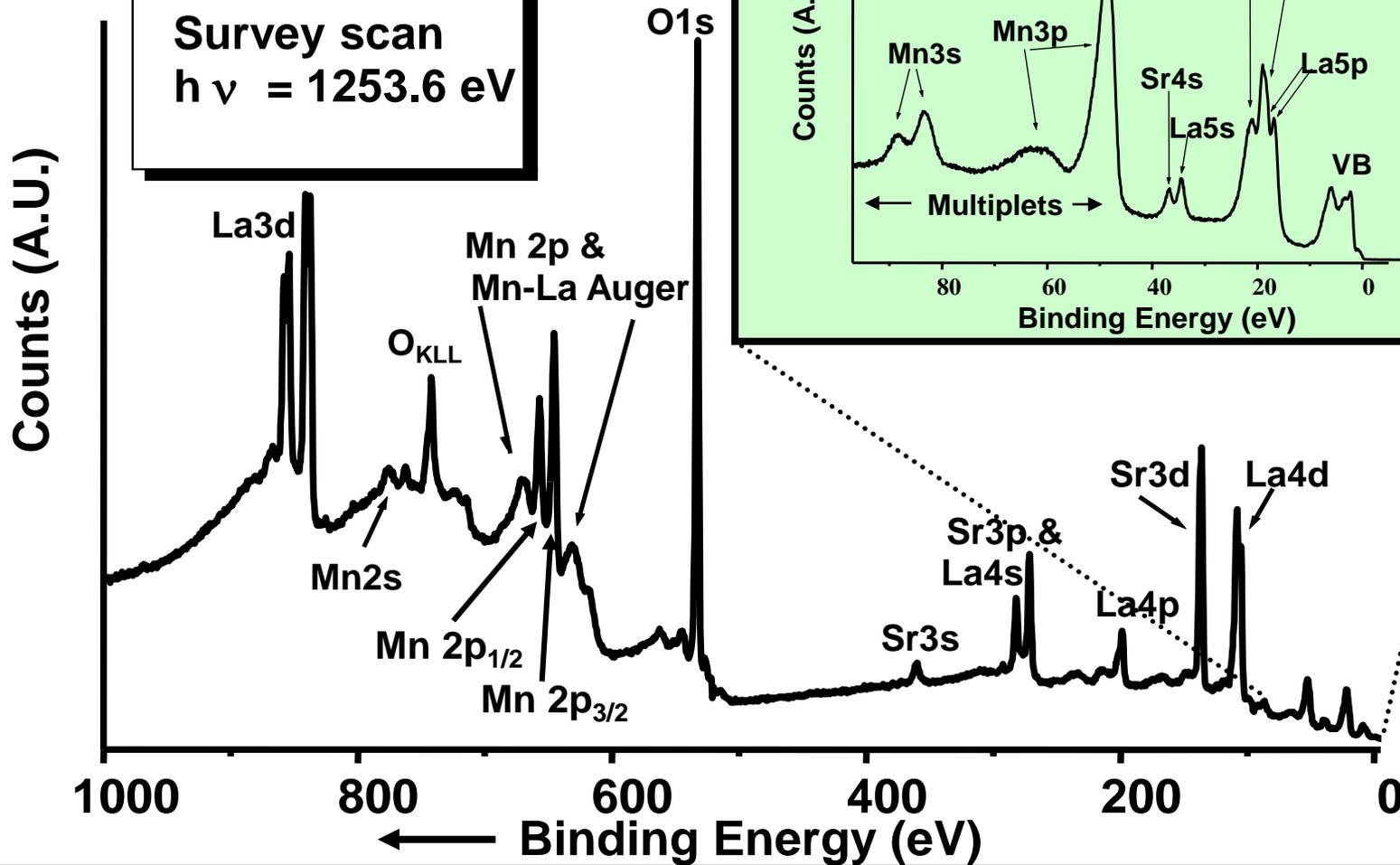


Typical experimental geometry for energy- and angle-resolved photoemission measurements





$x = 0.4$
Survey scan
 $h\nu = 1253.6 \text{ eV}$



Basic Concepts and Experiments

Core-Level Photoemission



Intensities and Quantitative Analysis, the 3-Step Model
Varying Surface and Bulk Sensitivity
Chemical shifts
Multiplet Splittings
Electron Screening and Satellite Structure
Magnetic and Non-Magnetic Dichroism
Resonant Photoemission
Photoelectron Diffraction and Holography

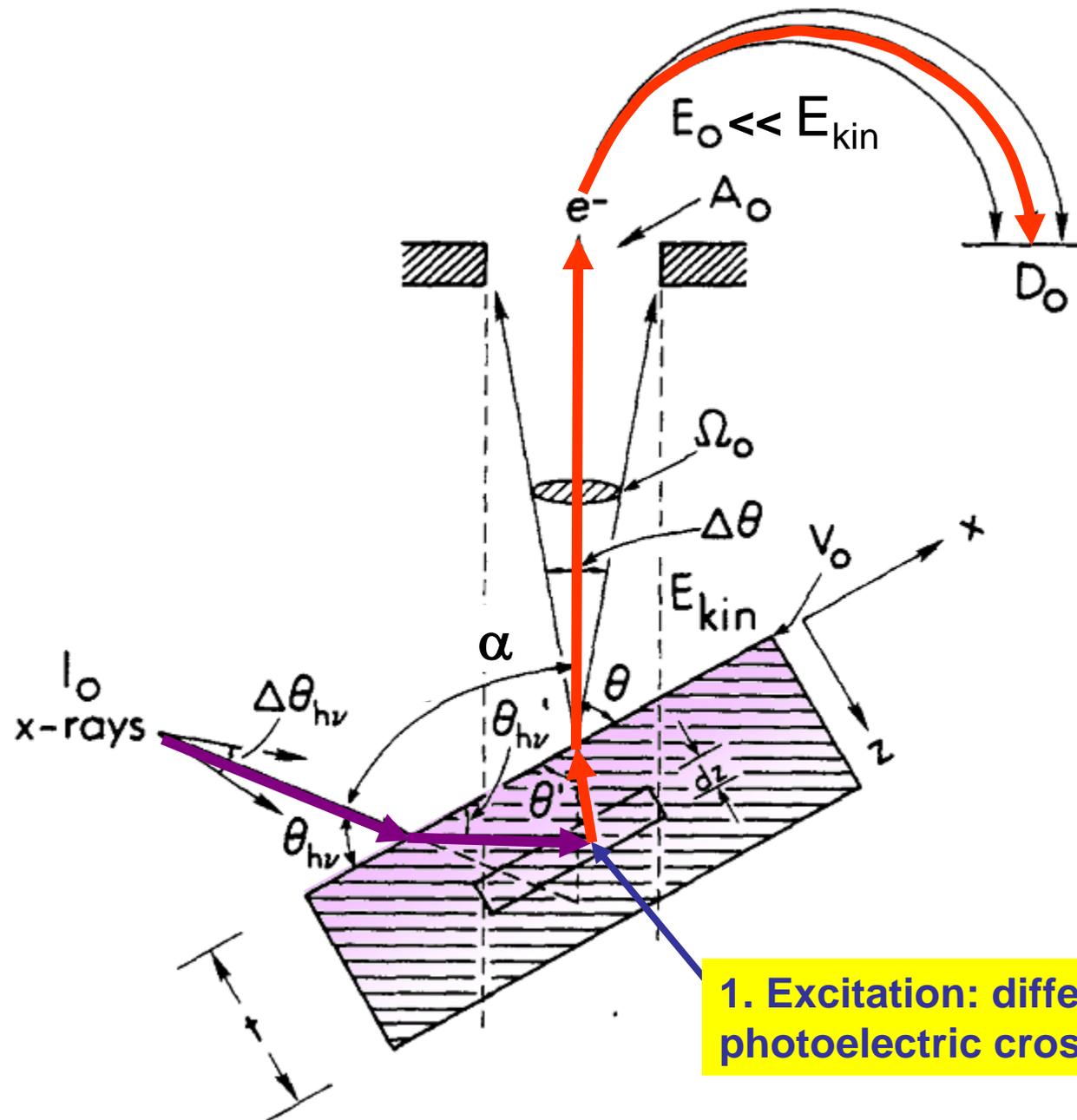
Valence-Level Photoemission

Band-Mapping in the Ultraviolet Photoemission Limit
Densities of States in the X-Ray Photoemission Limit

Some New Directions

Photoemission with Hard X-Rays (throughout lectures)
Photoemission with Standing Wave Excitation
Photoemission with Spatial Resolution/Photoelectron Microscopy
Temporal Resolution
@ Higher Pressures

PHOTOELECTRON INTENSITIES—THE 3-STEP MODEL



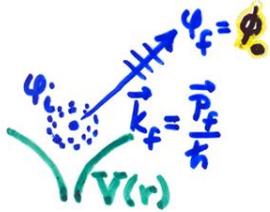
1. Excitation: differential photoelectric cross section ($d\sigma/d\Omega$)

PHOTOELECTRON EMISSION -

BASIC MATRIX ELEMENTS + SELECTION RULES:

- ATOMIC-LIKE (LOCALIZED) STATES \Rightarrow CORE:

$$\psi_i(\vec{r}) = \psi_{n_i, l_i, m_i}(r, \theta, \phi) = R_{n_i, l_i}(r) Y_{l_i, m_i}(\theta, \phi)$$



$$\psi_f(\vec{r}, \vec{k}_f) = \psi_{E_f}(\vec{r}, \vec{k}_f)$$

$$= 4\pi \sum_{l_f, m_f} i^{l_f} e^{-i\delta_{l_f}} Y_{l_f, m_f}^*(\theta, \phi) Y_{l_i, m_i}(\theta, \phi) R_{E_f, l_f}(r)$$

PHASE SHIFT OF l_f WAVE IN $V(r)$

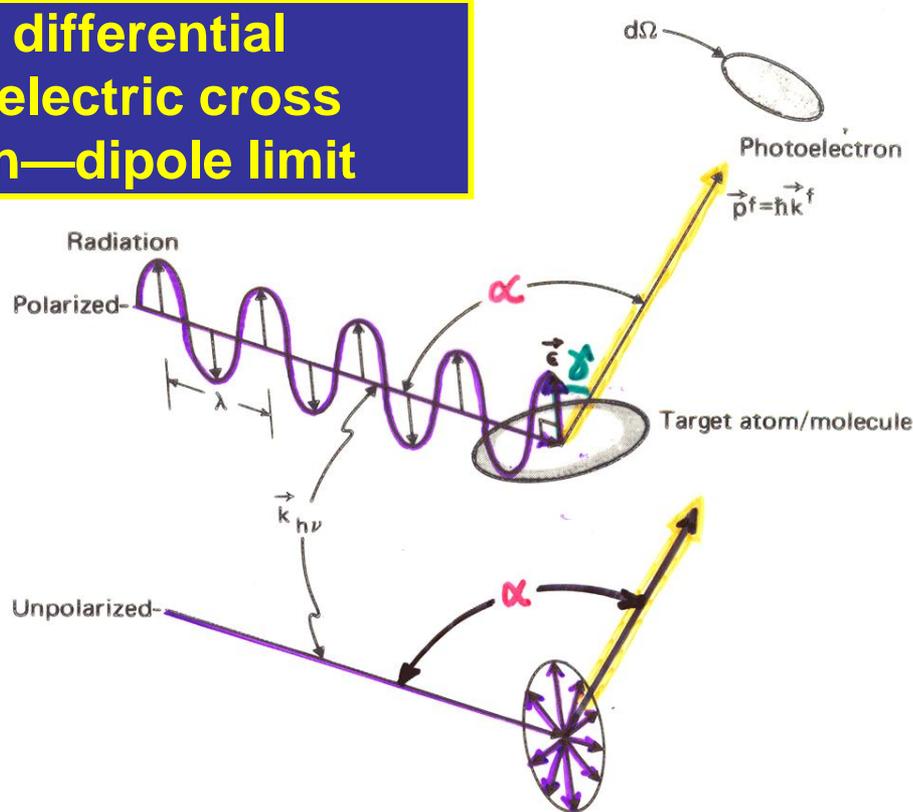
DIPOLE APPROX.: INT. $\propto |\langle \psi_f | \hat{E} \cdot \vec{r} | \psi_i \rangle|^2 = |\hat{E} \cdot \langle \psi_f | \vec{r} | \psi_i \rangle|^2$

EQUIVALENT
WITHIN CONSTANT
FACTOR



- < $\Delta l = l_f - l_i = \pm 1$
TWO CHANNELS
- < $\Delta m = m_f - m_i = 0, \pm 1$
LINEAR POLARIZ.
- < $\Delta m = \pm 1$, CIRCULAR POLARIZATION

The differential photoelectric cross section—dipole limit



FOR ATOMIC-LIKE EMISSION:

LIN.

$$\text{POLARIZED: } \frac{d\sigma_{nl}(E_f)}{d\Omega} = \frac{\sigma_{nl}(E_f)}{4\pi} \left[1 + \beta_{nl}(E_f) \left(\frac{3}{2} \cos^2 \gamma - \frac{1}{2} \right) \right]$$

$$\text{UNPOLARIZED: } \frac{d\sigma_{nl}(E_f)}{d\Omega} = \frac{\sigma_{nl}(E_f)}{4\pi} \left[1 + \frac{1}{2} \beta_{nl}(E_f) \left(\frac{3}{2} \sin^2 \alpha - 1 \right) \right]$$

Figure 7 -- General geometry for defining the differential cross section $d\sigma/d\Omega$, showing both polarized and unpolarized incident radiation. The polarization vector \vec{e} is parallel to the electric field \vec{E} of the radiation. In order for the dipole approximation to be valid, the radiation wave length λ should be much larger than typical target dimensions (that is, the opposite of what is shown here).

WITH:

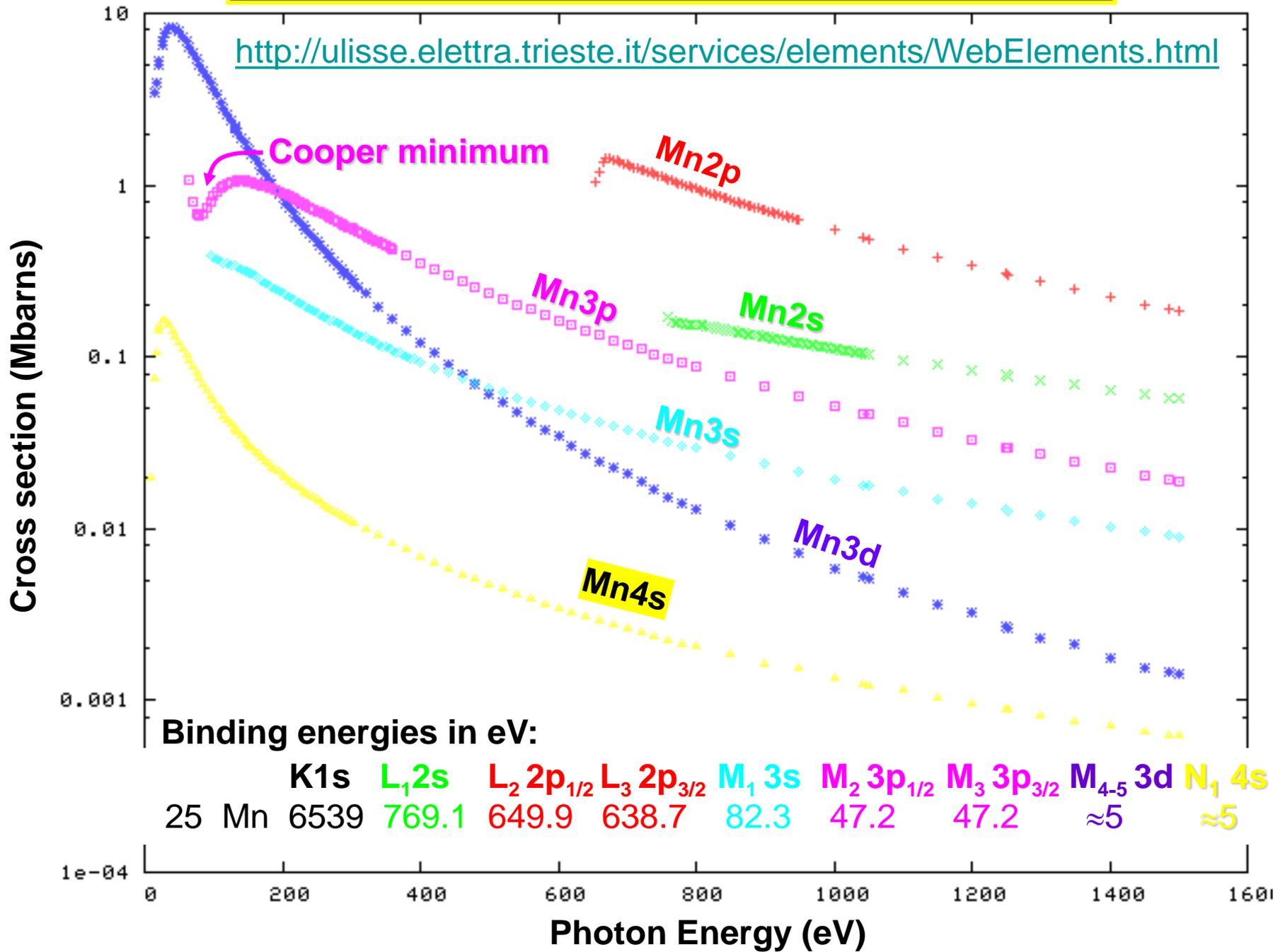
σ_{nl} = TOTAL CROSS SECTION

β_{nl} = ASYMMETRY PARAMETER

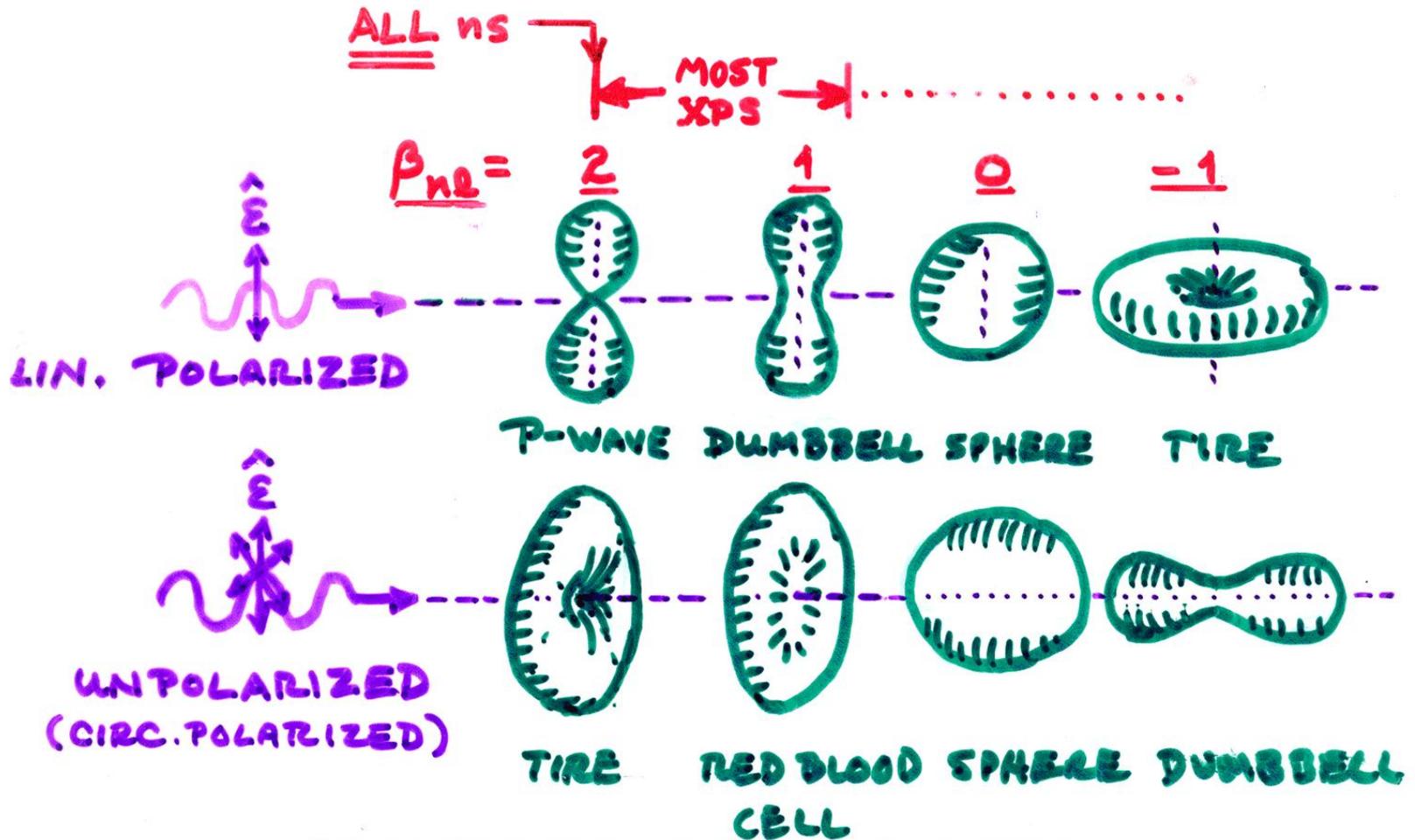
σ_{nl}, β_{nl} TABULATIONS IN: GOLDBERG ET AL., J. ELECT. SPECT.
 & YEH, LINDAU, AT. NUC. DATA 22, 1 ('85) \ 21, 285 ('91)

<http://ulisse.elettra.trieste.it/services/elements/WebElements.html>

PHOTOELECTRIC CROSS SECTIONS FOR Mn



The differential subshell photoelectric cross section—dipole limit



TRANGE OF SHAPES OF $\frac{d\sigma}{d\Omega}$

→ Higher energies: Nondipolar Angular Distributions

$$\frac{d\sigma(h\nu)}{d\Omega} = \frac{\sigma(h\nu)}{4\pi} \left[1 + \frac{\beta}{2} (3 \cos^2 \theta - 1) + (\delta + \gamma \cos^2 \theta) \sin \theta \cos \phi \right]$$

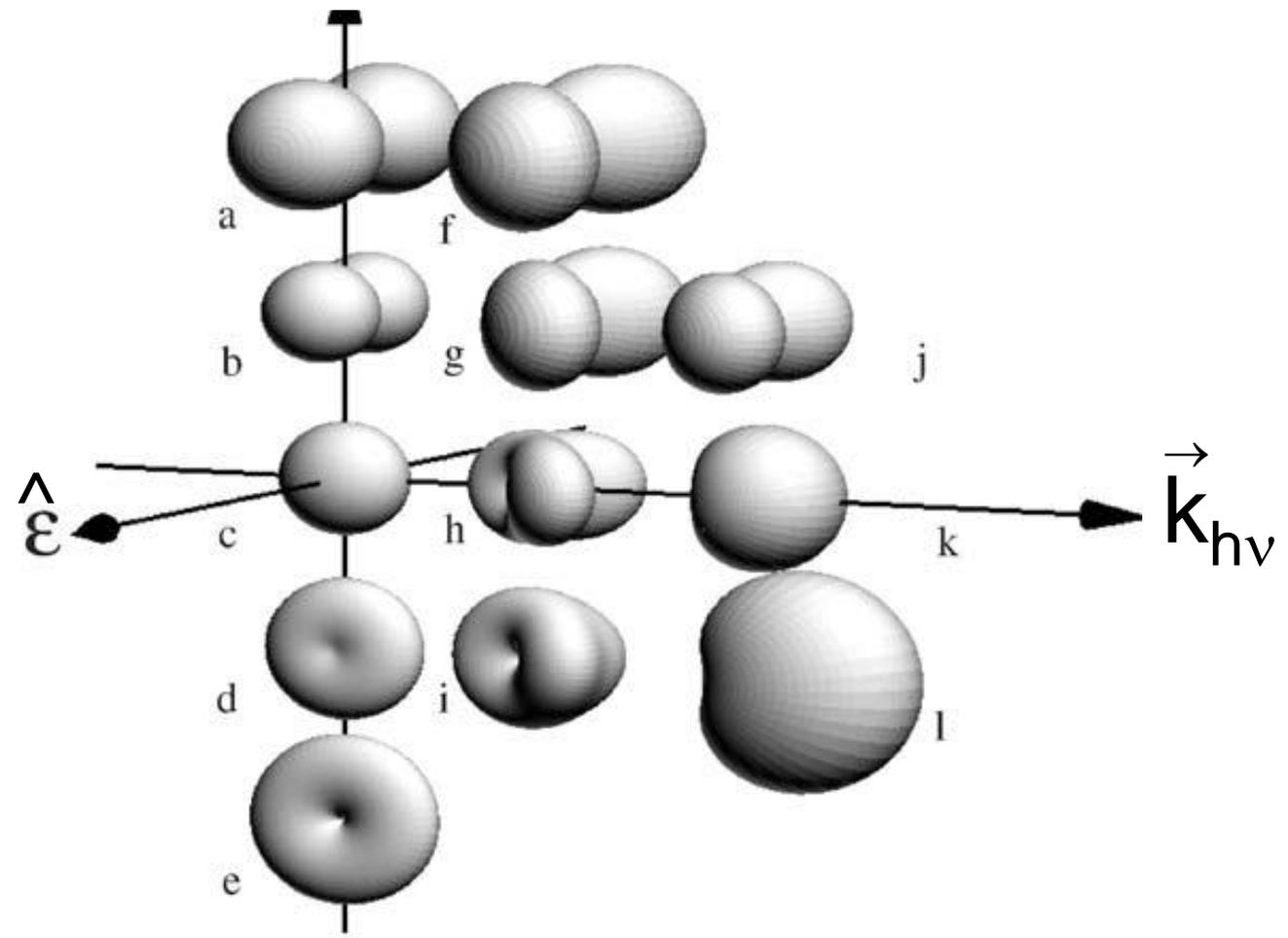
$\beta = 2$

1

0

-0.5

-1

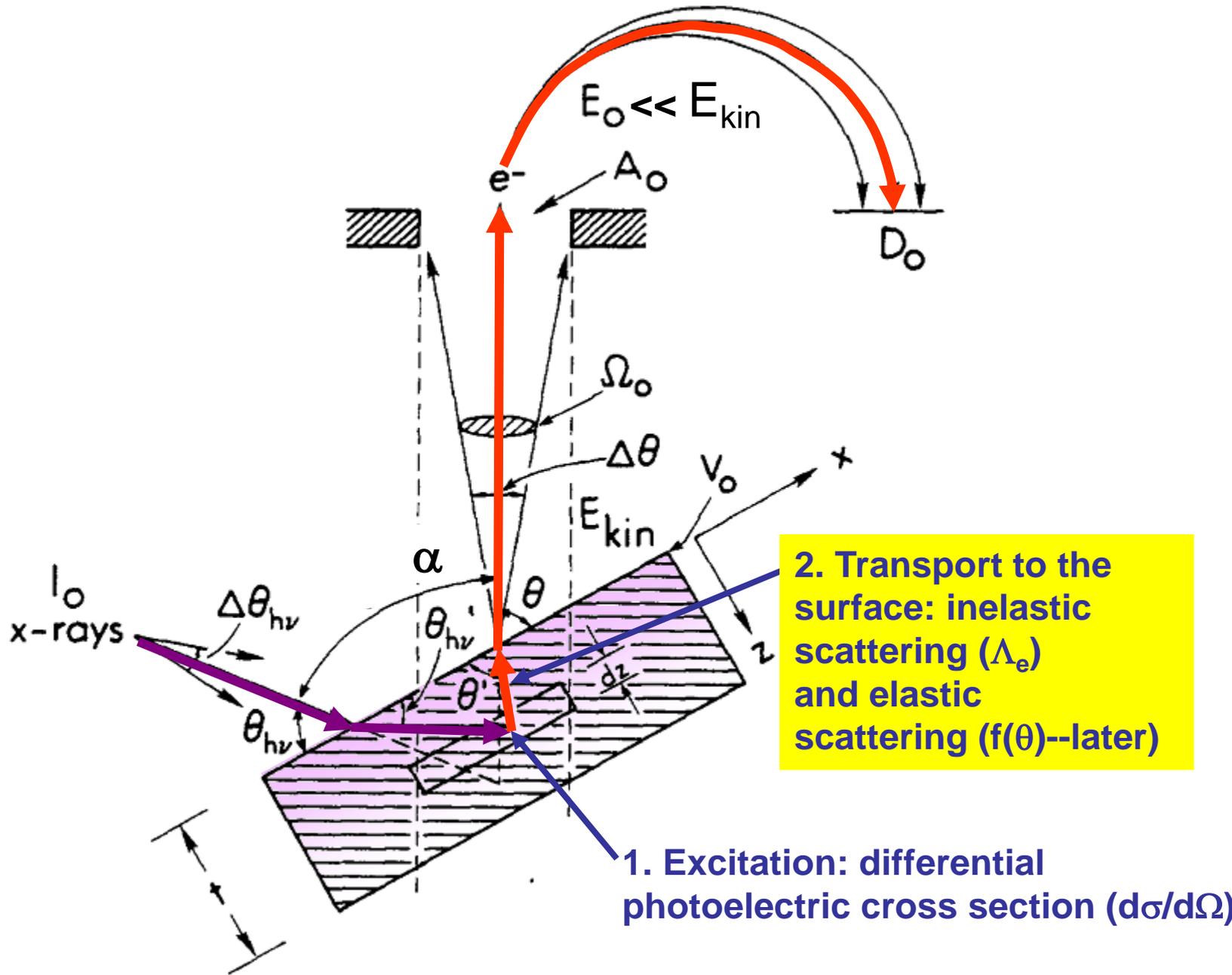


$\gamma, \delta = 0$

γ large

δ large

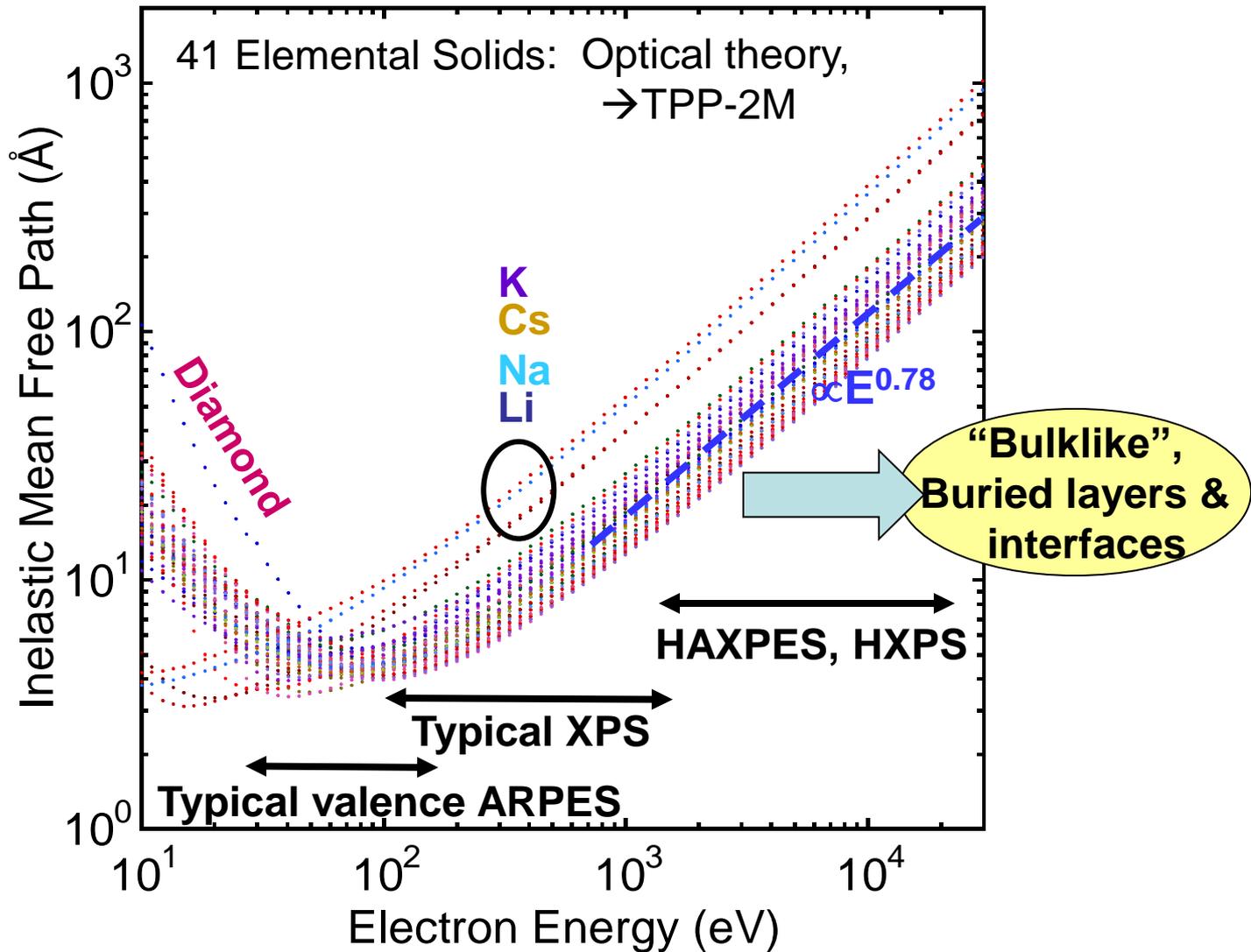
PHOTOELECTRON INTENSITIES—THE 3-STEP MODEL



2. Transport to the surface: inelastic scattering (Λ_e) and elastic scattering ($f(\theta)$ --later)

1. Excitation: differential photoelectric cross section ($d\sigma/d\Omega$)

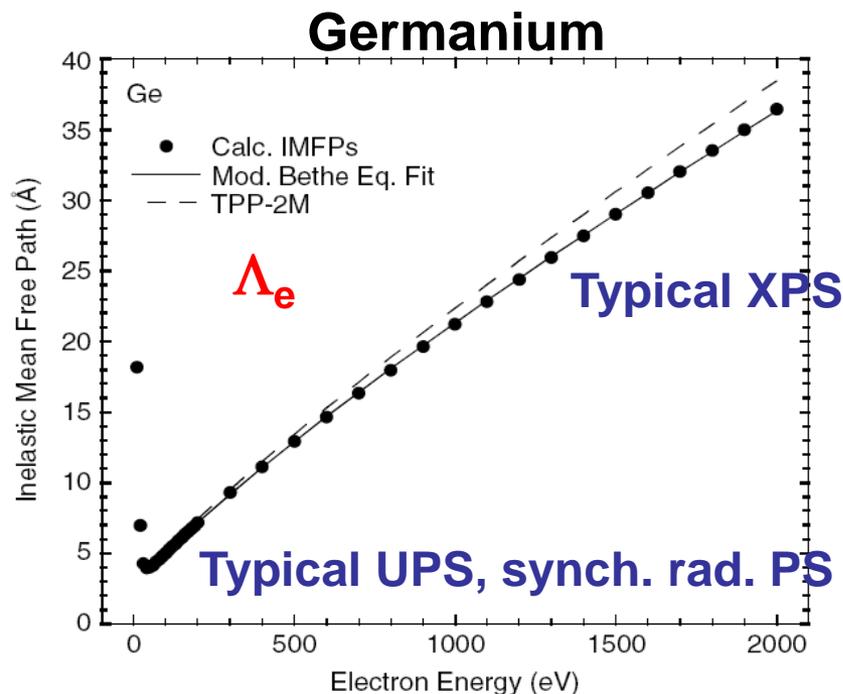
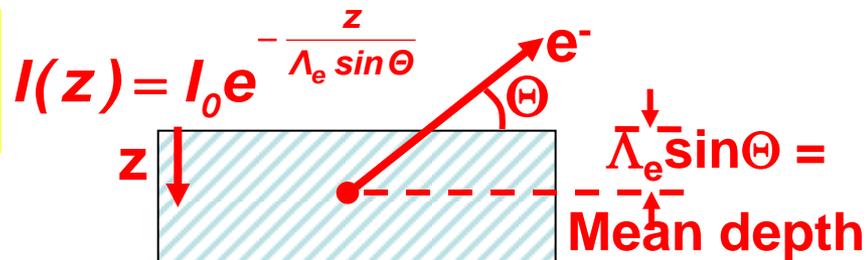
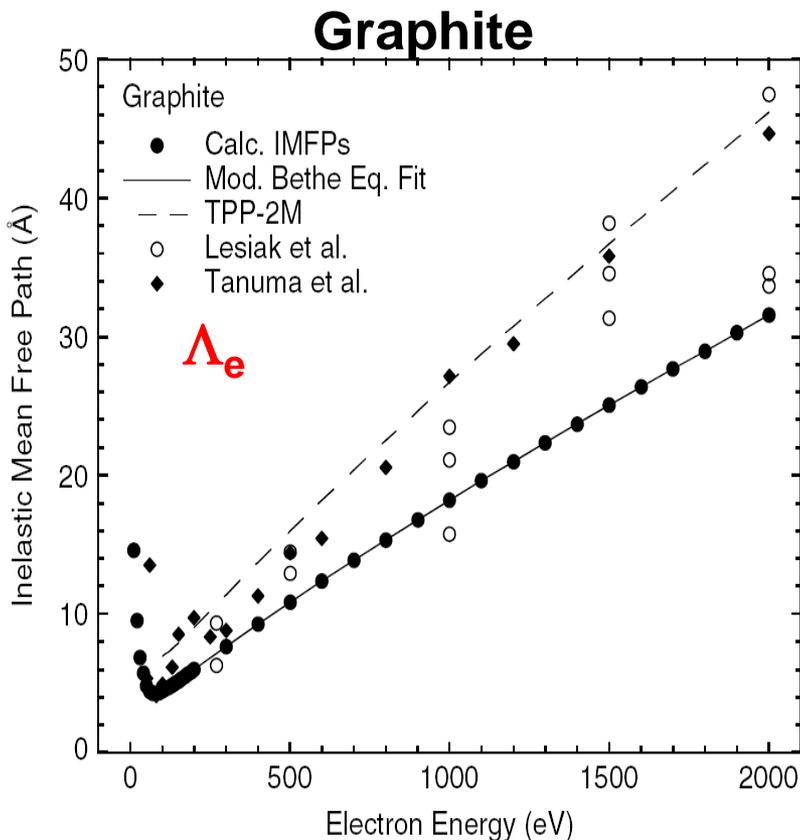
Surface sensitivity and why we may want to go to 5-10 keV in XPS



Electron inelastic attenuation length in solids—the “universal curve”

Photoemission is a surface sensitive experiment

Changing angle:
1st way to vary surface sensitivity



Changing photon energy:
2nd way to vary surface sensitivity

Basic Concepts and Experiments

Core-Level Photoemission



Intensities and Quantitative Analysis, the 3-Step Model
Varying Surface and Bulk Sensitivity
Chemical shifts
Multiplet Splittings
Electron Screening and Satellite Structure
Magnetic and Non-Magnetic Dichroism
Resonant Photoemission
Photoelectron Diffraction and Holography

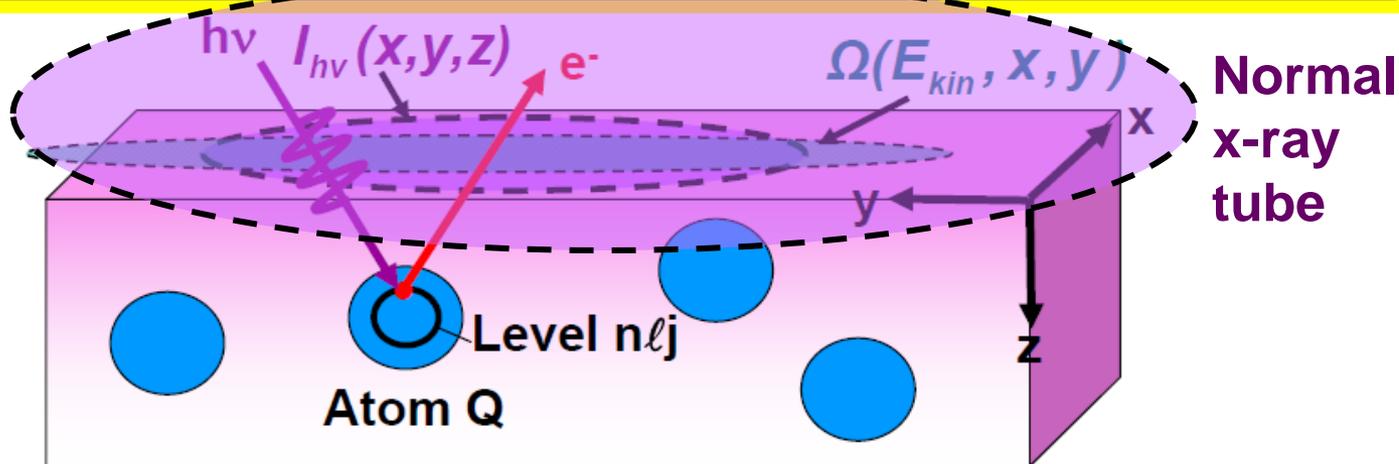
Valence-Level Photoemission

Band-Mapping in the Ultraviolet Photoemission Limit
Densities of States in the X-Ray Photoemission Limit

Some New Directions

Photoemission with Hard X-Rays (throughout lectures)
Photoemission with Standing Wave Excitation
Photoemission with Spatial Resolution/Photoelectron Microscopy
Temporal Resolution
@ Higher Pressures

CORE PHOTOELECTRON INTENSITIES AND COMPOSITION



$$I(Qn\ell j) =$$

$$C \int_0^{\infty} I_{h\nu}(x,y,z) \rho_Q(x,y,z) \frac{d\sigma_{Qn\ell j}(h\nu)}{d\Omega} \exp\left[-\frac{z}{\Lambda_e(E_{kin}) \sin\theta}\right] \Omega(E_{kin}, x, y) dx dy dz$$

$$I_{h\nu}(x,y,z) = \text{x-ray flux}$$

$\rho_Q(x,y,z)$ = density of atoms Q → quantitative analysis

$\frac{d\sigma_{Qn\ell j}(h\nu)}{d\Omega}$ = **energy-dependent** differential photoelectric cross section for subshell $Qn\ell j$

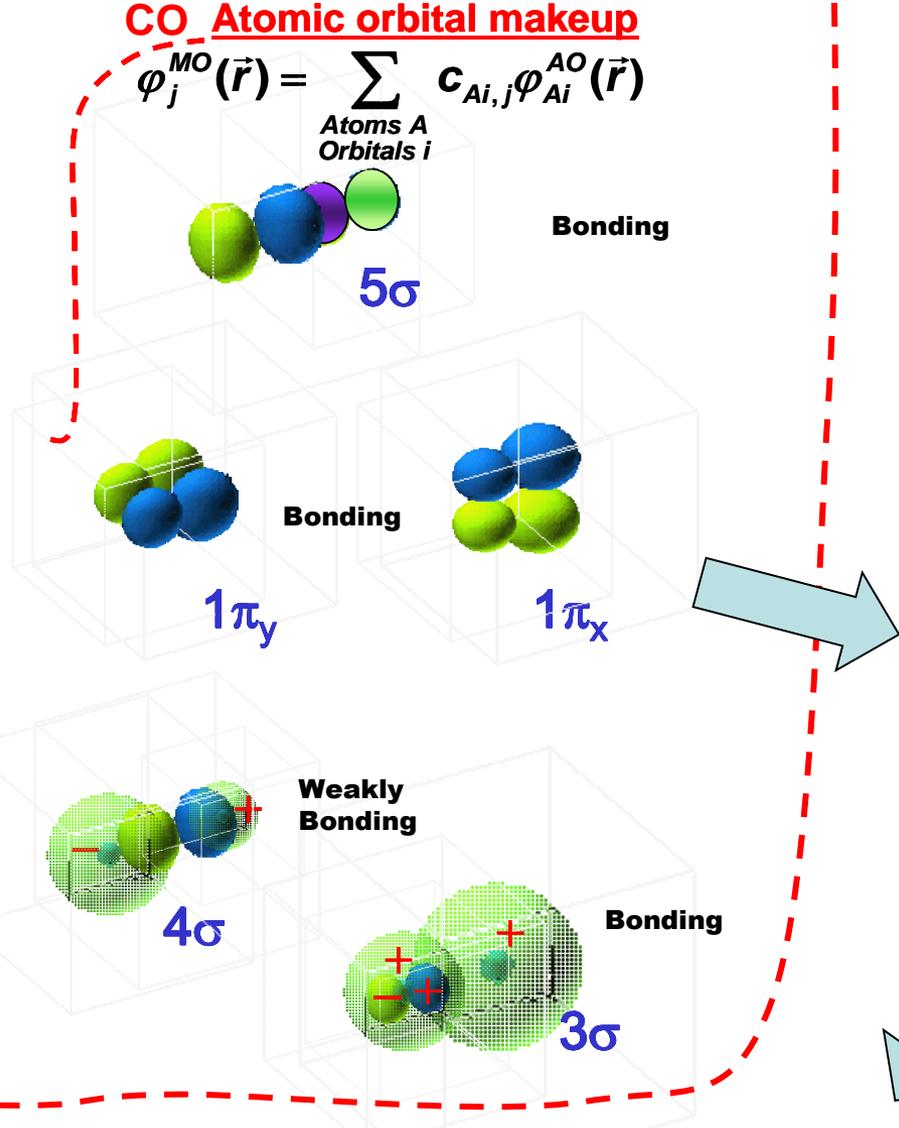
$\Lambda_e(E_{kin})$ = **energy-dependent** inelastic attenuation length

→ Effective Attenuation Length (EAL) → Mean Emission Depth (MED)

$\Omega(E_{kin}, x, y)$ = **energy-dependent** spectrometer acceptance solid angle

CO Atomic orbital makeup

$$\varphi_j^{MO}(\vec{r}) = \sum_{\text{Atoms } A} \sum_{\text{Orbitals } i} c_{Ai,j} \varphi_{Ai}^{AO}(\vec{r})$$



$$\varphi_{\vec{k}}^{TB}(\vec{r}) = \text{a Bloch function} = u_{\vec{k}}^{TB}(\vec{r}) e^{i\vec{k} \cdot \vec{r}}$$

$$\propto \frac{1}{N^{1/2}}$$

$$\sum_{j=1 \dots N \text{ unit cells at } \vec{R}_j}$$

$$e^{i\vec{k} \cdot \vec{R}_j}$$

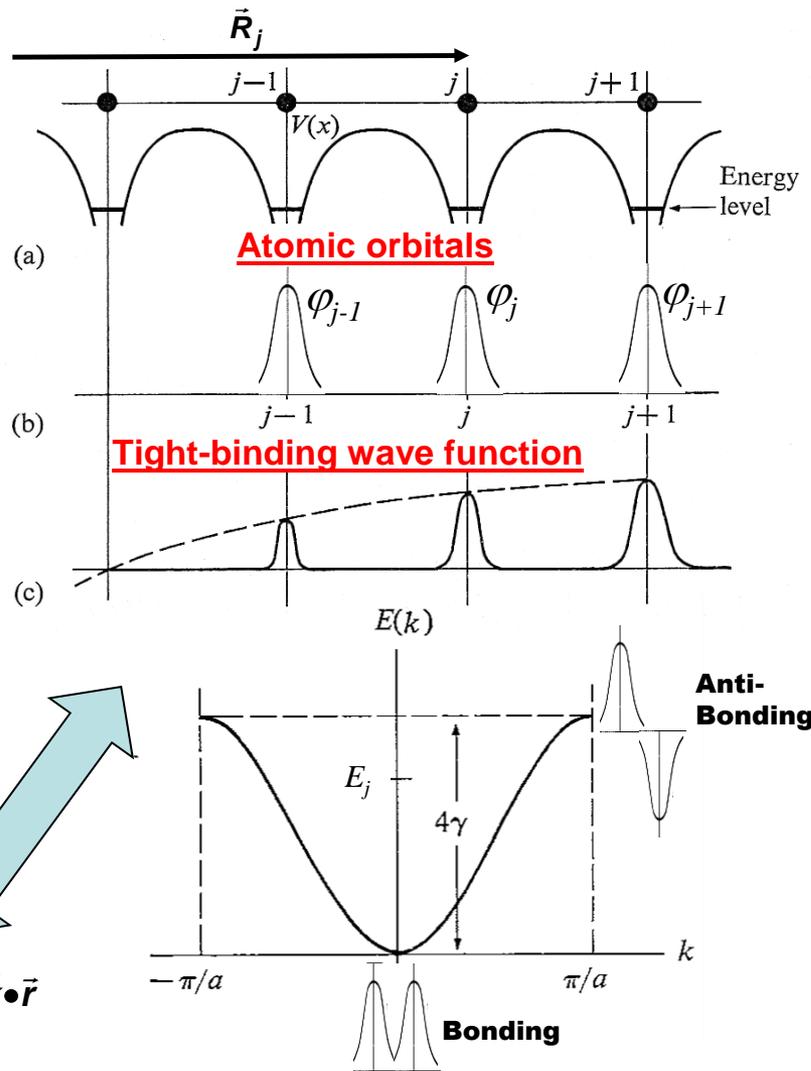
$$\sum_{Ai}$$

$$c_{Ai,\vec{k}} \varphi_{Ai}^{AO}(\vec{r} - \vec{R}_j)$$

Ai = basis set of AOs in unit cell

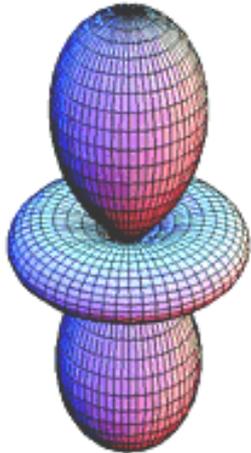
Solid state tight-binding approach

Crystal potential-1D

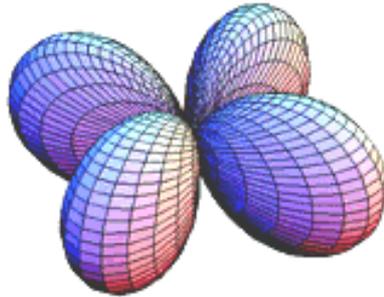


The d orbitals

e_g

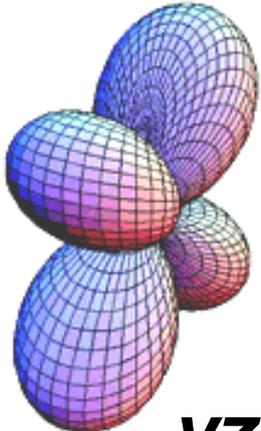


$3z^2-r^2$

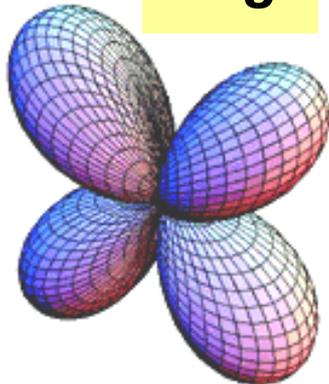


x^2-y^2

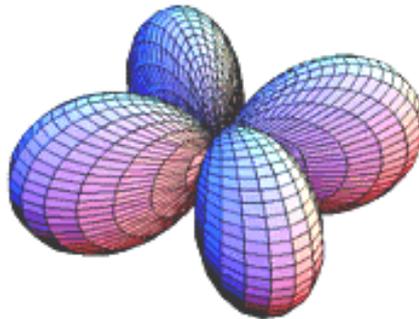
t_{2g}



yz

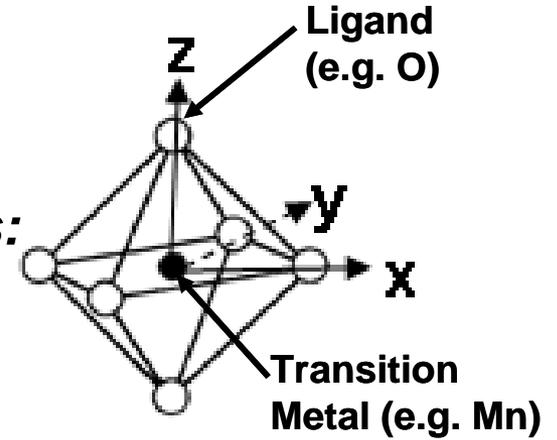


zx



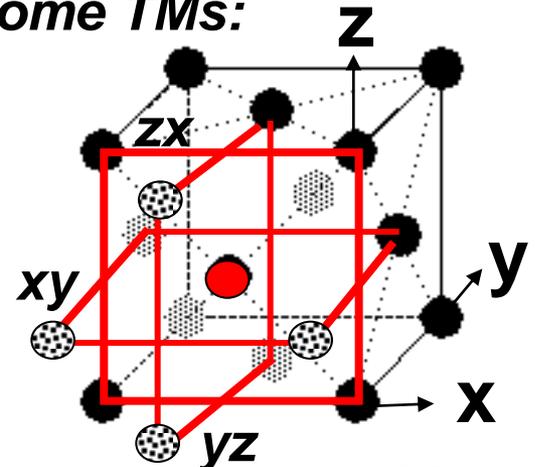
xy

Some Transition-Metal (TM) compounds:



e_g and t_{2g} not equivalent in octahedral (cubic) environment

Some TMs:



Face-centered cubic—12 nearest neighbors

Electronic bands and density of states for a transition metal-Copper

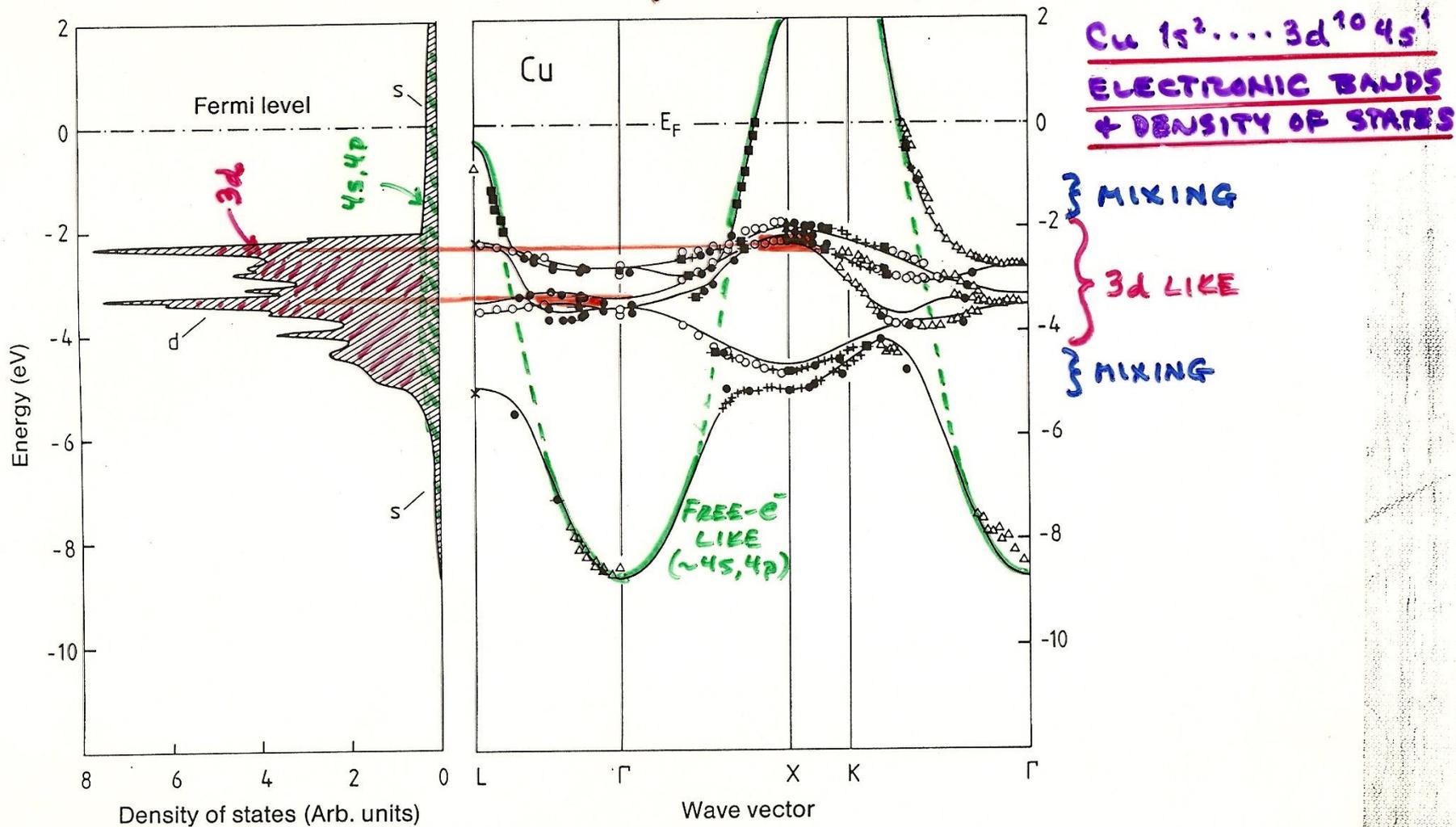
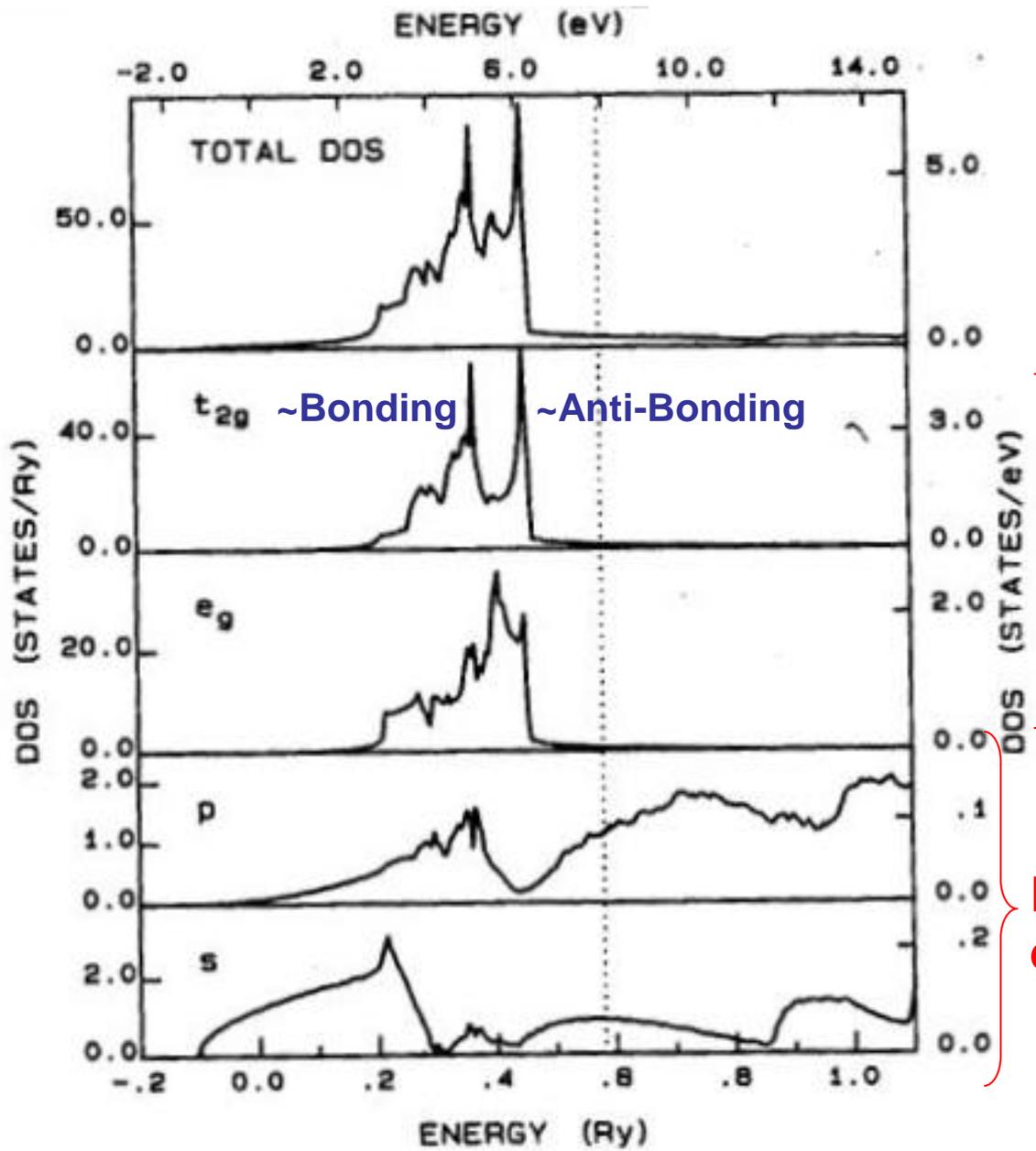


Fig. 7.12. Bandstructure $E(k)$ for copper along directions of high crystal symmetry (right). The experimental data were measured by various authors and were presented collectively by Courths and Hüfner [7.4]. The full lines showing the calculated energy bands and the density of states (left) are from [7.5]. The experimental data agree very well, not only among themselves, but also with the calculation

Copper projected densities of states-total and by orbital type

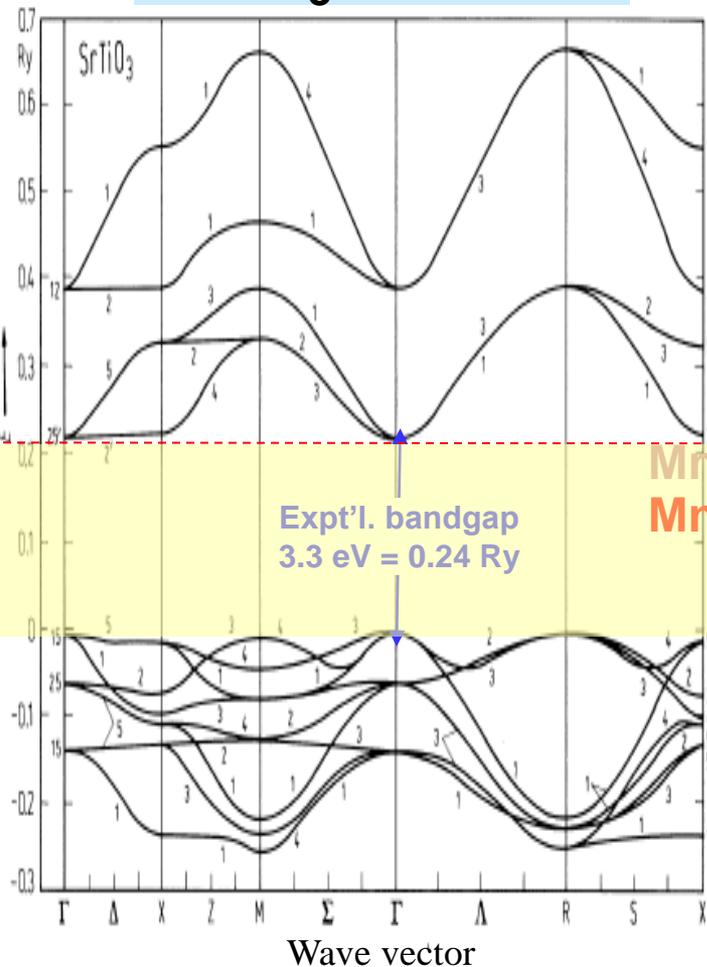


More localized 3d-like

Delocalized free electron-like

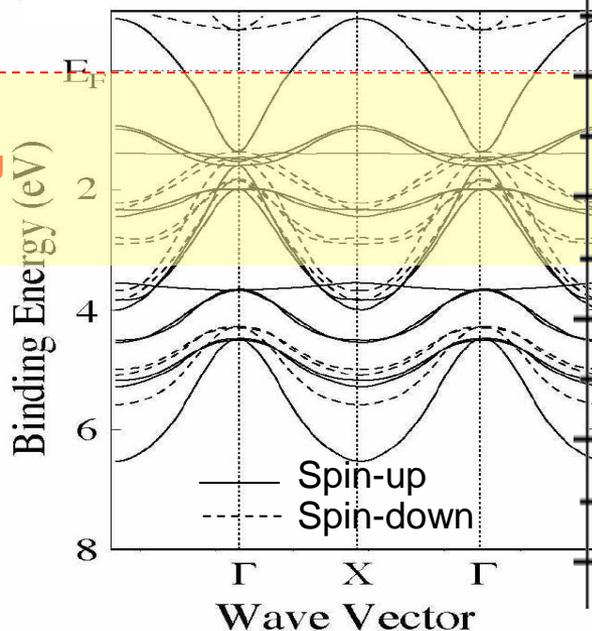
SrTiO₃ and La_{0.67}Sr_{0.33}MnO₃ band structures and DOS

SrTiO₃-insulator



Mattheiss, PRB 6, 4718 (1972)

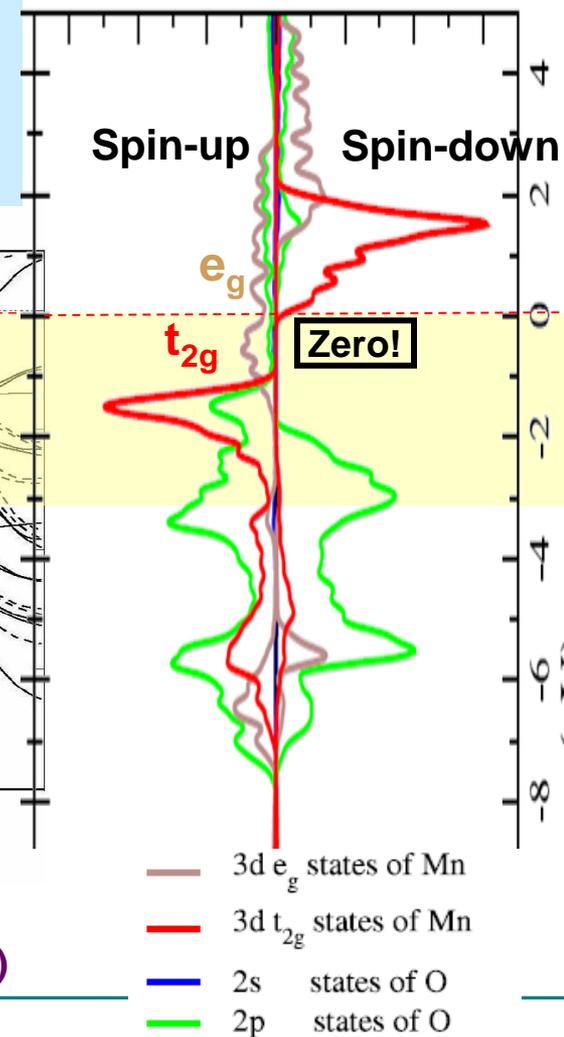
La_{0.67}Sr_{0.33}MnO₃- Half-Metallic Ferromagnetic metal



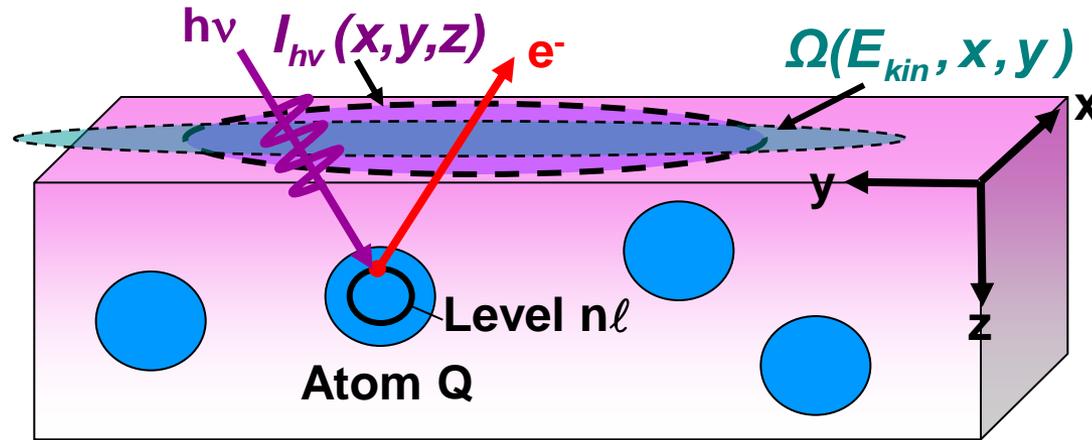
Chikamatsu et al.,
PRB 73, 195105 (2006)

Zheng, Binggeli, J. Phys.
Cond. Matt. 21, 115602 (2009)

Projected DOSs



VALENCE-BAND PHOTOELECTRON INTENSITIES AND DENSITIES OF STATES



$$I(E_{kin}, Qn\ell) =$$

$$C' \int_0^{\infty} I_{hv}(x, y, z) \rho_{Qn\ell}(E_b, x, y, z) \frac{d\sigma_{Qn\ell}(hv)}{d\Omega} \exp\left[-\frac{z}{\Lambda_e(E_{kin}) \sin\theta}\right] \Omega(E_{kin}, x, y) dx dy dz$$

$I_{hv}(x, y, z)$ = x-ray flux

$\rho_{Qn\ell}(E_b, x, y, z)$ = density of states, projected onto $Qn\ell$ character

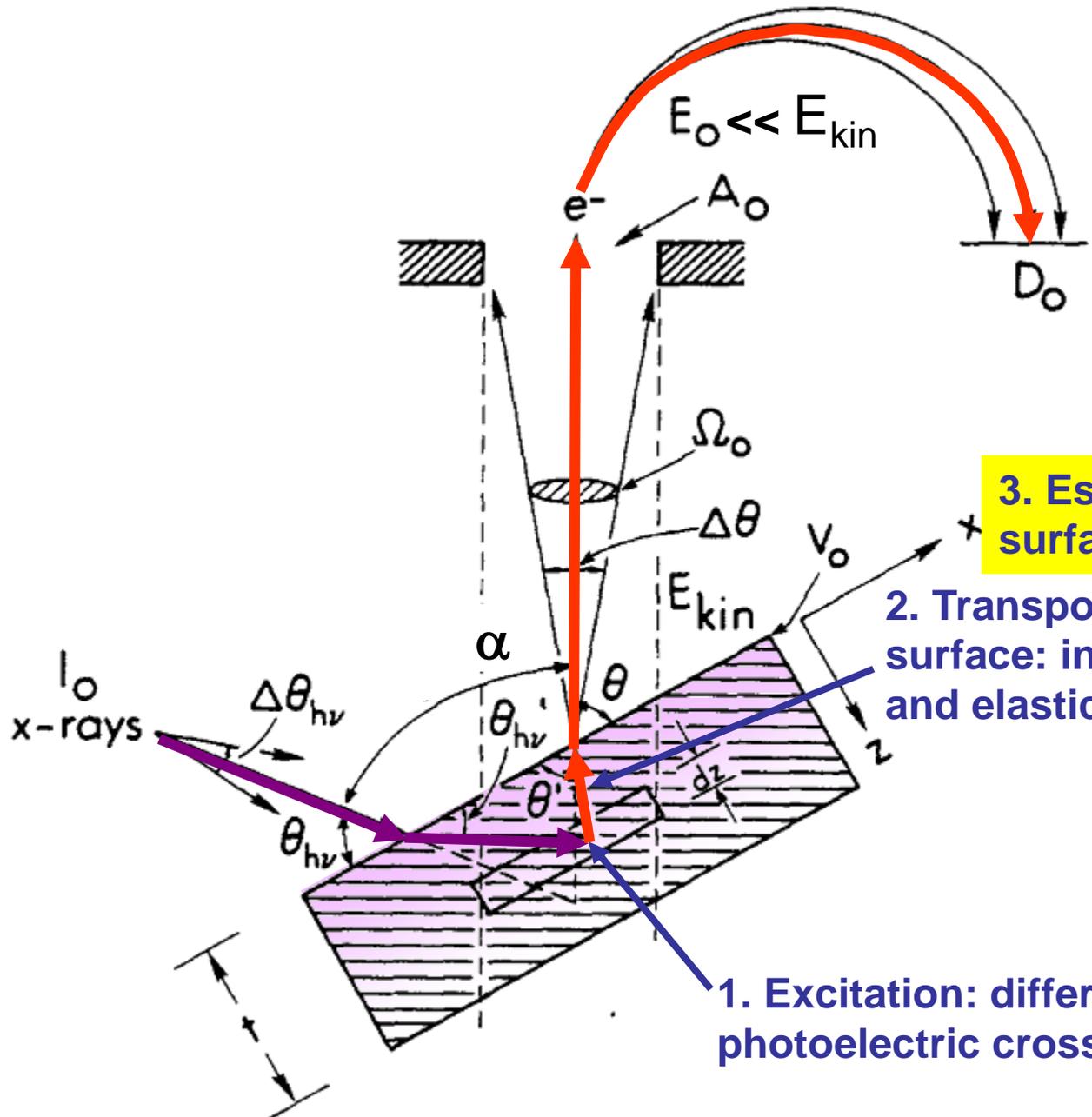
$\frac{d\sigma_{Qn\ell}(hv)}{d\Omega}$ = **energy-dependent** differential photoelectric cross section for subshell $Qn\ell$

$\Lambda_e(E_{kin})$ = **energy-dependent** inelastic attenuation length

→ Mean Emission Depth

$\Omega(E_{kin}, x, y)$ = **energy-dependent** spectrometer acceptance solid angle

PHOTOELECTRON INTENSITIES—THE 3-STEP MODEL

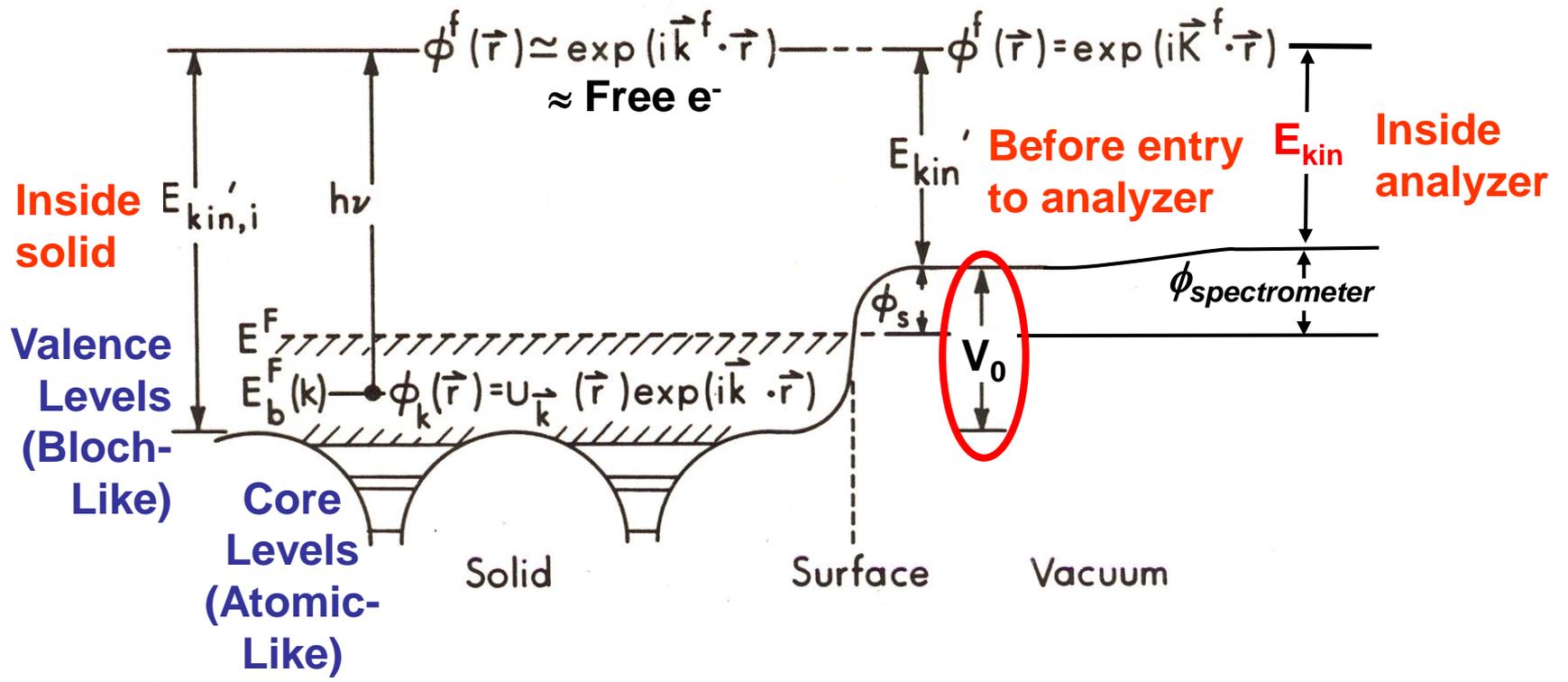


3. Escape across the surface barrier (V_0)

2. Transport to the surface: inelastic (Λ_e) and elastic ($f(\theta)$) scattering

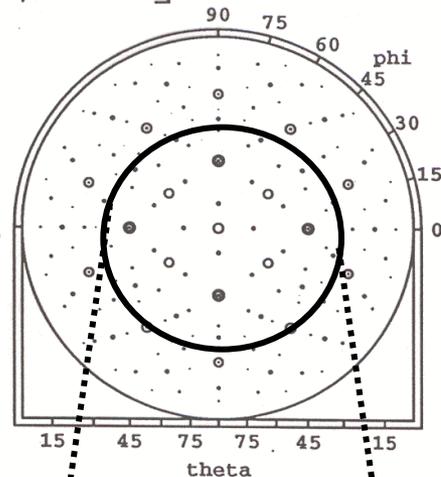
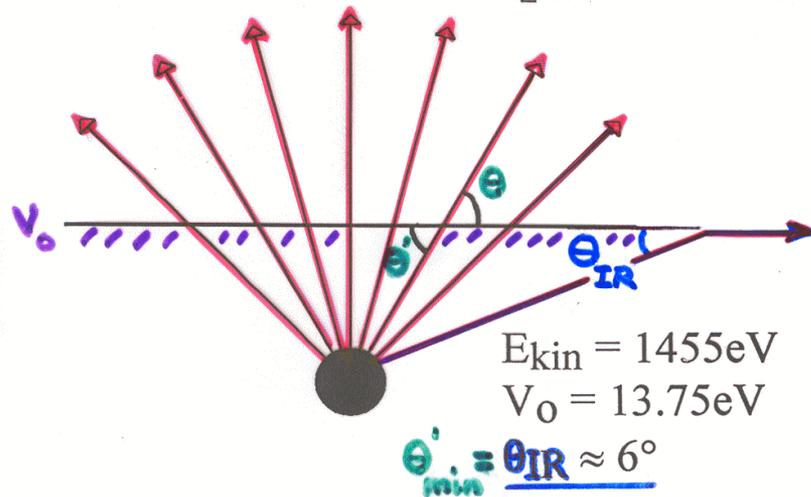
1. Excitation: differential photoelectric cross section ($d\sigma/d\Omega$)

One-Electron Picture of Photoemission from a Surface

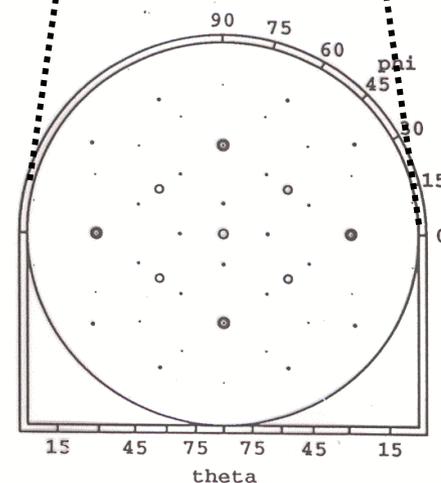
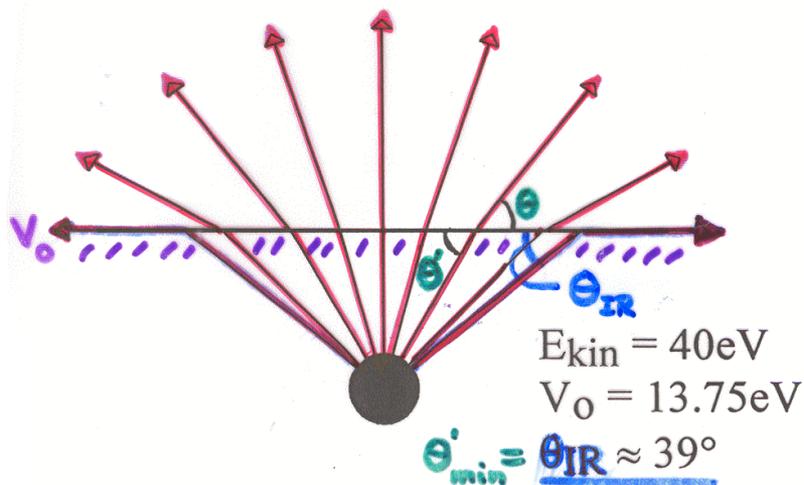


Electron Refraction at the Surface Due to the Inner Potential

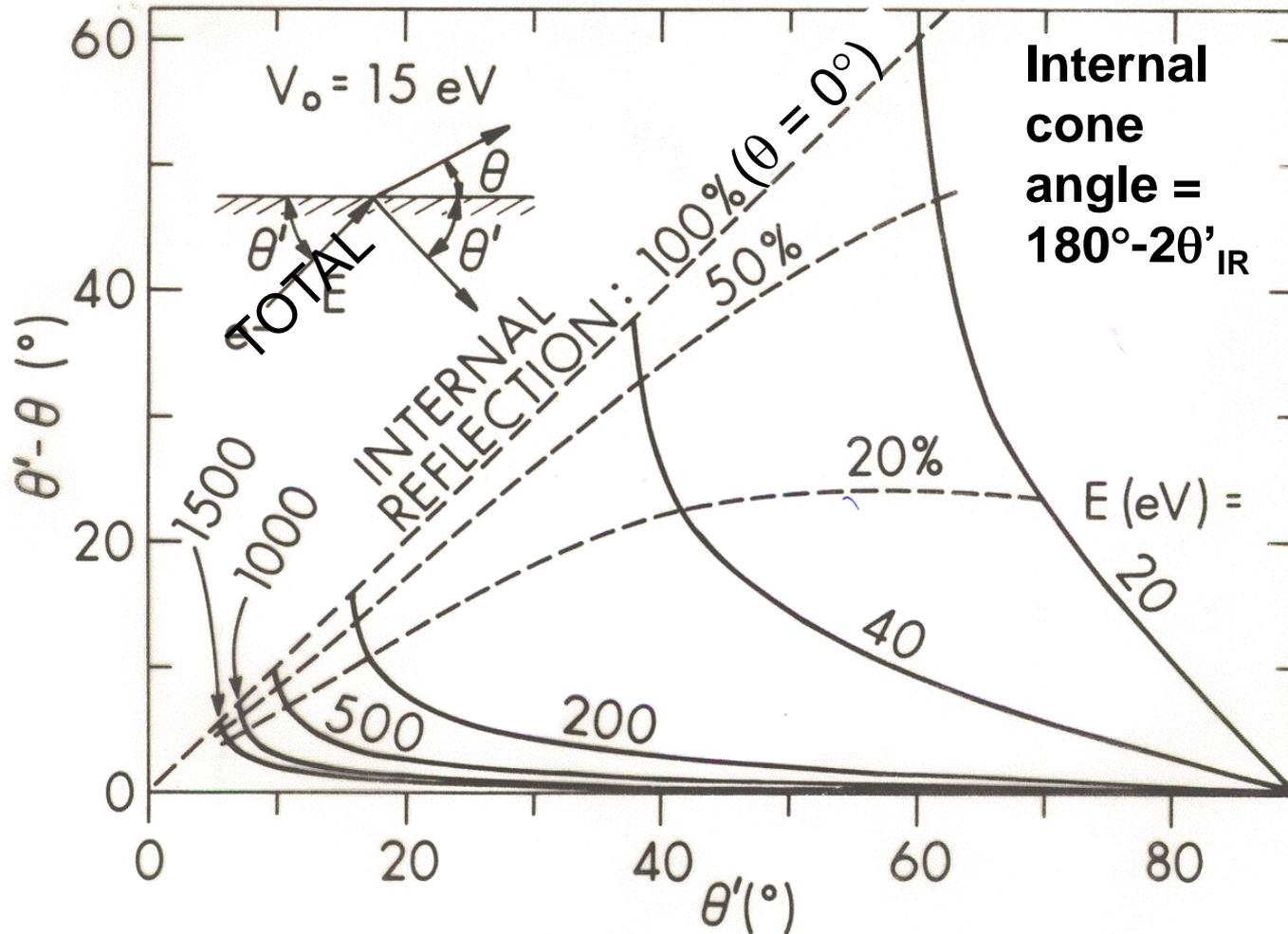
$$\theta = \tan^{-1} \left[\sqrt{\sin^2 \theta' - \frac{V_0}{E_{kin}}} / \cos \theta' \right], \theta_{IR} = \text{total internal reflection}$$



Observed
Low-Index
Directions
Above
W(110)



Electron Refraction at the Surface Due to the Inner Potential

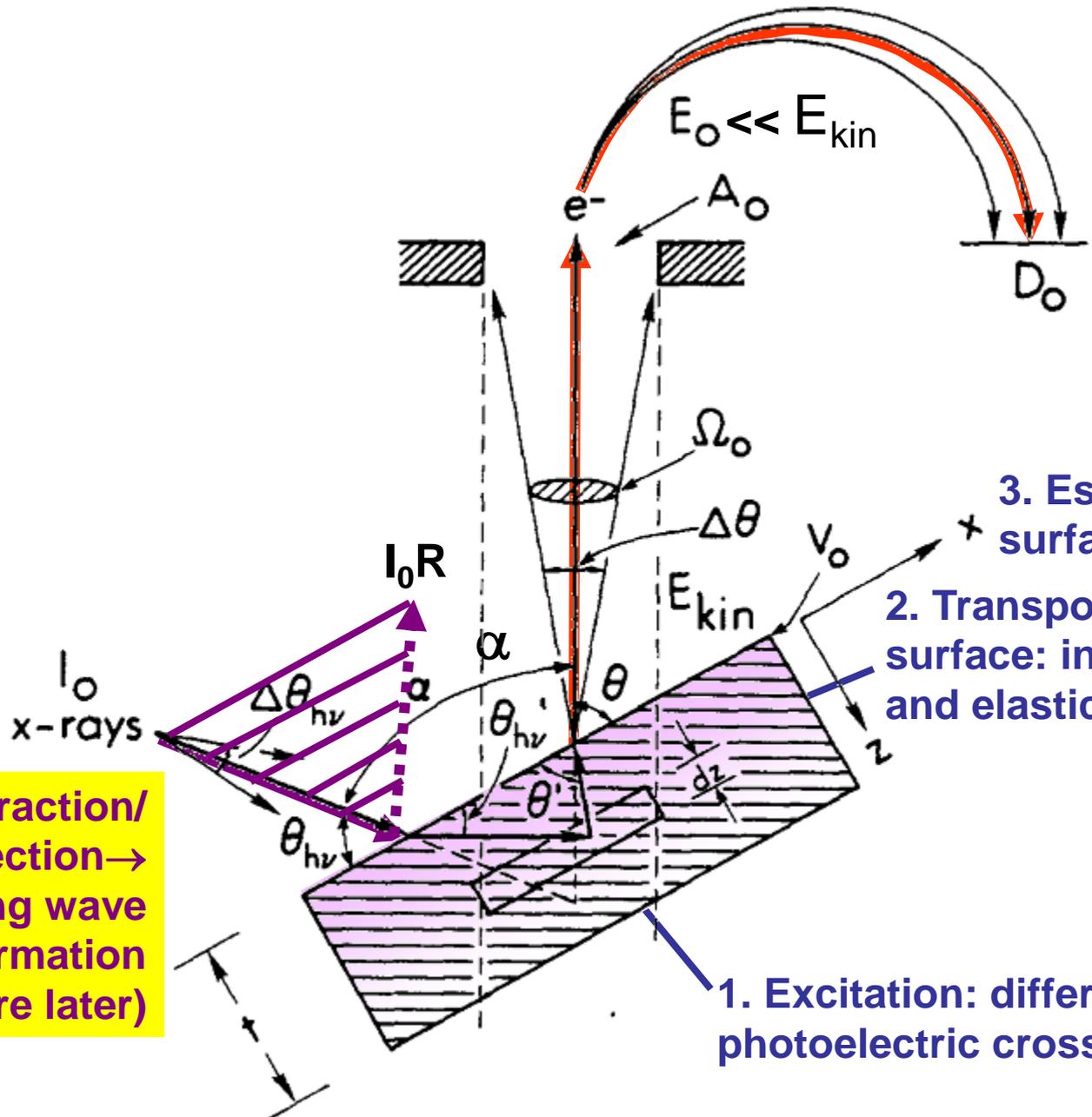


Plus total reflection for lower energies as:

<u>E(eV)</u>	<u>θ'_{IR}</u>	<u>Int. Cone Ang.</u>
1500	5.5°	169°
1000	7°	166°
500	9.5°	161°
200	15.5°	149°
40	37.5°	105°
20	60°	60°
19	63°	54°
18	66°	48°
17	70°	40°
16	76°	28°
15.1	86°	8°
15	90°	0°

(Nothing gets out!)

PHOTOELECTRON INTENSITIES—THE 3-STEP MODEL

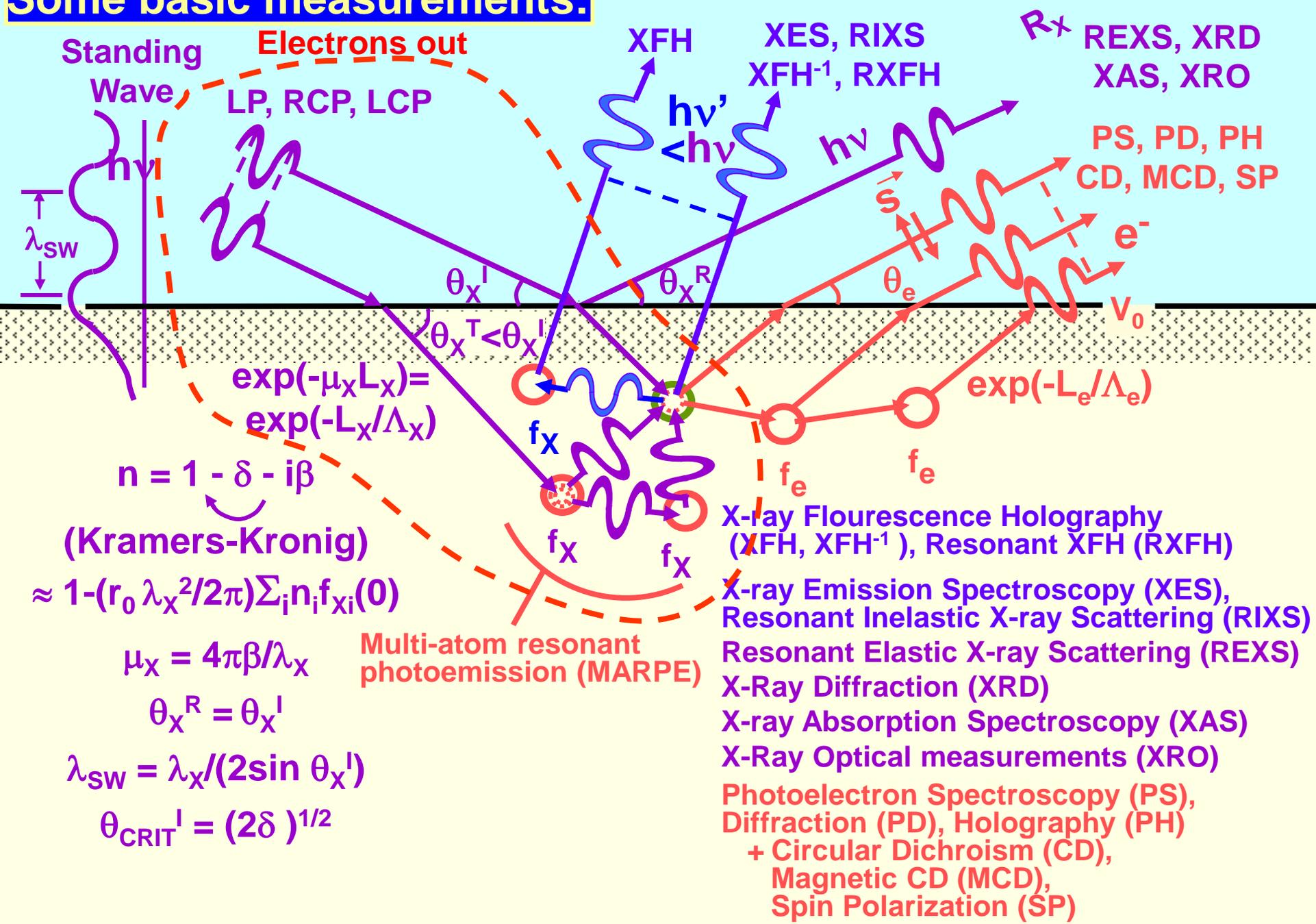


+X-ray refraction/
reflection →
standing wave
formation
(more later)

- 3. Escape across the surface barrier (V_0)
- 2. Transport to the surface: inelastic (Λ_e) and elastic ($f(\theta)$) scattering

1. Excitation: differential photoelectric cross section ($d\sigma/d\Omega$)

Some basic measurements:



X-ray scattering factor:
 $f_i = \text{Re}f_i + i(\text{Im}f_i)$

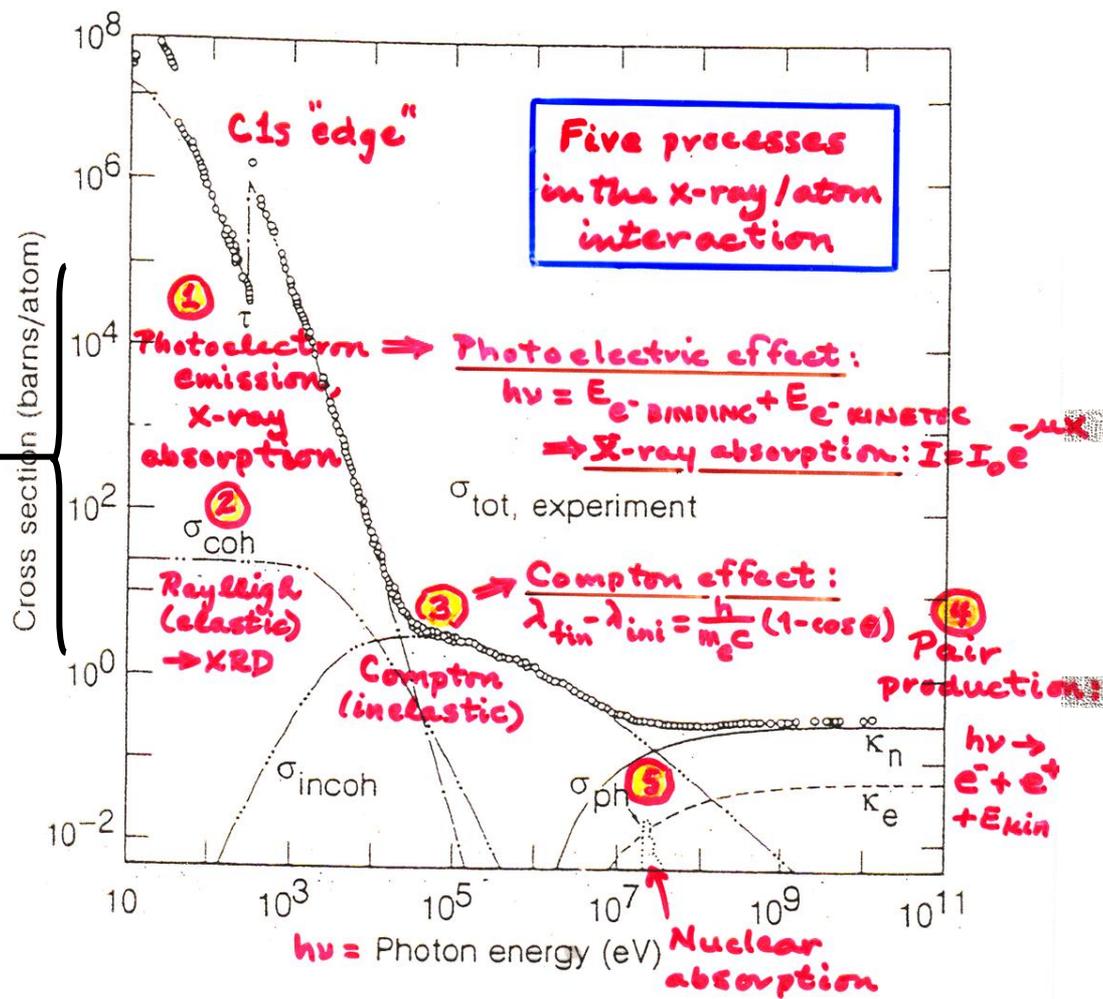
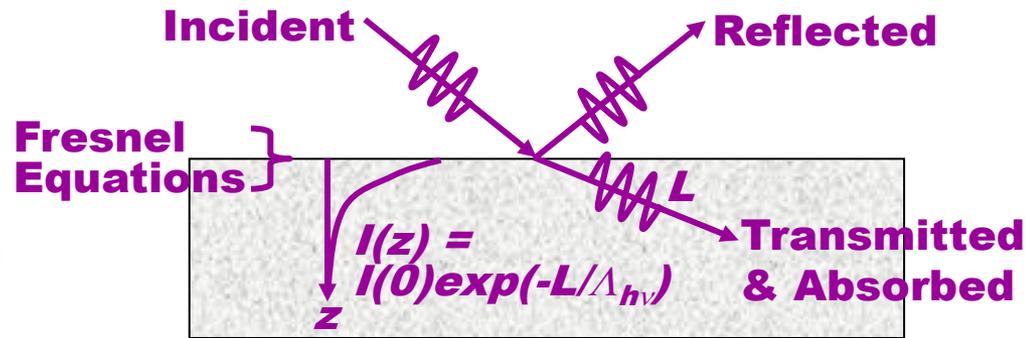


Fig. 3-1. Total photon cross section σ_{tot} in carbon, as a function of energy, showing the contributions of different processes: τ , atomic photo-effect (electron ejection, photon absorption); σ_{coh} , coherent scattering (Rayleigh scattering—atom neither ionized nor excited); σ_{incoh} , incoherent scattering (Compton scattering off an electron); κ_n , pair production, nuclear field; κ_e , pair production, electron field; σ_{ph} , photonuclear absorption (nuclear absorption usually followed by emission of a neutron or other particle). (From Ref. 3; figure courtesy of J. H. Hubbell.)

A LITTLE X-RAY OPTICS



(E.G. See pp. 1-38, 1-44, 5-18-5-19 in X-Ray Data Booklet)

$$\text{Index of refraction} = n = 1 - \delta - i\beta$$

$\delta = + \text{no.} = \text{refractive decrement} \ll 1$ (Sometimes negative through absorption resonances)

$\beta = + \text{no.} = \text{absorptive decrement} \ll 1$

δ and β linked by Kramers-Kronig transform

$$n \text{ also} = 1 - (r_e/2\pi)\lambda_{hv}^2 \sum n_i f_i (0 = \text{fwd. scatt.})$$

$$r_e = \text{classical electron radius} \\ = e^2/4\pi\epsilon_0 m_e e^2 = 2.817 \times 10^{-15} \text{ m} \\ \lambda_{hv} = \text{x-ray wavelength}$$

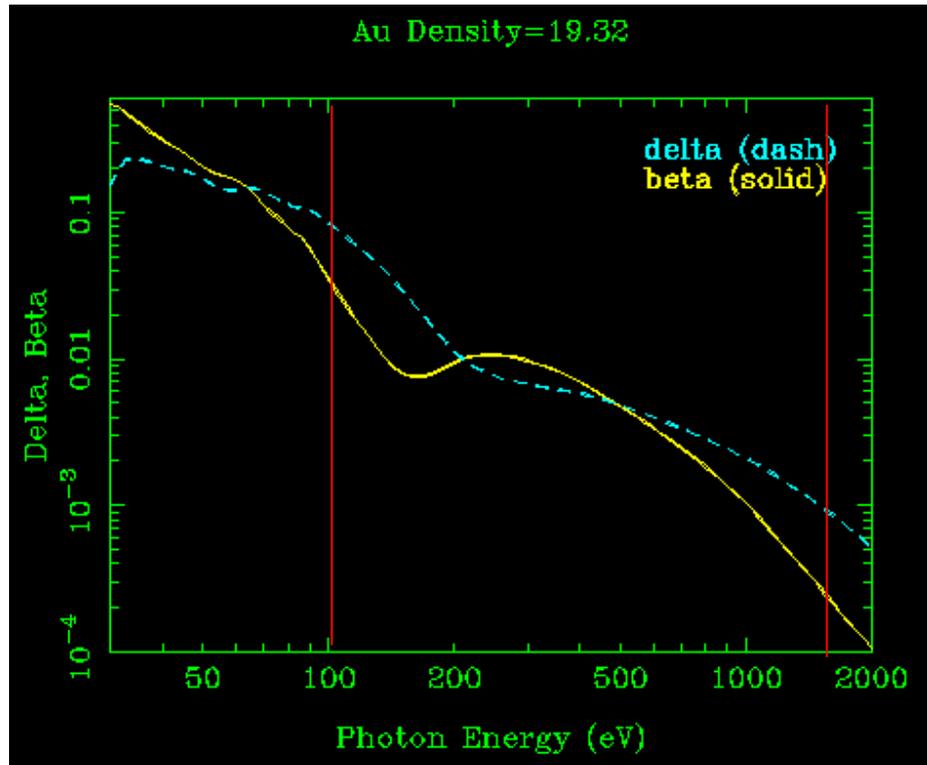
$n_i = \text{no. } i \text{ atoms per unit volume}$

$f_i = \text{x-ray scattering factor for } i\text{th type of atom, in forward direction}$

$$\text{Exponential absorption length} = l_{\text{abs}} = \lambda_{hv}/(4\pi\beta) = \Lambda_{hv}$$

$$\theta_{\text{CRIT}} = \text{critical grazing angle at which reflectivity begins (} R \approx 0.20 \text{)} \\ = [2\delta]^{0.5}$$

Online data and calculations at:
http://www-cxro.lbl.gov/optical_constants/



X-Ray Interactions with Matter



Contents

- [Introduction](#)
- Access the [atomic scattering factor](#) files.
- Look up [x-ray properties of the elements](#).
- The [index of refraction](#) for a compound material.
- The x-ray [attenuation length](#) of a solid.
- X-ray transmission
 - Of a [solid](#).
 - Of a [gas](#).
- X-ray reflectivity
 - Of a [thick mirror](#).
 - Of a [single layer](#).
 - Of a [bilayer](#).
 - Of a [multilayer](#).
- The diffraction efficiency of a [transmission grating](#).
- Related calculations:
 - Synchrotron [bend magnet radiation](#).

NEW! [What's New?](#)

[Other x-ray web resources](#).

These pages utilize *JavaScript*, but the [decaffeinated versions](#) are still available.

Reference

B.L. Henke, E.M. Gullikson, and J.C. Davis. *X-ray interactions: photoabsorption, scattering, transmission, and reflection at E- 50-30000 eV, Z- 1-92*, Atomic Data and Nuclear Data Tables Vol. **54** (no.2), 181-342 (July 1993).

| [CXRO](#) | [ALS](#) |

By Eric Gullikson. Please direct any comments to EMGullikson@lbl.gov
[Server Statistics](#) © 1995-2001

Website

SOME X-RAY OPTICAL EFFECTS: REDUCED PENETRATION DEPTHS AND INCREASED REFLECTIVITY AT GRAZING INCIDENCE ANGLES

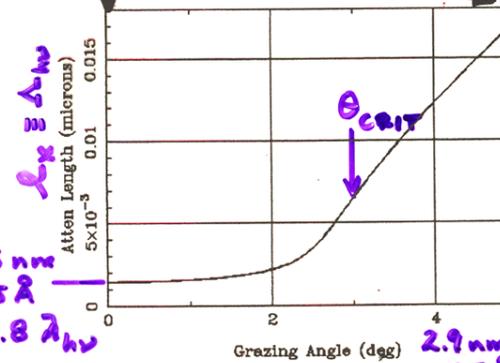
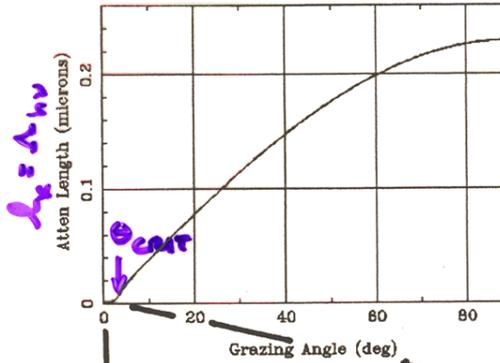
θ_{CRIT} = Grazing angle at which reflectivity begins ($R \approx 0.20$)
 $= [2\delta]^{0.5}$

ENHANCED SURFACE SENSITIVITY @ GRAZING INCIDENCE



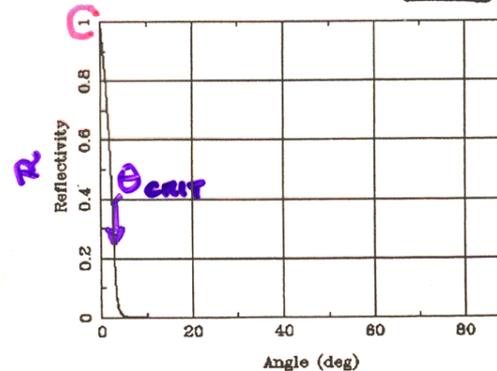
X-Ray Attenuation Length

Au Density=19.32, Energy=1487.eV



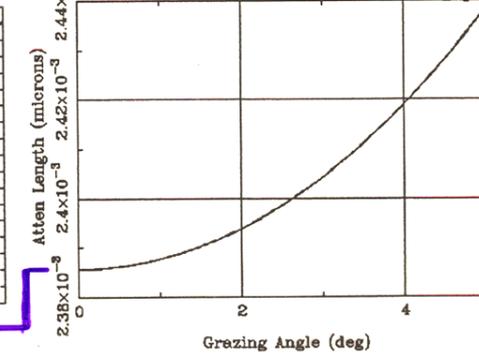
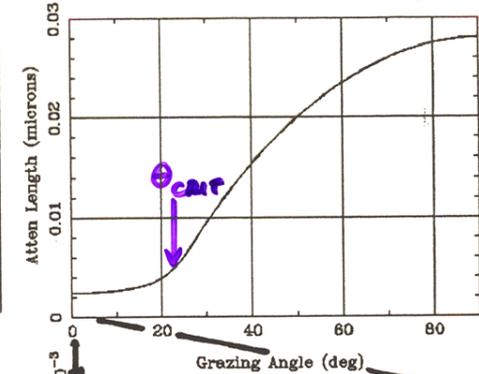
Mirror Reflectivity

Au Rho=19.32, Sig=0.nm, P=-1., E=1487.eV



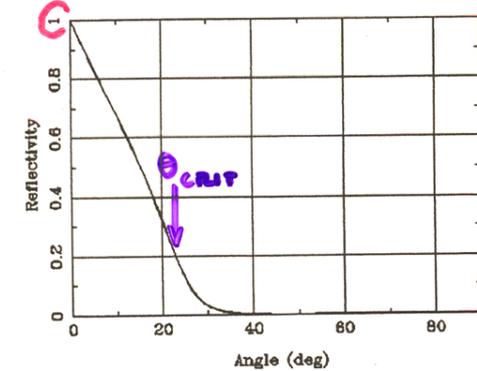
X-Ray Attenuation Length

Au Density=19.32, Energy=100.eV

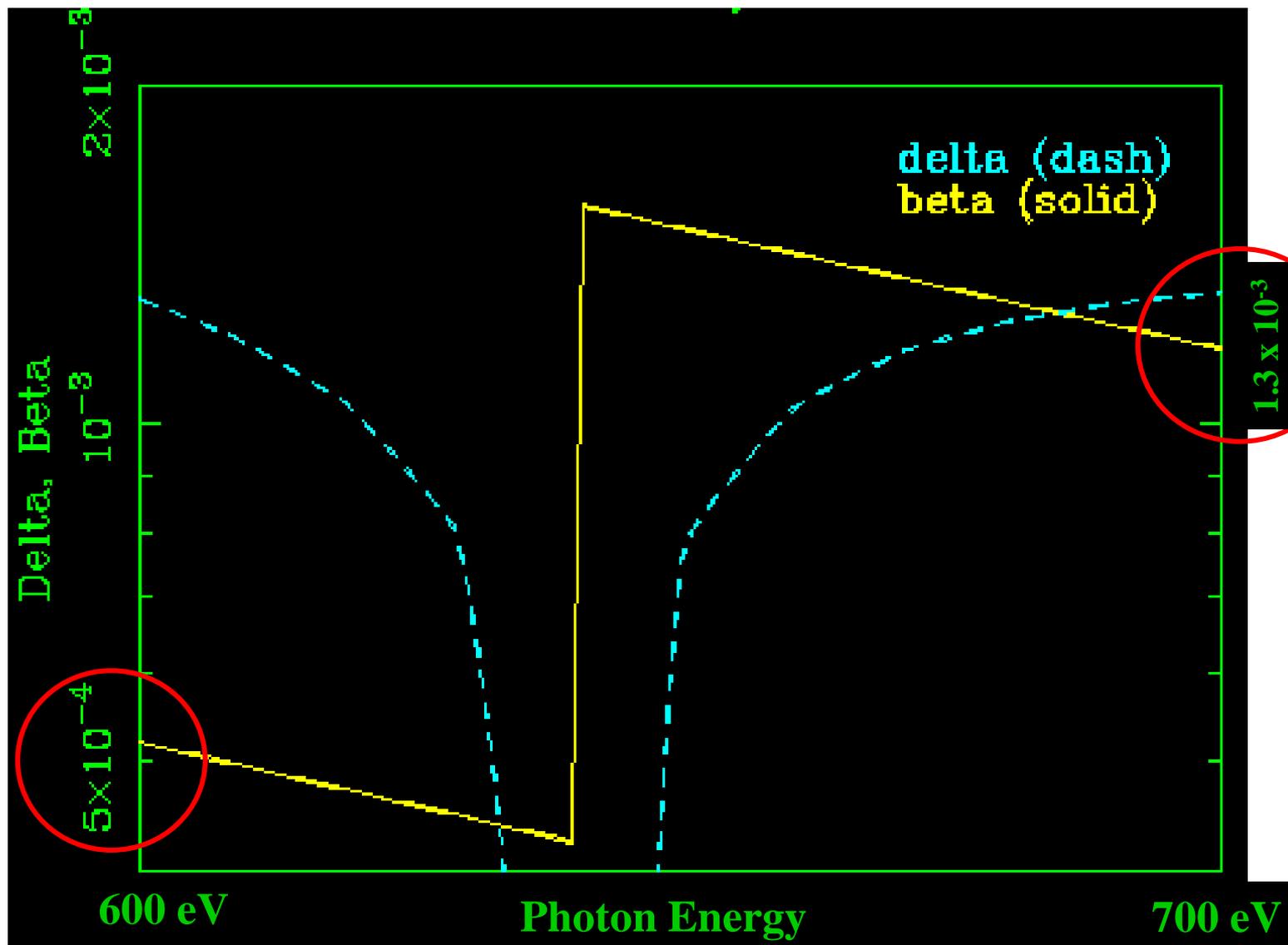


Mirror Reflectivity

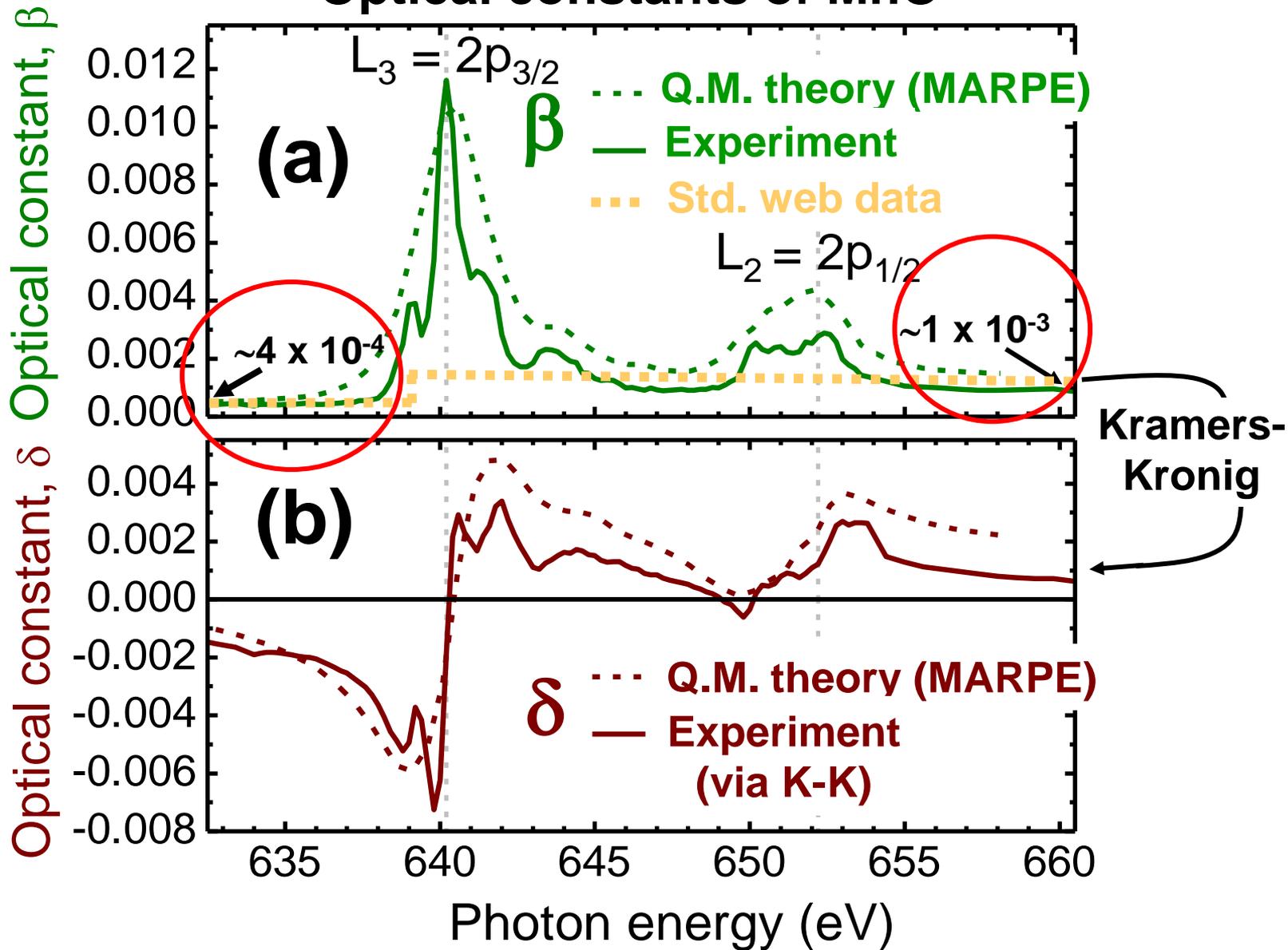
Au Rho=19.32, Sig=0.nm, P=-1., E=100.eV



Optical constants through Mn 2p edges of MnO— Web data without absorption peaks

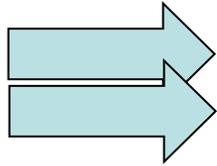


Optical constants of MnO



Basic Concepts and Experiments

Core-Level Photoemission



Intensities and Quantitative Analysis, the 3-Step Model
Varying Surface and Bulk Sensitivity
Chemical shifts
Multiplet Splittings
Electron Screening and Satellite Structure
Magnetic and Non-Magnetic Dichroism
Resonant Photoemission
Photoelectron Diffraction and Holography

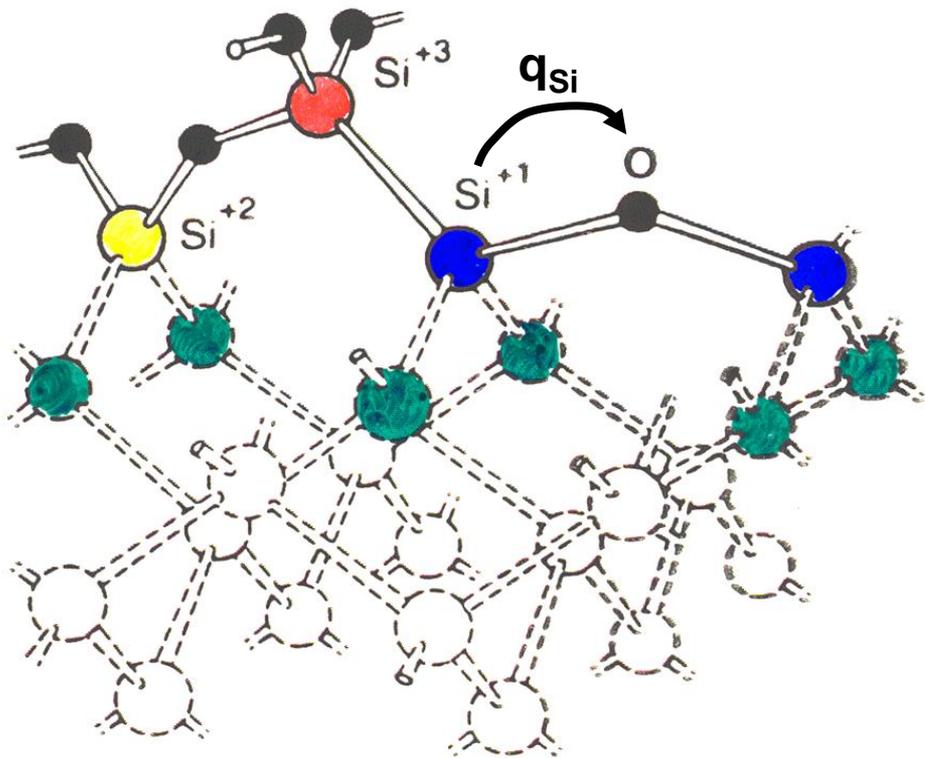
Valence-Level Photoemission

Band-Mapping in the Ultraviolet Photoemission Limit
Densities of States in the X-Ray Photoemission Limit

Some New Directions

Photoemission with Hard X-Rays (throughout lectures)
Photoemission with Standing Wave Excitation
Photoemission with Spatial Resolution/Photoelectron Microscopy
Temporal Resolution
@ Higher Pressures

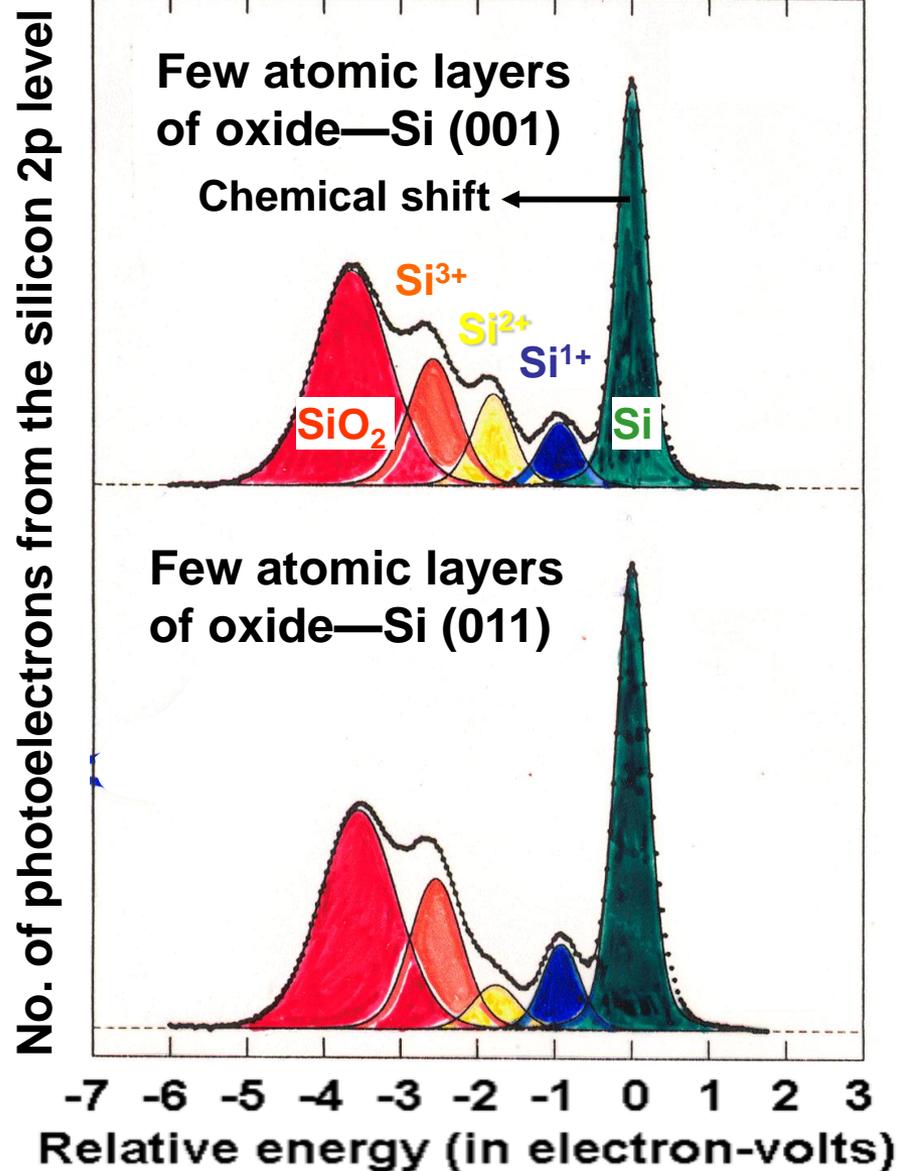
Chemical shifts: Looking into the silicon dioxide layer with photoelectron spectroscopy



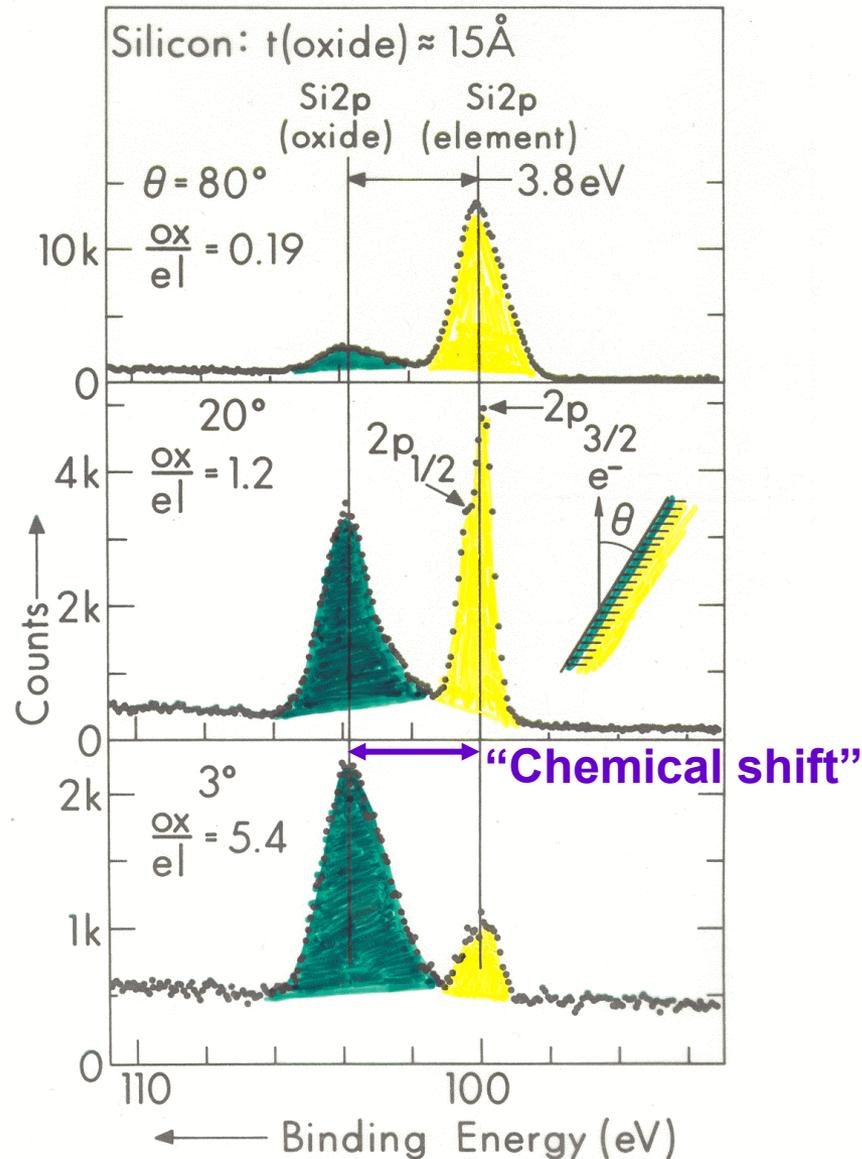
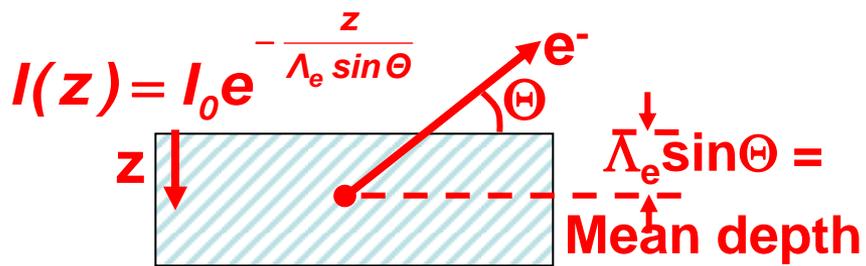
Charge transfer, $e^- - e^-$ coulomb integral:

$$\text{Shift} \approx q_{Si} K_{Si2p, Si3p} = q_{Si} \int \varphi_{2p}^*(\vec{r}_1) \varphi_{3p}^*(\vec{r}_2) \frac{e^2}{r_{12}} \varphi_{2p}(\vec{r}_1) \varphi_{3p}(\vec{r}_2) dV_1 dV_2$$

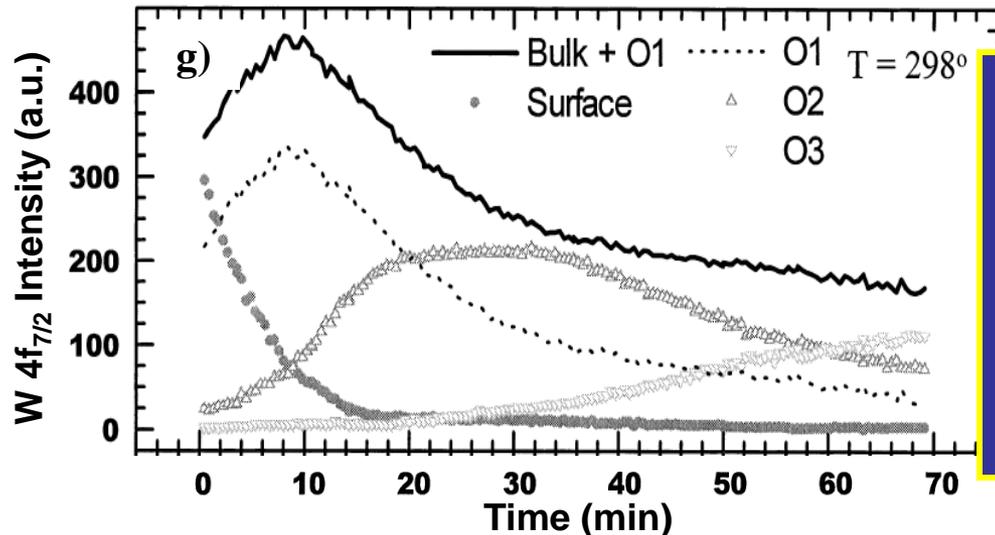
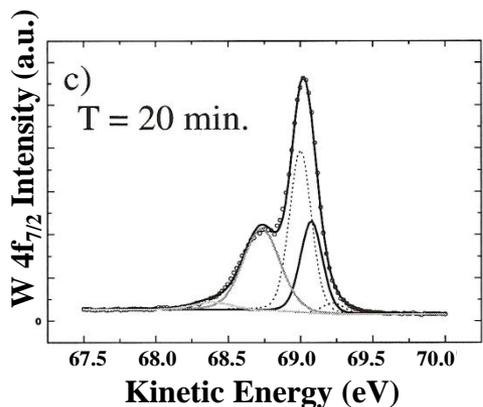
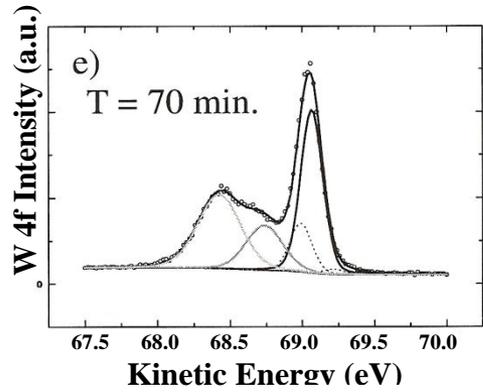
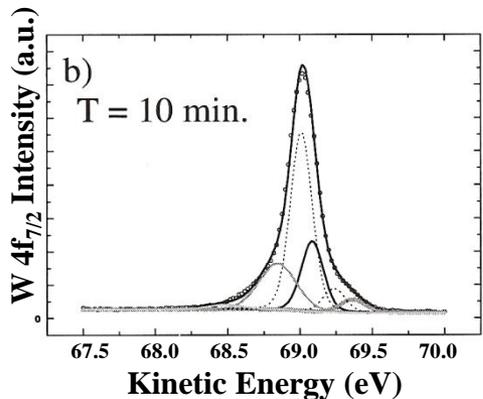
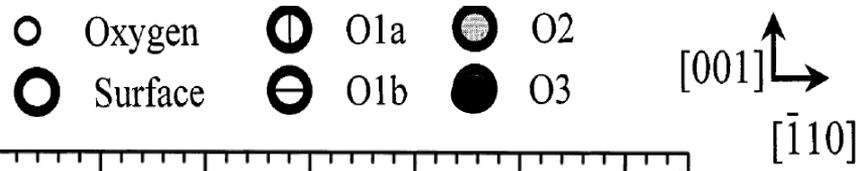
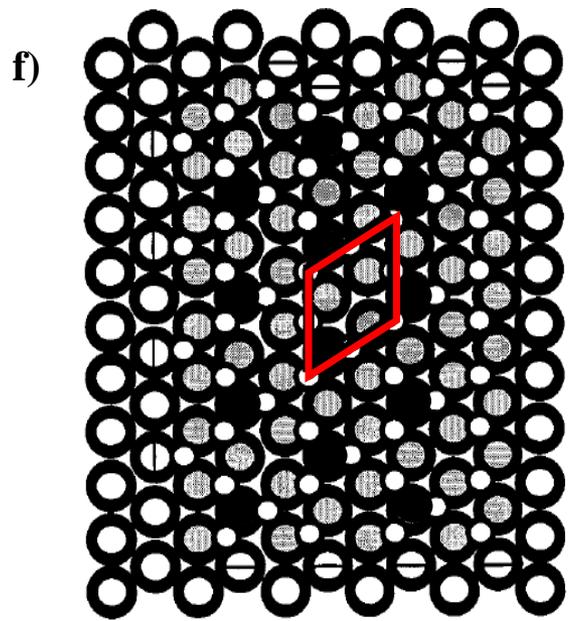
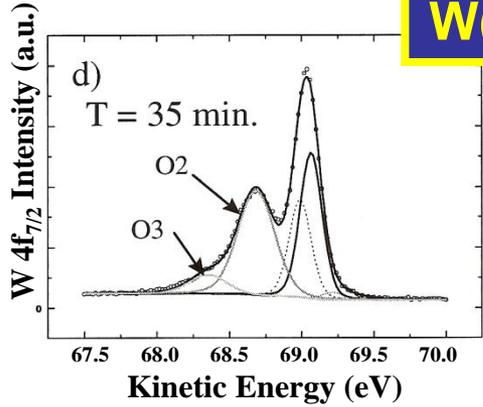
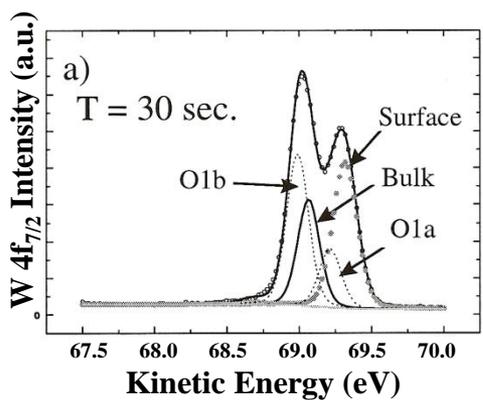
Himpsel et al., Phys. Rev. B 38, 6086 ('88)



Enhancing surface sensitivity at grazing exit angles



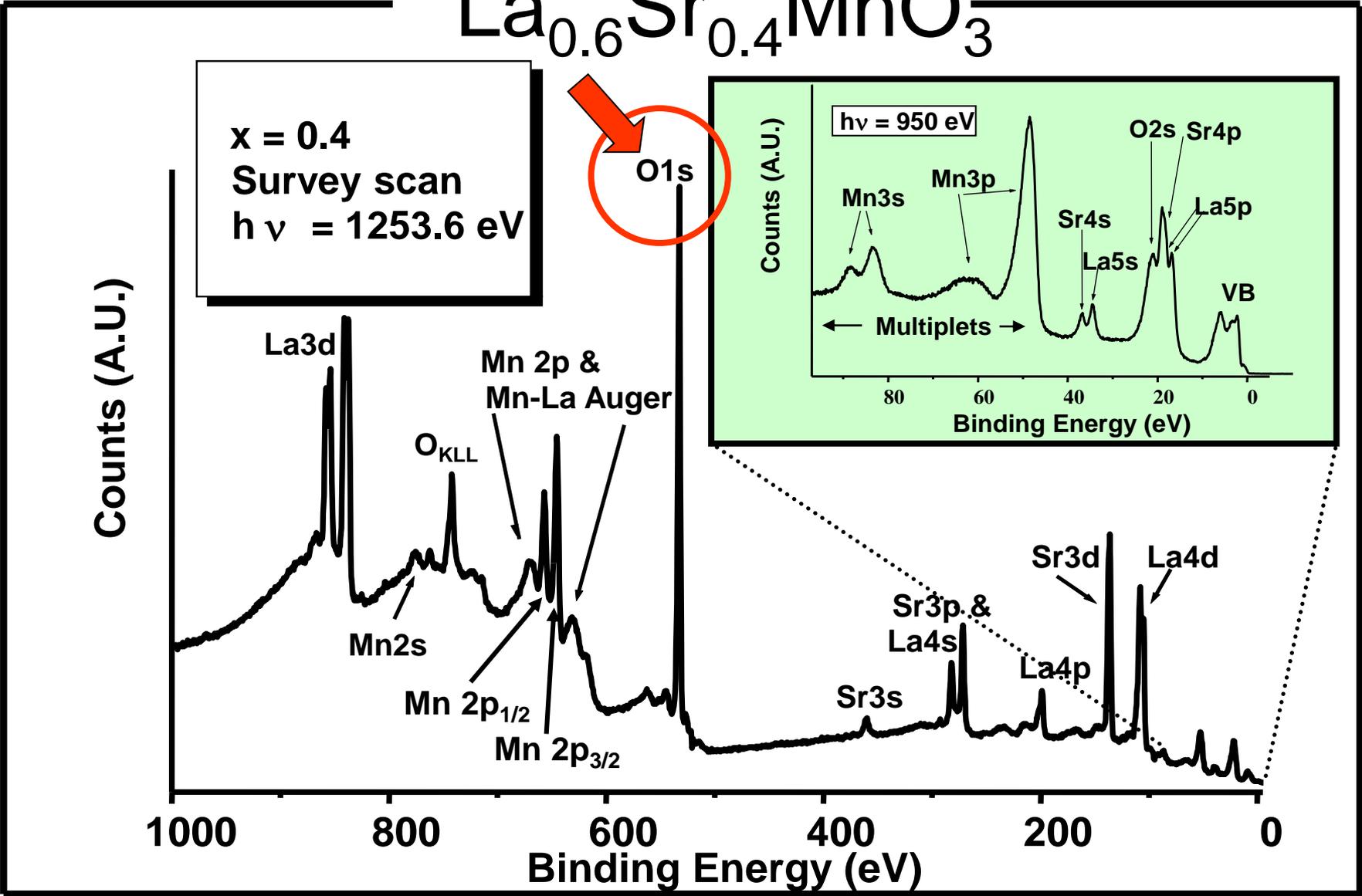
W(110)/O—W 4f_{7/2} Chemical Shifts



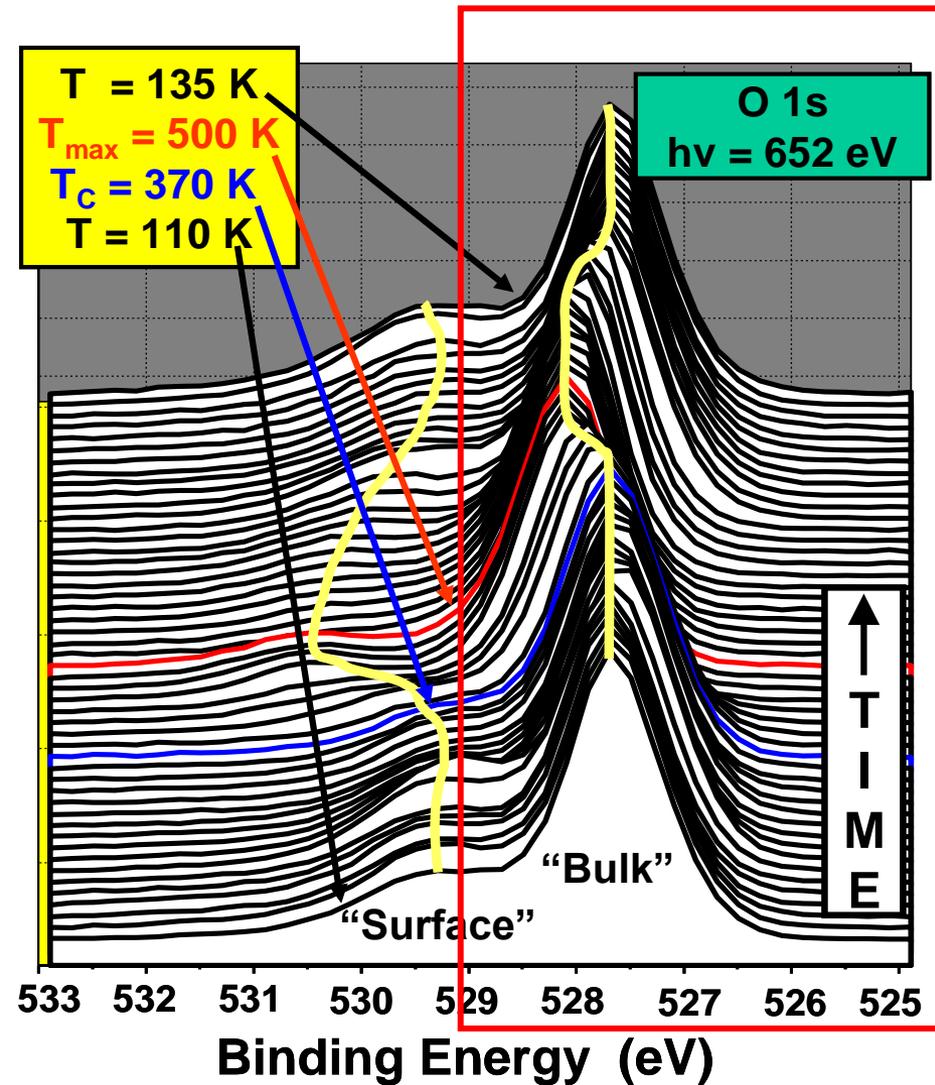
An early time-resolved reaction study (more later)

Ynzunza et al.,
 Surf. Sci. 459 (2000) 69

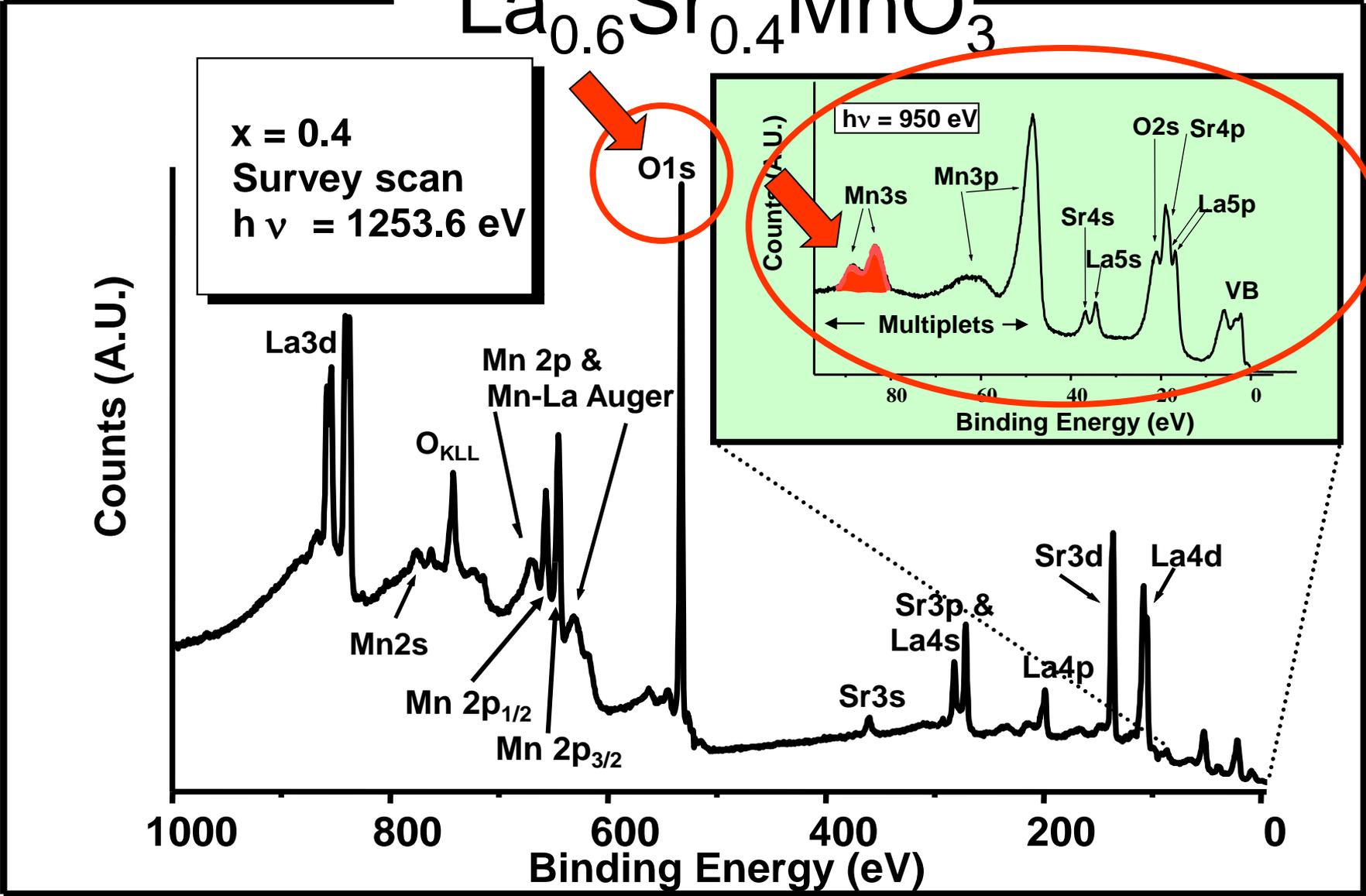
Core and valence photoemission



Temperature dependence of Mn3s and O1s spectra: $\text{La}_{0.7}\text{Sr}_{0.3}\text{MnO}_3$



Core and valence photoemission



Basic Concepts and Experiments

Core-Level Photoemission

Intensities and Quantitative Analysis, the 3-Step Model

Varying Surface and Bulk Sensitivity

Chemical shifts



Multiplet Splittings

Electron Screening and Satellite Structure

Magnetic and Non-Magnetic Dichroism

Resonant Photoemission

Photoelectron Diffraction and Holography

Valence-Level Photoemission

Band-Mapping in the Ultraviolet Photoemission Limit

Densities of States in the X-Ray Photoemission Limit

Some New Directions

Photoemission with Hard X-Rays (throughout lectures)

Photoemission with Standing Wave Excitation

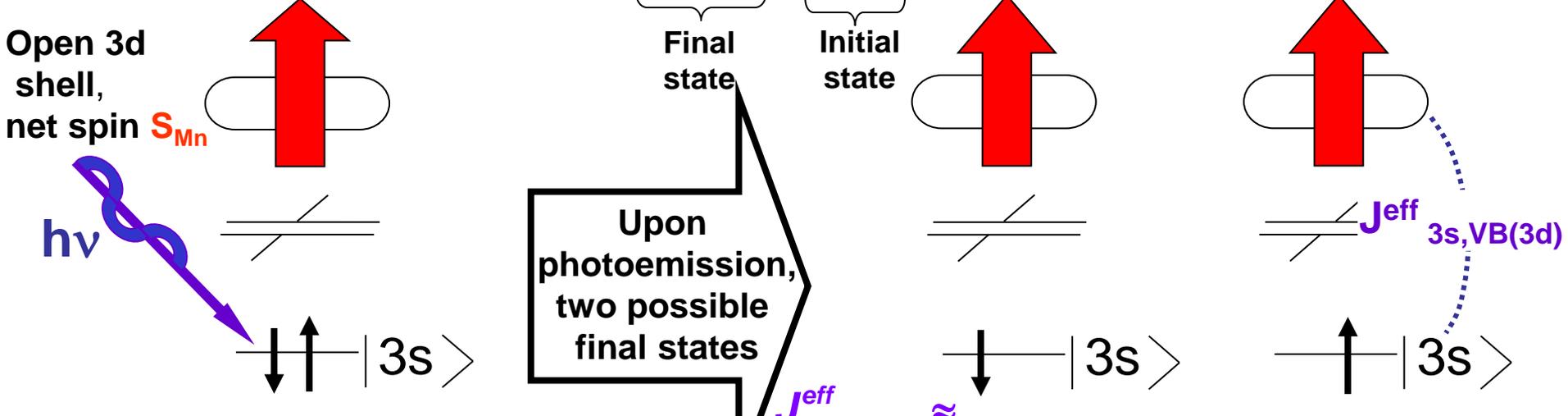
Photoemission with Spatial Resolution/Photoelectron Microscopy

Temporal Resolution

@ Higher Pressures

Multiplet splitting in core levels of transition metal oxides

$$BE = h\nu - E_{kin} = \underbrace{E^*(N-1)}_{\text{Final state}} - \underbrace{E_0(N)}_{\text{Initial state}}$$



The splitting between the two peaks is given by

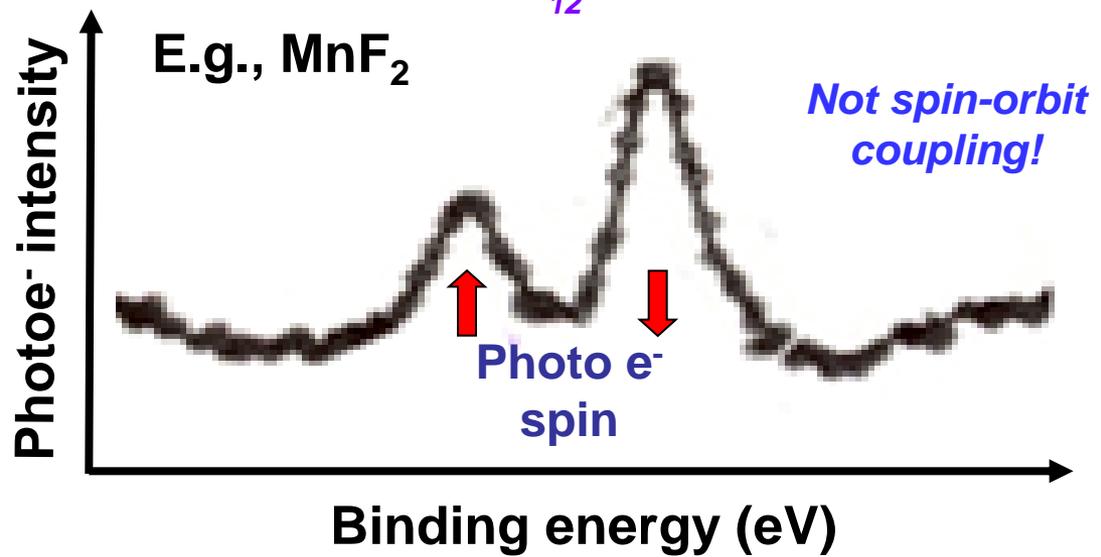
$$\Delta E_{3s} \approx (2S_{Mn} + 1) J_{3s,VB(3d)}^{eff}$$

(Van Vleck Theorem)

For the cubic manganites in simplest doping model,

$$S_{Mn} = 1/2(4-x) \rightarrow \Delta E_{3s} \approx [5-x] J_{3s,VB(3d)}^{eff}$$

with $J_{3s,VB}^{eff} \approx 1.1 \text{ eV}$



INTENSITIES IN PHOTOELECTRON SPECTRA:

- GENERAL: FINAL STATE K (k -SUBSHELL + ALL OTHER DESIG.)

$$\text{INT.}_K \propto |\hat{e} \cdot \langle \Psi_{\text{tot}}^f(N, K) | \sum_{i=1}^N \vec{r}_i | \Psi_i^i(N) \rangle|^2 \quad (\text{DIPOLE APPROX.})$$

- BORN-OPPENHEIMER: e^- 's FAST, VIBRATIONS SLOW

$$\text{INT.}_K \propto \underbrace{|\langle \Psi_{\text{vib}, \nu}^f | \Psi_{\text{vib}, \nu}^i \rangle|^2}_{\text{FRANK-CONDON FACTOR}} |\hat{e} \cdot \langle \Psi_e^f(N, K) | \sum_{i=1}^N \vec{r}_i | \Psi_e^i(N) \rangle|^2$$

- SUDDEN APPROXIMATION: $\Psi_k \rightarrow \Psi_f = \text{PHOTO}^-$ (FAST)



$$\text{INT.}_K \propto |\langle \Psi_{\text{vib}, \nu}^f | \Psi_{\text{vib}, \nu}^i \rangle|^2 |\langle \Psi_e^f(N-1, K) | \Psi_e^i(N-1, K) \rangle|^2$$

$$|\hat{e} \cdot \langle \psi_f | \vec{r} | \psi_k \rangle|^2 \quad \text{SAME SUBSHELL COUPLING + TOTAL L, S} \rightarrow \text{"MONOPOLE"}$$

\hookrightarrow NORMAL $\frac{dG_K}{d\Omega}$

- SLATER DETS. FOR $\Psi_e^f = \det(\psi_1^f, \psi_2^f, \dots, \psi_{k-1}^f, \psi_{k+1}^f, \dots, \psi_N^f)$

$$\Psi_e^i = \det(\psi_1^i, \psi_2^i, \dots, \psi_{k-1}^i, \psi_{k+1}^i, \dots, \psi_N^i)$$

$$\text{INT.}_K \propto |\langle \Psi_{\text{vib}, \nu}^f | \Psi_{\text{vib}, \nu}^i \rangle|^2 |\langle \psi_1^f | \psi_1^i \rangle|^2 |\langle \psi_2^f | \psi_2^i \rangle|^2 \dots$$

$$|\langle \psi_{k-1}^f | \psi_{k-1}^i \rangle|^2 |\langle \psi_{k+1}^f | \psi_{k+1}^i \rangle|^2 \dots |\langle \psi_N^f | \psi_N^i \rangle|^2$$

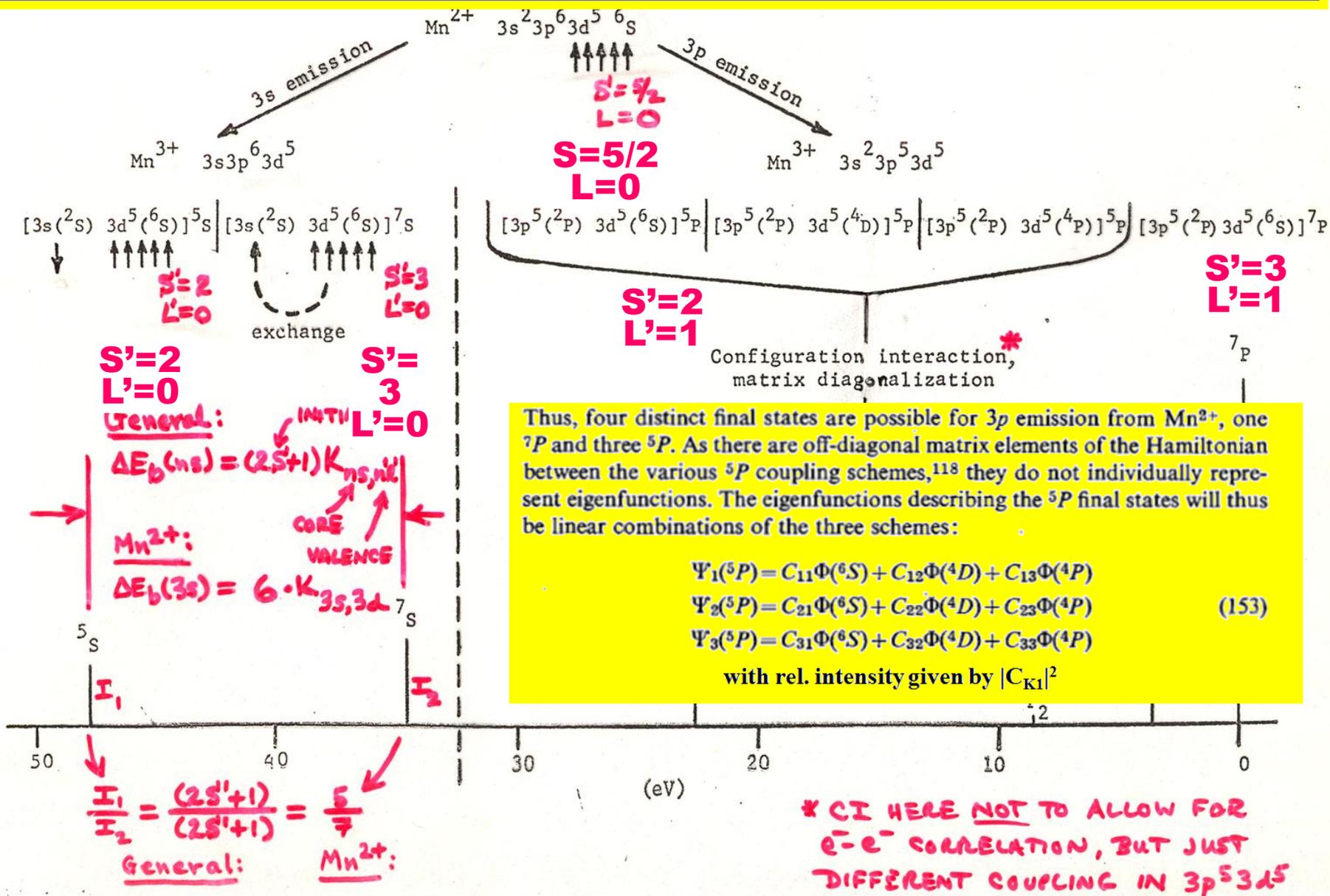
$$|\hat{e} \cdot \langle \psi_f | \vec{r} | \psi_k \rangle|^2$$

1e- DIPOLE $\rightarrow d\sigma/d\Omega$

**(N-1)e- SHAKE-UP/
SHAKE-OFF** \rightarrow
"MONOPOLE"

- PLUS DIFFRACTION EFFECTS IN Ψ_f ESCAPE

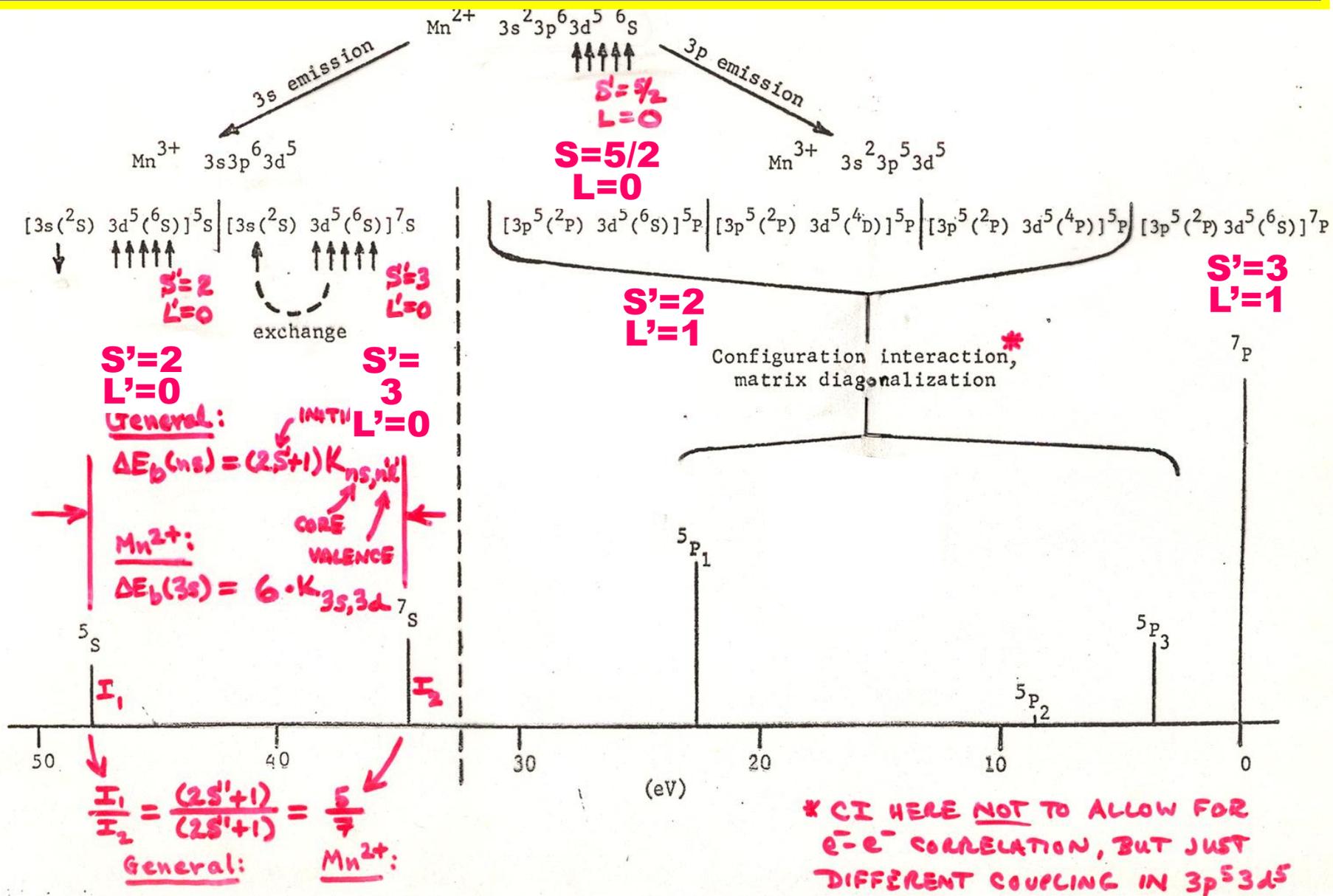
ORIGIN OF MULTIPLIET SPLITTINGS IN Mn²⁺: "ONE-ELECTRON" THEORY



General Mn²⁺ "Basic Concepts of XPS"

Figure 30

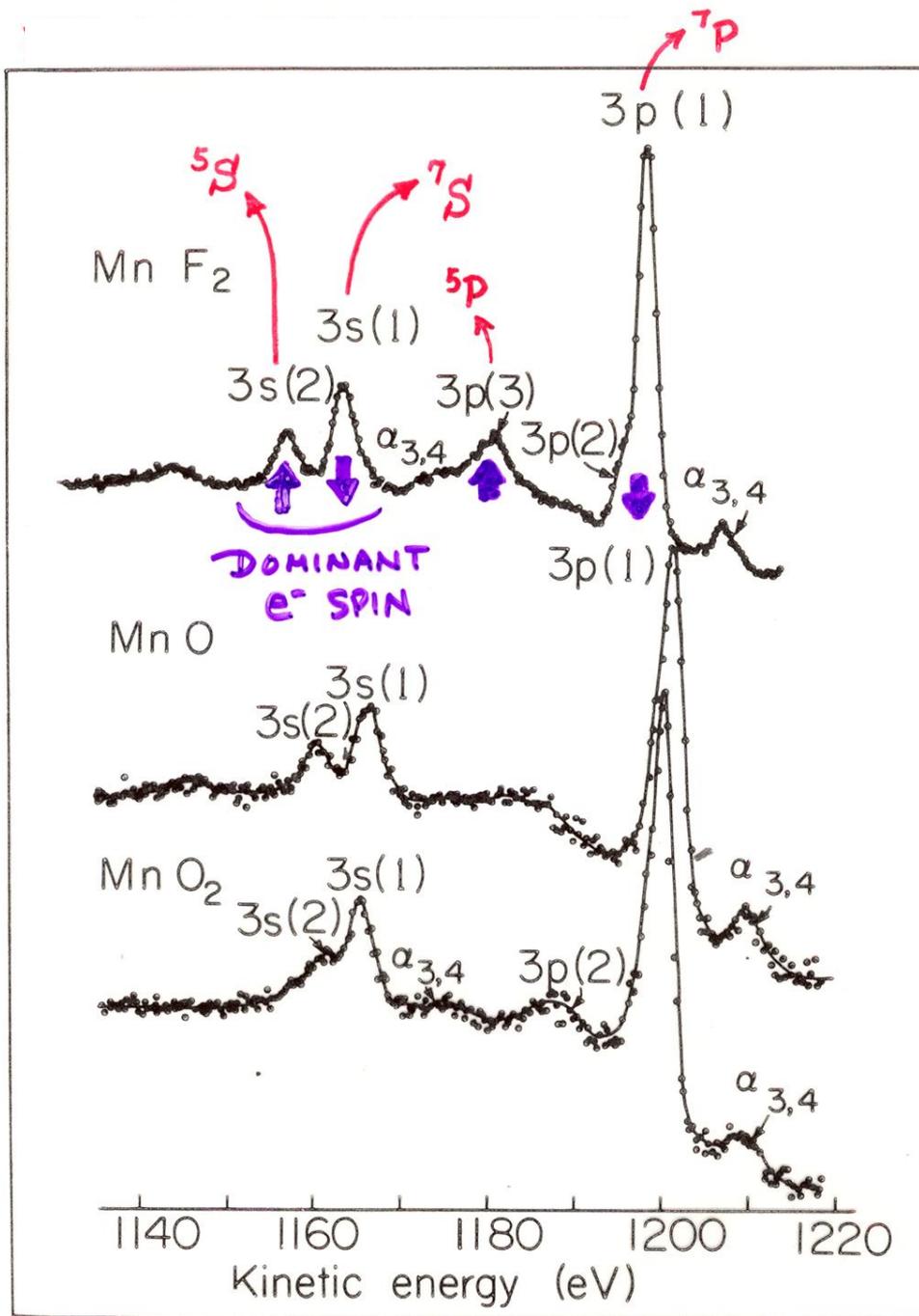
ORIGIN OF MULTIPLIET SPLITTINGS IN Mn²⁺: "ONE-ELECTRON" THEORY



General Mn²⁺ "Basic Concepts of XPS"

Figure 30

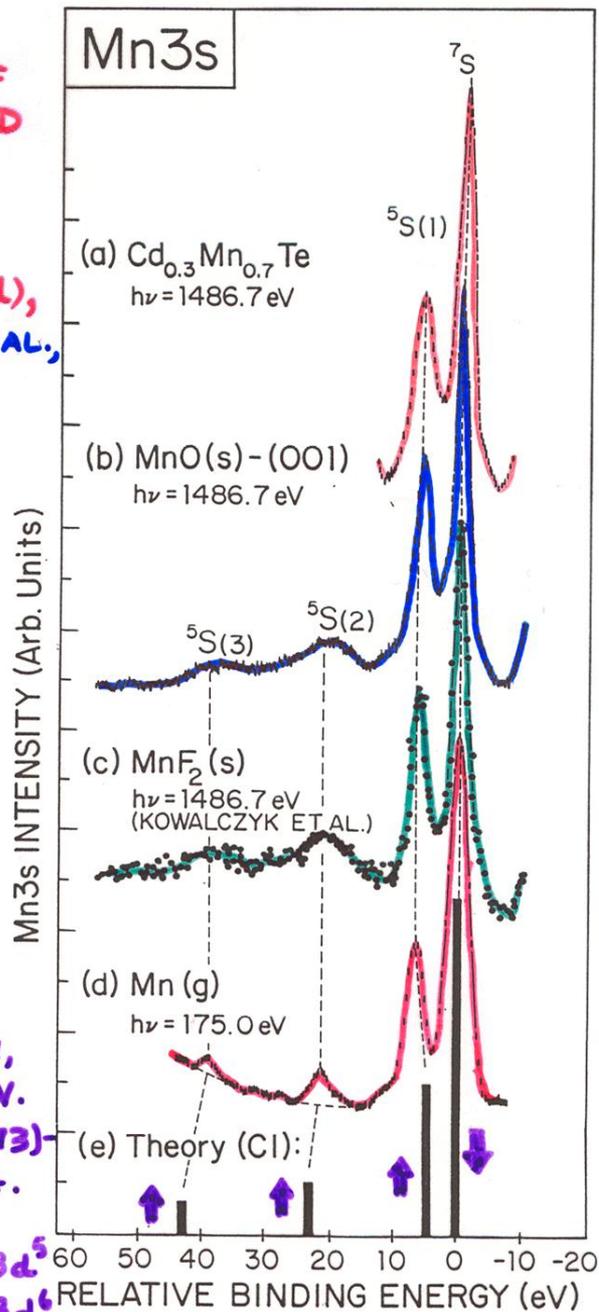
CORE-LEVEL MULTIPLY SPLITTINGS IN Mn COMPOUNDS



“Basic Concepts of XPS”
Figure 31

**COMPARISON OF
GAS-PHASE AND
SOLID-STATE
SPECTRA**

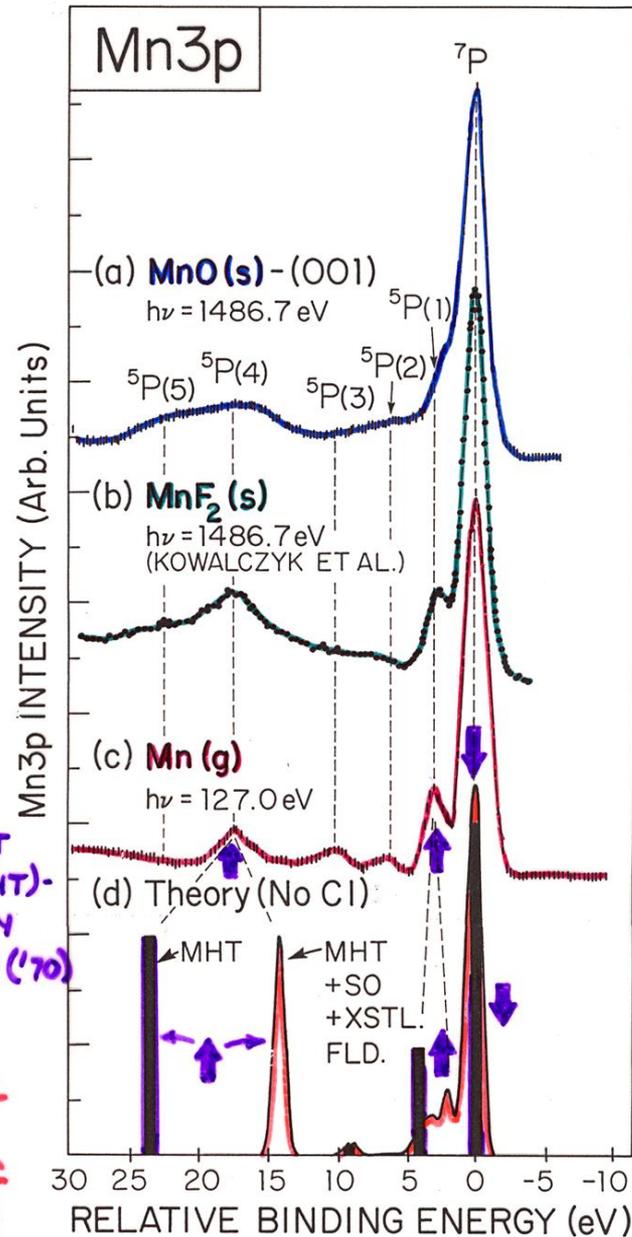
**EXPT. : (a), (b), (d),
HERMSMEIER ET AL.,
PHYS. REV. LETT.
61, 2592 (1988)
(OUR GROUP)**



**Correlation
CI effects:
anti-parallel
electrons**

**THEORY:
BAGUS, FREEMAN,
SASAKI, PHYS. REV.
LETT. 30, 850 (1973)
ATOMIC CONFIG.
INT. IN
 $\text{Mn}^{2+} \dots 3s^2 \dots 3d^5$
 $+ \text{Mn}^{3+} \dots 3s^2 3p^4 3d^6$**

**"Basic Concepts of XPS"
Figure 33**



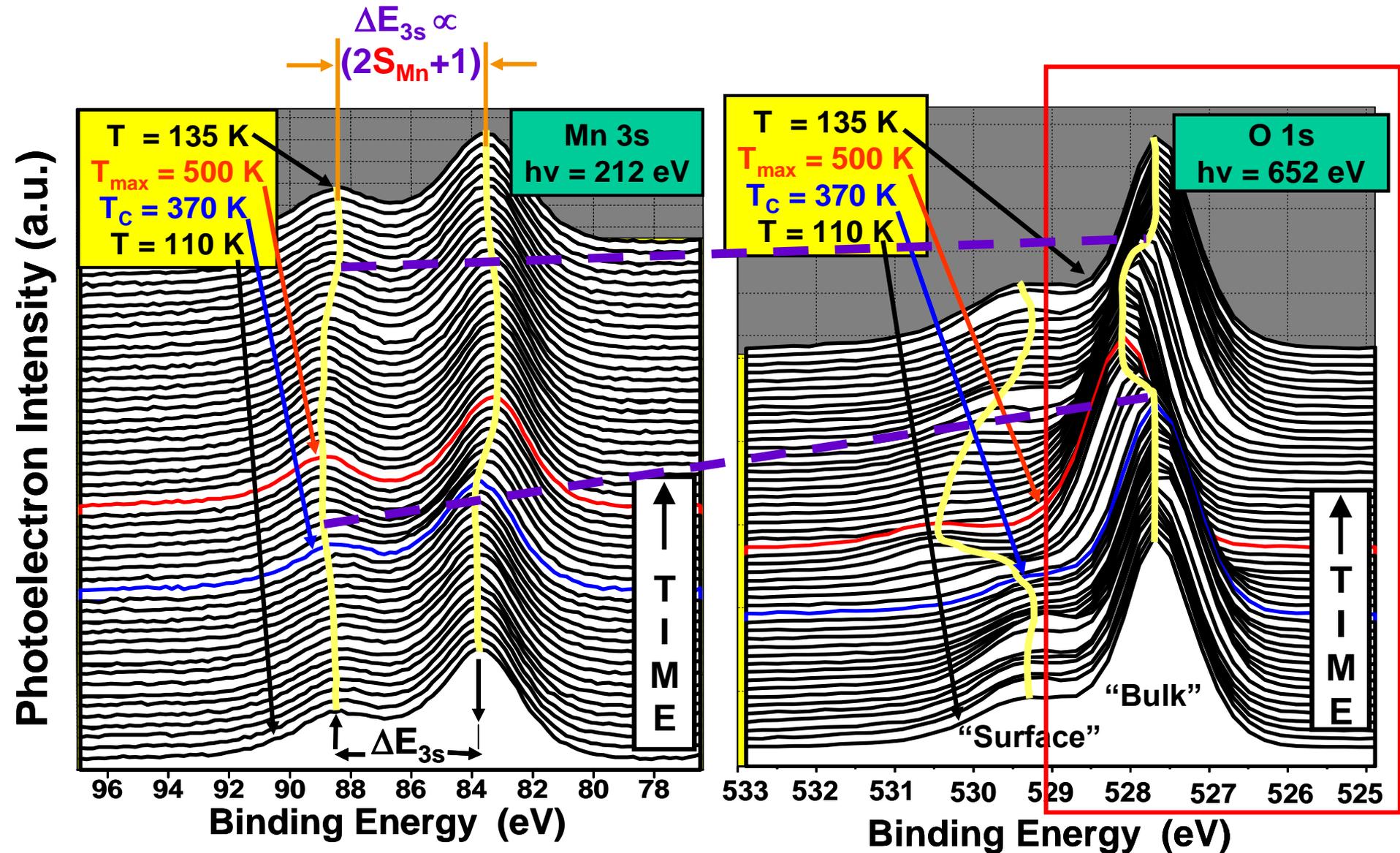
THEORY: NO CI

**SIMPLE MULTIPLY
 HOLE THEORY (MHT)-
 FADLEY, SHIRLEY
 PHYS. REV. A2, 1109 ('70)**

**EMPIRICAL
 MHT WITH SPIN
 ORBIT + CRYSTAL
 FIELD - SUGANO
 ET AL., J. PHYS. C
 15, 2625 (1982)**

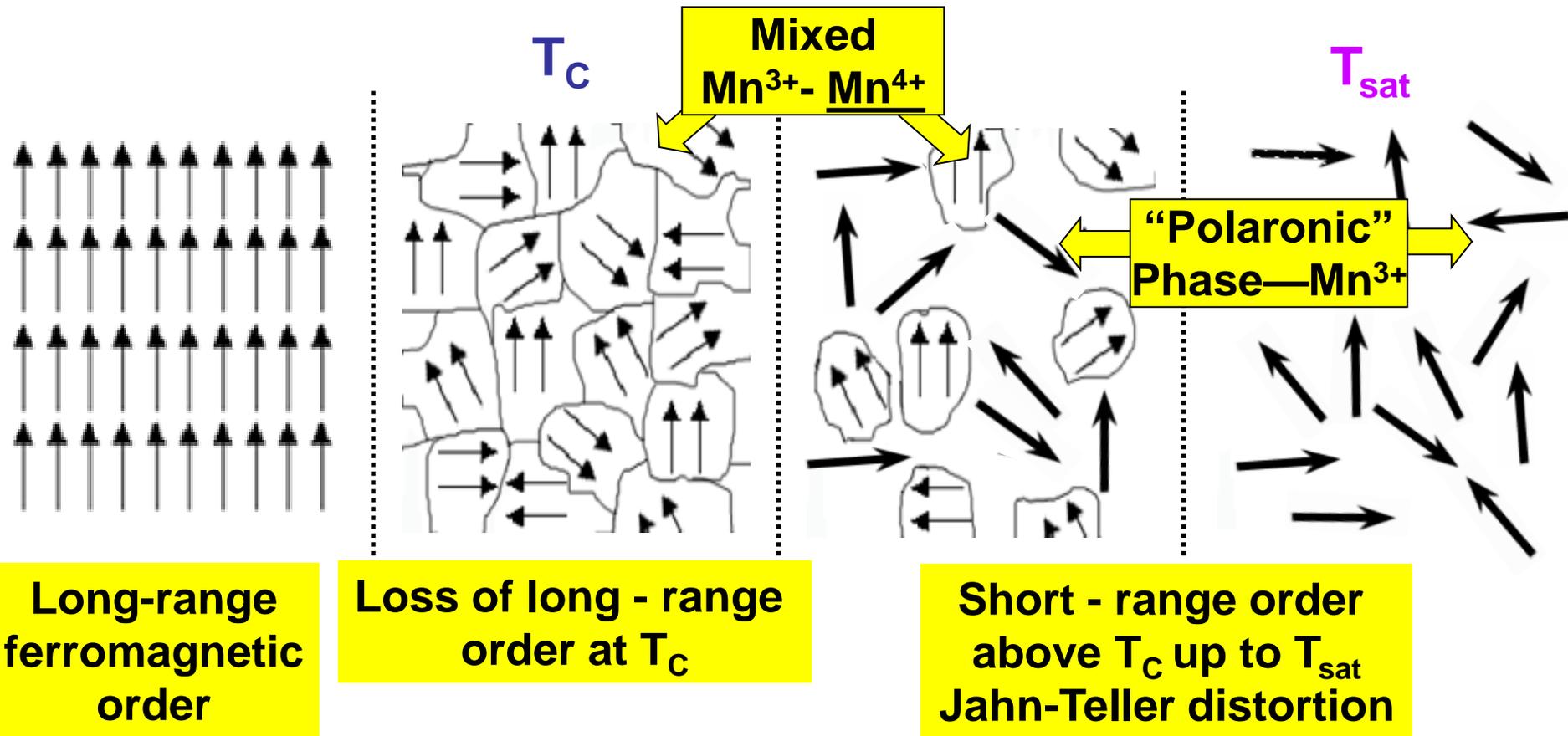
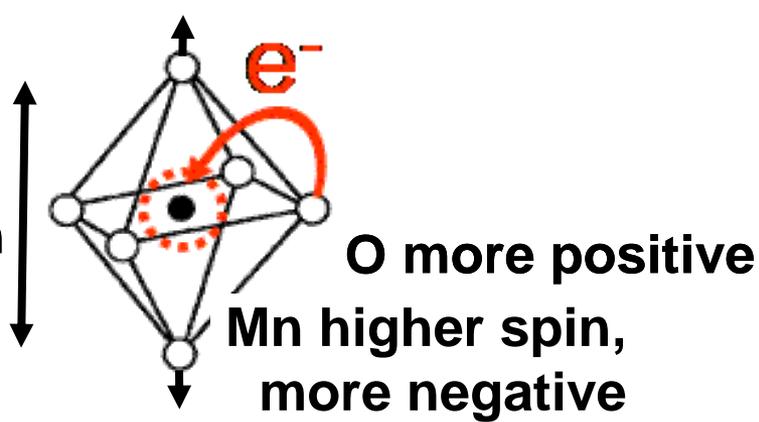
**HERMSMEIER
 ET AL.,
 P. R. L. 61, 2592 ('88)**

Temperature dependence of Mn3s and O1s spectra: $\text{La}_{0.7}\text{Sr}_{0.3}\text{MnO}_3$



Suggested scenario

Jahn-Teller distortion



Basic Concepts and Experiments

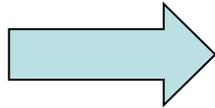
Core-Level Photoemission

Intensities and Quantitative Analysis, the 3-Step Model

Varying Surface and Bulk Sensitivity

Chemical shifts

Multiplet Splittings



Electron Screening and Satellite Structure

Magnetic and Non-Magnetic Dichroism

Resonant Photoemission

Photoelectron Diffraction and Holography

Valence-Level Photoemission

Band-Mapping in the Ultraviolet Photoemission Limit

Densities of States in the X-Ray Photoemission Limit

Some New Directions

Photoemission with Hard X-Rays (throughout lectures)

Photoemission with Standing Wave Excitation

Photoemission with Spatial Resolution/Photoelectron Microscopy

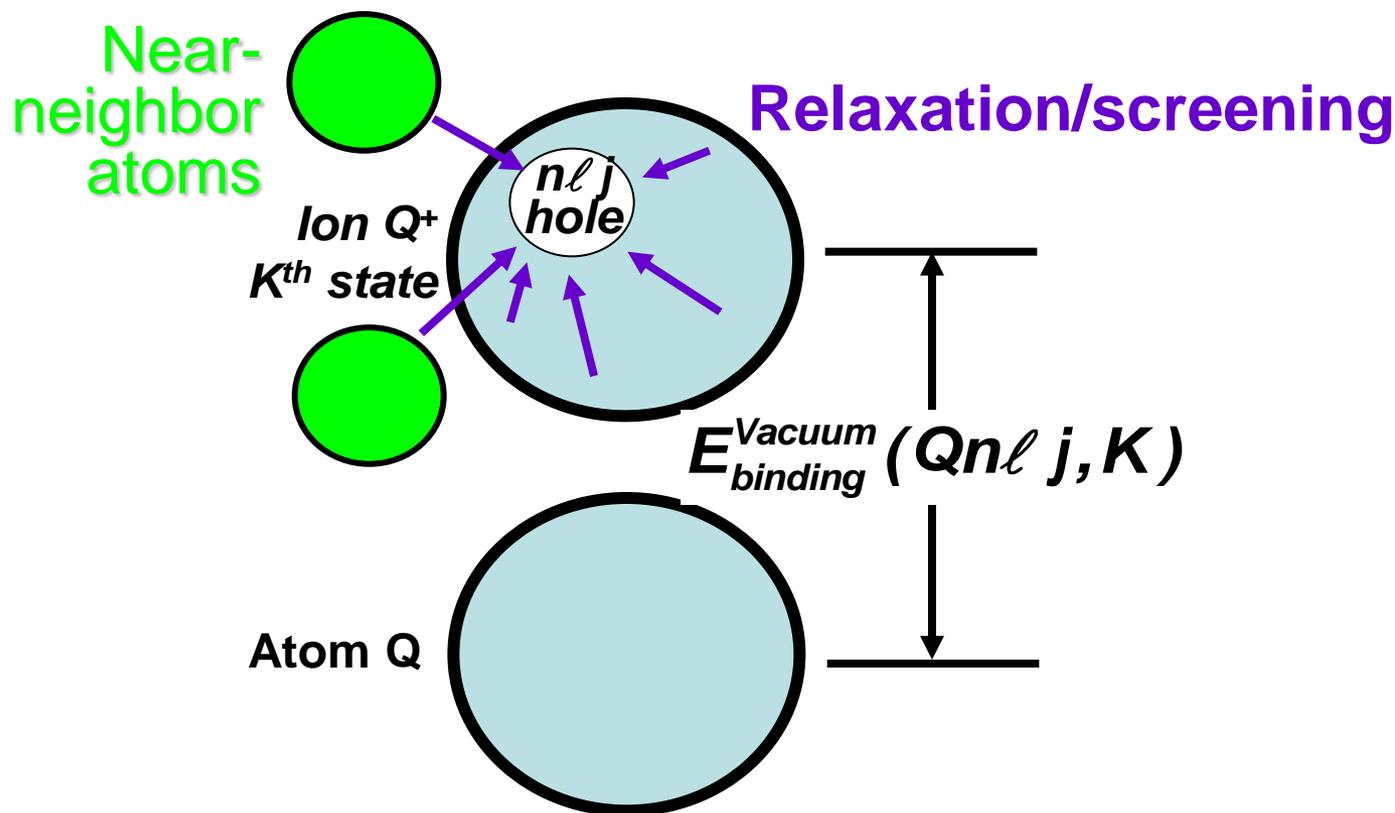
Temporal Resolution

@ Higher Pressures

Basic energetics—Many e⁻ & many atom picture

$$h\nu = E_{\text{binding}}^{\text{Vacuum}} + E_{\text{kinetic}} = E_{\text{binding}}^{\text{Fermi}} + \varphi_{\text{spectrometer}} + E_{\text{kinetic}}$$

$$E_{\text{binding}}^{\text{Vacuum}}(Qn\ell j, K) = E_{\text{final}}(N-1, Qn\ell j \text{ hole}, K) - E_{\text{initial}}(N)$$



INTENSITIES IN PHOTOELECTRON SPECTRA:

- GENERAL: FINAL STATE K (k -SUBSHELL + ALL OTHER DESIG.)

$$\text{INT.}_K \propto |\hat{e} \cdot \langle \Psi_{\text{tot}}^f(N, K) | \sum_{i=1}^N \vec{r}_i | \Psi_i^i(N) \rangle|^2 \quad (\text{DIPOLE APPROX.})$$

- BORN-OPPENHEIMER: e^- 's FAST, VIBRATIONS SLOW

$$\text{INT.}_K \propto \underbrace{|\langle \Psi_{\text{vib}, \nu}^f | \Psi_{\text{vib}, \nu}^i \rangle|^2}_{\text{FRANK-CONDON FACTOR}} |\hat{e} \cdot \langle \Psi_e^f(N, K) | \sum_{i=1}^N \vec{r}_i | \Psi_e^i(N) \rangle|^2$$

- SUDDEN APPROXIMATION: $\psi_k \rightarrow \psi_f = \text{PHOTO}^-$ (FAST)



$$\text{INT.}_K \propto |\langle \Psi_{\text{vib}, \nu}^f | \Psi_{\text{vib}, \nu}^i \rangle|^2 |\langle \Psi_e^f(N-1, K) | \Psi_e^i(N-1, K) \rangle|^2$$

$$|\hat{e} \cdot \langle \psi_f | \vec{r} | \psi_k \rangle|^2 \quad \text{SAME SUBSHELL COUPLING + TOTAL L, S} \rightarrow \text{"MONOPOLE"}$$

↳ NORMAL $\frac{d\sigma_K}{d\Omega}$

- SLATER DETS. FOR $\Psi_e^- = \det(\psi'_1, \psi'_2, \dots, \psi'_{k-1}, \psi'_{k+1}, \dots, \psi'_N)$

$$\Psi_e = \det(\psi_1, \psi_2, \dots, \psi_{k-1}, \psi_{k+1}, \dots, \psi_N)$$

$$\text{INT.}_K \propto |\langle \Psi_{\text{vib}, \nu}^f | \Psi_{\text{vib}, \nu}^i \rangle|^2 |\langle \psi'_1 | \psi_1 \rangle|^2 |\langle \psi'_2 | \psi_2 \rangle|^2 \dots$$

$$|\langle \psi'_{k-1} | \psi_{k-1} \rangle|^2 |\langle \psi'_{k+1} | \psi_{k+1} \rangle|^2 \dots |\langle \psi'_N | \psi_N \rangle|^2$$

$$|\hat{e} \cdot \langle \psi_f | \vec{r} | \psi_k \rangle|^2$$

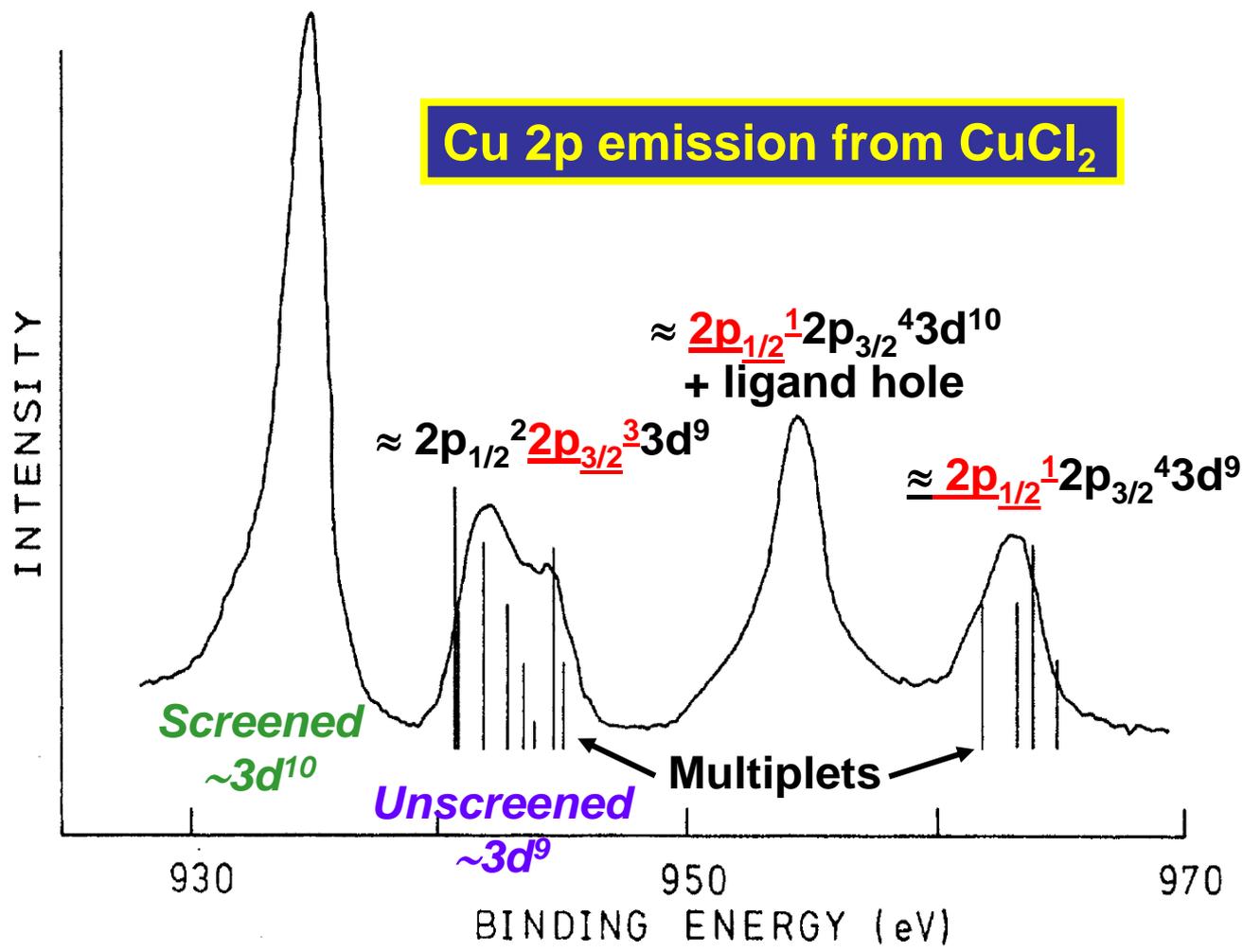
1e- DIPOLE $\rightarrow d\sigma/d\Omega$

(N-1)e- SHAKE-UP/
SHAKE-OFF \rightarrow
"MONOPOLE"

- PLUS DIFFRACTION EFFECTS IN ψ_f ESCAPE

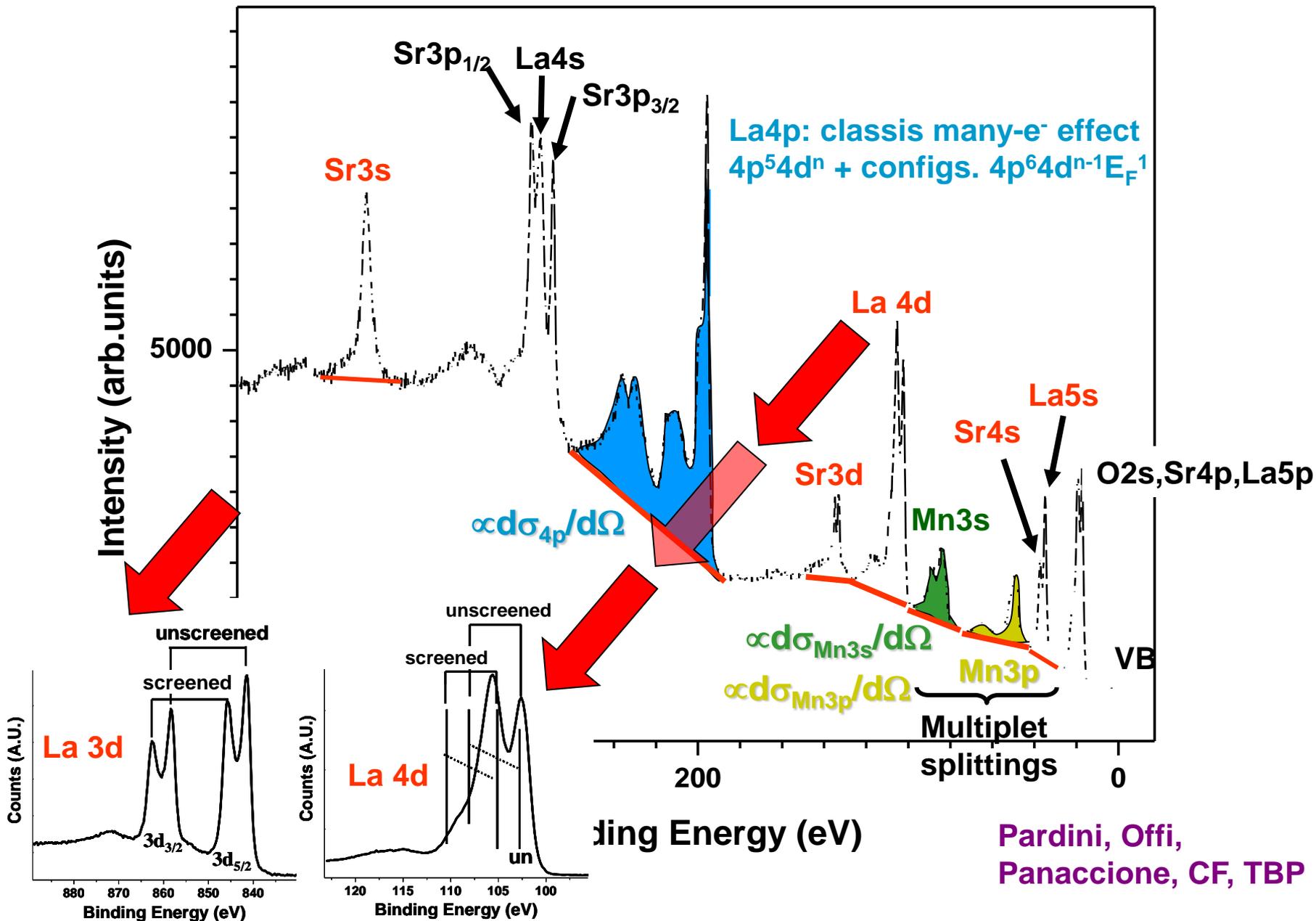
$\approx 2p_{1/2}^2 \underline{2p_{3/2}^3} 3d^{10}$
+ ligand hole

Cu 2p emission from CuCl₂



$$\Psi_{final,K}(N-1) = C_{1,K} \Phi_1(2p_{1/2}^2 \underline{2p_{3/2}^3} 3d^{10} + C_{\ell} \text{ hole}) + C_{2,K} \Phi_2(2p_{1/2}^2 \underline{2p_{3/2}^3} 3d^9)$$

Many-electron and screening effects: $\text{La}_{0.7}\text{Sr}_{0.3}\text{MnO}_3$, $h\nu = 7700 \text{ eV}$



Basic Concepts and Experiments

Core-Level Photoemission

Intensities and Quantitative Analysis, the 3-Step Model

Varying Surface and Bulk Sensitivity

Chemical shifts

Multiplet Splittings

Electron Screening and Satellite Structure



Magnetic and Non-Magnetic Dichroism

Resonant Photoemission

Photoelectron Diffraction and Holography

Valence-Level Photoemission

Band-Mapping in the Ultraviolet Photoemission Limit

Densities of States in the X-Ray Photoemission Limit

Some New Directions

Photoemission with Hard X-Rays (throughout lectures)

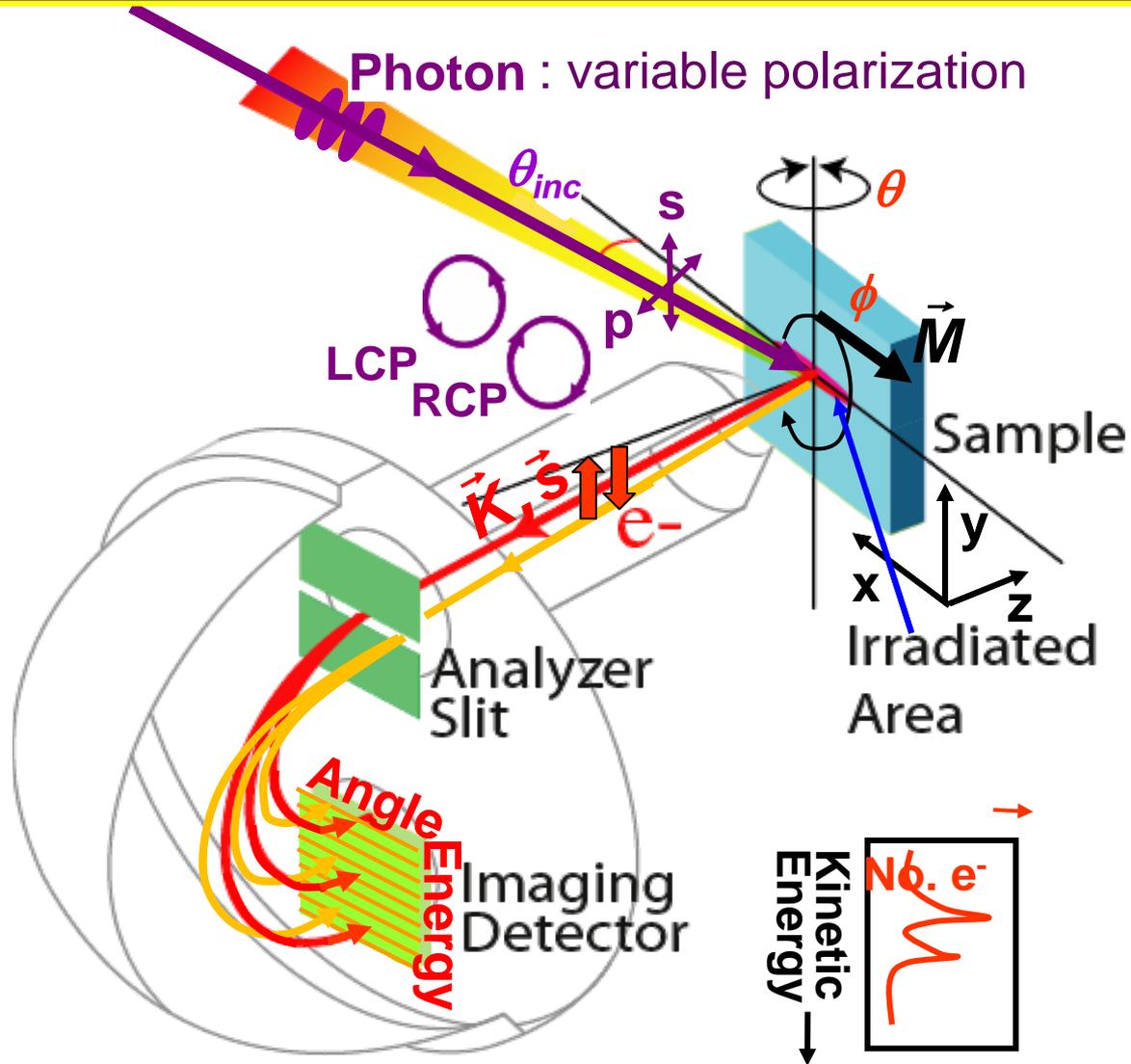
Photoemission with Standing Wave Excitation

Photoemission with Spatial Resolution/Photoelectron Microscopy

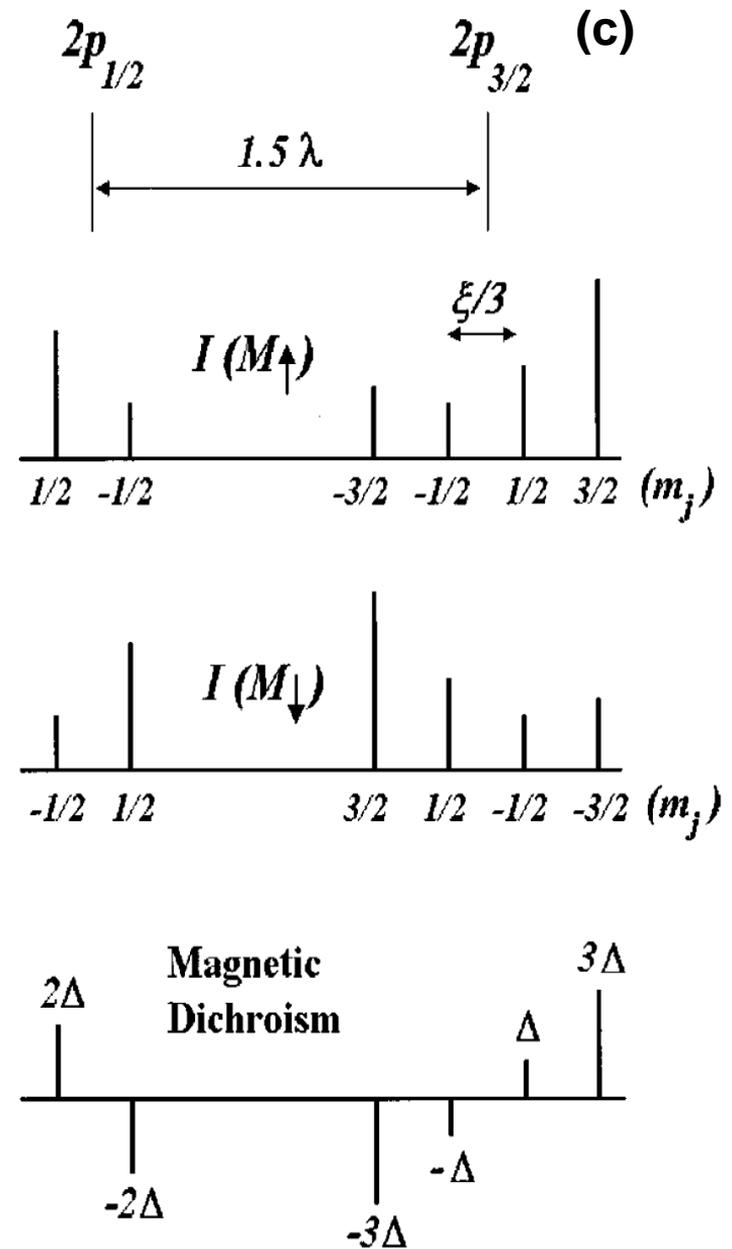
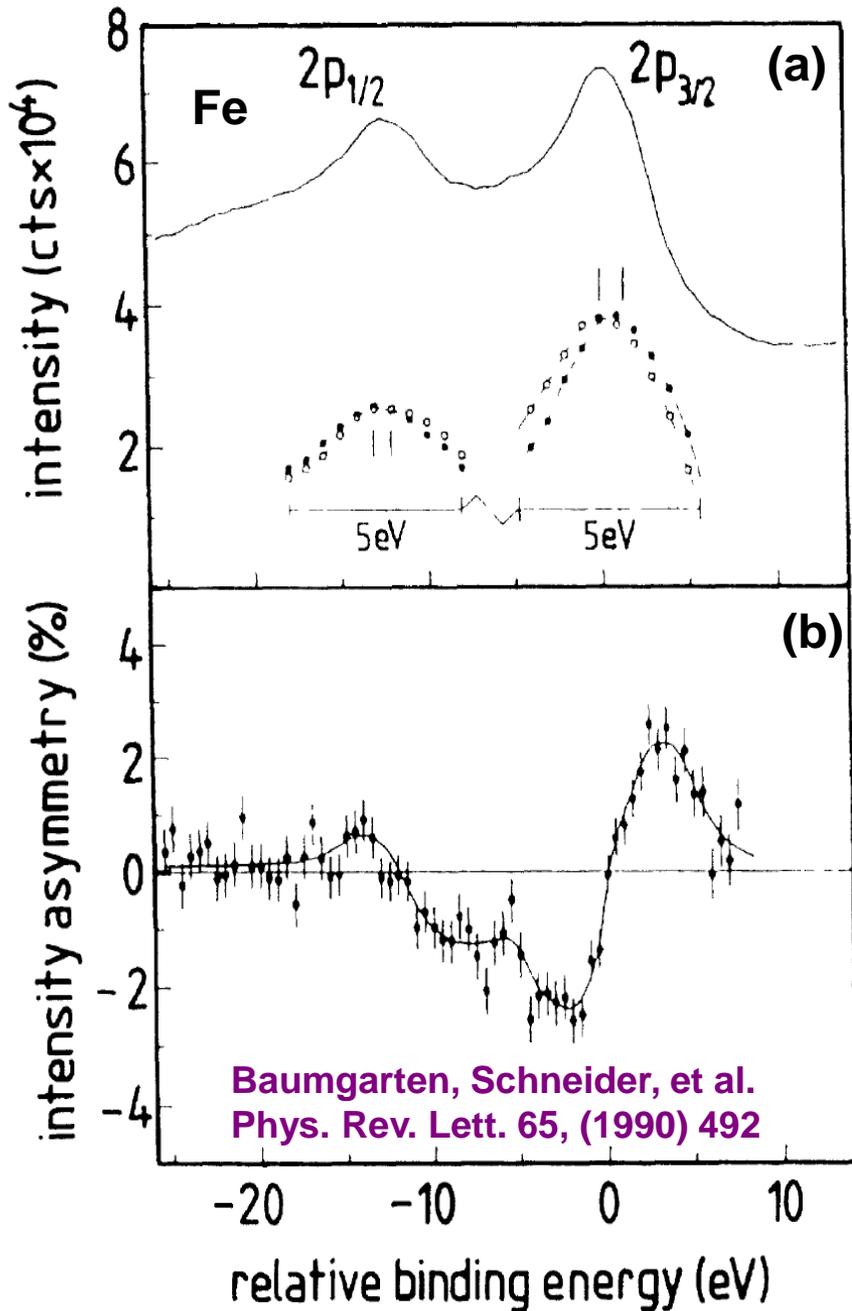
Temporal Resolution

@ Higher Pressures

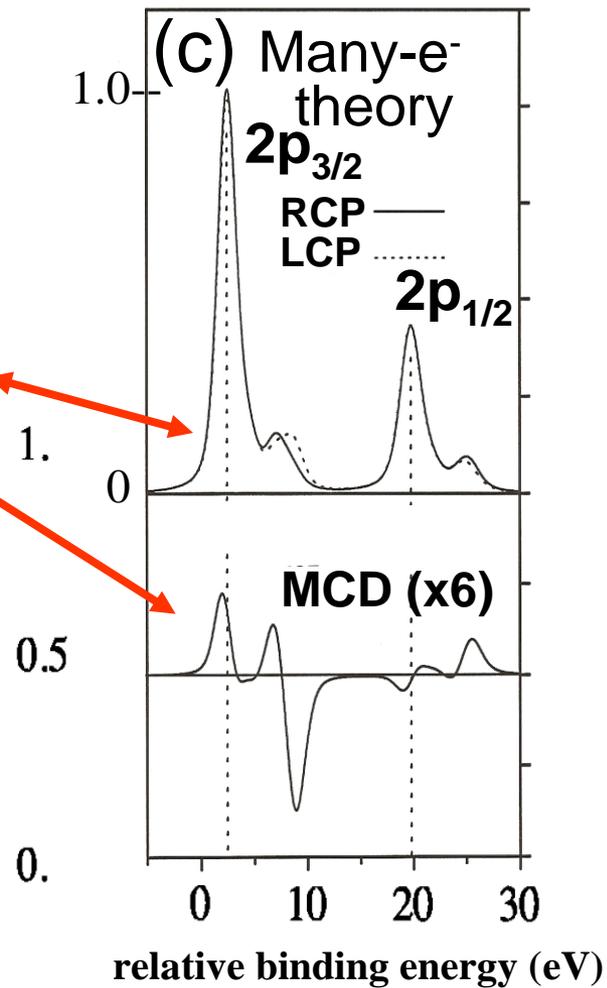
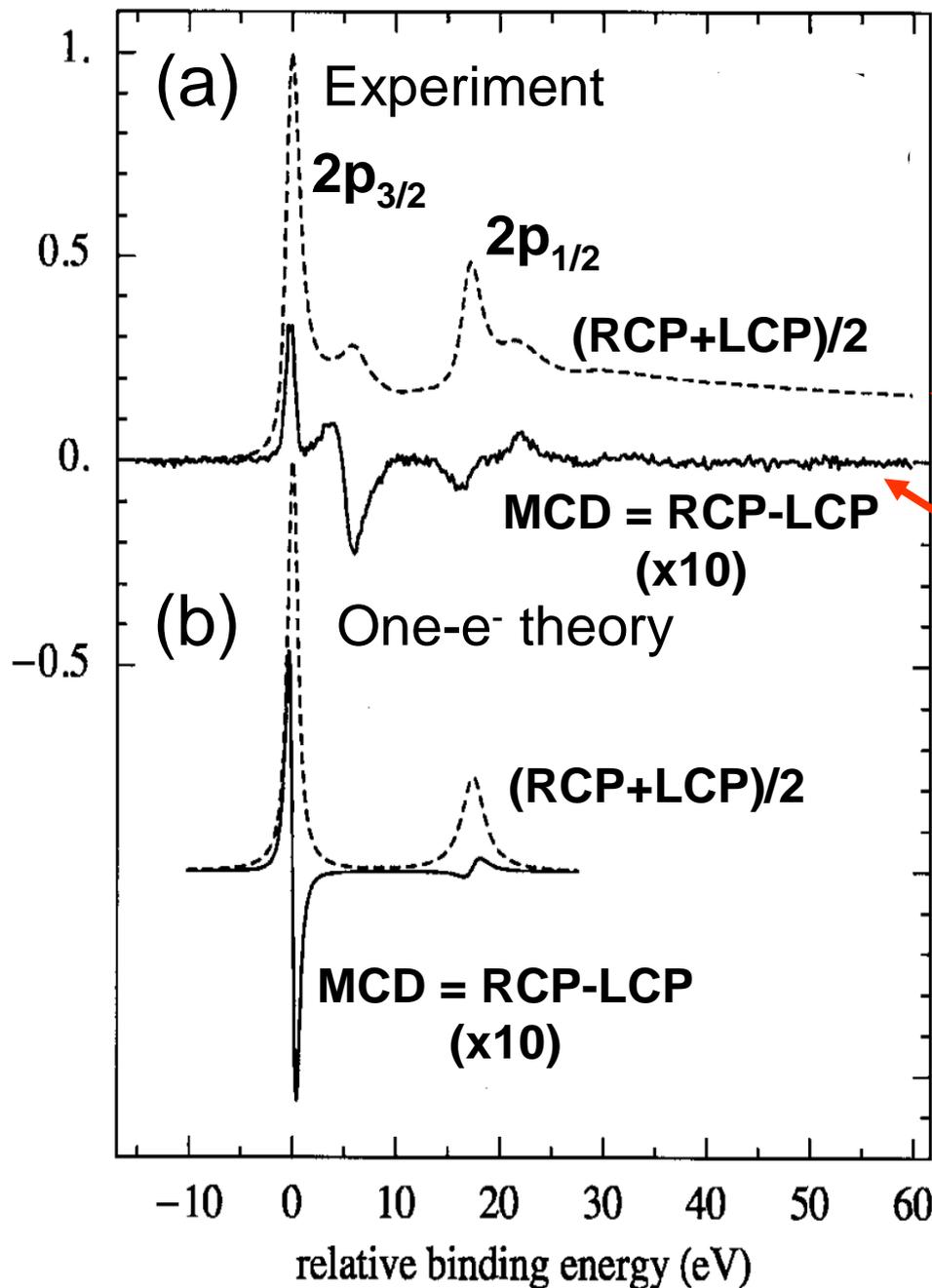
Typical experimental geometry for energy- and angle-resolved photoemission measurements



1st Expt. Magnetic Circular Dichroism 1-e⁻ Theo.



Menchero, Phys. Rev. B 57 (1998) 993

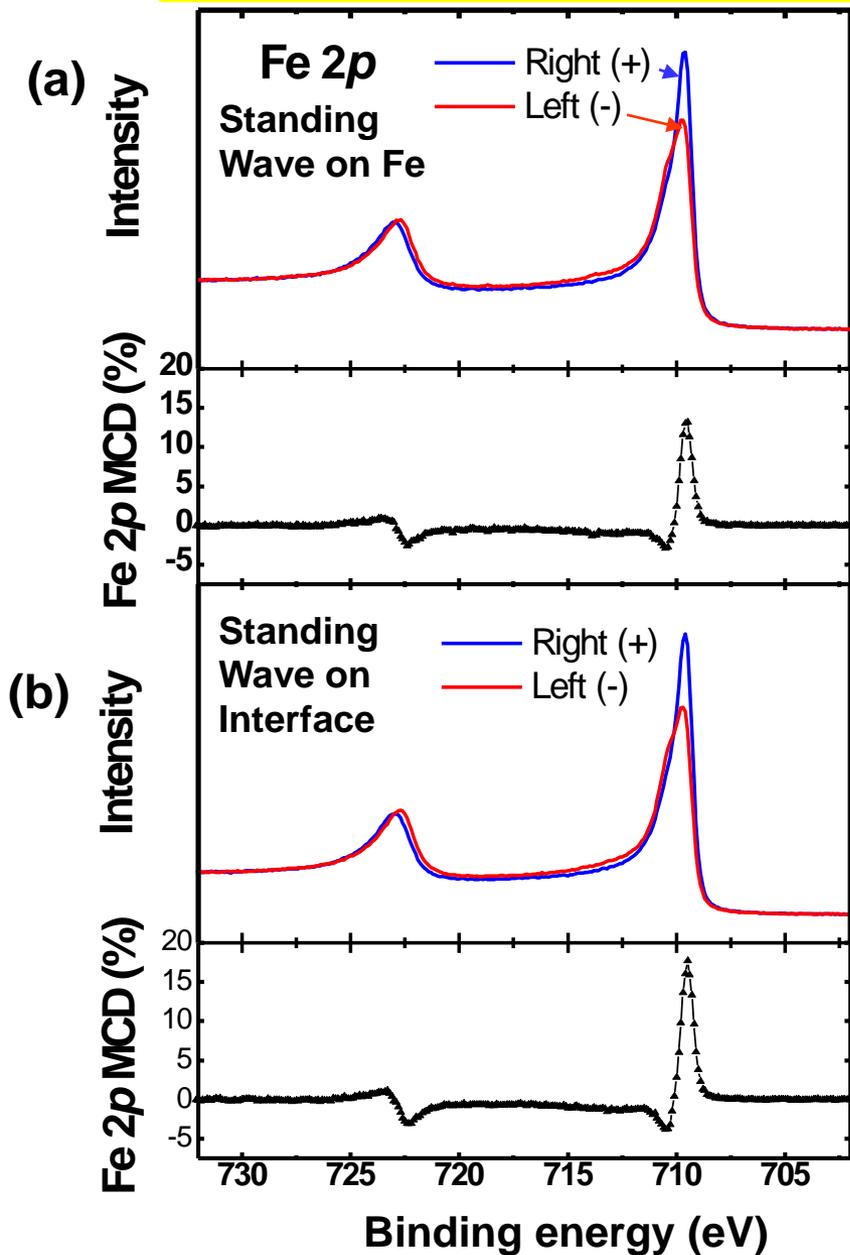


Ni 2p-MCD

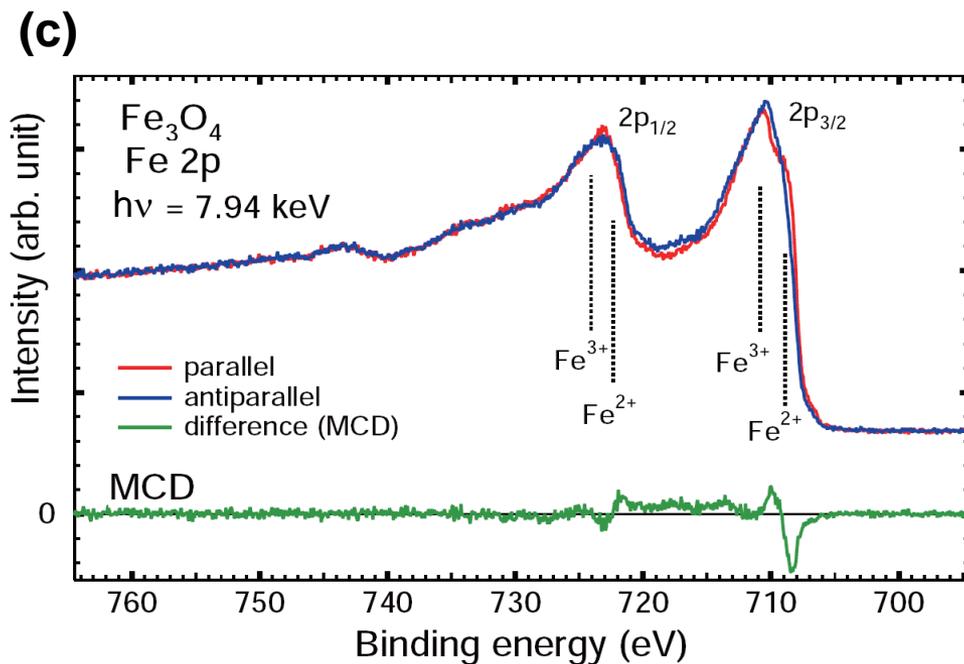
Menchero, Phys. Rev. Lett. 76 (1996) 3208

Van der Laan et al., J. Phys. Condens. Matter 12 (2000) L275

Magnetic circular dichroism with soft and hard x-rays



Magnetic circular dichroism in Fe 2p emission. (a) and (b) MCD data for Fe emission from a 16Å overlayer of Fe on a Cr wedge, excited by a soft X-ray standing wave at 825 eV that has been positioned either (a) maximally on Fe or (b) maximally on Cr near the Fe/Cr interface. (c) MCD for Fe emission from hematite at 7.94 keV, demonstrating for the first time that these effects can be measured with hard X-rays.



Basic Concepts and Experiments

Core-Level Photoemission

Intensities and Quantitative Analysis, the 3-Step Model

Varying Surface and Bulk Sensitivity

Chemical shifts

Multiplet Splittings

Electron Screening and Satellite Structure



Magnetic and **Non-Magnetic** Dichroism

Resonant Photoemission

Photoelectron Diffraction and Holography

Valence-Level Photoemission

Band-Mapping in the Ultraviolet Photoemission Limit

Densities of States in the X-Ray Photoemission Limit

Some New Directions

Photoemission with Hard X-Rays (throughout lectures)

Photoemission with Standing Wave Excitation

Photoemission with Spatial Resolution/Photoelectron Microscopy

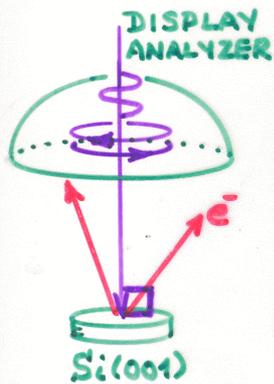
Temporal Resolution

@ Higher Pressures

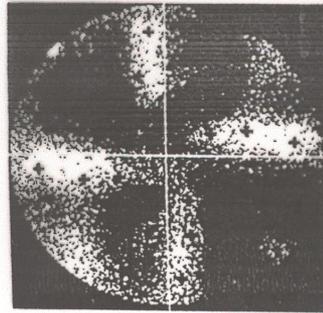
CIRCULAR DICHOISM - NON-MAGNETIC SYSTEMS

Si2p -- 250eV = E_{kin}

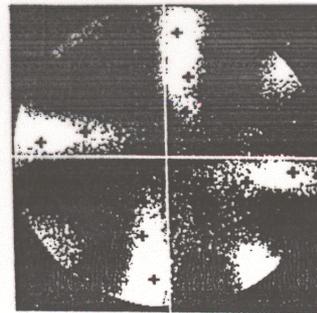
EXPERIMENT



(a) LCP



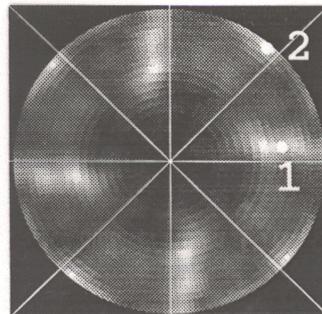
(b) RCP



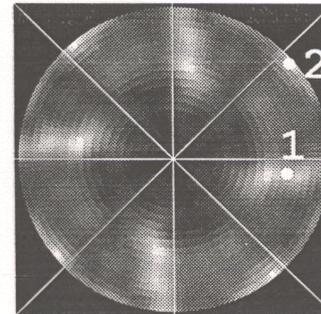
DAIMON ET AL.
JPN. J. APPL. PHYS.
32, L1480 ('93)

THEORY

(c) LCP

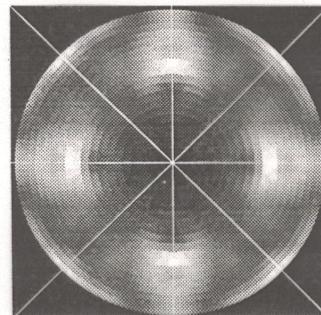


(d) RCP



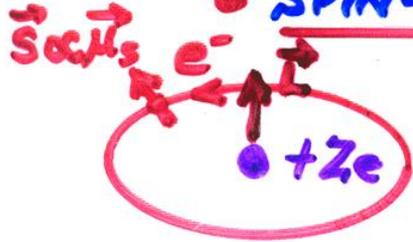
ILADUWELA ET AL.
P. R. B 50, 6203 ('94)

(e) UNPOLARIZED



Adding relativity to the Hamiltonian—spin-orbit coupling

• SPIN-ORBIT SPLITTING OF LEVELS:



⇒ EFFECTIVE \vec{B} (NUCLEUS AROUND e^-) $\propto \vec{L}$

$$\hat{H}_{s.o.} = \xi(r) \vec{L} \cdot \vec{S}$$

- SPLITS ALL nl LEVELS $2(2l+1)$
 - $nl_j = l + 1/2 \rightarrow 2l+2$
 - $nl_j = l - 1/2 \rightarrow 2l$

• MIXES SPIN + ORBITAL ANGULAR MOM.::

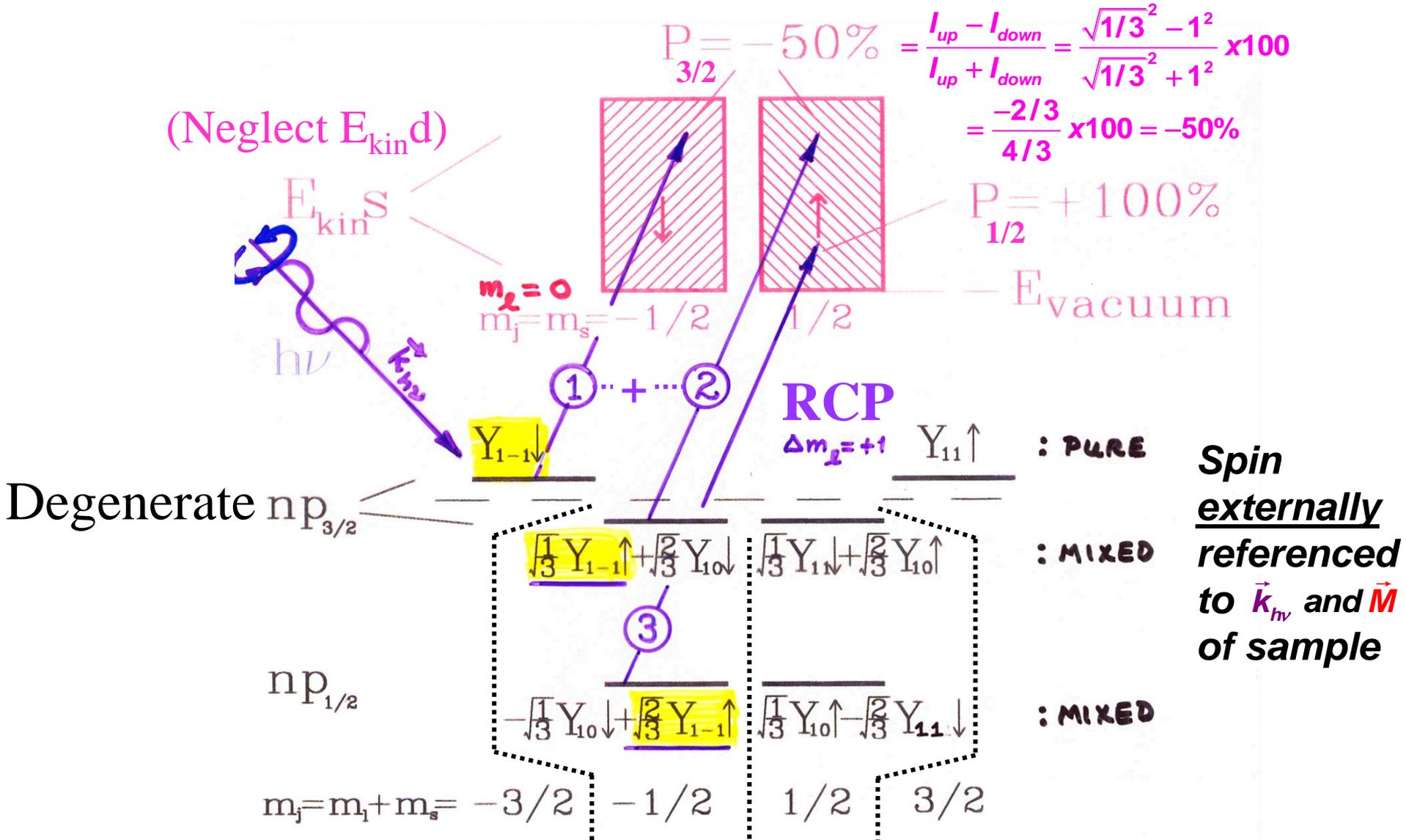
$$\psi_{nljm_j} = C_1 \psi_{nl, m_j - 1/2} \begin{pmatrix} 1 \\ 0 \end{pmatrix} + C_2 \psi_{nl, m_j + 1/2} \begin{pmatrix} 0 \\ 1 \end{pmatrix}$$

\parallel
 $m_s = +1/2$
 \parallel
 \uparrow

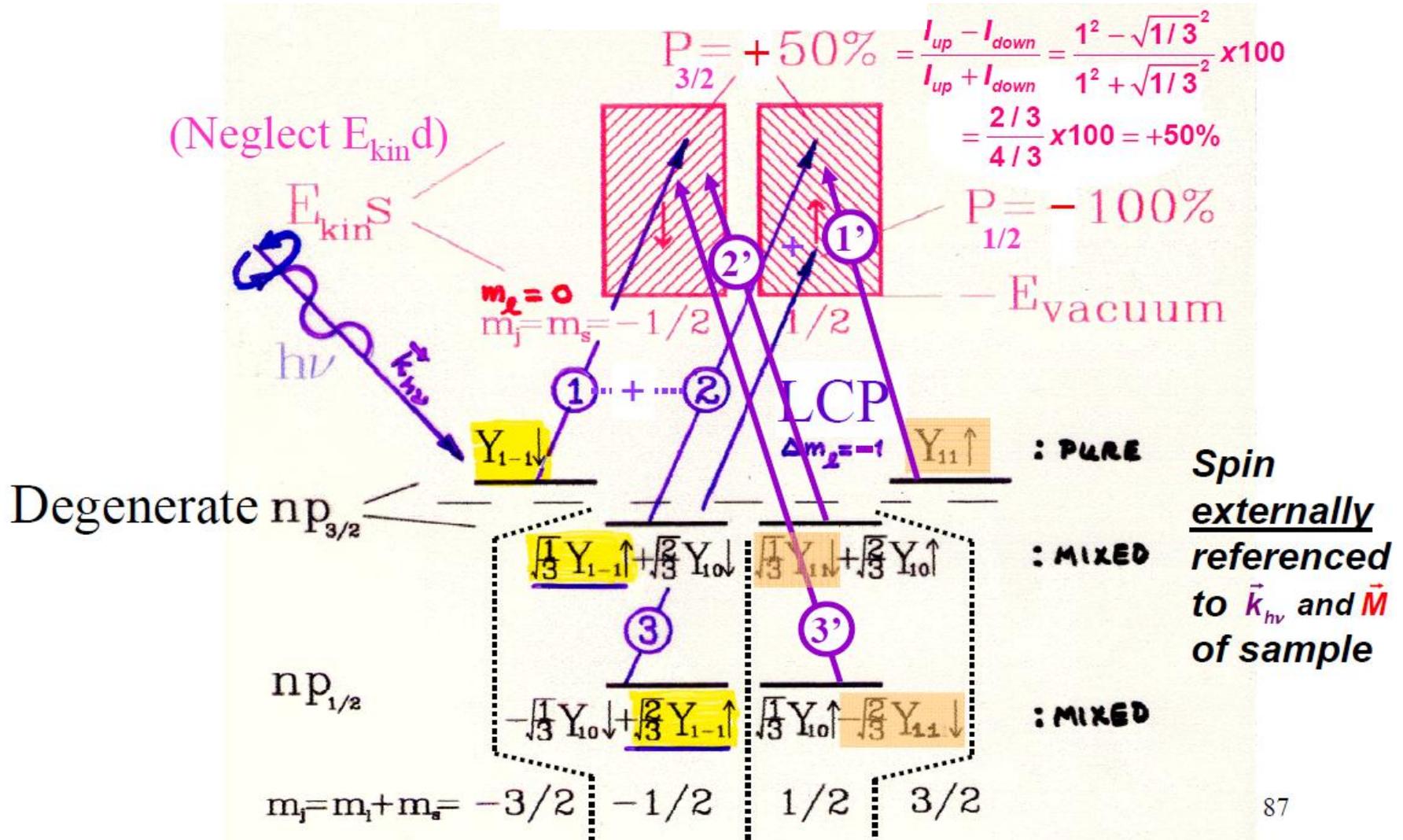
\parallel
 $m_s = -1/2$
 \parallel
 \downarrow

WITH C_1 AND C_2 TABULATED CLEBSCH-GORDAN OR WIGNER $3j$ SYMBOLS

Photoelectron spin polarization from spin-orbit coupling and circularly-polarized radiation—The Fano Effect



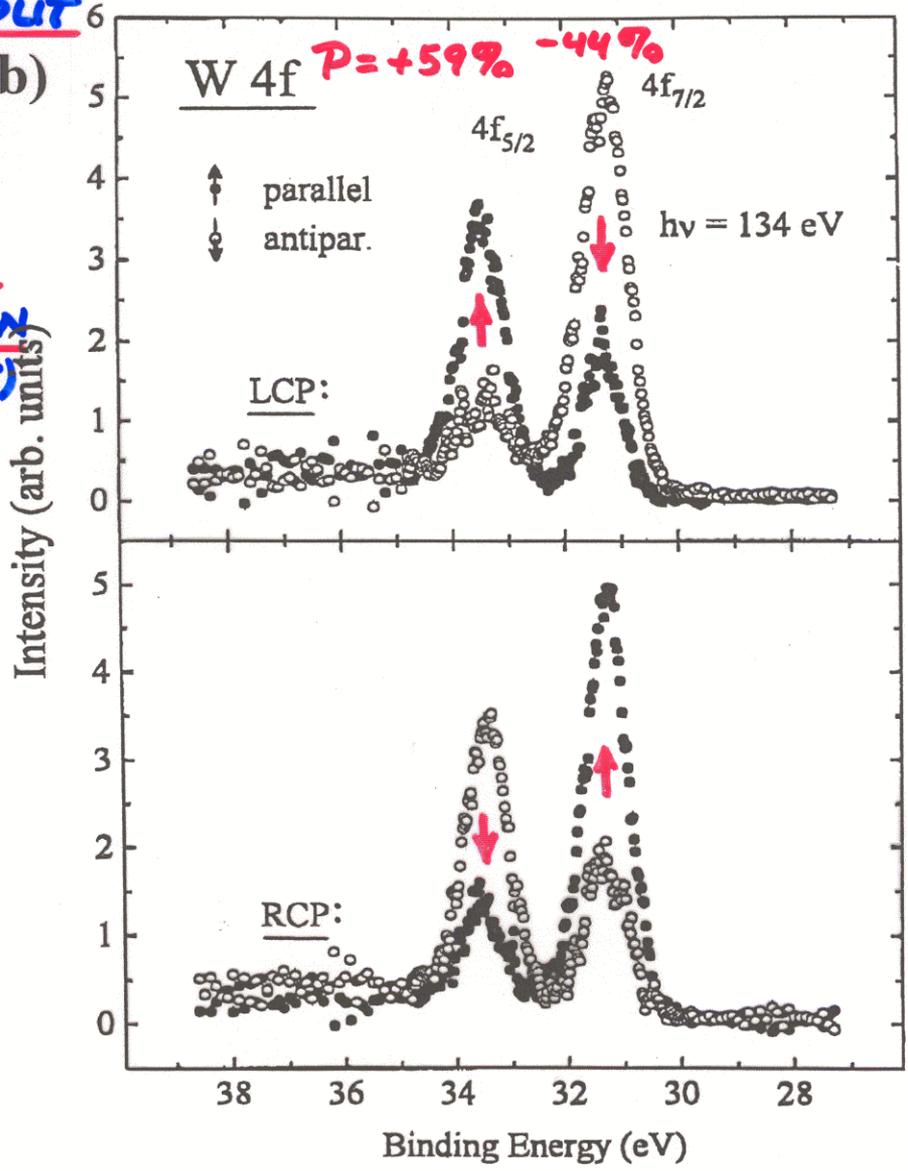
Photoelectron spin polarization from spin-orbit coupling and circularly-polarized radiation—The Fano Effect



Fano effect and Spin polarization in core photoelect

SPIN-ORBIT SPLIT

LEVEL (b)
EXCITED
WITH
CIRCULAR
POLARIZATION
(FANO EFFECT)



EXPT. - STARKE ET AL.
PAB 53, R10544
(1996)

Basic Concepts and Experiments

Core-Level Photoemission

Intensities and Quantitative Analysis, the 3-Step Model

Varying Surface and Bulk Sensitivity

Chemical shifts

Multiplet Splittings

Electron Screening and Satellite Structure

Magnetic and Non-Magnetic Dichroism



Resonant Photoemission

Photoelectron Diffraction and Holography

Valence-Level Photoemission

Band-Mapping in the Ultraviolet Photoemission Limit

Densities of States in the X-Ray Photoemission Limit

Some New Directions

Photoemission with Hard X-Rays (throughout lectures)

Photoemission with Standing Wave Excitation

Photoemission with Spatial Resolution/Photoelectron Microscopy

Temporal Resolution

@ Higher Pressures

Resonant Photoemission—Mn 3d

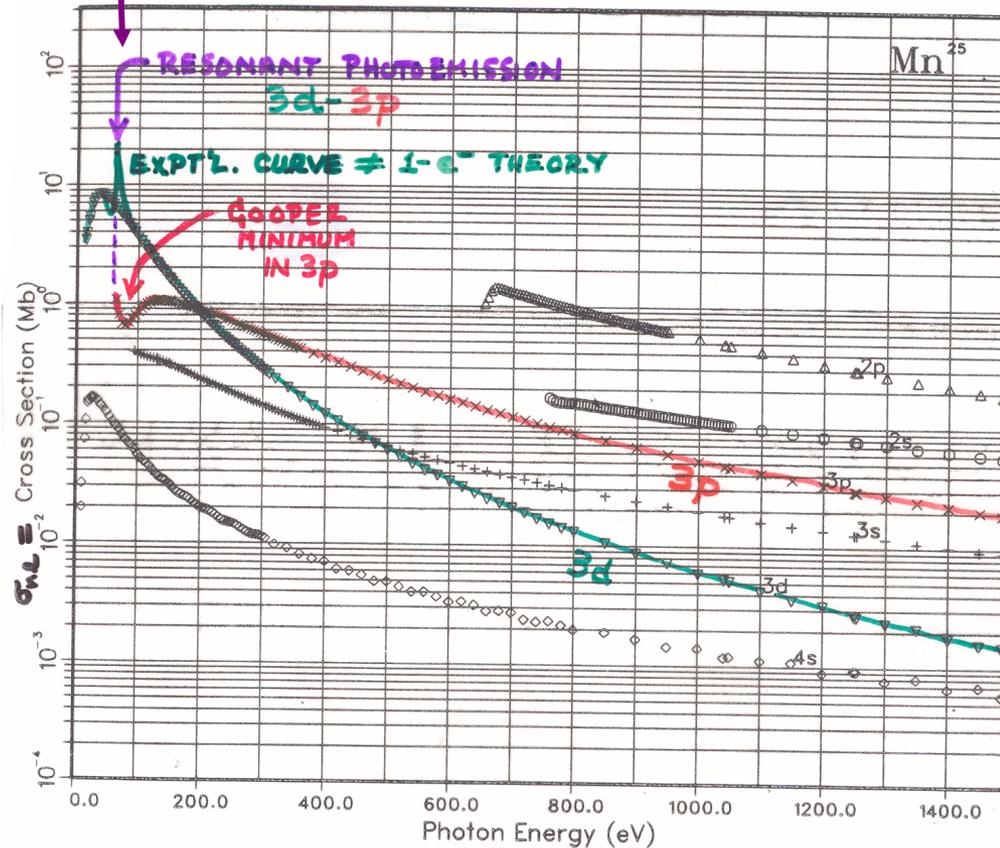
J. J. YEh and I. LINDAU Subshell Photoionization Cross Sections

ATOMIC & NUCLEAR DATA TABLES 32, 45 (1985)

GRAPH I. Atomic Subshell Photoionization Cross Sections for 0–1500 eV, $1 \leq Z \leq 103$

See page 6 for Explanation of Graphs

THEORETICAL ATOMIC CROSS SECTIONS: ENTIRE PERIODIC TABLE

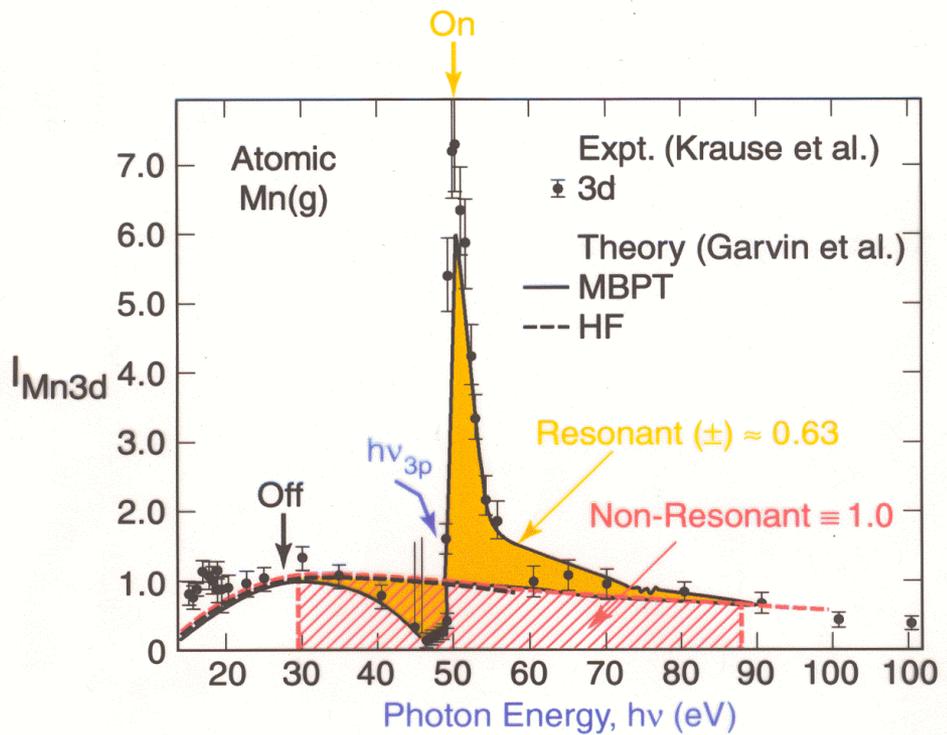
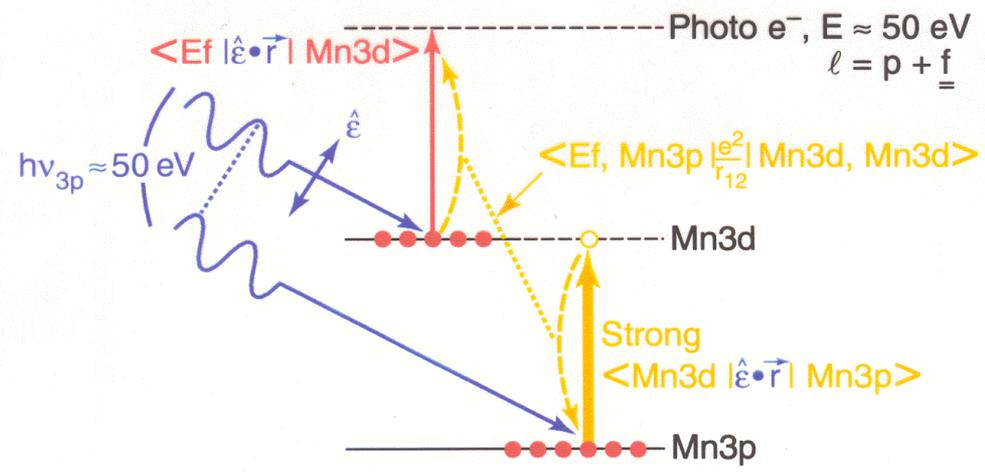


Mn binding energies(eV) are:

1s (2) 6455.26	2s (2) 755.155	2p (6) 653.681
3s (2) 90.8814	3p (6) 60.9150	4s (2) 7.14671
3d (5) 12.0486		

Resonant Photoemission—Mn 3d

Ex. – Mn atom: Mn3d emission, resonance with Mn3p

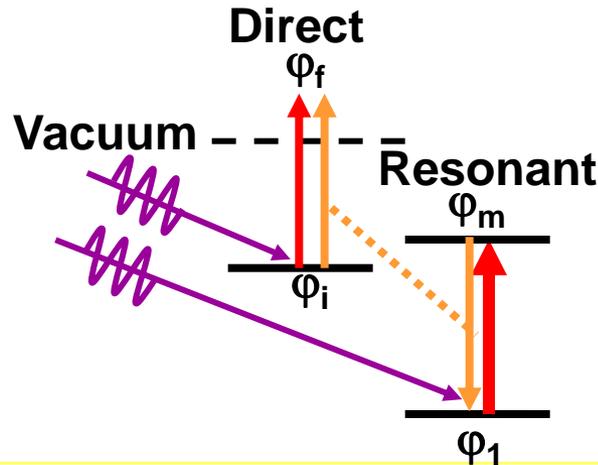


MATRIX ELEMENTS IN THE SOFT X-RAY SPECTROSCOPIES: RESONANT EFFECTS

• Resonant photoemission:

$$I \propto \left| \langle \varphi_f(\mathbf{1}) | \hat{\mathbf{e}} \cdot \vec{r} | \varphi_i(\mathbf{1}) \rangle + \sum_m \langle \varphi_f(\mathbf{1}) \varphi_1(\mathbf{2}) | \frac{e^2}{r_{12}} | \varphi_i(\mathbf{1}) \varphi_m(\mathbf{2}) \rangle \langle \varphi_m(\mathbf{1}) | \hat{\mathbf{e}} \cdot \vec{r} | \varphi_1(\mathbf{1}) \rangle \right|^2$$

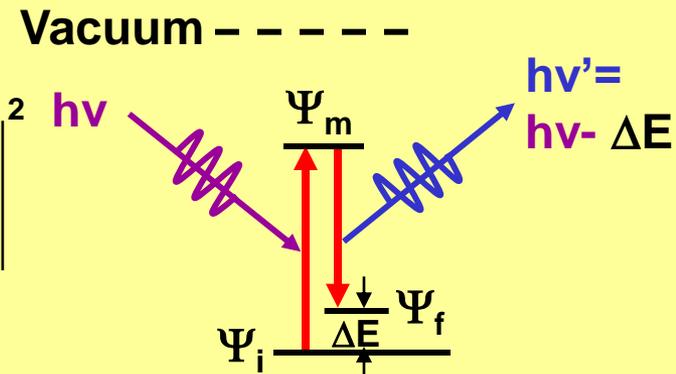
$$\times \delta(h\nu - (E_m - E_1))$$



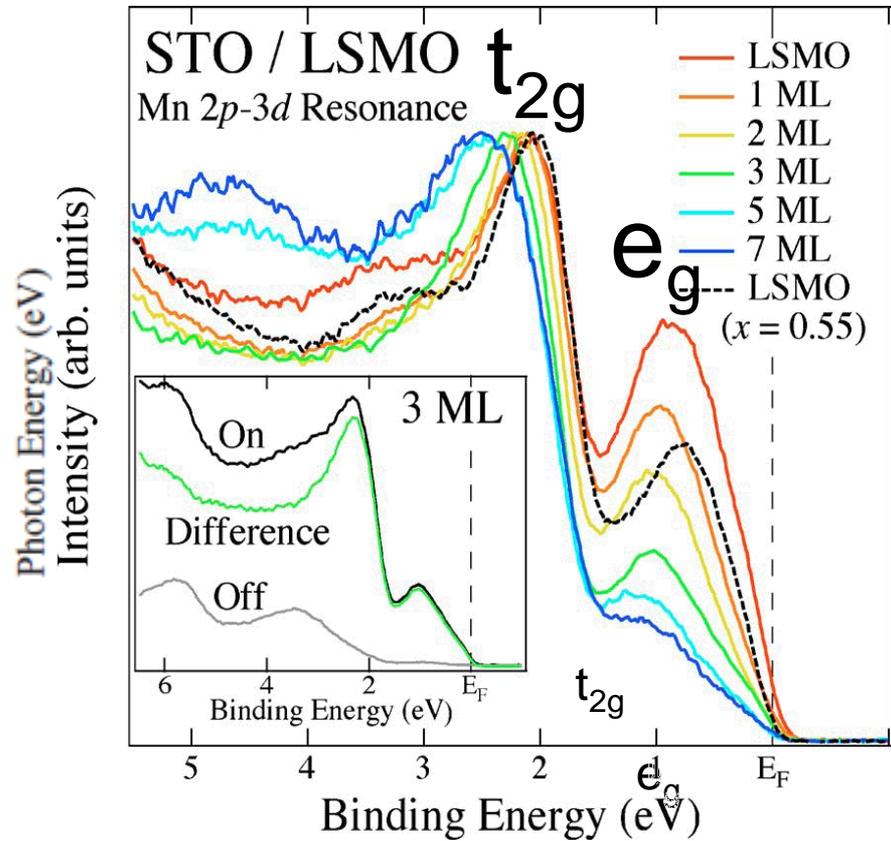
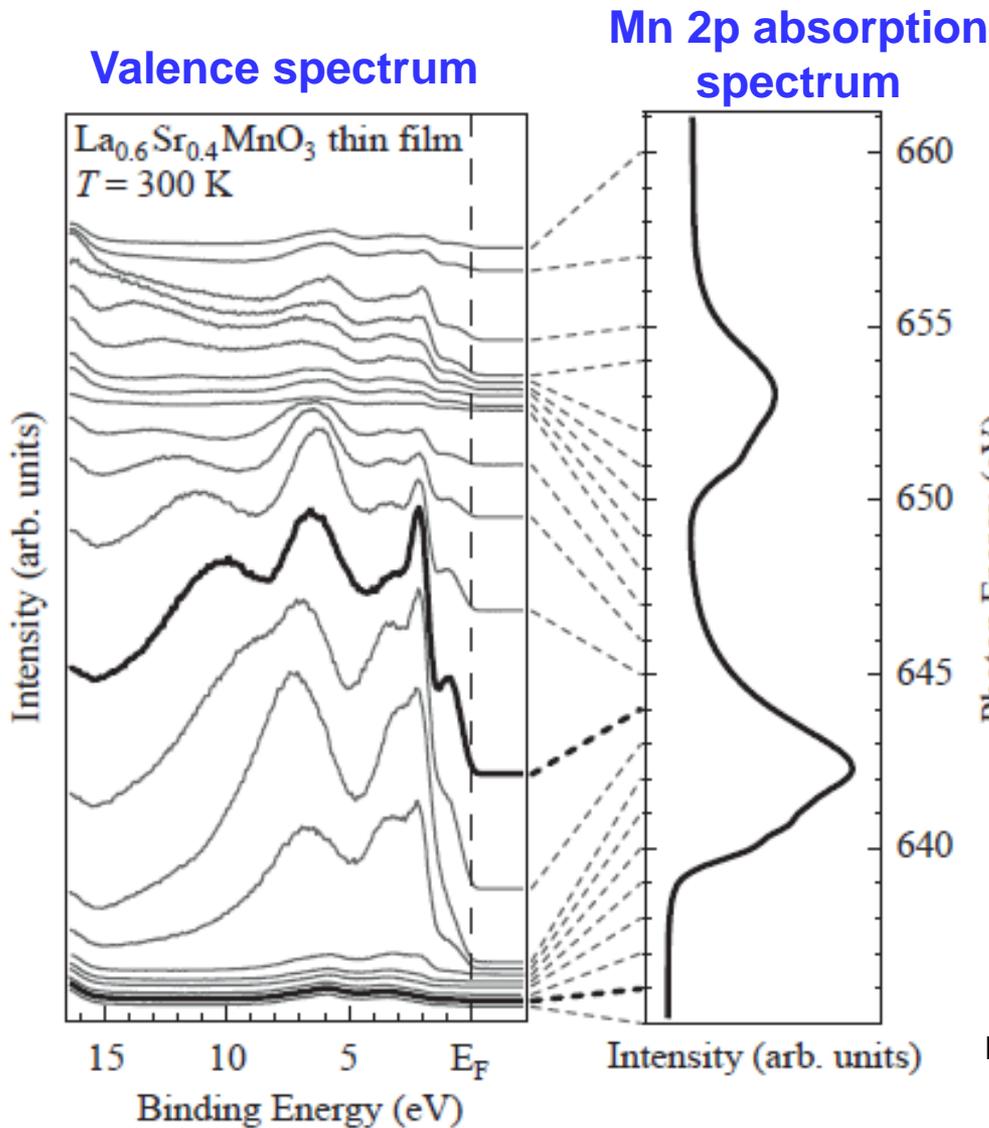
• Resonant inelastic x-ray scattering:

$$I \propto \sum_f \left| \sum_m \frac{\langle \Psi_f(N) | \hat{\mathbf{e}} \cdot \vec{r} | \Psi_m(N) \rangle \langle \Psi_m(N) | \hat{\mathbf{e}} \cdot \vec{r} | \Psi_i(N) \rangle}{h\nu + E_i(N) - E_m(N) - i\Gamma} \right|^2$$

$$\times \delta(h\nu - (E_m(N) - E_i(N)))$$



Resonant Photoemission— $\text{La}_{0.6}\text{Sr}_{0.4}\text{MnO}_3$, Mn 3d with Mn 2p



Prior resonant PS: Fujimori et al., J.A.P 99, 08S903 (2006)

Basic Concepts and Experiments

Core-Level Photoemission

Intensities and Quantitative Analysis, the 3-Step Model

Varying Surface and Bulk Sensitivity

Chemical shifts

Multiplet Splittings

Electron Screening and Satellite Structure

Magnetic and Non-Magnetic Dichroism

Resonant Photoemission



Photoelectron Diffraction and Holography

Valence-Level Photoemission

Band-Mapping in the Ultraviolet Photoemission Limit

Densities of States in the X-Ray Photoemission Limit

Some New Directions

Photoemission with Hard X-Rays (throughout lectures)

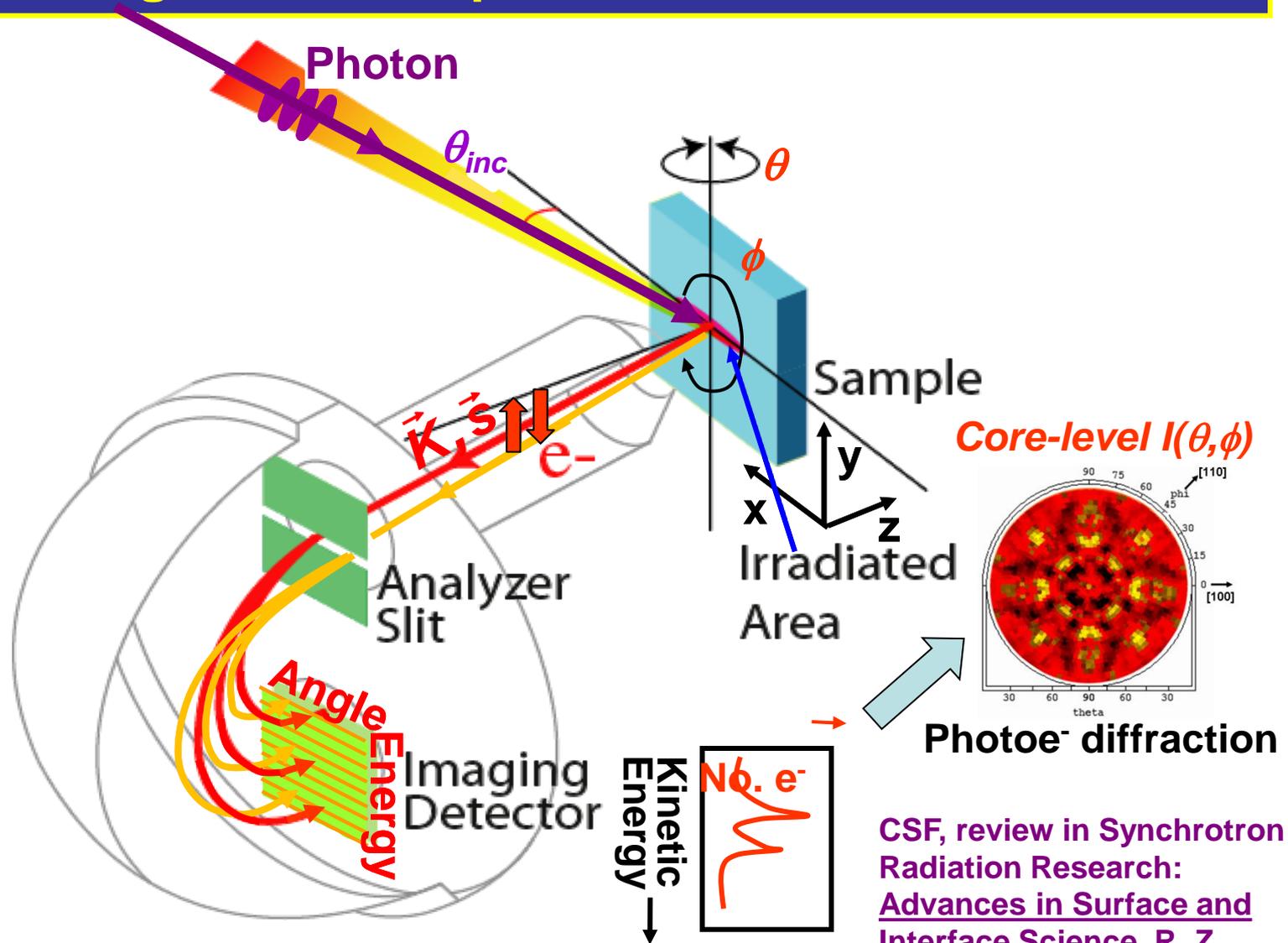
Photoemission with Standing Wave Excitation

Photoemission with Spatial Resolution/Photoelectron Microscopy

Temporal Resolution

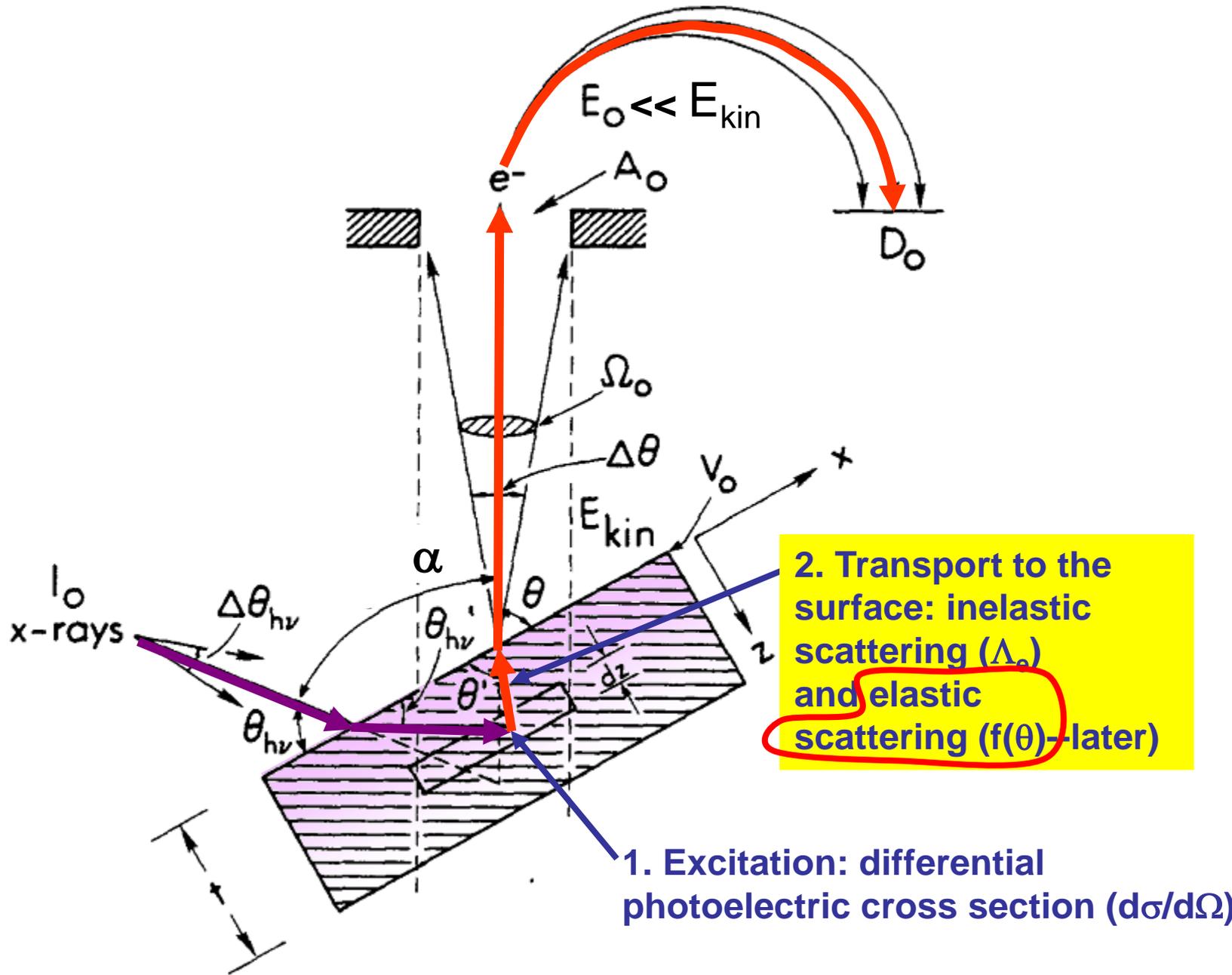
@ Higher Pressures

Typical experimental geometry for energy- and angle-resolved photoemission measurements



CSF, review in *Synchrotron Radiation Research: Advances in Surface and Interface Science*, R. Z. Bachrach, Ed. (Plenum Press, New York, 1992).

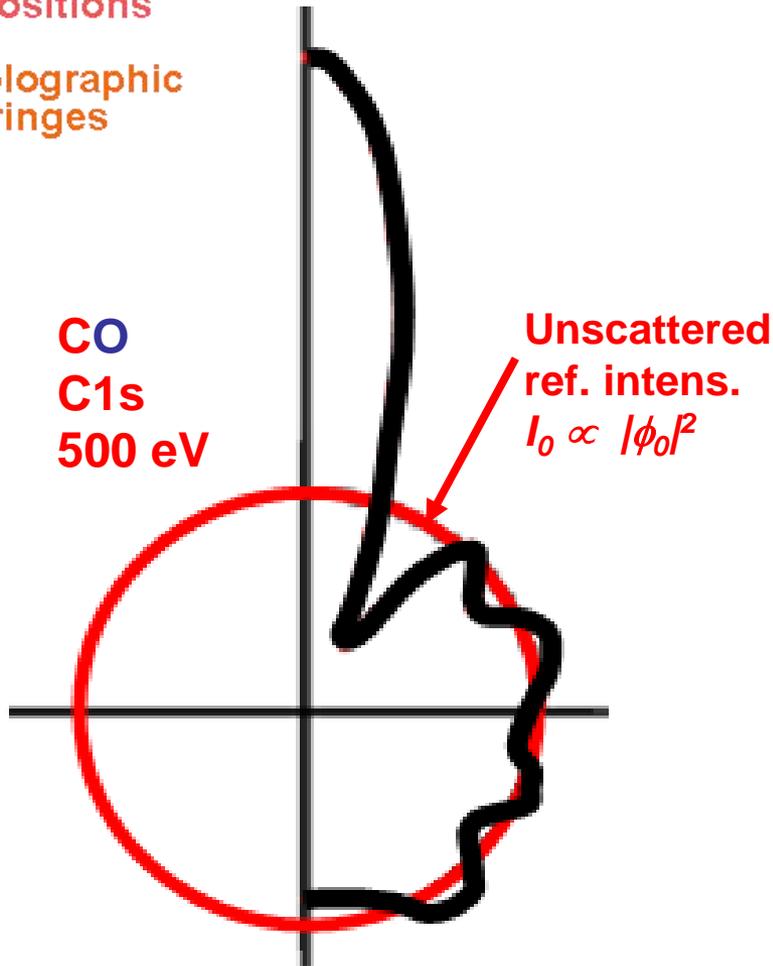
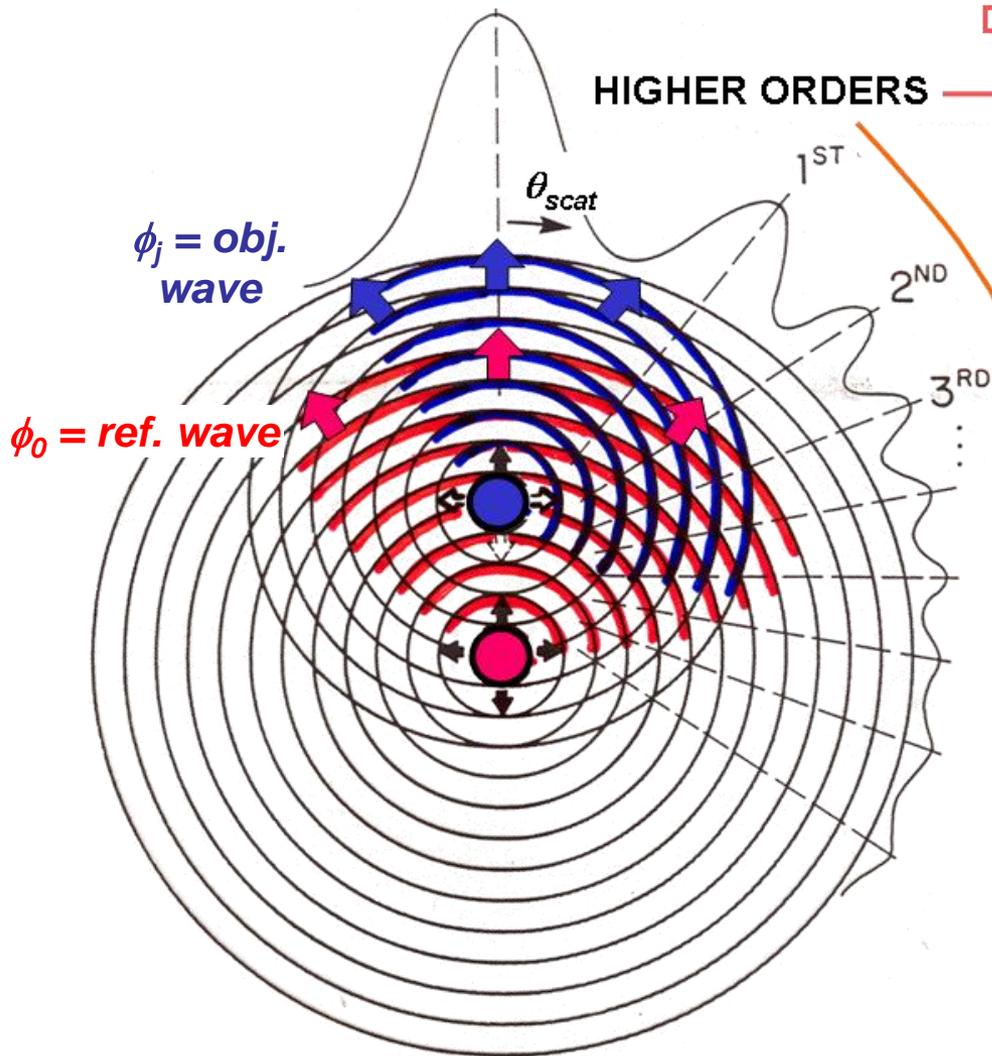
PHOTOELECTRON INTENSITIES—THE 3-STEP MODEL



FORWARD SCATT. = "0TH ORDER" → Bond & Low-Index Directions

HIGHER ORDERS → Bond Lengths & Atomic Positions

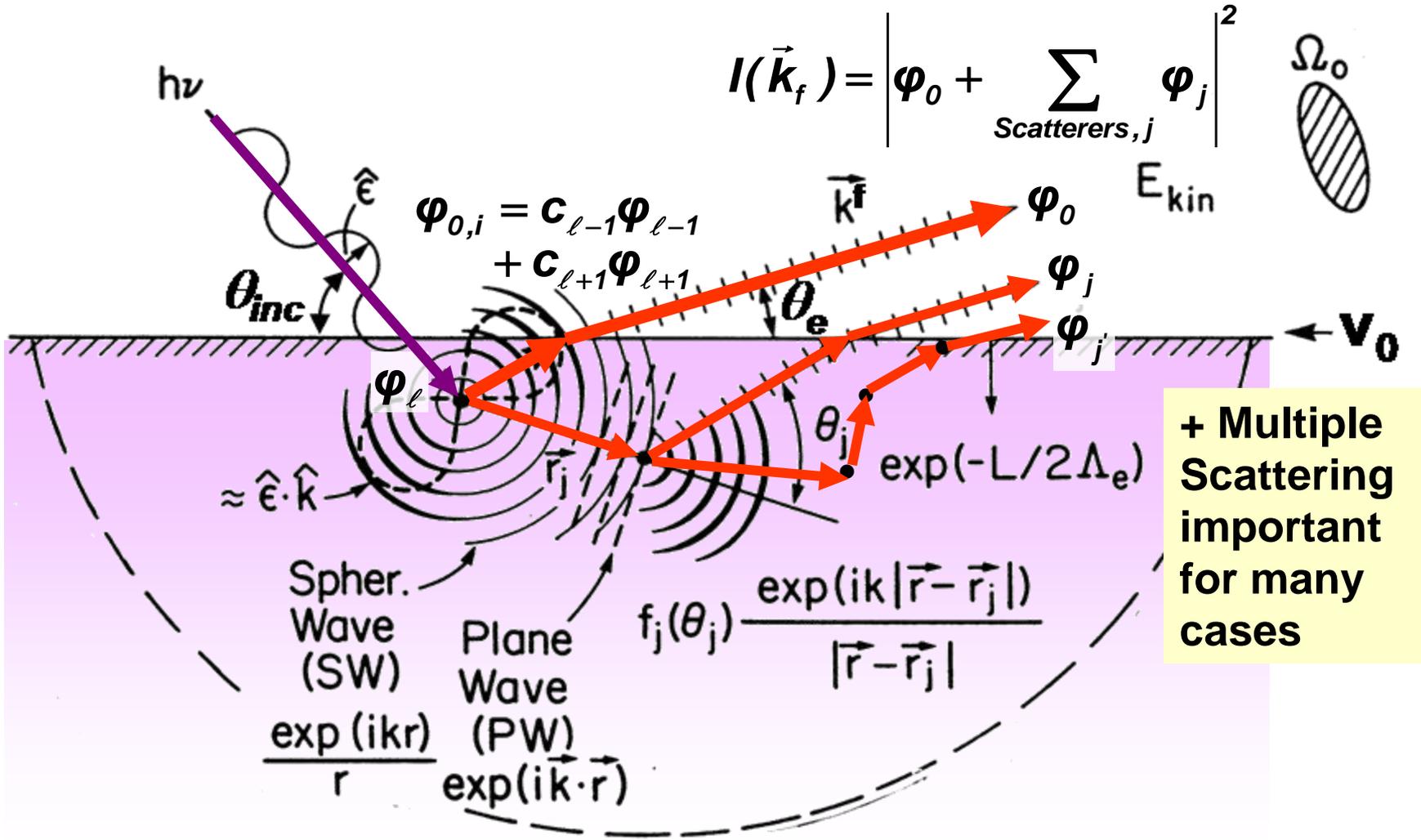
→ Holographic fringes



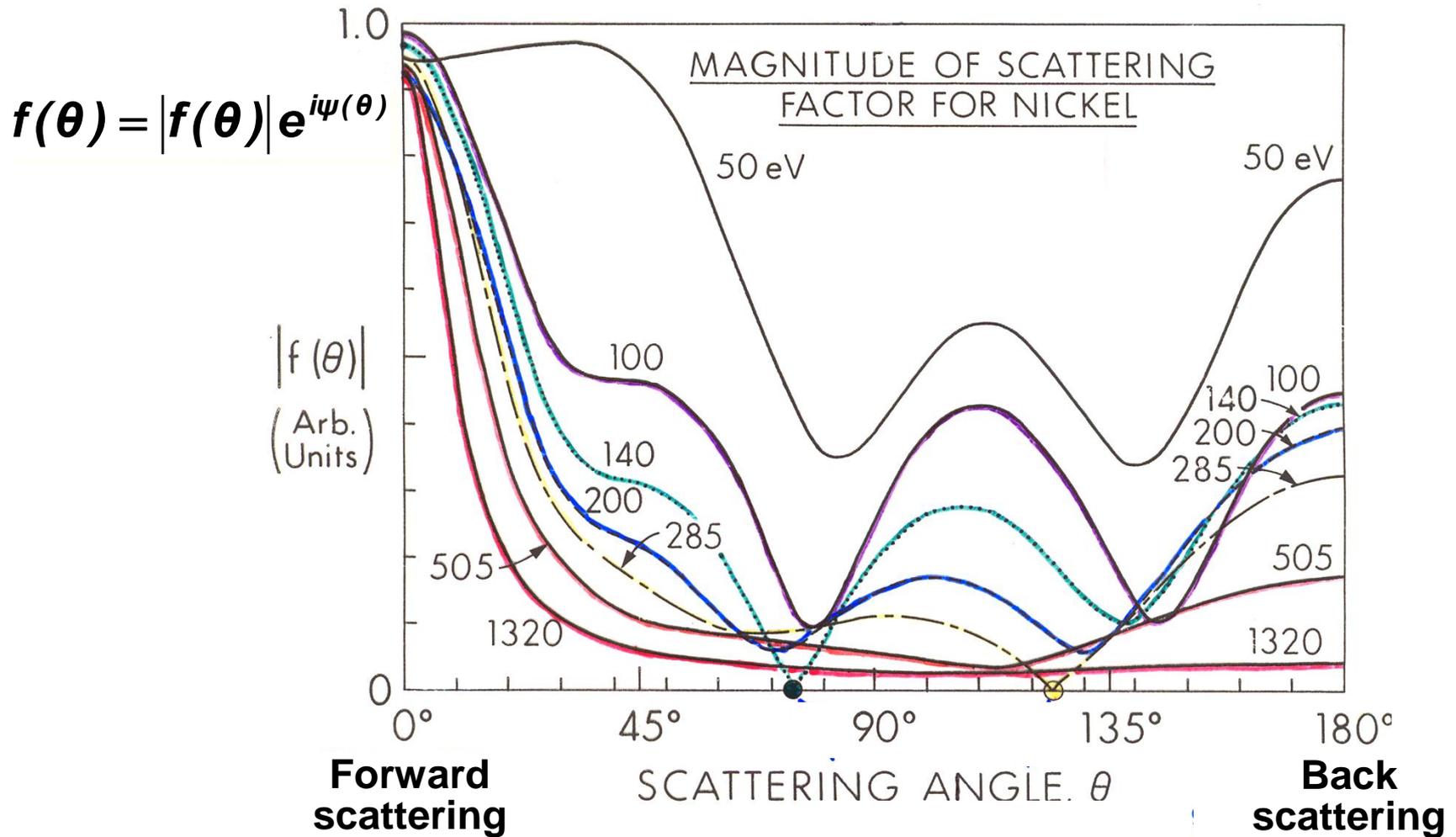
CO
C1s
500 eV

**Photoelectron
Diffraction**

Photoelectron Diffraction: Single Scattering Theory

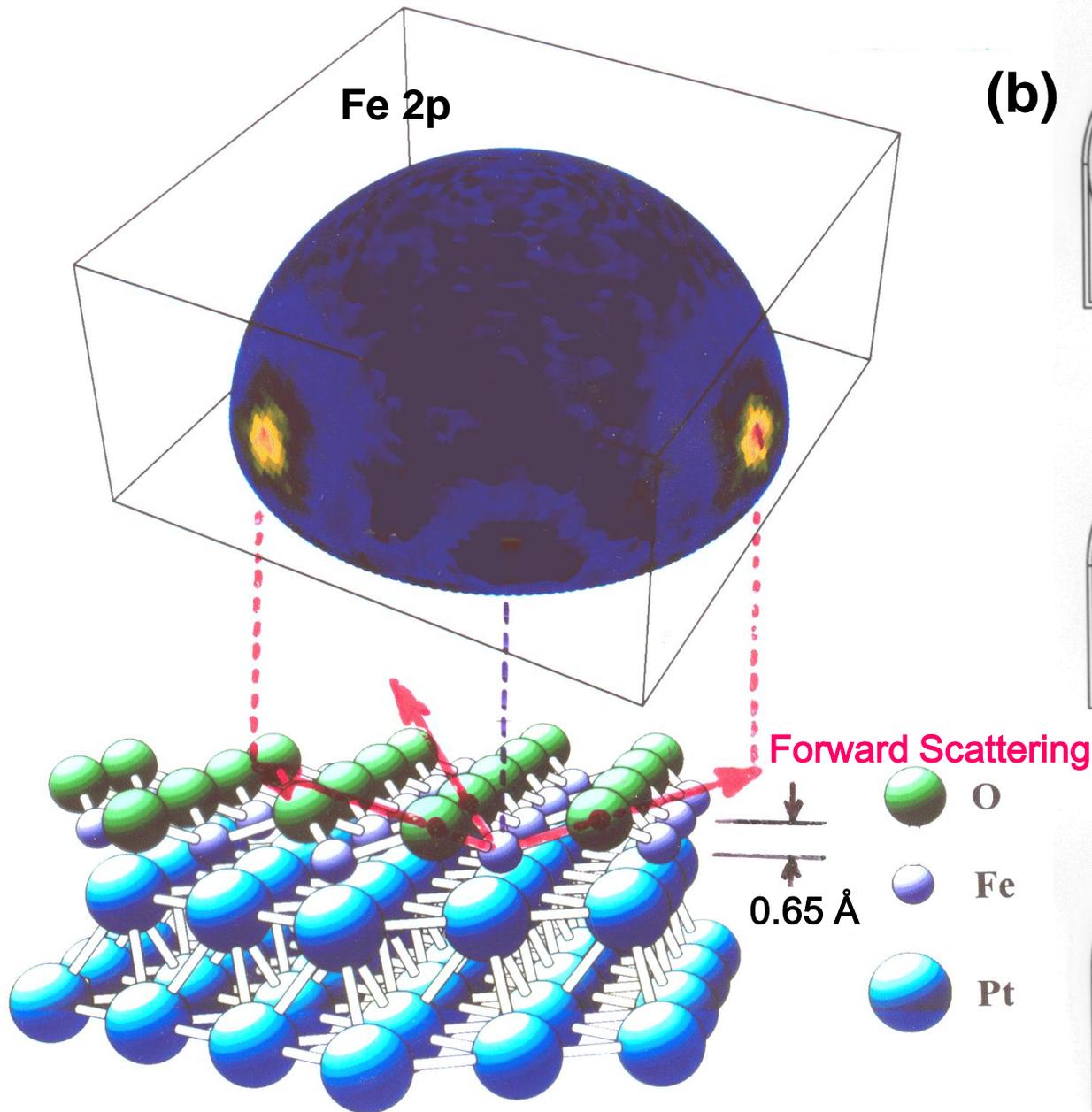


ENERGY DEPENDENCE OF ELECTRON ELASTIC SCATTERING

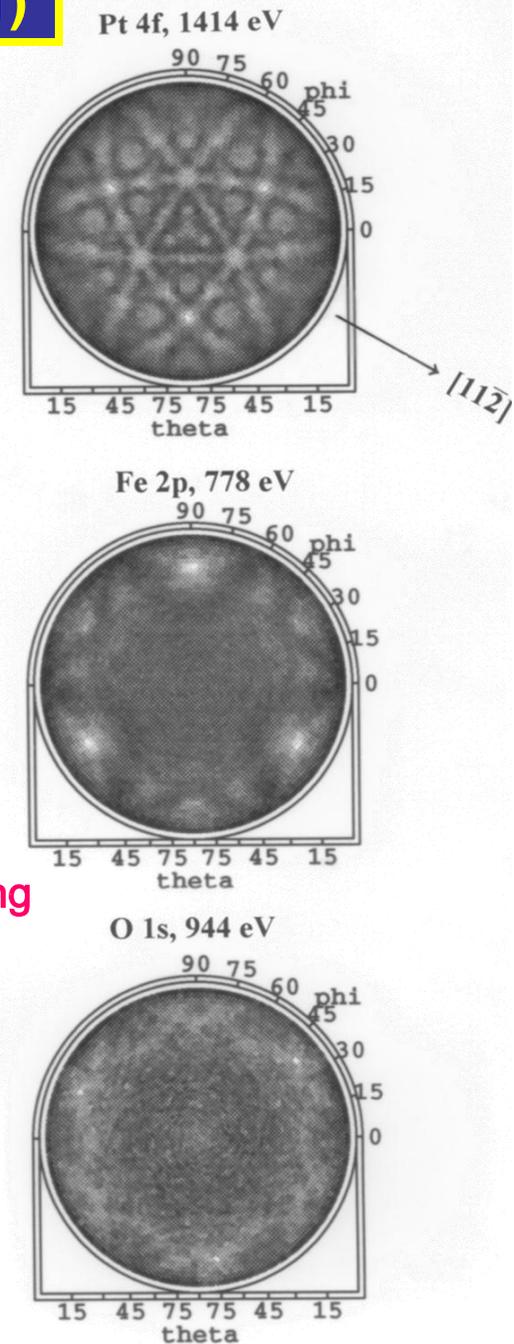


X-ray Photoelectron Diffraction: 1ML FeO on Pt(111)

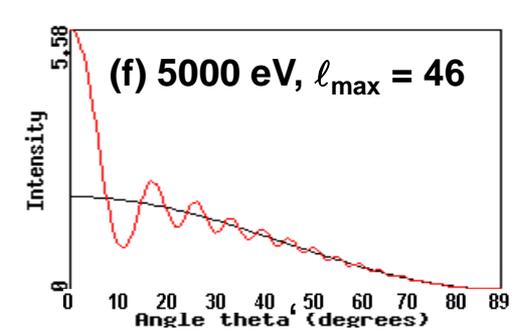
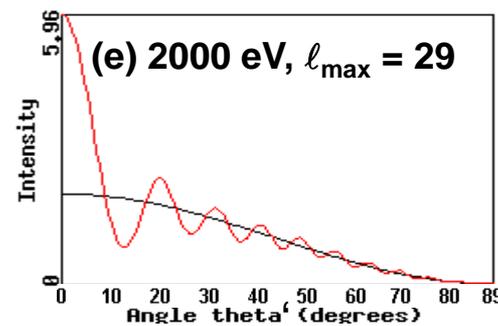
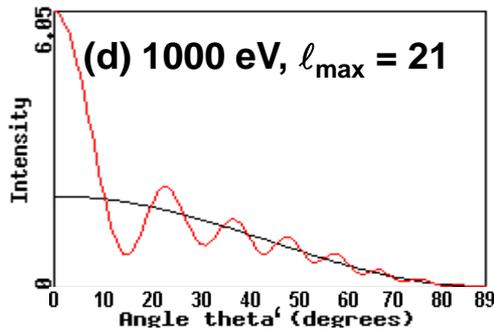
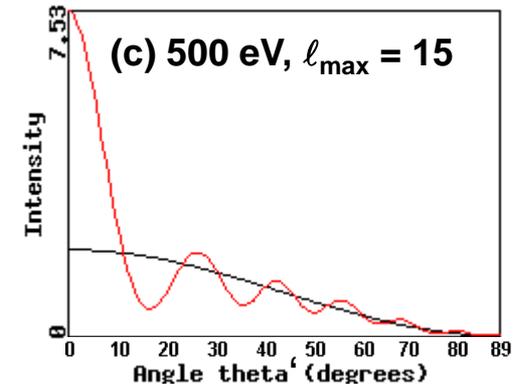
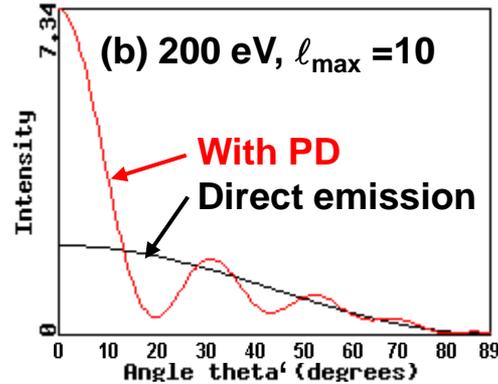
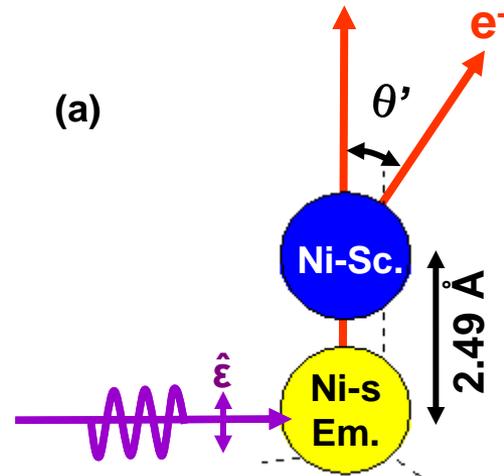
(a)



(b)



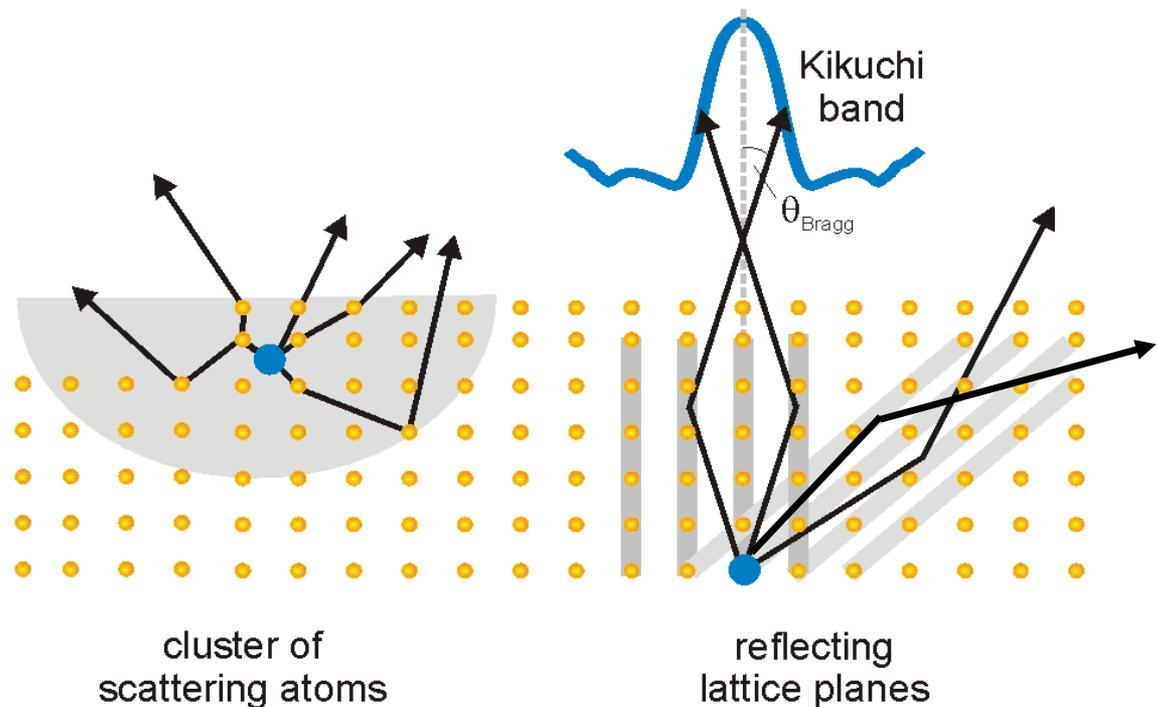
Energy dependence of photoelectron diffraction: Theory



Much more forward peaked at higher energy, weaker in directions away from forward

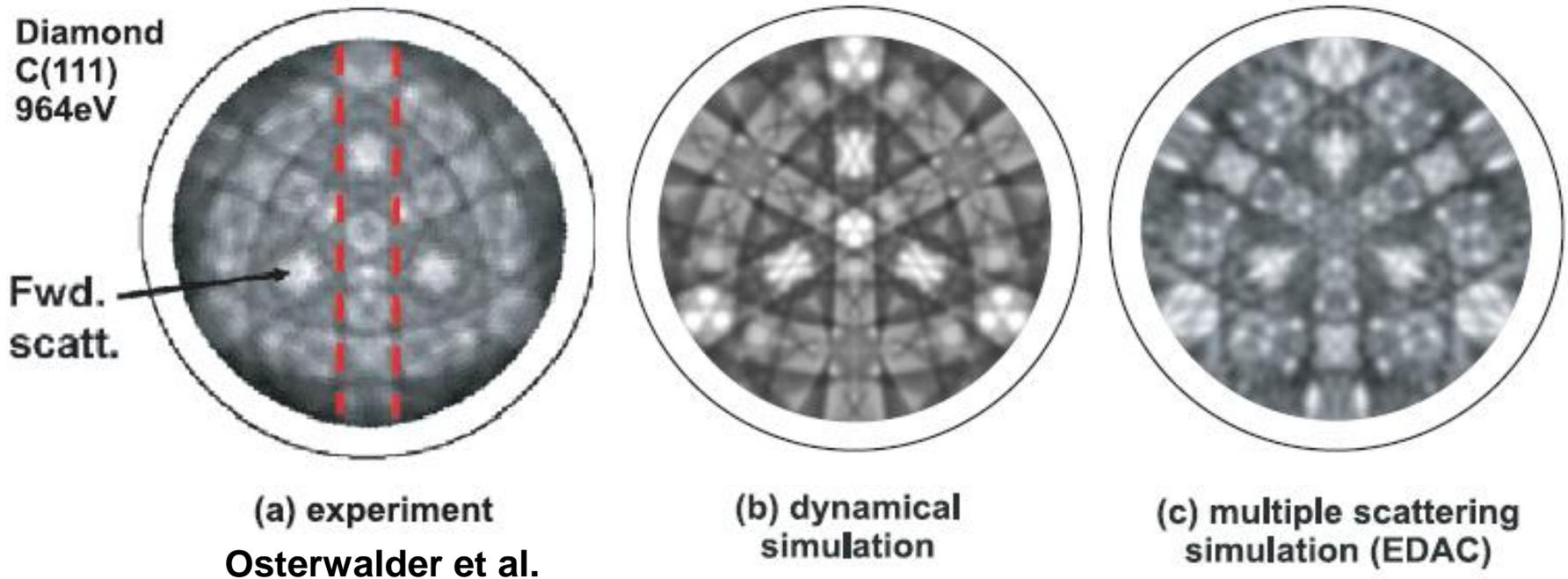
Photoelectron Diffraction with soft and hard x-ray excitation: two viewpoints

The scattering of photoelectrons from localized sources can be described in real space (multiple scattering cluster) and reciprocal space (dynamical theory of electron diffraction)



Soft x-ray excitation \longrightarrow Hard x-ray excitation

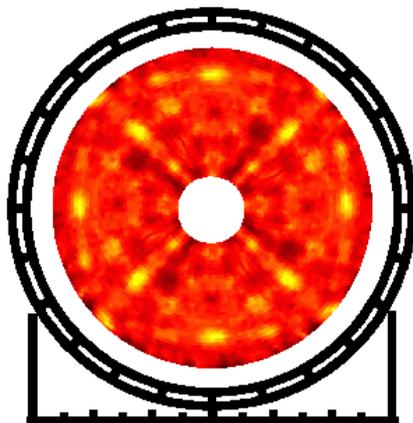
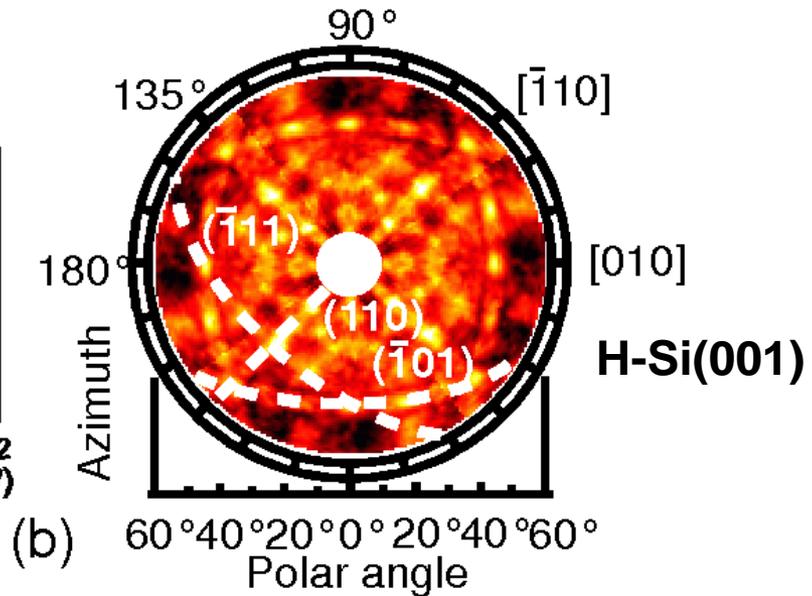
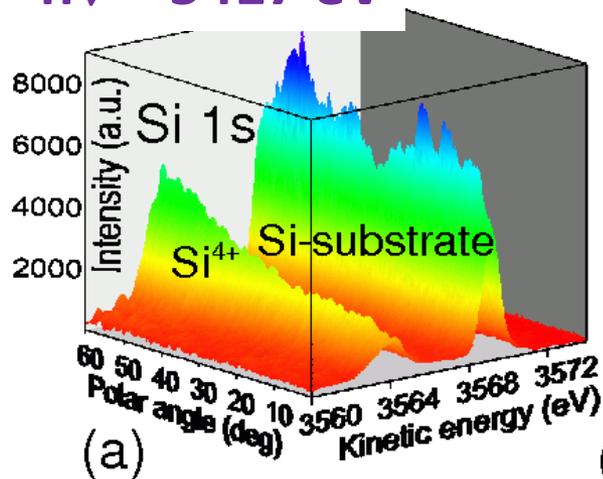
Photoelectron Diffraction with soft and hard x-ray excitation: two viewpoints, expt. vs. theory



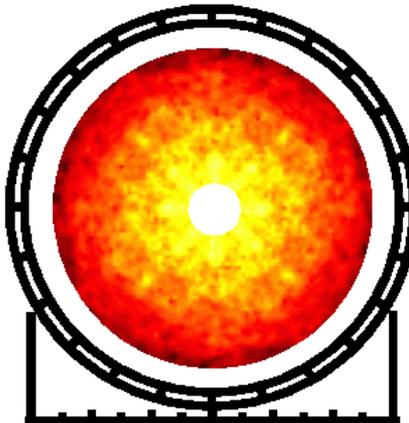
Bulk sensitive HXPD of Si 1s

Native SiO₂
/Si(001)

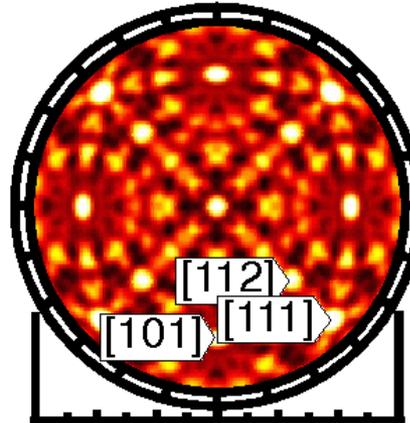
$h\nu = 5417 \text{ eV}$



(c) 4.1 nm SiO₂/Si(001)



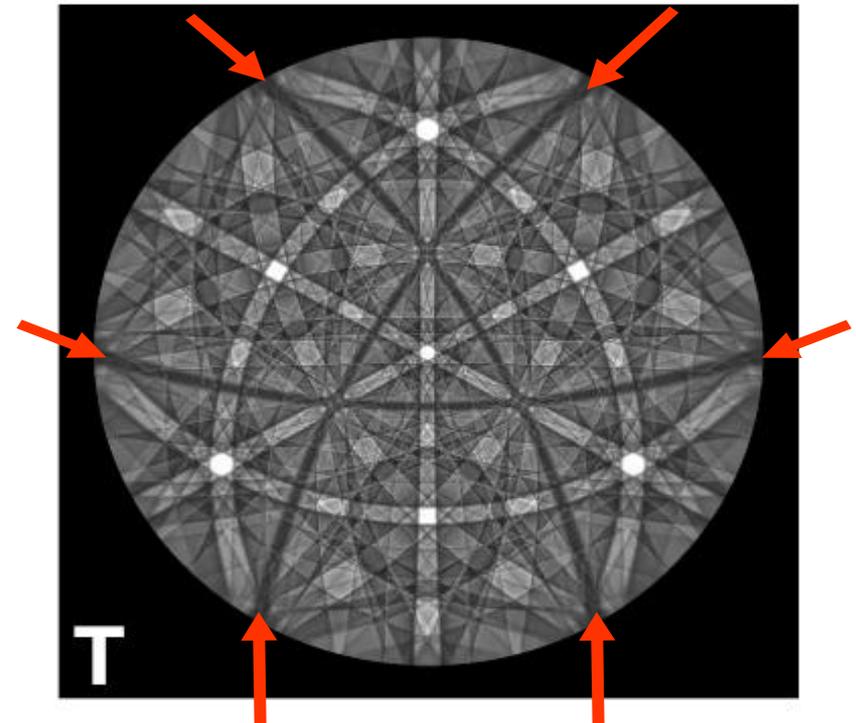
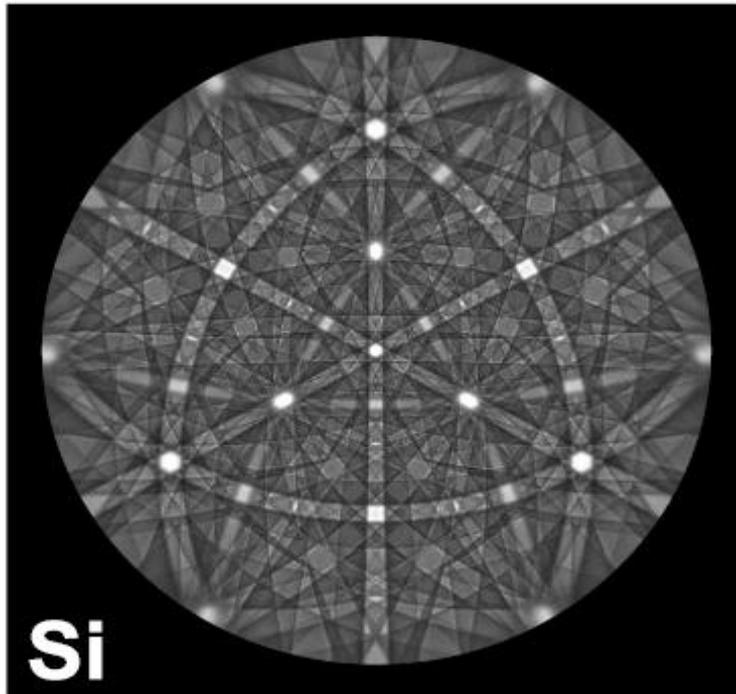
(d) 7.0 nm SiO₂/Si(001)



(e) Multiple scattering cluster simulation

Hard x-ray photoelectron diffraction--Theory: Sensitivity to lattice distortions and atomic site type?

Si(111)-6 keV: Impurity atom on lattice site (Si) vs. tetrahedral interstitial (T)



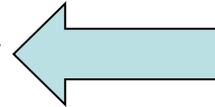
A. Winkelman, J. Garcia de Abajo,
MPI Halle, CF, New Journal of
Physics 10 (2008) 113002

Missing Kikuchi bands-->"forbidden reflections"

Basic Concepts and Experiments

Core-Level Photoemission

Intensities and Quantitative Analysis, the 3-Step Model
Varying Surface and Bulk Sensitivity
Chemical shifts
Multiplet Splittings
Electron Screening and Satellite Structure
Magnetic and Non-Magnetic Dichroism
Resonant Photoemission
Photoelectron Diffraction and Holography



Valence-Level Photoemission

Band-Mapping in the Ultraviolet Photoemission Limit
Densities of States in the X-Ray Photoemission Limit

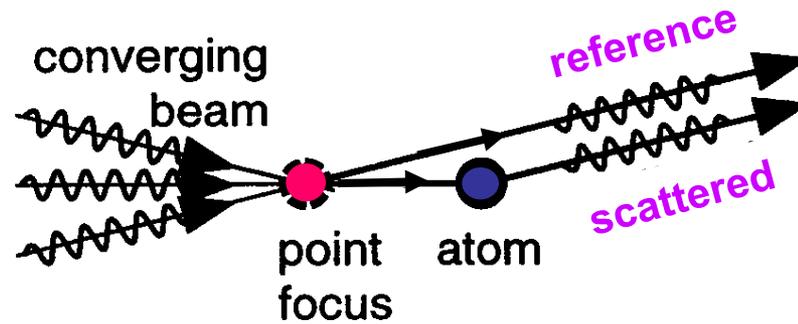
Some New Directions

Photoemission with Hard X-Rays (throughout lectures)
Photoemission with Standing Wave Excitation
Photoemission with Spatial Resolution/Photoelectron Microscopy
Temporal Resolution
@ Higher Pressures

Diffraction and Holography: What's the Difference?

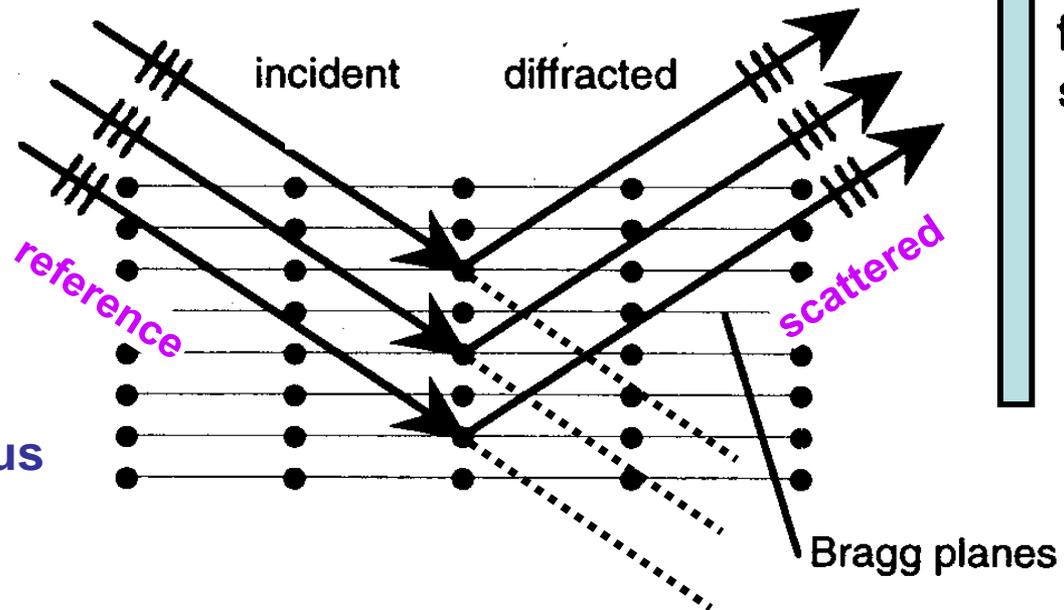
Holography:

Reference and scattered interfere → phase relation preserved



Diffraction:

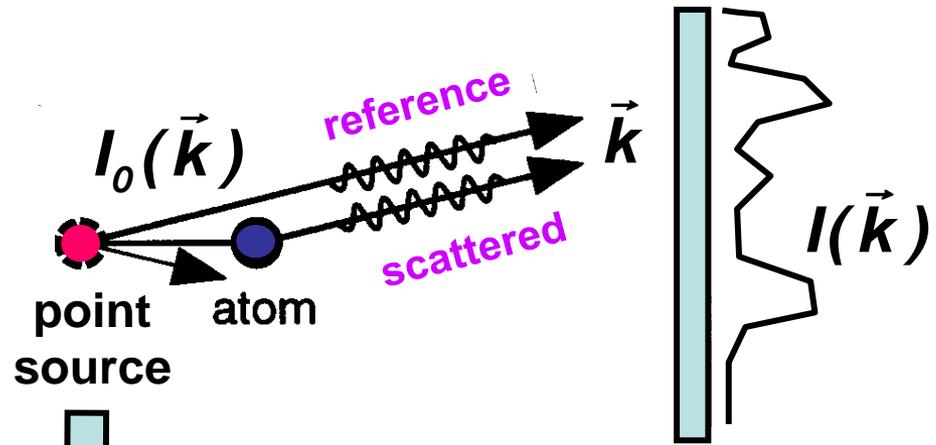
Reference lost → "The phase problem": solved for x-rays via "direct methods", multiple anomalous dispersion, isomorphous replacement



Photoelectron Holography: An Additional Trick

Holography:

Reference and scattered interfere → phase relation preserved



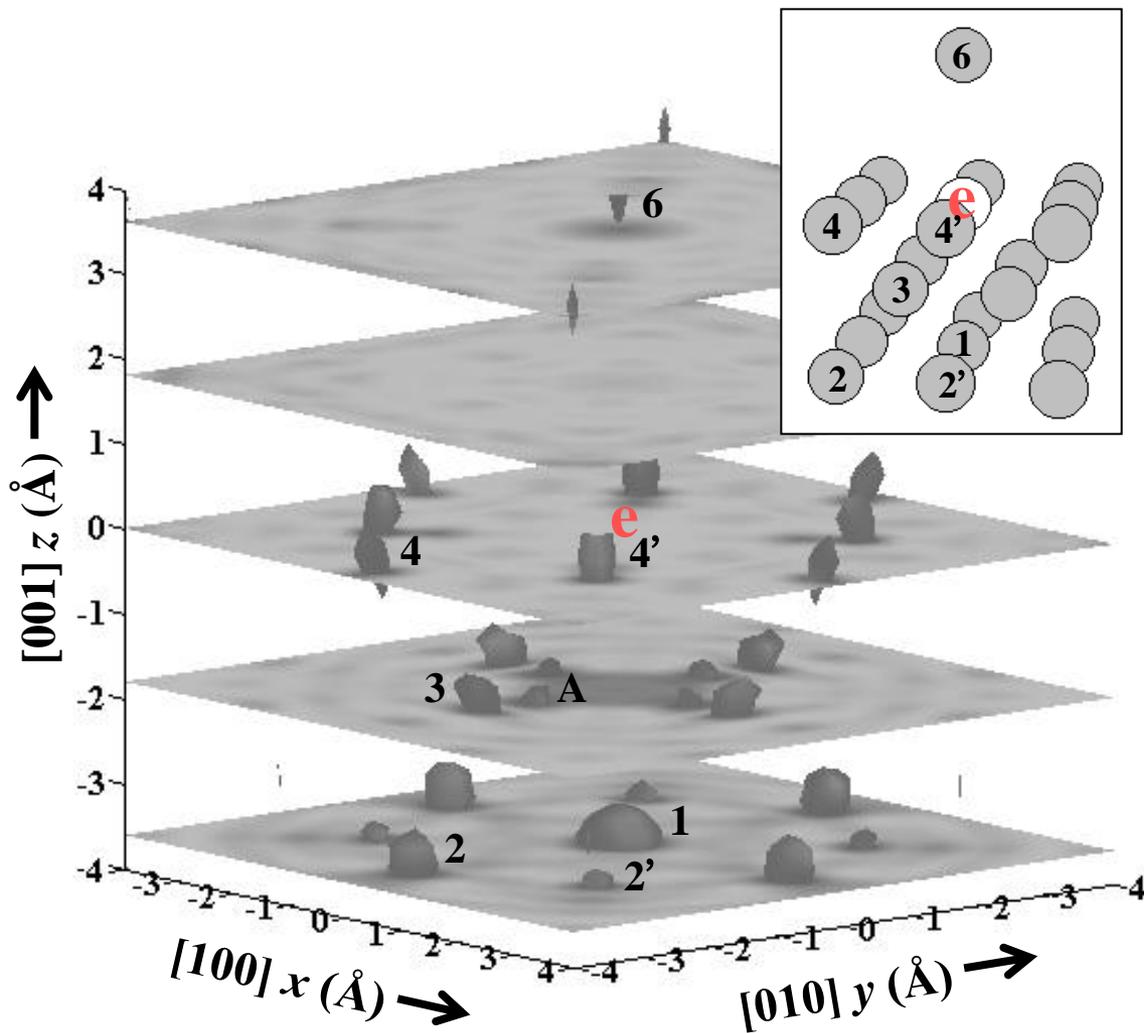
Use an “inside-source”:
atomic or nuclear emitter
(Szöke, in “Short Wavelength
Radiation: Generation and
Applications”, D. Attwood
and J. Bokor, Eds.,
AIP Conference Proceedings
No. 147 (1986))

$$\chi(\vec{k}) = [I(\vec{k}) - I_0(\vec{k})] / I_0(\vec{k})$$

Holographic image of scatterers =

$$U(\vec{r}) = \left| \iiint \chi(\vec{k}) \exp[i\vec{k} \cdot \vec{r} - ikr] d^3 k \right|$$

Differential photoelectron holography in Cu(001) —Cu 3p emission



Derivative photoelectron holography: As and Si emission from As/Si(111):

$$U(\vec{r}) = \left| \iiint \chi(\vec{k}) \exp[i\vec{k} \cdot \vec{r} - ikr] d^3k \right|$$

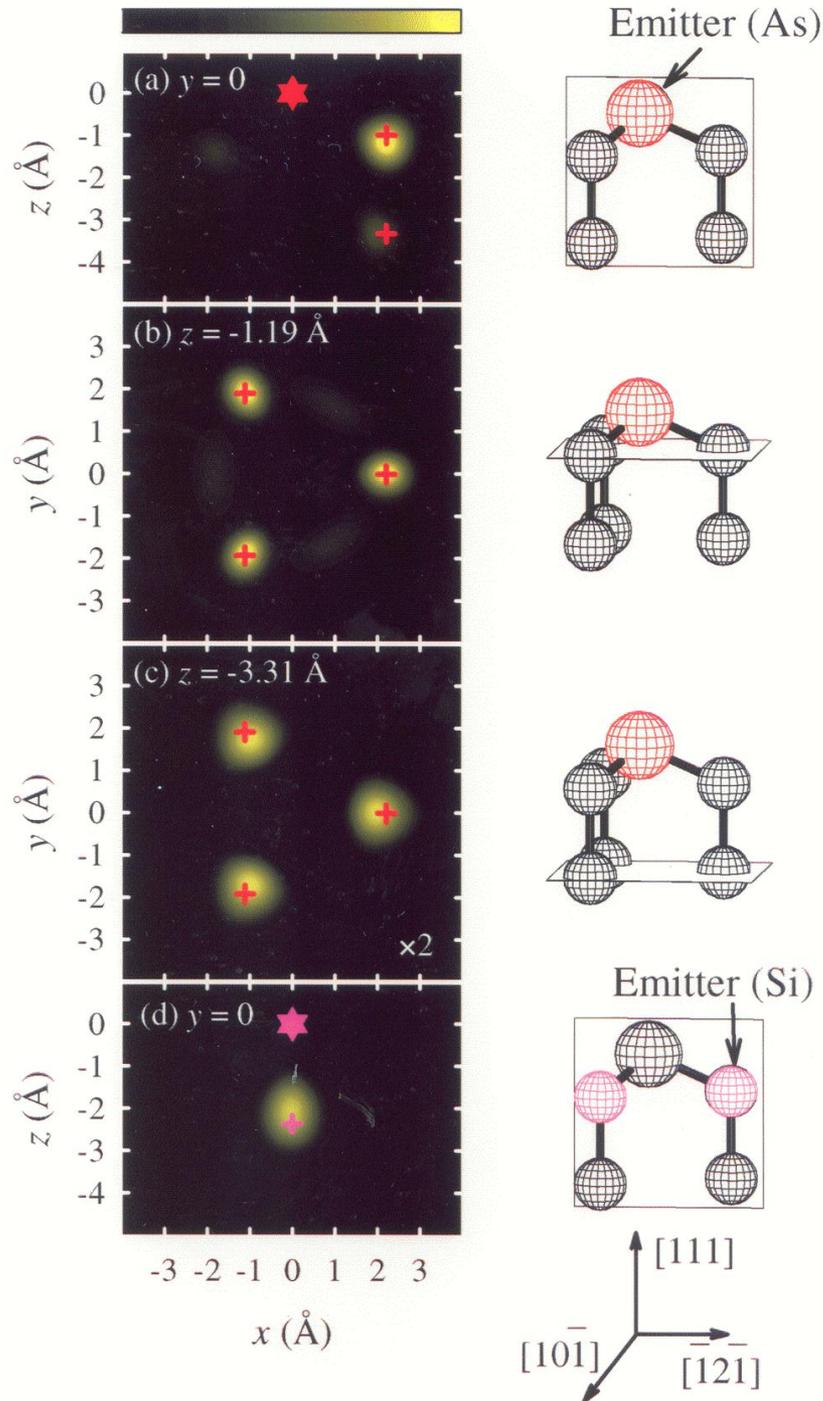
$$\text{with } \chi = \frac{I(\vec{k}) - I_0}{I_0}$$

and $I(\vec{k})$ from integration of log arithmetic derivative

$$L(h\nu, \hat{k}) = \frac{I(h\nu + \delta, \hat{k}) - I(h\nu - \delta, \hat{k})}{[I(h\nu + \delta, \hat{k}) + I(h\nu - \delta, \hat{k})] \delta}$$

$$I(\vec{k}) \equiv I(k, \hat{k}) = A \int L(h\nu, \hat{k}) d^3k$$

Luh et al. Phys. Rev. Lett.,
81, 4160 (1998)



Basic Concepts and Experiments

Core-Level Photoemission

Intensities and Quantitative Analysis, the 3-Step Model
Varying Surface and Bulk Sensitivity
Chemical shifts
Multiplet Splittings
Electron Screening and Satellite Structure
Magnetic and Non-Magnetic Dichroism
Resonant Photoemission
Photoelectron Diffraction and Holography

Valence-Level Photoemission

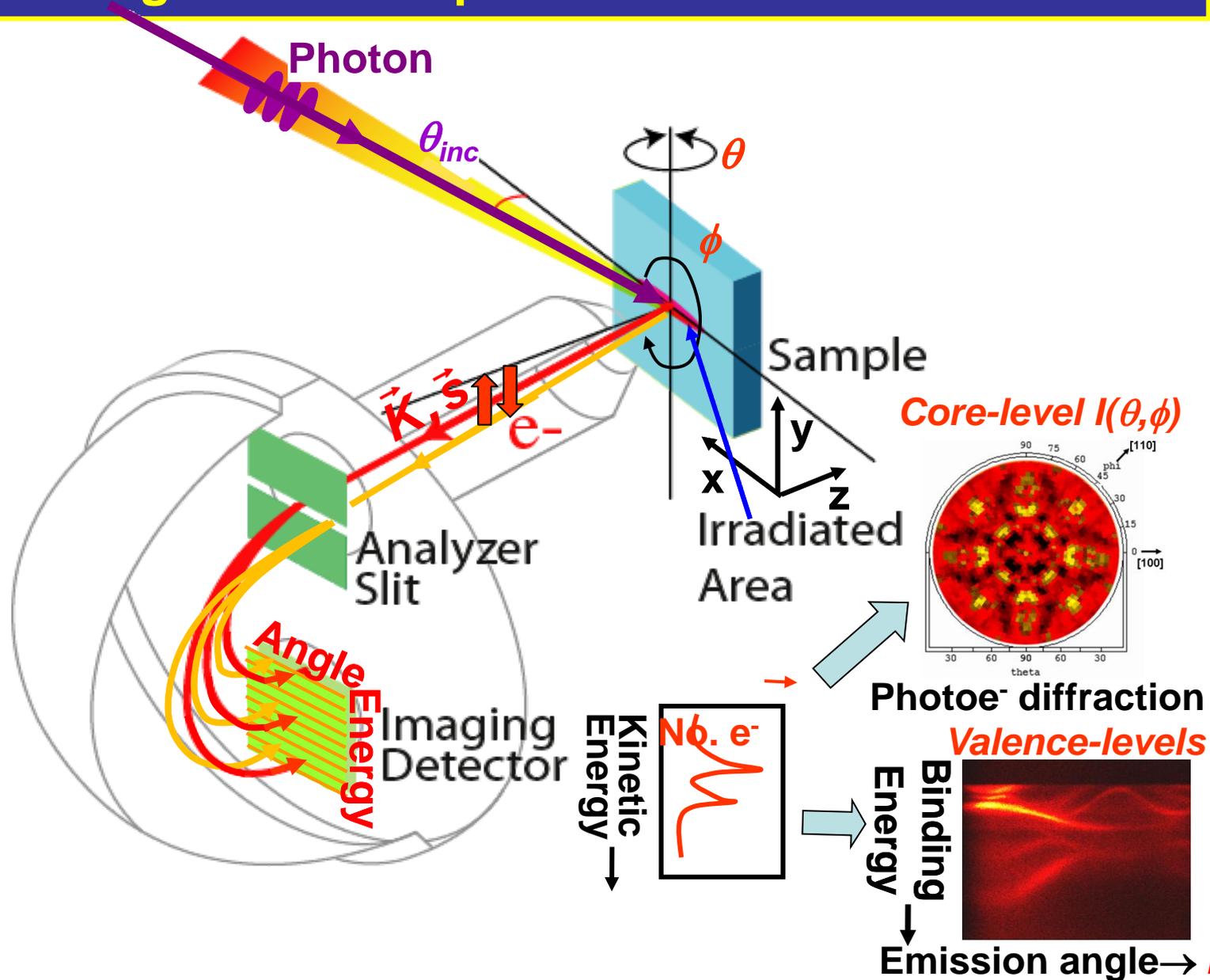


Band-Mapping in the Ultraviolet Photoemission Limit
Densities of States in the X-Ray Photoemission Limit

Some New Directions

Photoemission with Hard X-Rays (throughout lectures)
Photoemission with Standing Wave Excitation
Photoemission with Spatial Resolution/Photoelectron Microscopy
Temporal Resolution
@ Higher Pressures

Typical experimental geometry for energy- and angle-resolved photoemission measurements

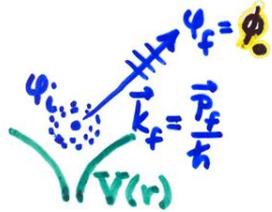


PHOTOELECTRON EMISSION -

BASIC MATRIX ELEMENTS + SELECTION RULES:

- ATOMIC-LIKE (LOCALIZED) STATES \Rightarrow CORE:

$$\psi_i(\vec{r}) = \psi_{n_i, l_i, m_i}(r, \theta, \phi) = R_{n_i, l_i}(r) Y_{l_i, m_i}(\theta, \phi)$$



$$\psi_f(\vec{r}, \vec{k}_f) = \psi_{E_f}(\vec{r}, \vec{k}_f)$$

$$= 4\pi \sum_{l_f, m_f} i^{l_f} e^{-i\delta_{l_f}} Y_{l_f, m_f}^*(\theta, \phi) Y_{l_f, m_f}(\theta, \phi) R_{E_f, l_f}(r)$$

PHASE SHIFT OF l_f WAVE IN $V(r)$

DIPOLE APPROX.: INT. $\propto |\langle \psi_f | \hat{E} \cdot \vec{r} | \psi_i \rangle|^2 = |\hat{E} \cdot \langle \psi_f | \vec{r} | \psi_i \rangle|^2$

EQUIVALENT WITHIN CONSTANT FACTOR



- $\langle \Delta l = l_f - l_i = \pm 1$
TWO CHANNELS
- $\langle \Delta m = m_f - m_i = 0, \pm 1$
LINEAR POLARIZ.
- $\langle \Delta m = \pm 1$, CIRCULAR POLARIZATION

- BLOCH-FUNCTION (DELOCALIZED) STATES \Rightarrow VALENCE:

$$\psi_i(\vec{r}) = u_{\vec{k}_i}(\vec{r}) e^{i\vec{k}_i \cdot \vec{r}}$$

$$\psi_f(\vec{r}) = u_{\vec{k}_f}(\vec{r}) e^{i\vec{k}_f \cdot \vec{r}}; E_f = \frac{p_f^2}{2m} = \frac{\hbar^2 k_f^2}{2m}$$

USUALLY NEGLIG.



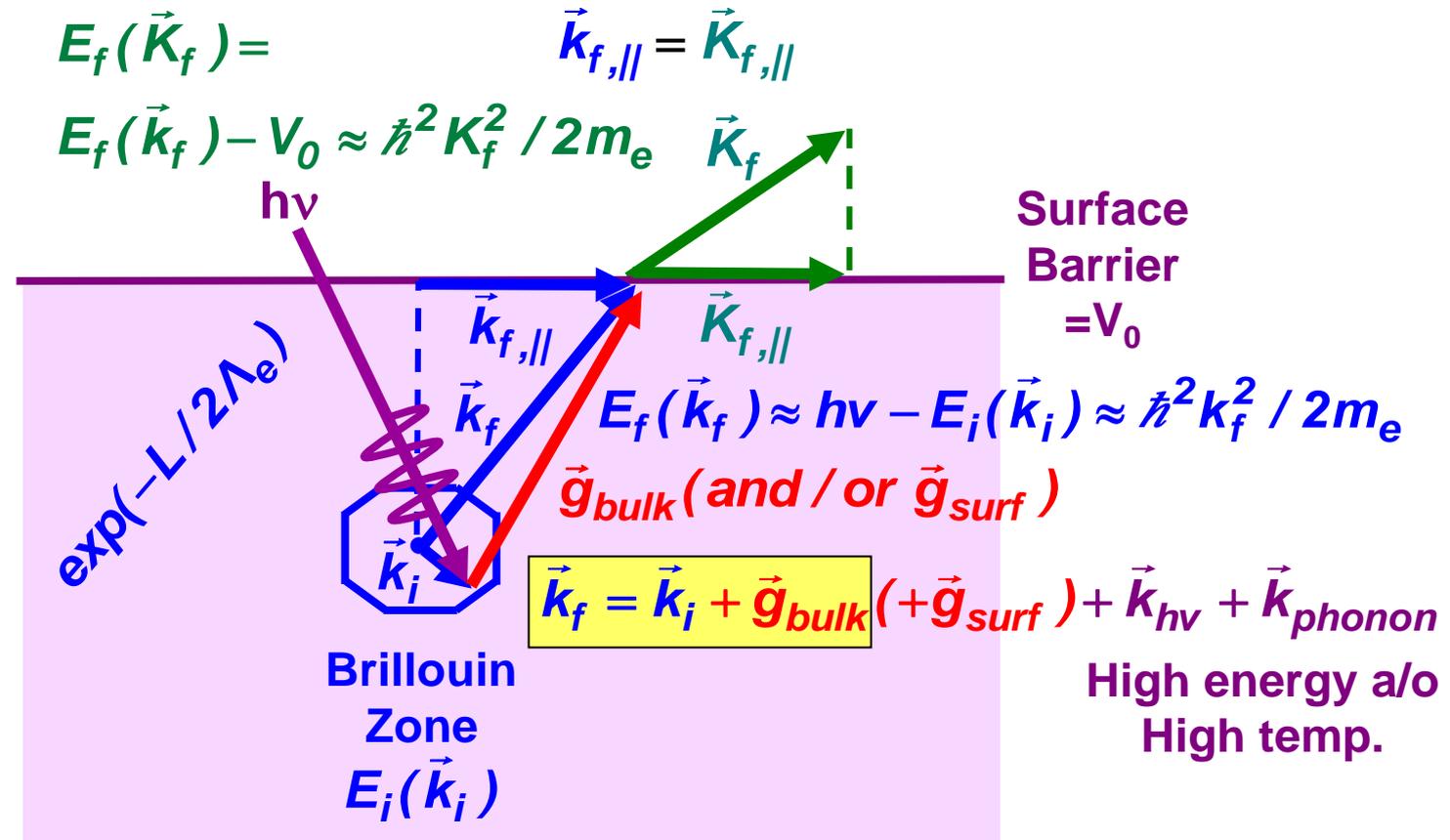
$$|\langle \psi_f | \hat{E} \cdot \vec{p} | \psi_i \rangle|^2 = |\hat{E} \cdot \langle \psi_f | \vec{p} | \psi_i \rangle|^2 \Rightarrow \Delta \vec{k} = \vec{k}_f - \vec{k}_i - \vec{k}_{\text{PHONON}}$$

$$= \vec{g}_{\text{BULK}} \text{ (or } \vec{g}_{\text{SURF}})$$

"DIRECT" TRANSITIONS

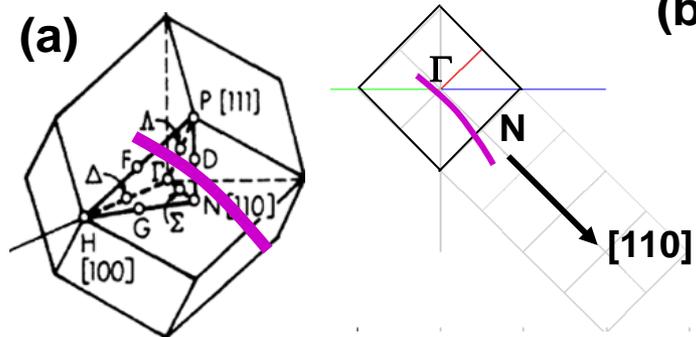
BUT LATTICE VIBRATIONS \Rightarrow SUM OVER \vec{k}_{PHONON}
 \Rightarrow FRACTION DIRECT = DEBYE-WALLER FACTOR
 $= \exp[-g^2 \bar{u}^2]$

Valence-band photoemission: Angle-Resolved Photoemission (ARPES)

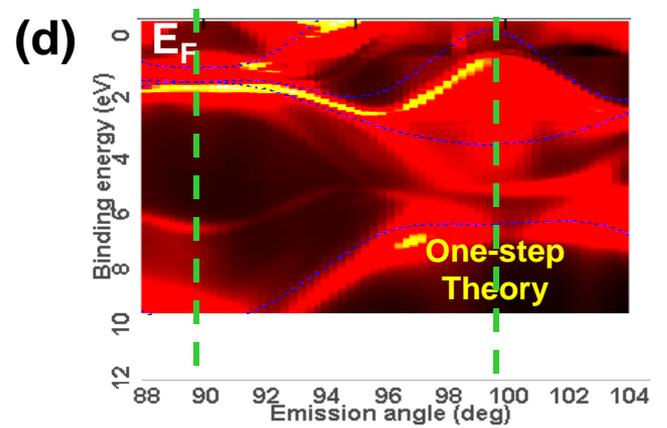
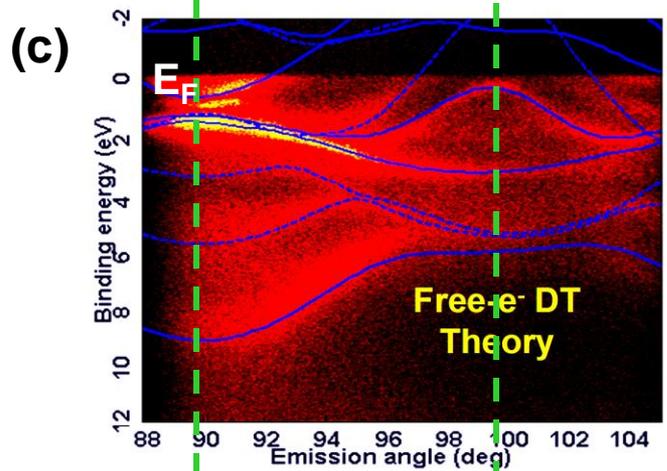
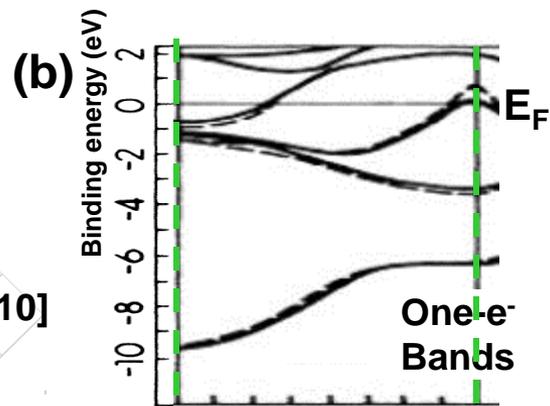


$$I(E_f, \vec{k}_f) \propto \left| \hat{\epsilon} \cdot \left\langle \varphi_{photoe}(E_f = h\nu + E_i, \vec{k}_f = \vec{k}_i + \vec{g}) \middle| \vec{r} \middle| \varphi(E_i, \vec{k}_i) \right\rangle \right|^2$$

“Direct” or k-conserving transitions



**Angle-Resolved
Photoemission
from W(110)**
 $h\nu = 260 \text{ eV}$
 $90^\circ = \text{normal}$
 $\Theta_{\text{Debye}} = 400\text{K}$

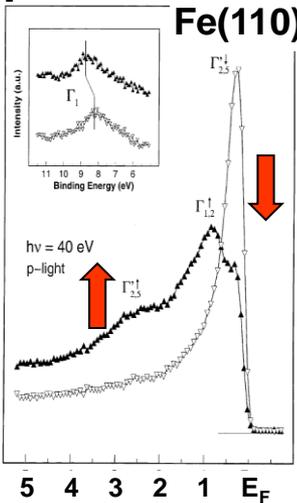


Plucinski, Minar, et al.
 PRB 78, 035108 (2008)

Typical experimental geometry for energy- and angle-resolved photoemission measurements

Spin-resolution

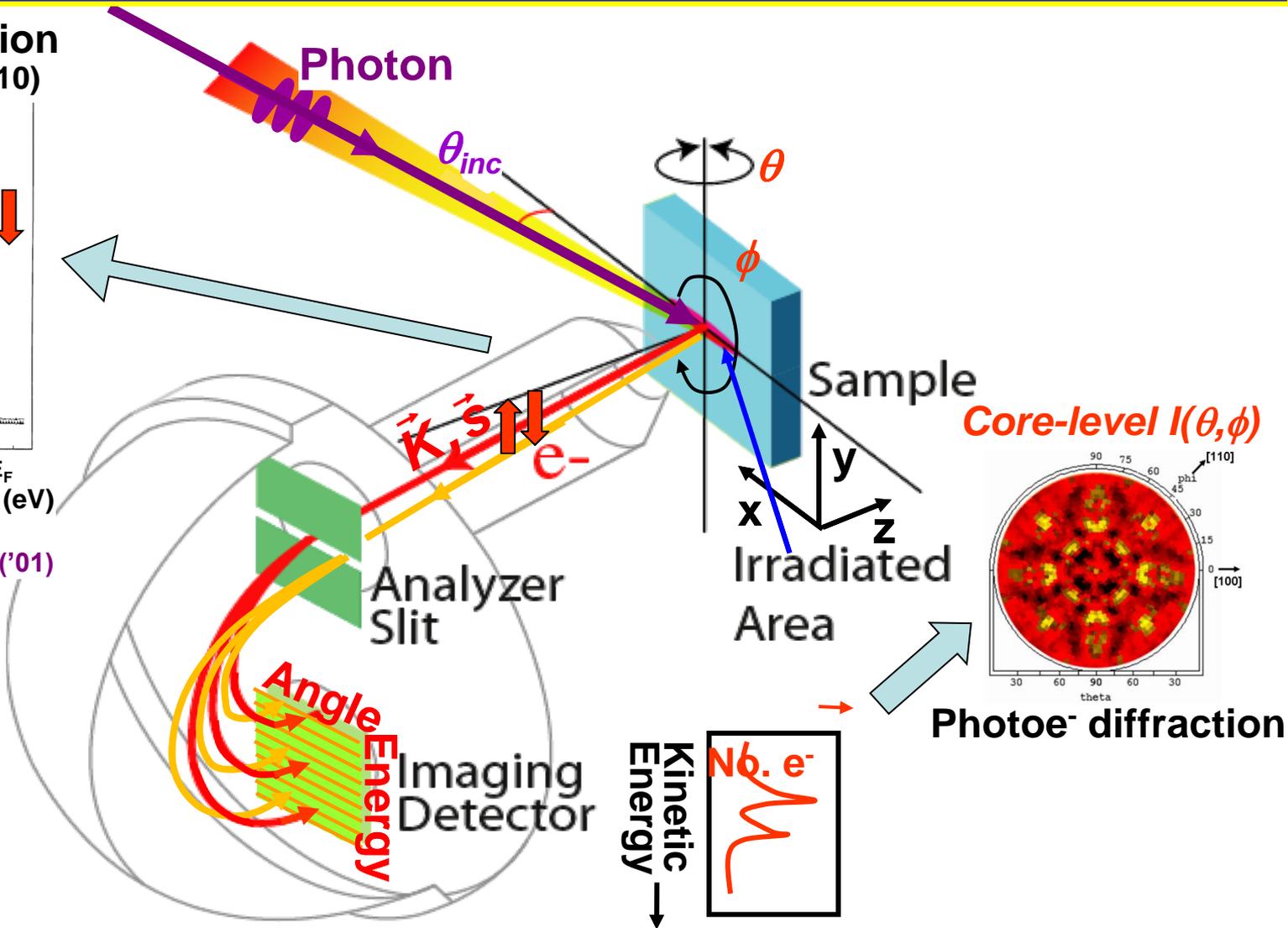
Spin-resolved intensity (a.u.)



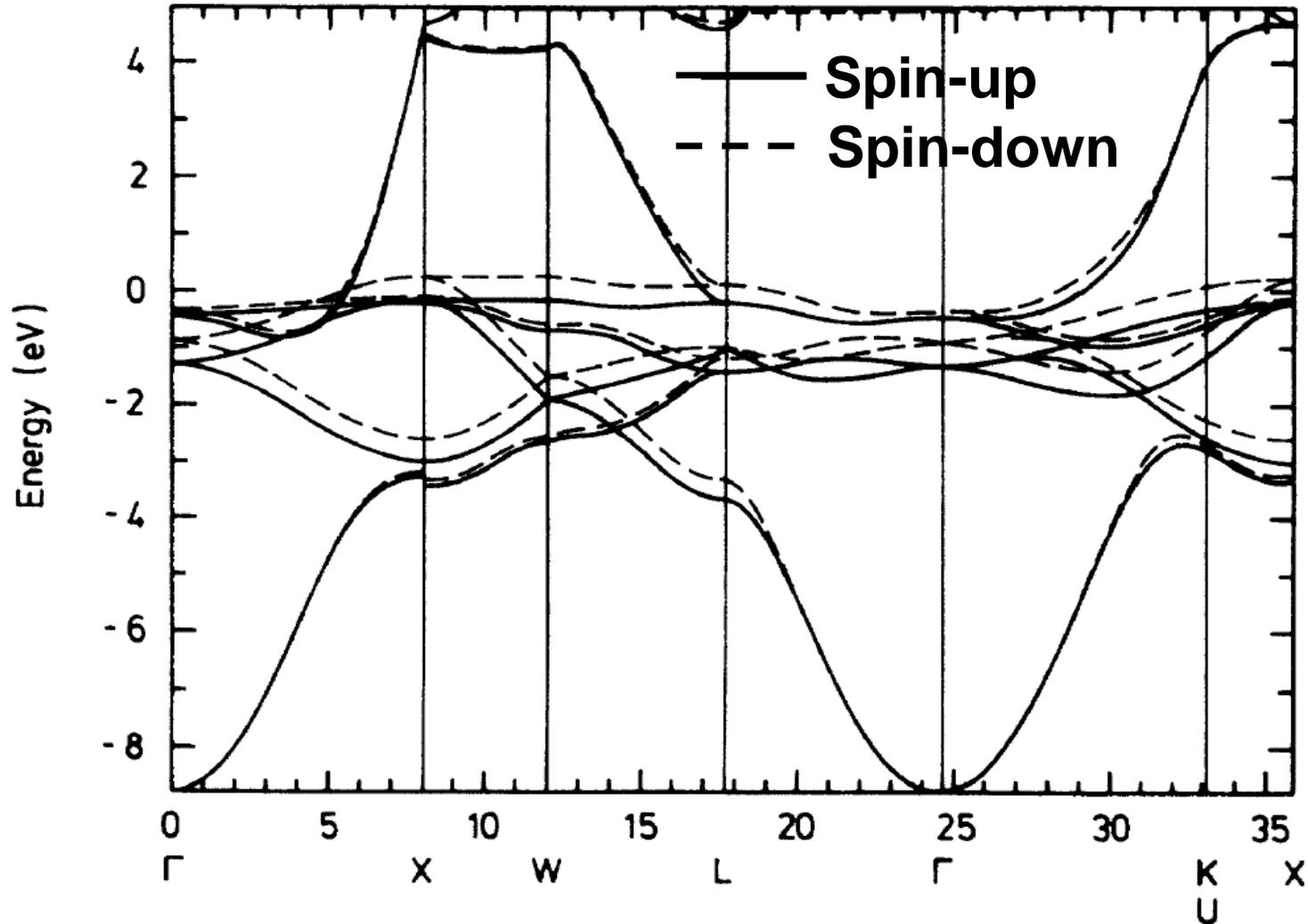
← Binding Energy (eV)

Kim et al.

Surf. Sci. **478**, 193 ('01)

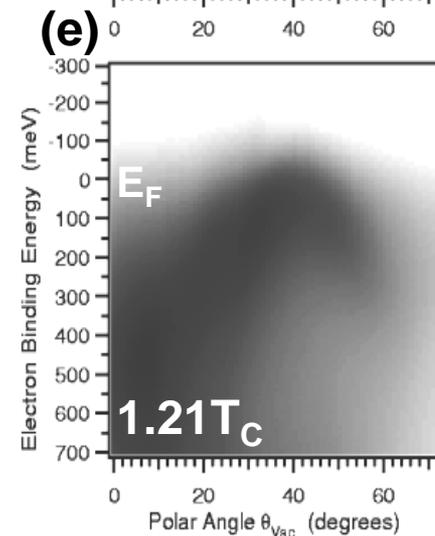
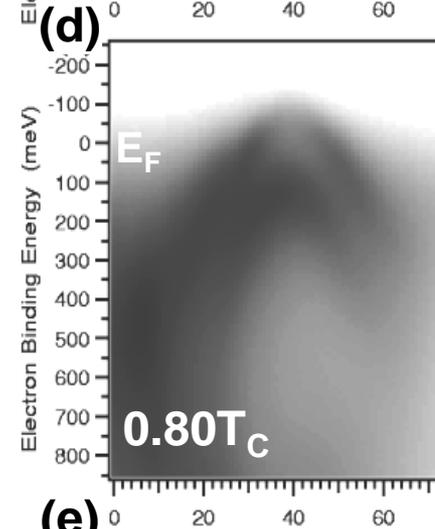
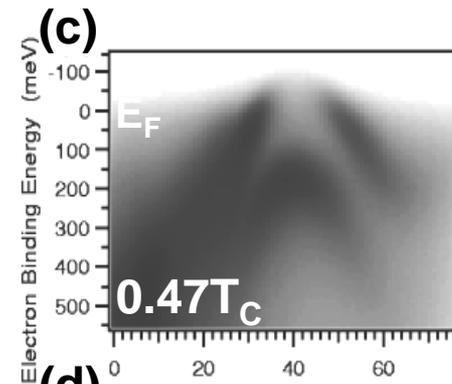
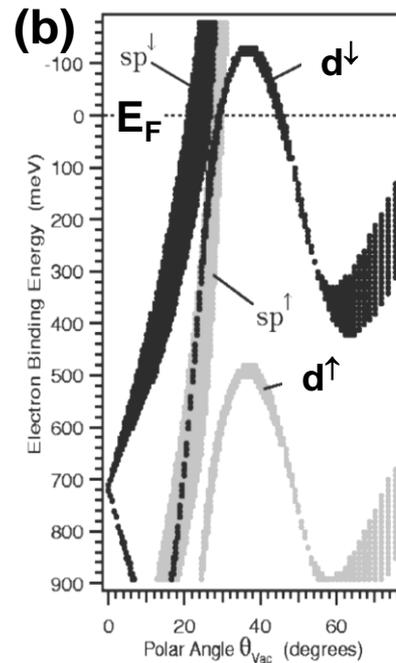
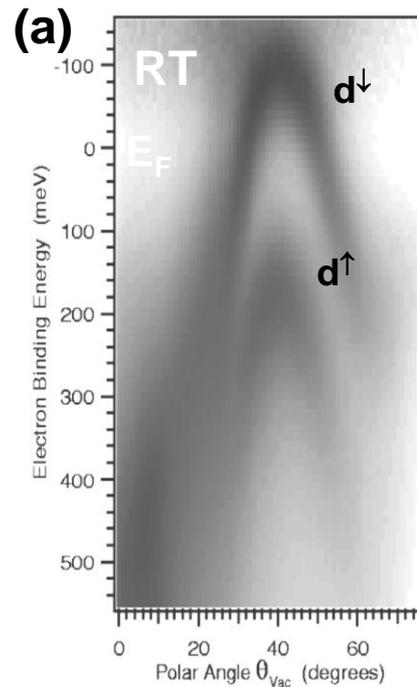


Band structure of Nickel



Huebner, Phys. Rev. B 42, 11553 (1990)

**Angle-Resolved
Photoemission
from ferromagnetic
Ni(111)
 $h\nu = 21.2$ eV
Spin-split
bands**



Basic Concepts and Experiments

Core-Level Photoemission

Intensities and Quantitative Analysis, the 3-Step Model
Varying Surface and Bulk Sensitivity
Chemical shifts
Multiplet Splittings
Electron Screening and Satellite Structure
Magnetic and Non-Magnetic Dichroism
Resonant Photoemission
Photoelectron Diffraction and Holography

Valence-Level Photoemission

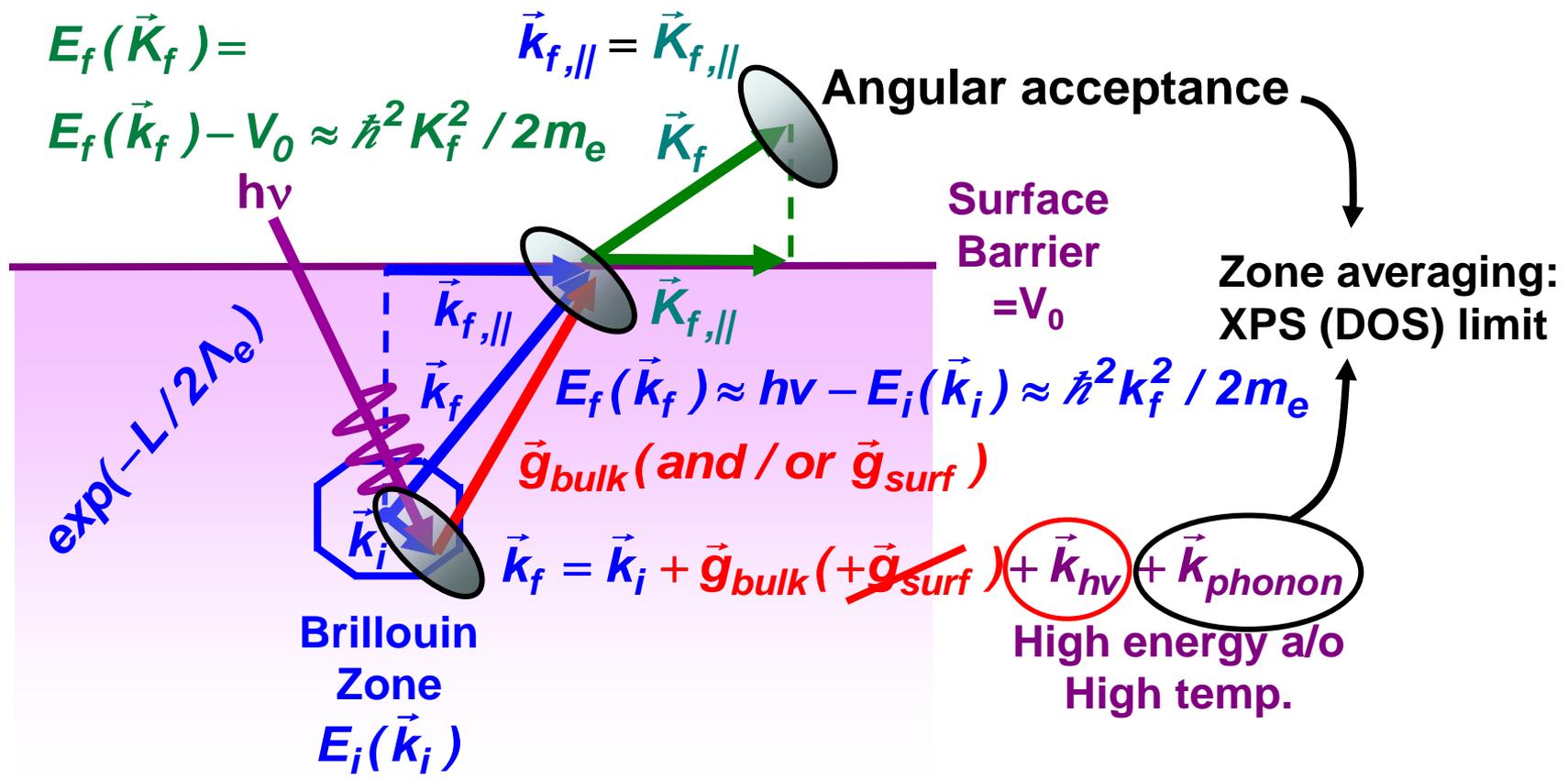


Band-Mapping in the Ultraviolet Photoemission Limit
Densities of States in the X-Ray Photoemission Limit

Some New Directions

Photoemission with Hard X-Rays (throughout lectures)
Photoemission with Standing Wave Excitation
Photoemission with Spatial Resolution/Photoelectron Microscopy
Temporal Resolution
@ Higher Pressures

ARPES—How high can we go in energy and temperature?



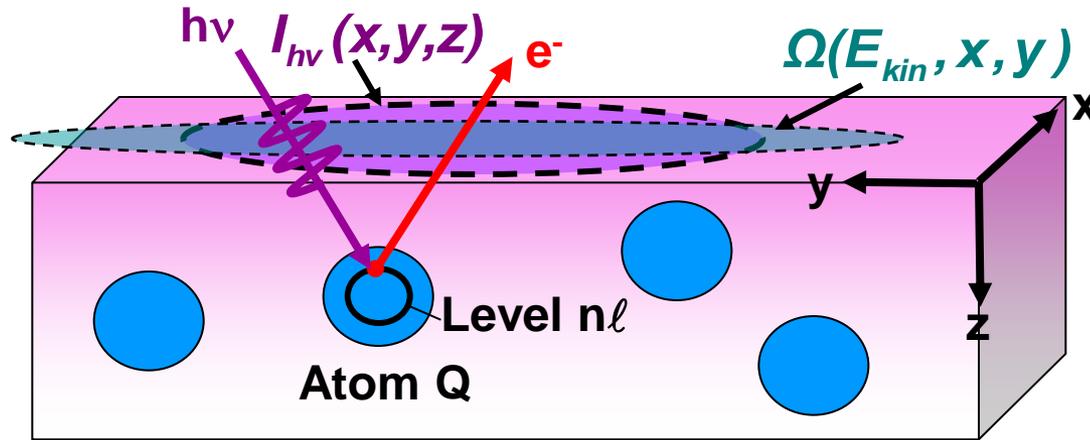
Fraction DTs \approx Debye-Waller factor = $W(T) \approx \exp[-(k^f)^2 \langle u^2(T) \rangle]$
 $\approx \exp[-C_1 (k^f)^2 T / (m \Theta_D^2)] \approx \exp(-C_2 E_{kin} T)$

ARPES \rightarrow bands, quasiparticles
 (Low $h\nu$, Low T , High angul. Res.)

Shevchik, Phys. Rev. B 16, 3428 (1977)
 Hussain....CF, Phys. Rev. B 34 (1986) 5226

XPS \rightarrow DOS
 (High $h\nu$, High T , Low angul. Res.)

VALENCE-BAND PHOTOELECTRON INTENSITIES AND DENSITIES OF STATES



$$I(E_{kin}, Qn\ell) =$$

$$C' \int_0^{\infty} I_{h\nu}(x, y, z) \rho_{Qn\ell}(E_b, x, y, z) \frac{d\sigma_{Qn\ell}(h\nu)}{d\Omega} \exp\left[-\frac{z}{\Lambda_e(E_{kin}) \sin\theta}\right] \Omega(E_{kin}, x, y) dx dy dz$$

$I_{h\nu}(x, y, z)$ = x-ray flux

$\rho_{Qn\ell}(E_b, x, y, z)$ = density of states, projected onto $Qn\ell$ character

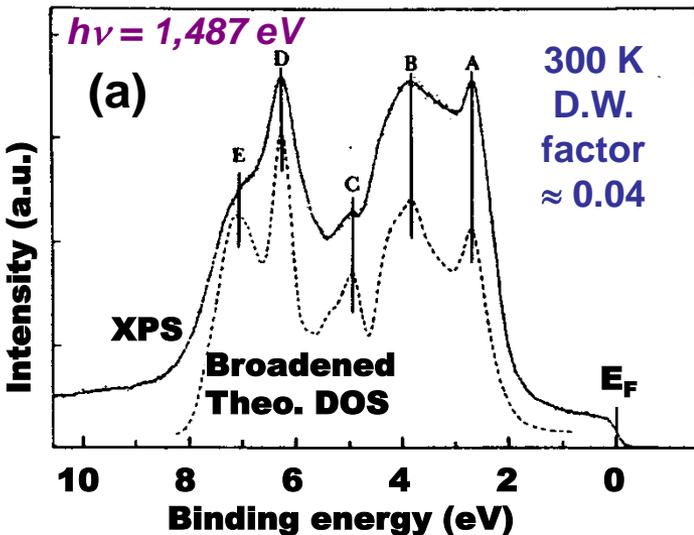
$\frac{d\sigma_{Qn\ell}(h\nu)}{d\Omega}$ = **energy-dependent** differential photoelectric cross section for subshell $Qn\ell$

$\Lambda_e(E_{kin})$ = **energy-dependent** inelastic attenuation length

→ Mean Emission Depth

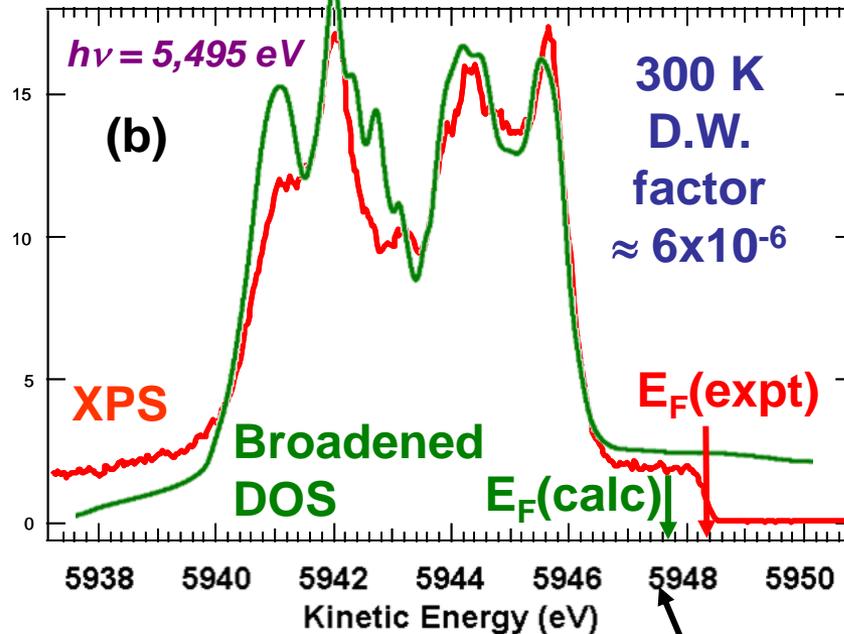
$\Omega(E_{kin}, x, y)$ = **energy-dependent** spectrometer acceptance solid angle

Gold Valence Spectrum

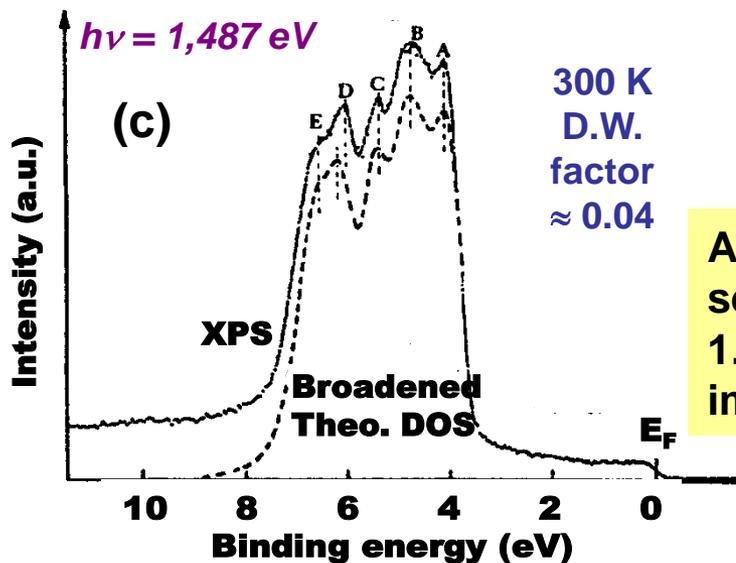


Valence spectra in the XPS Limit

Gold Valence Spectrum



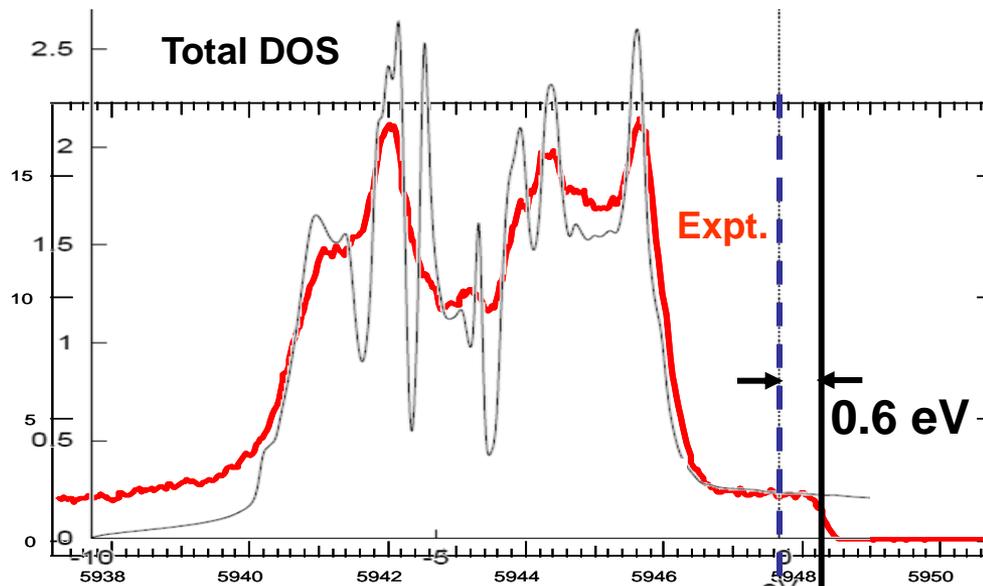
Silver Valence Spectrum



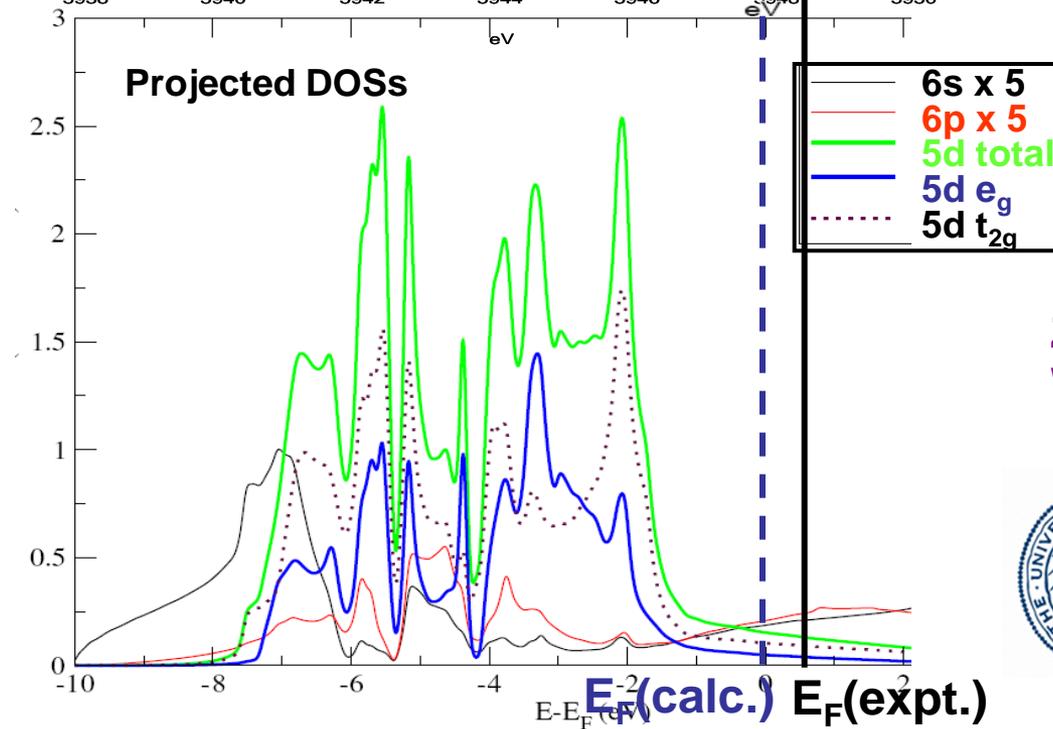
Au subshell photoelectric cross section ratios: $\text{Au}6s/\text{Au}5d = 0.012$ at 1.5 keV and 0.028 at 6 keV \rightarrow 6s more imp. at 6 keV x 2.5x

Screening/
self-energy
correction

**In the XPS
Limit—
Comparison to
theoretical
DOSs**



Takata et al.,
SPRING8 @
5.495 keV

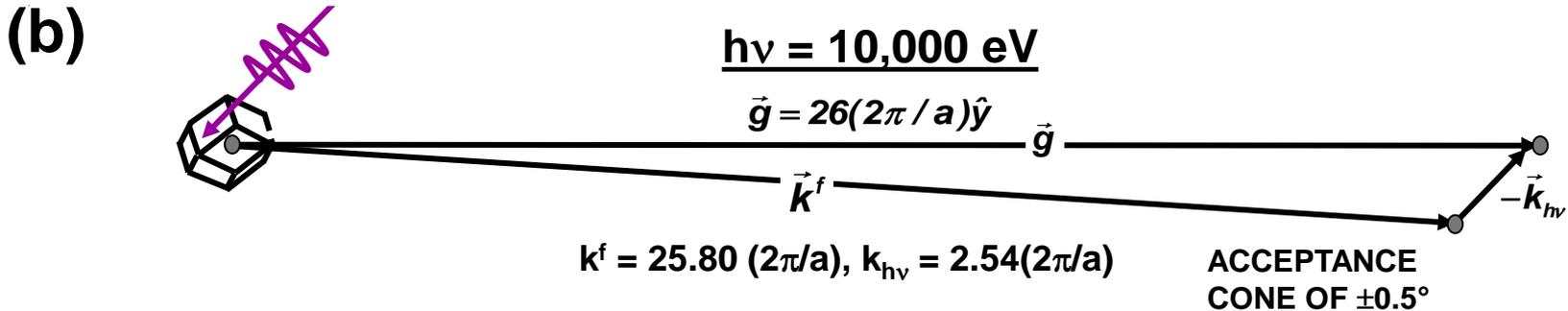
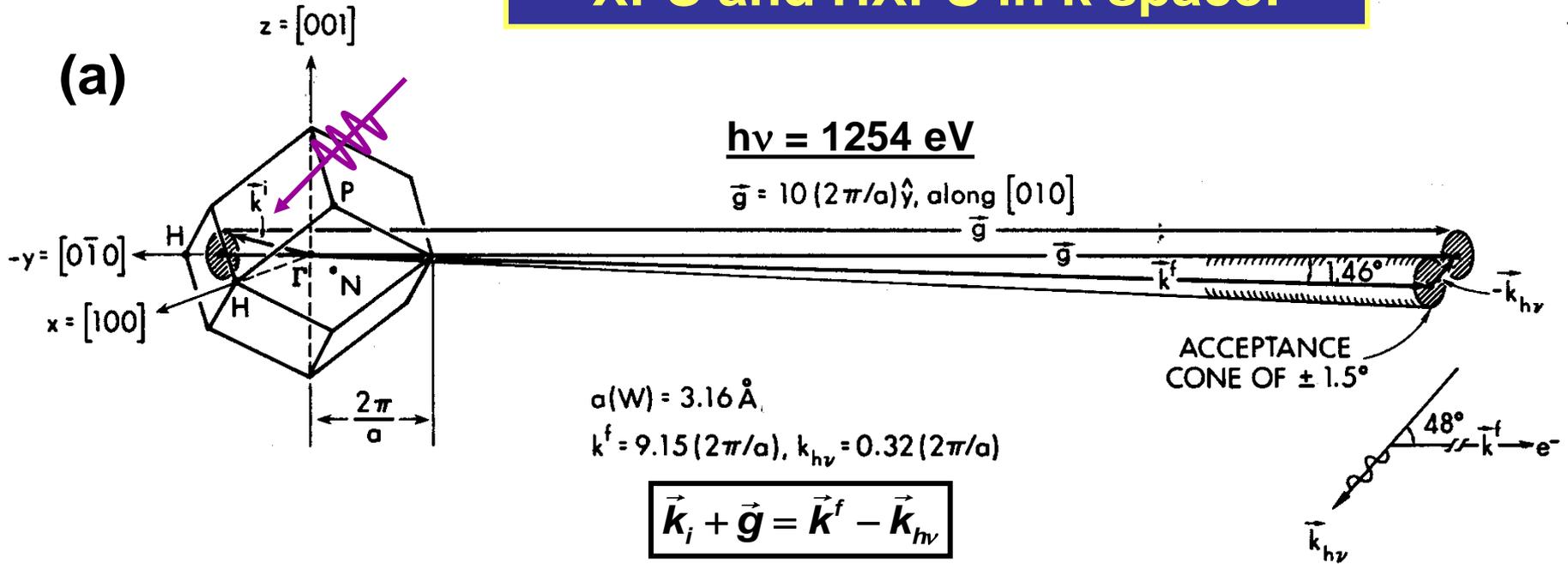


Z. Yin
W.E. Pickett



**Au subshell
photoelectric cross
section ratios:
Au6s/Au5d = 0.012
at 1.5 keV and 0.028
at 6 keV → 6s more
imp. at 6 keV x 2.5x**

XPS and HXPS in k space:



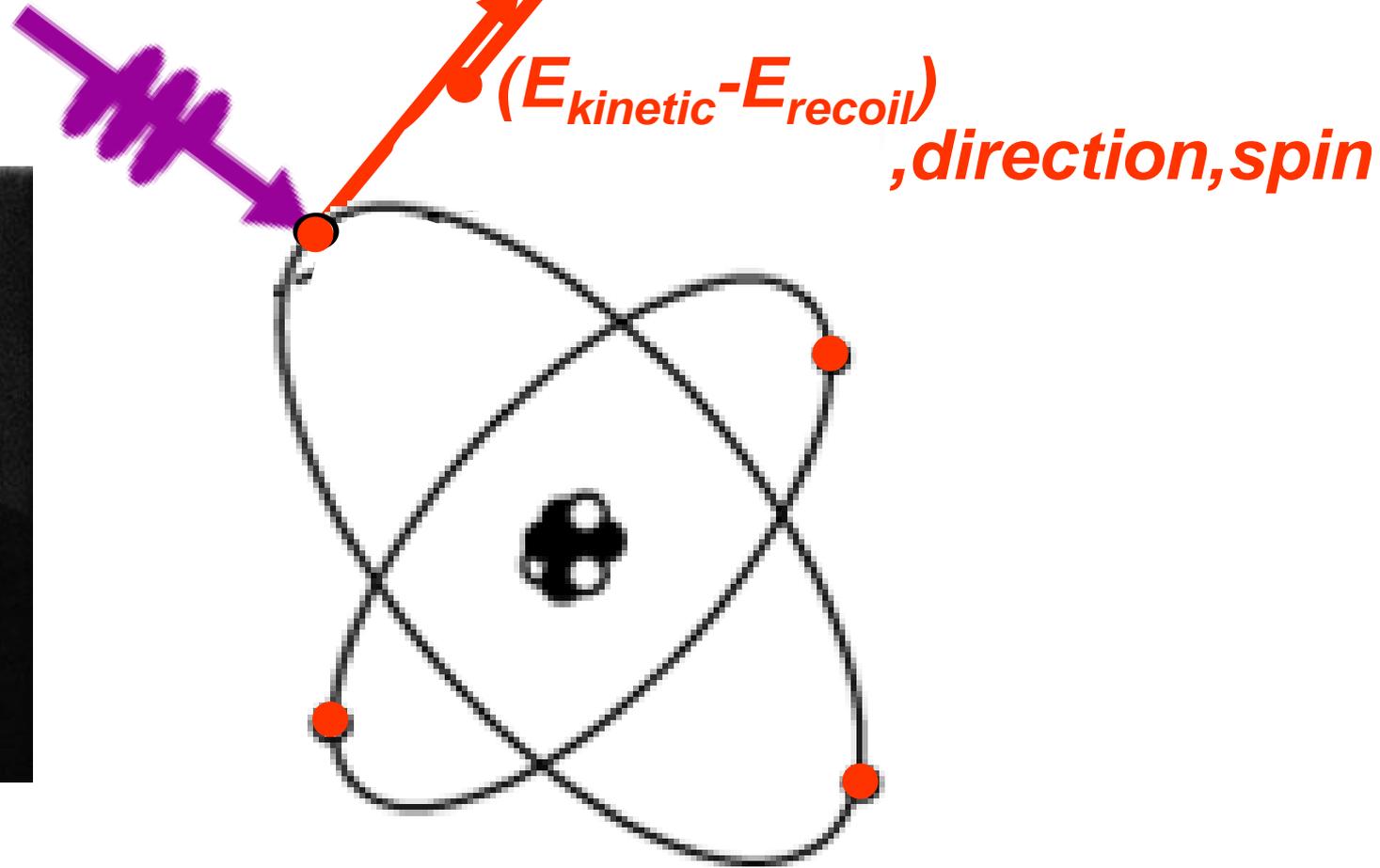
Phonon effects: Approximate fraction of “good” direct transitions

$$\approx \text{Debye-Waller factor} = W(T) \approx \exp[-g^2 \langle u^2(T) \rangle]$$

= Mean-squared
vibrational displacement

The Photoelectric Effect, Einstein, 1905

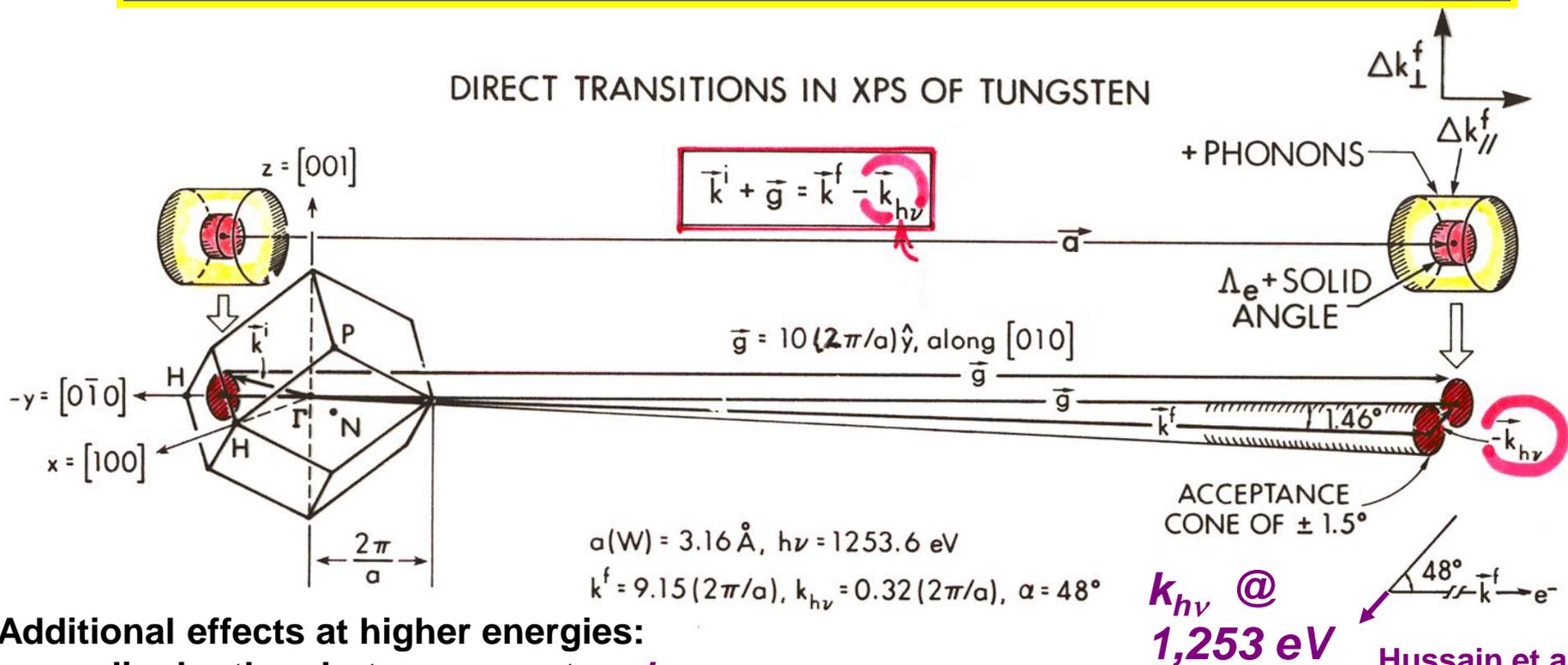
Light can behave like a Particle!



$$h\nu = E_{initial} - E_{final} = E_{binding} + (E_{kinetic} - E_{recoil})$$

Angle-Resolved Photoemission at High Energy-- How high can we go?:

DIRECT TRANSITIONS IN XPS OF TUNGSTEN



$k_{h\nu}$ @
1,253 eV

Hussain et al.,
Phys. Rev. B
22, 3750 (1980)

Takata et al.,
Phys. Rev. B
75, 233404
(2007)

Additional effects at higher energies:

- non-dipole--the photon momentum $k_{h\nu}$
- angular acceptance \rightarrow B.Z. averaging
- lattice recoil, phonon creation \rightarrow more

B.Z. averaging, \sim Debye-Waller factor = $W(T) \approx \exp[-(k^f)^2 \langle u^2(T) \rangle]$
 $\approx \exp[-C_1 (k^f)^2 T / (m \Theta_D^2)] \approx \exp(-C_2 E_{kin} T)$

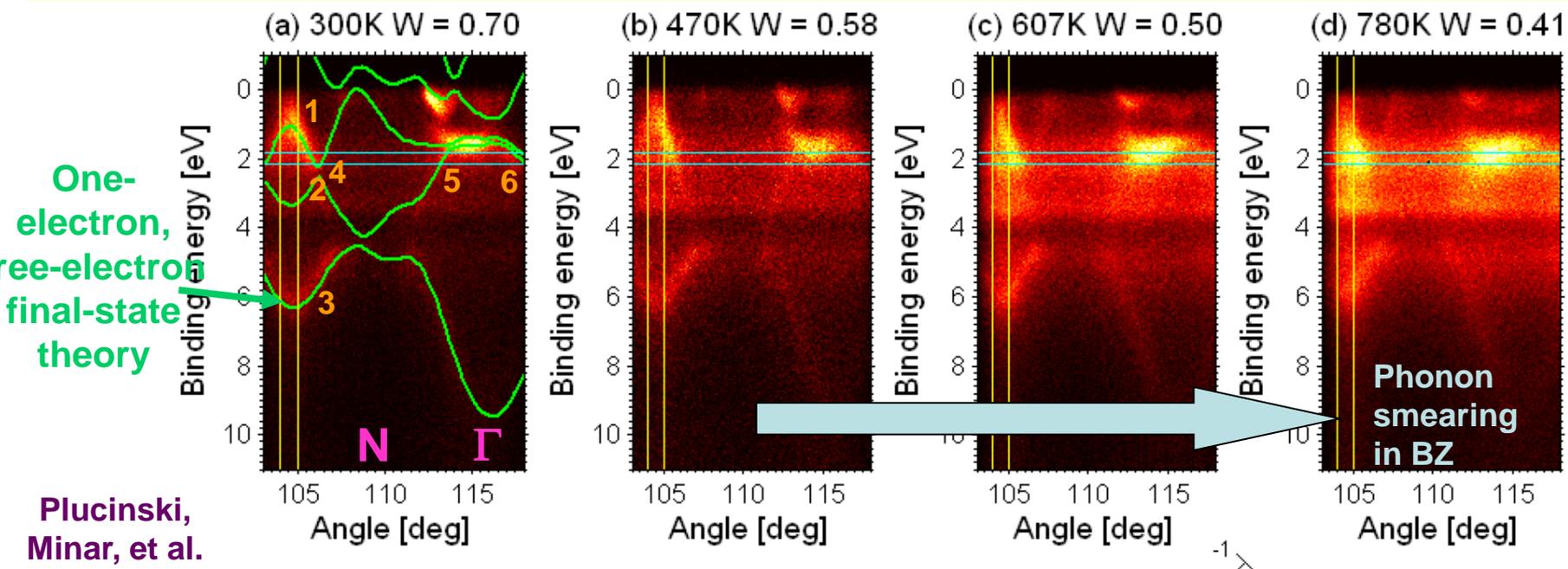
...the "XPS limit" of full B.Z. averaging and

D.O.S. sensitivity \rightarrow core-like photoelectron diffraction

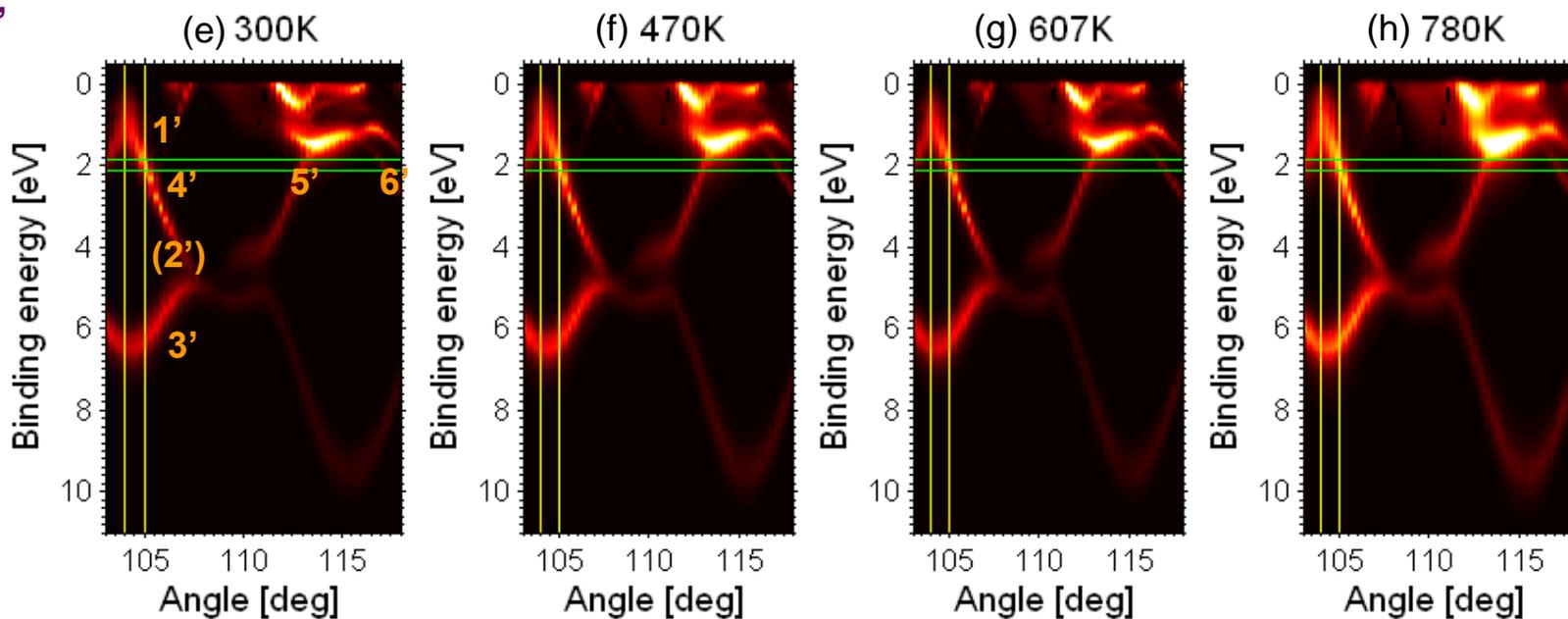
- recoil leads to peak shifts and broadening: $E_{recoil} \text{ (eV)} \approx \left[\frac{m_e}{M} \right] E_{kin} \approx 5.5 \times 10^{-4} \left[\frac{E_{kin} \text{ (eV)}}{M \text{ (amu)}} \right]$

Alvarez et al., PRB 54, 14703 (1996)

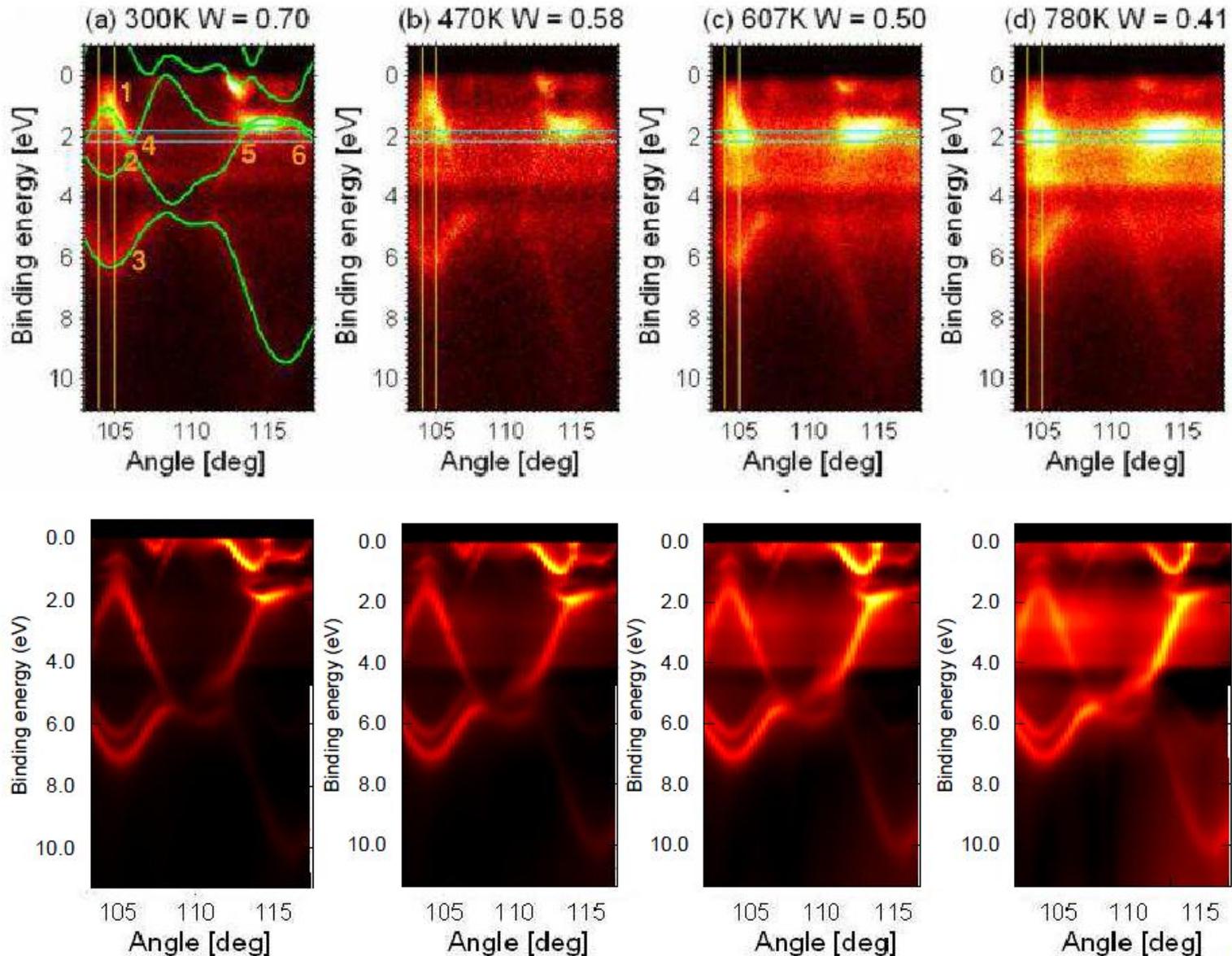
Angle-Resolved Photoemission (ARPES) with soft x-rays: W(110) at 860 eV



Plucinski,
Minar, et al.
PRB 78,
035108
(2008)

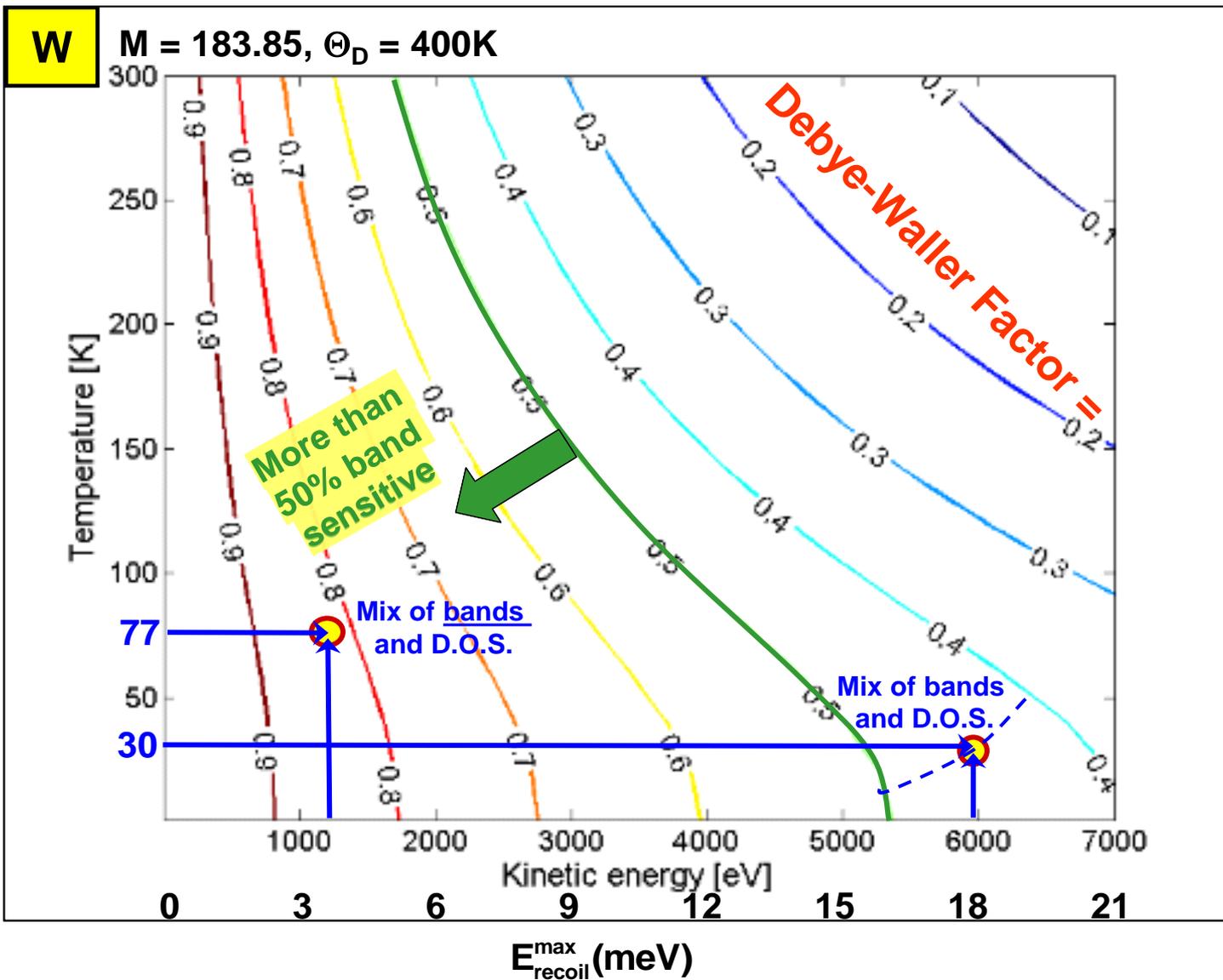


Experiment and theory: W(110)--860 eV

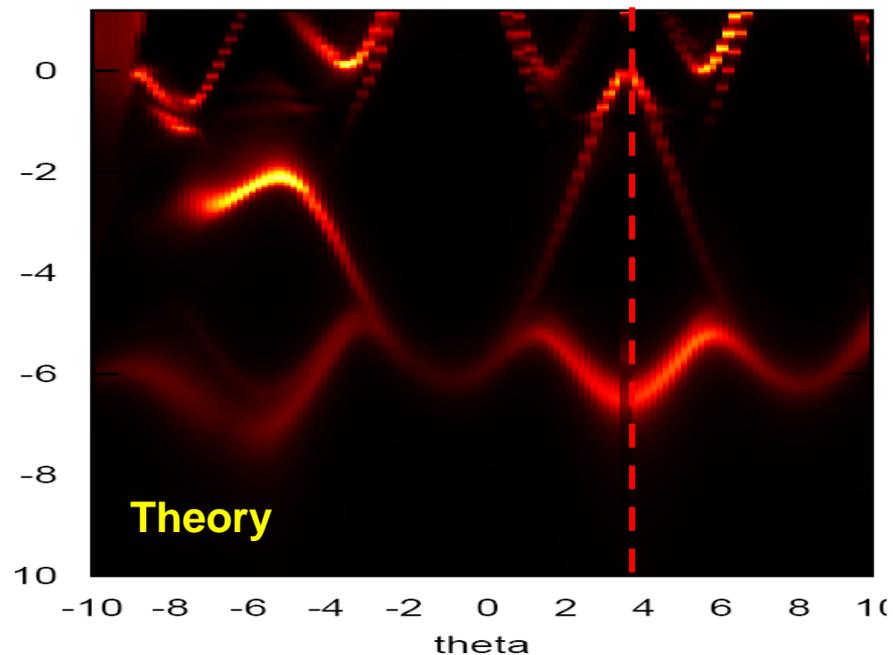
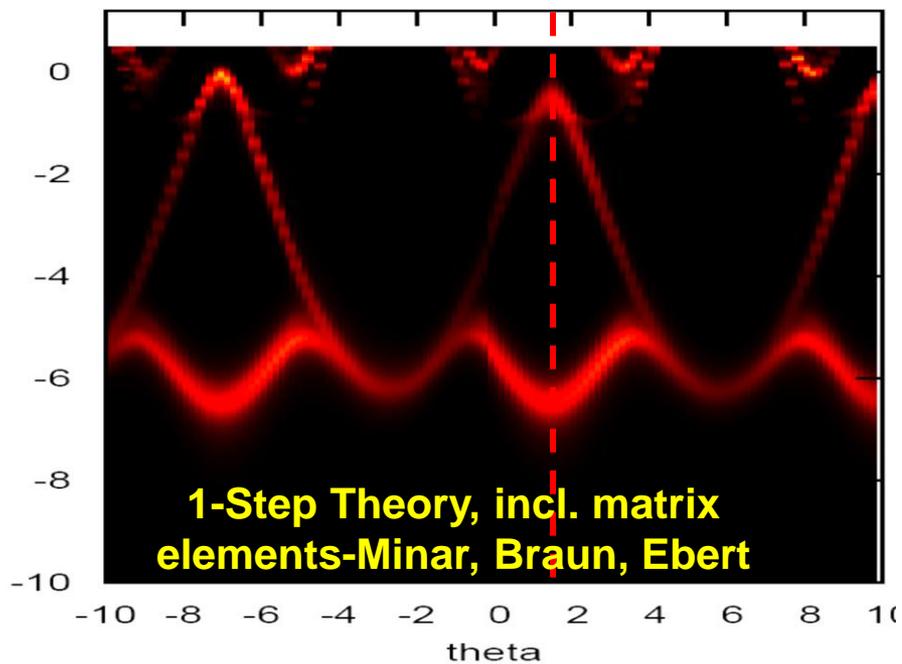
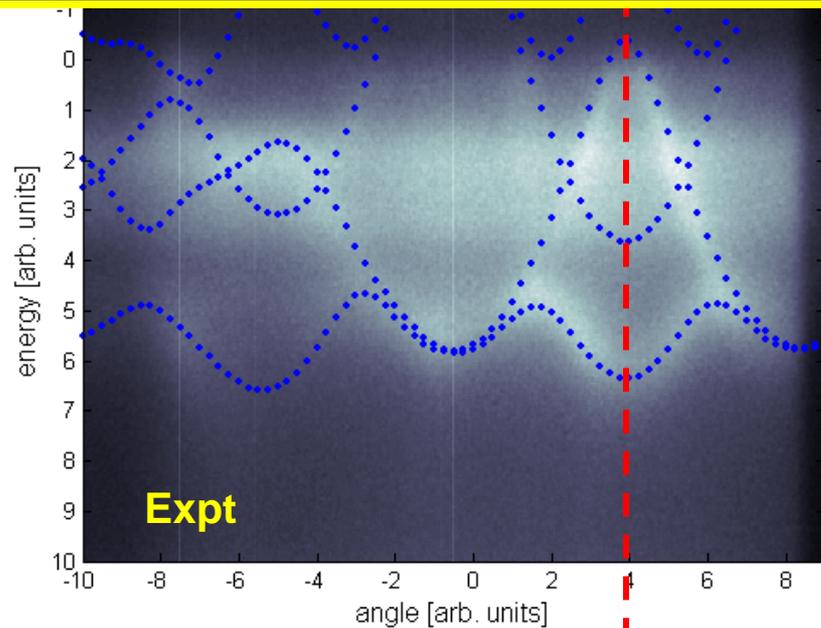
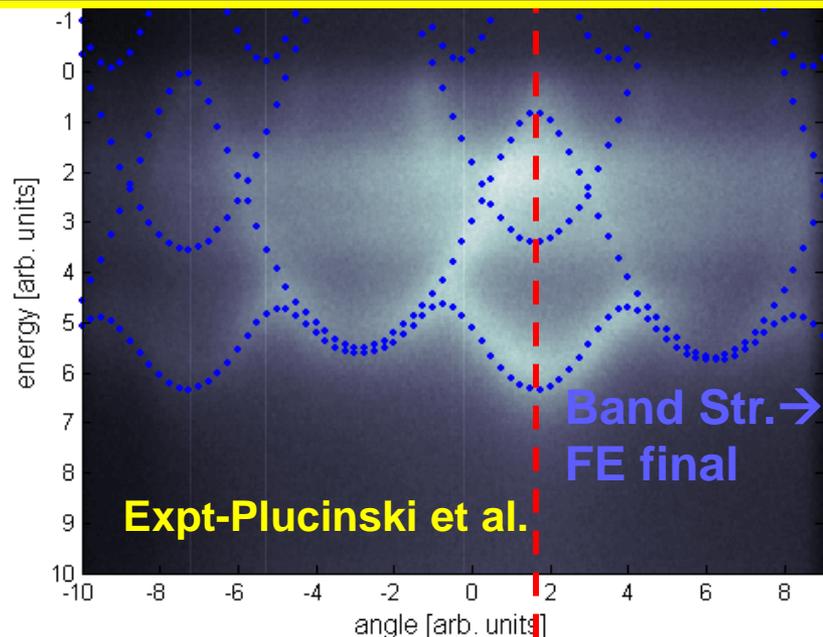


The Debye-Waller Factor

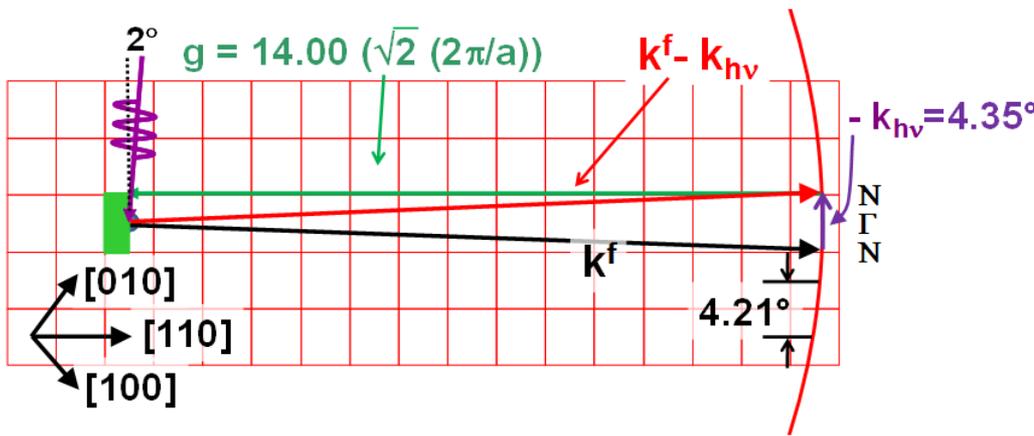
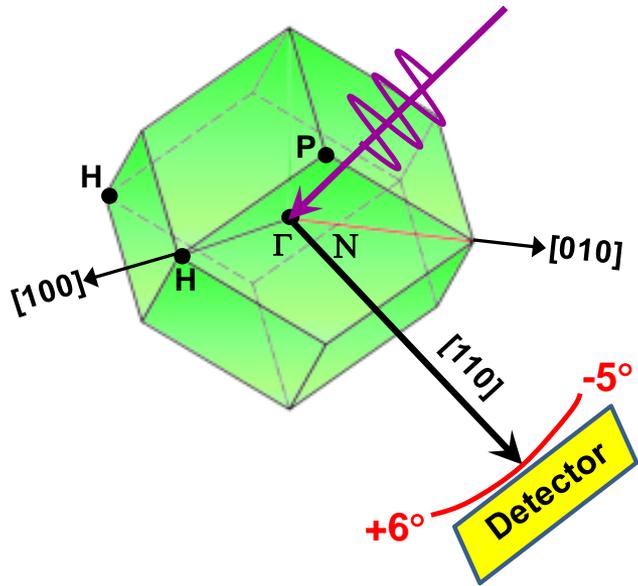
Estimate of the Fraction of Direct Transitions and Max. Recoil Effect



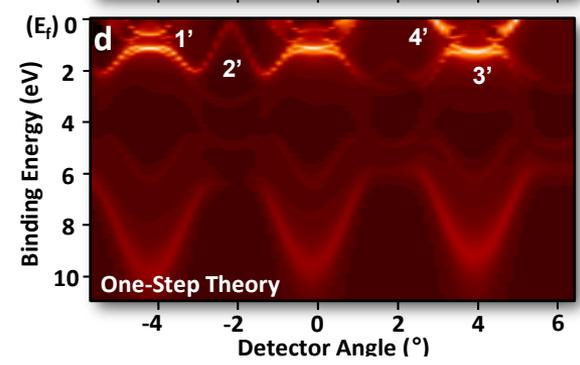
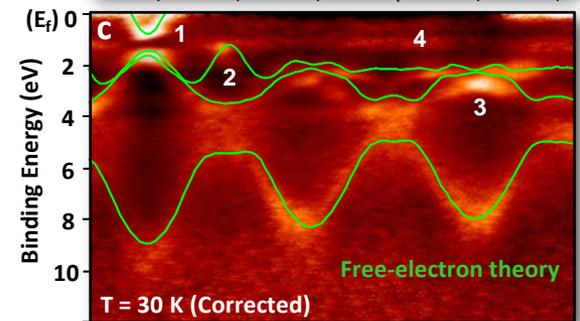
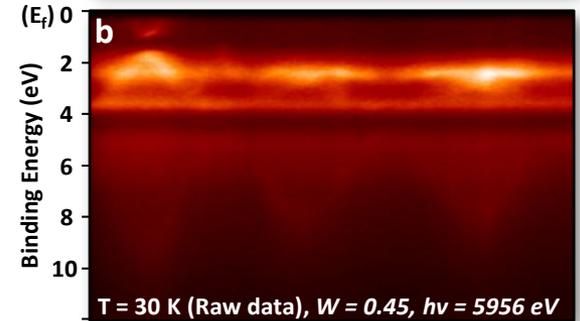
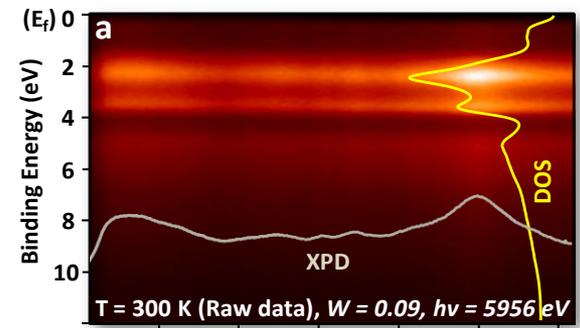
ARPES with a non-monochromatized lab. x-ray source: $h\nu = 1253.6$ eV, $T = \sim 77$ K, $W = 0.82$



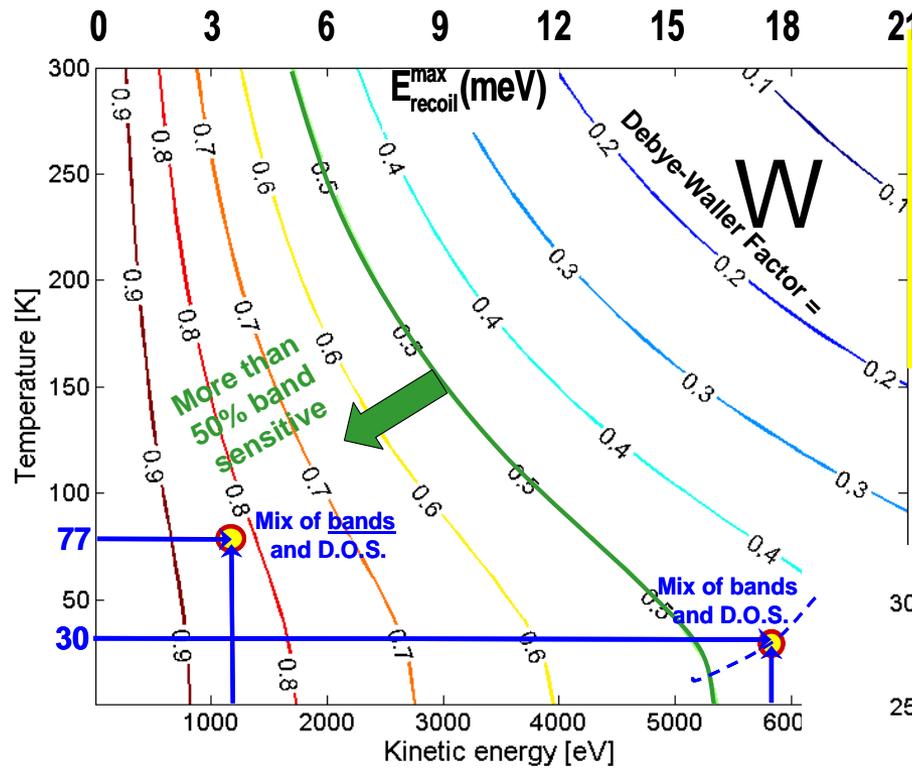
Hard x-ray ARPES for W(110): 6.0 keV, 300K and 30K



Gray, Minar et al., Nature Materials 10, 759 (2011)

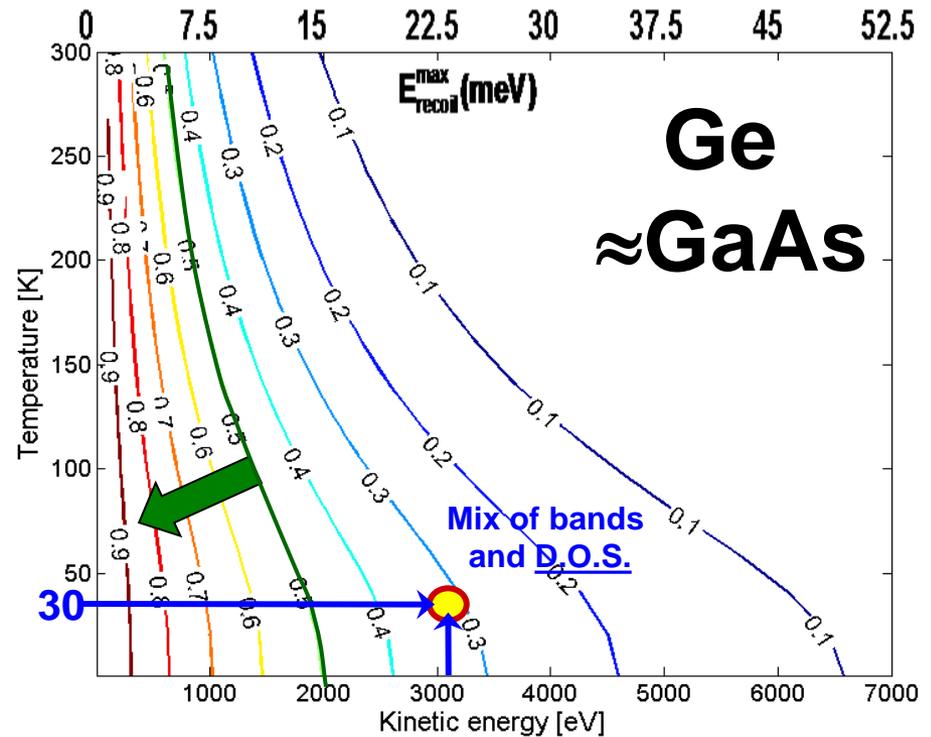


M = 183.9



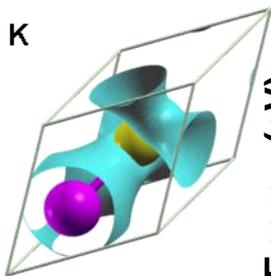
$W(T, E_{kin}) =$ Approximate recoil-free fraction \rightarrow fraction direct transitions

M = 72.6

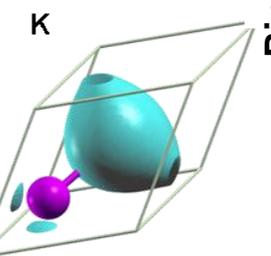


Plucinski

Hard x-ray ARPES from GaAs(001)-3.2 keV, 30 K, $W = 0.31$



Band-like

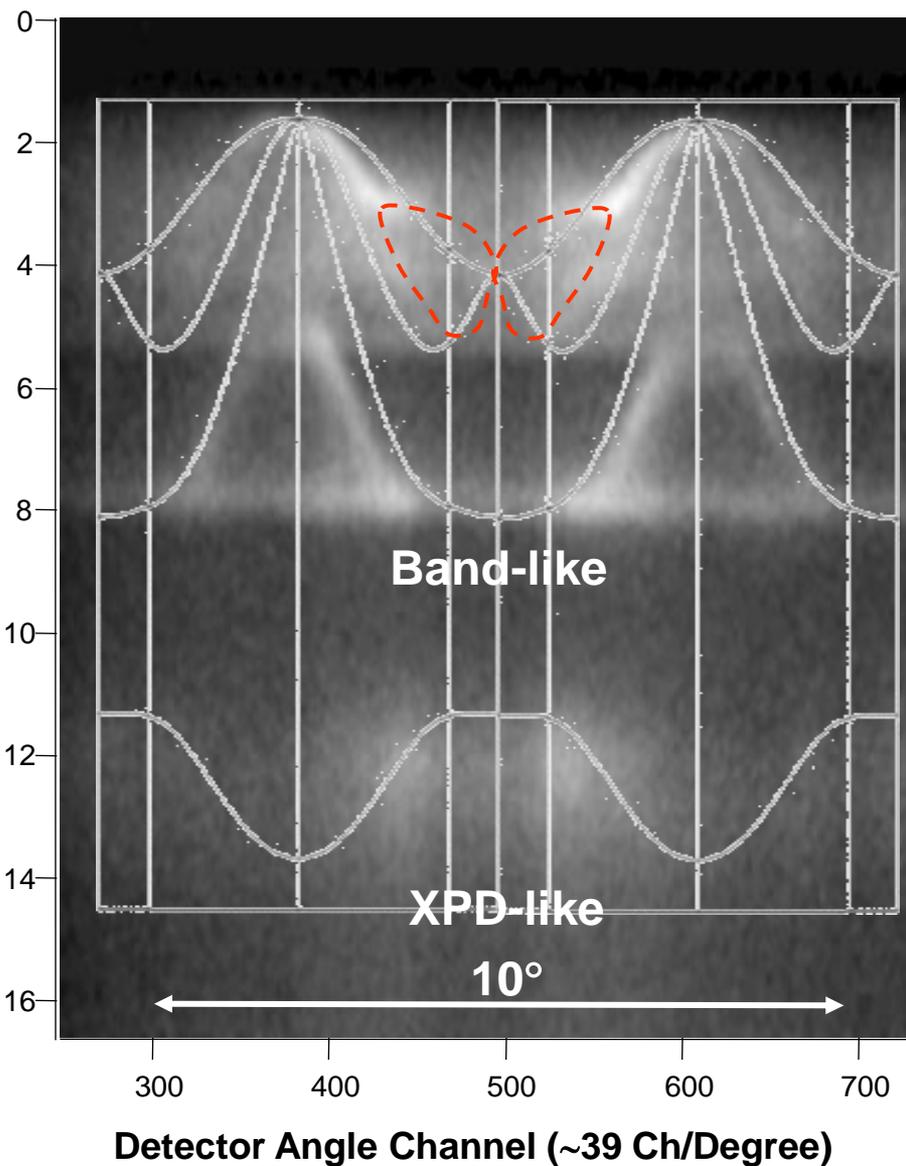


Core-like
As 4s

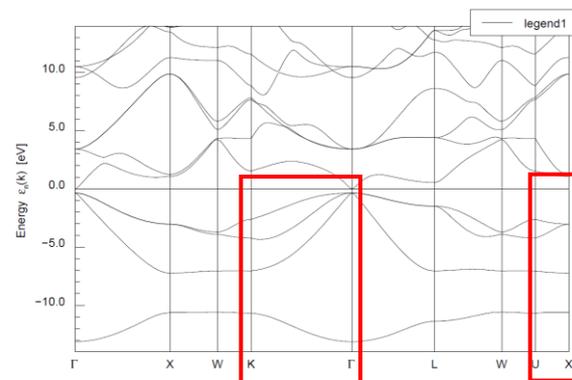
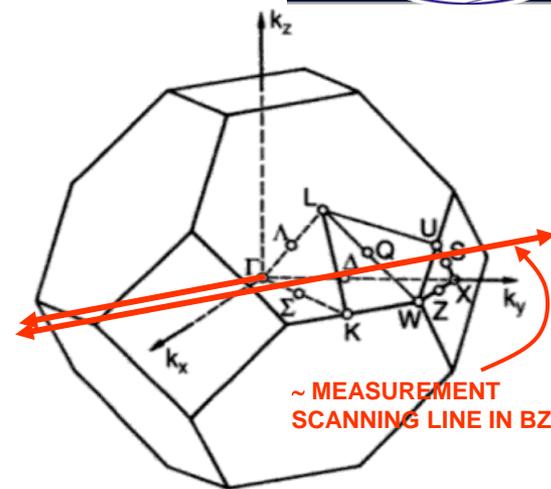


Charge
densities

Binding Energy (eV)

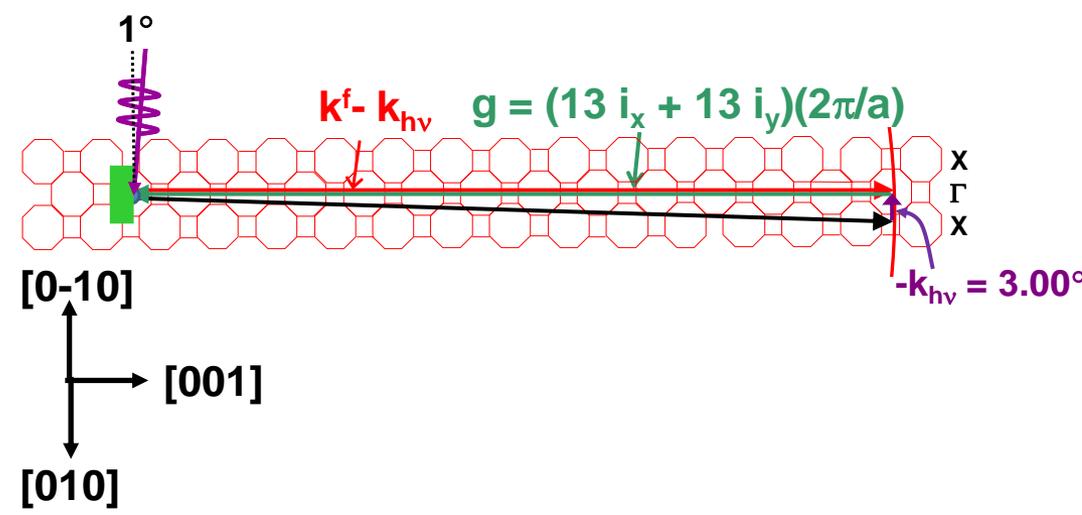
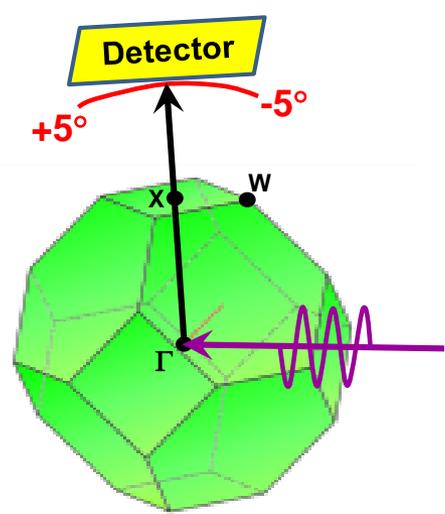


SPring-8

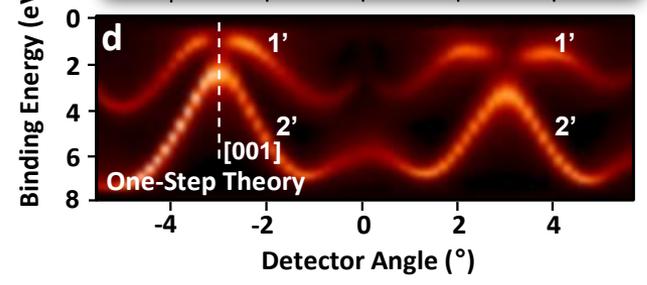
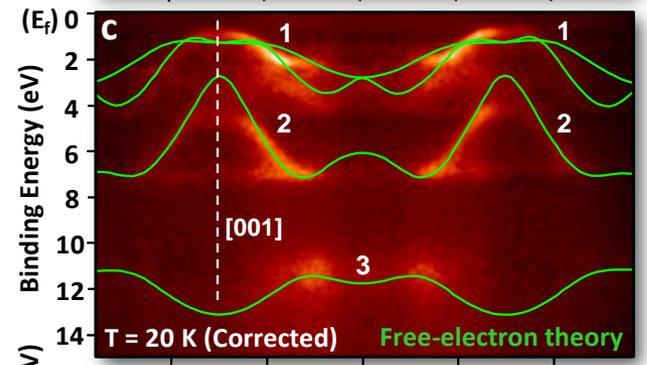
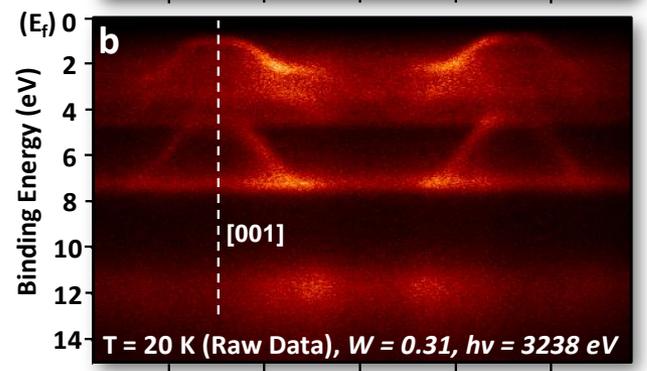
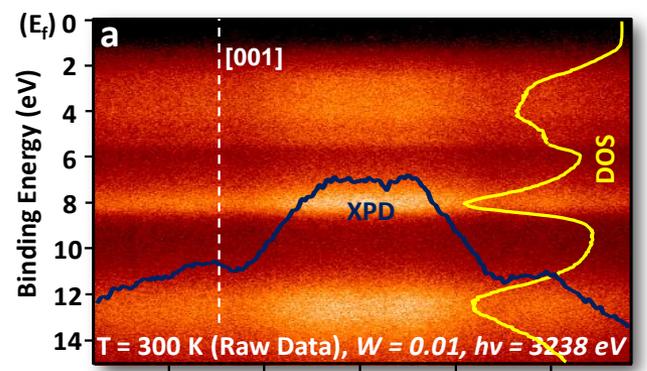


Expt.-Gray, Papp, Ueda, Yamashita,
Kobayashi
Theory- Pickett, Ylvisaker

Hard x-ray ARPES for GaAs(001): 3.2 keV

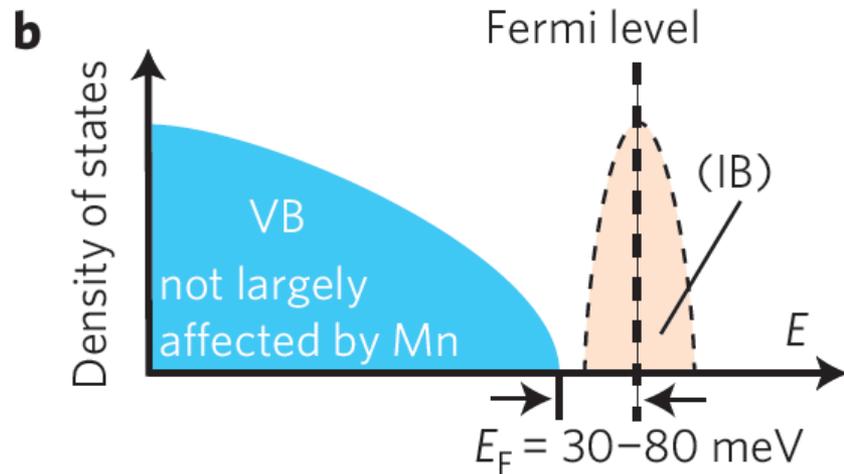
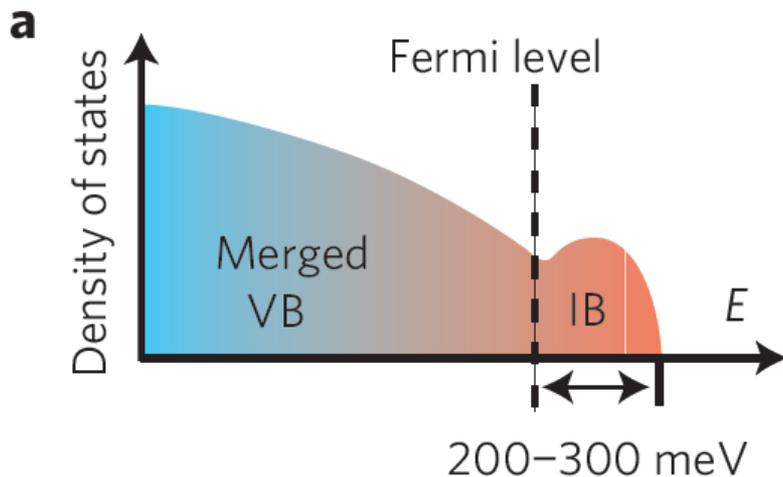


Gray, Minar et al., Nature Materials 10, 759 (2011)



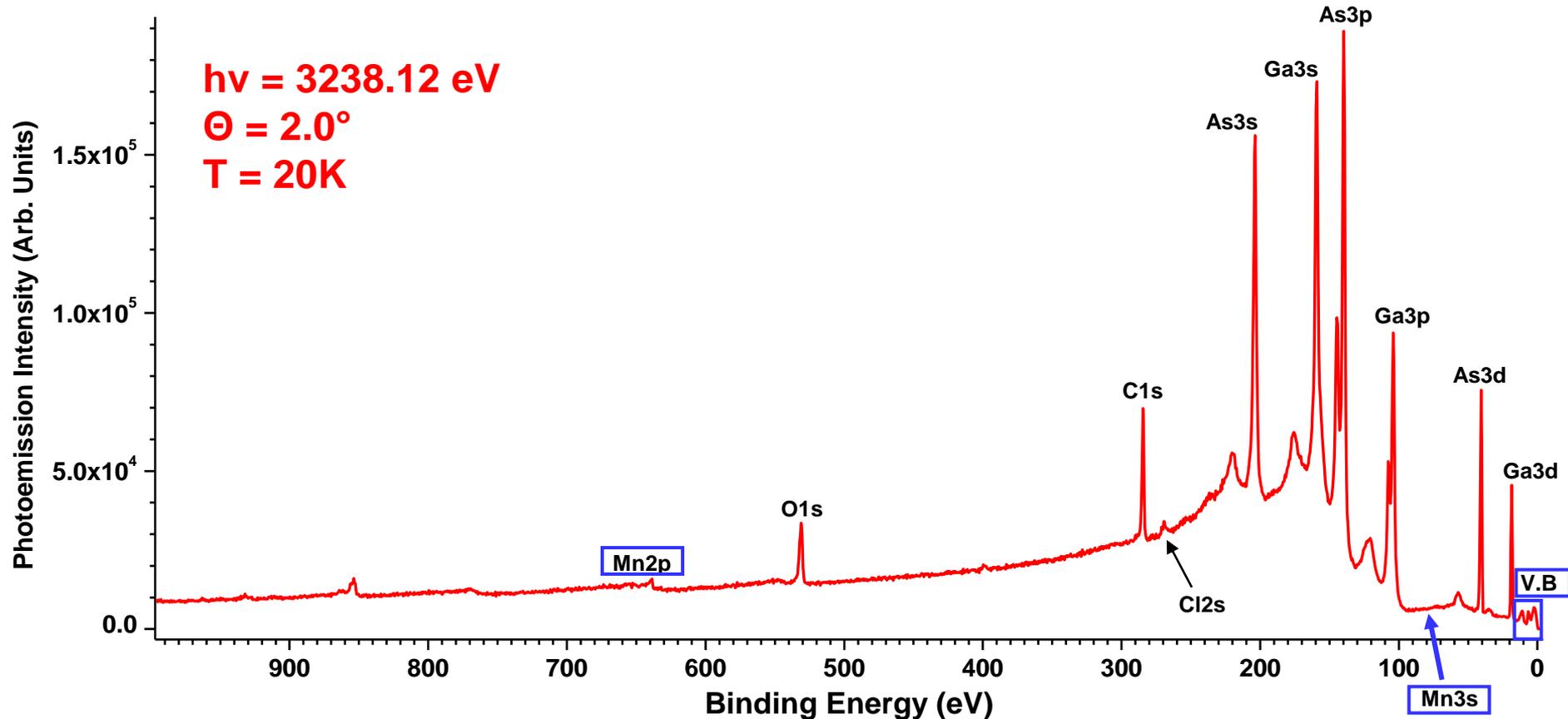
Hard x-ray ARPES--GaAs and the dilute magnetic semiconductor $\text{Ga}_{0.97}\text{Mn}_{0.03}\text{As}$

How does Mn alter the electronic structure so as to produce ferromagnetic coupling? Two differing views:



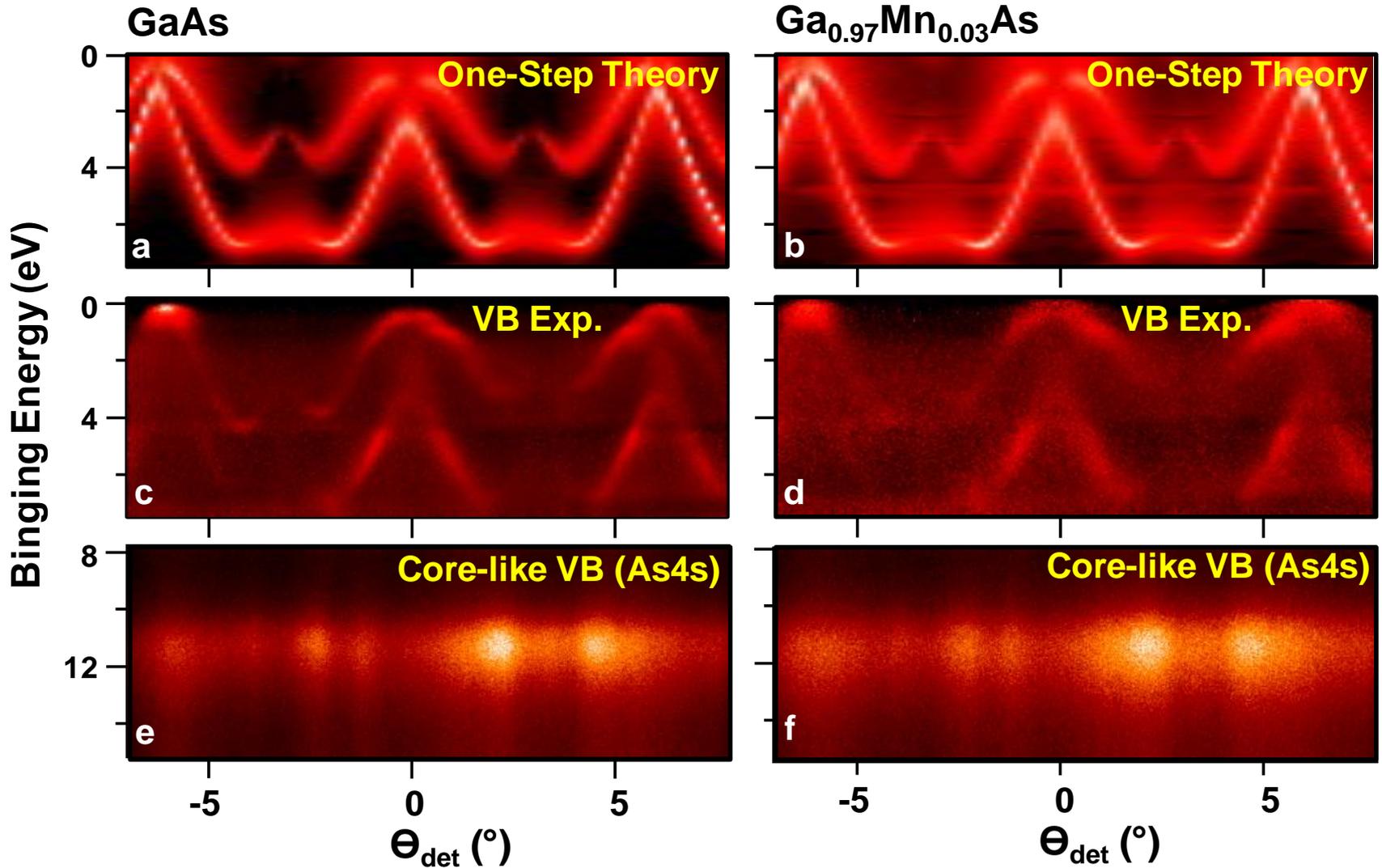
GaAs doped with Mn: a magnetic semiconductor

$\text{Ga}_{0.96}\text{Mn}_{0.04}\text{As}$ --HXPS: T = 20K, Broad Survey



Samples: Stone, Dubon
Expt.-Gray, Papp, Ueda, Yamashita, Kobayashi
Theory- Pickett, Ylvisaker, Minar, Braun, Ebert

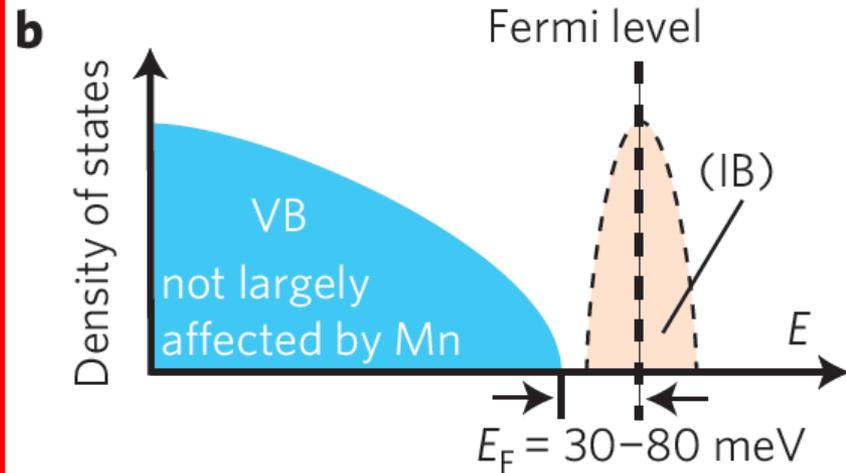
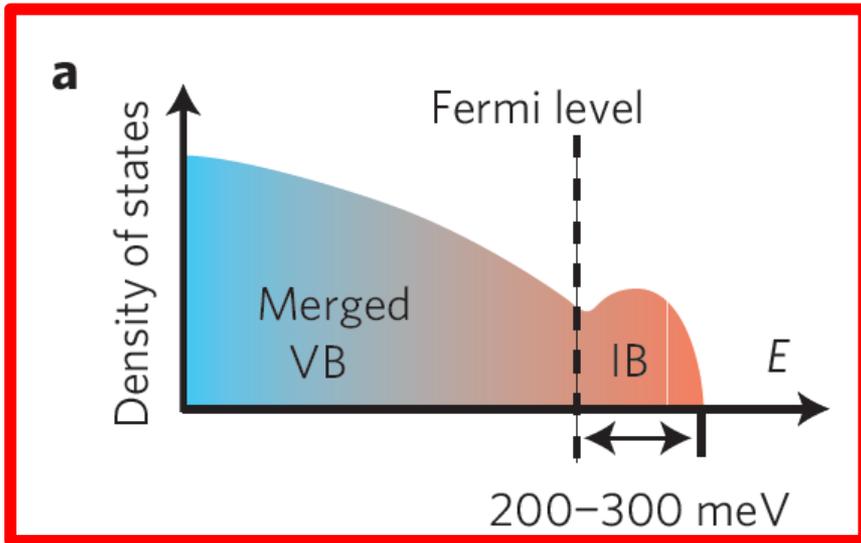
Hard x-ray ARPES--GaAs and DMS $\text{Ga}_{0.97}\text{Mn}_{0.03}\text{As}$ Comparing Experiment and One-Step KKR Theory

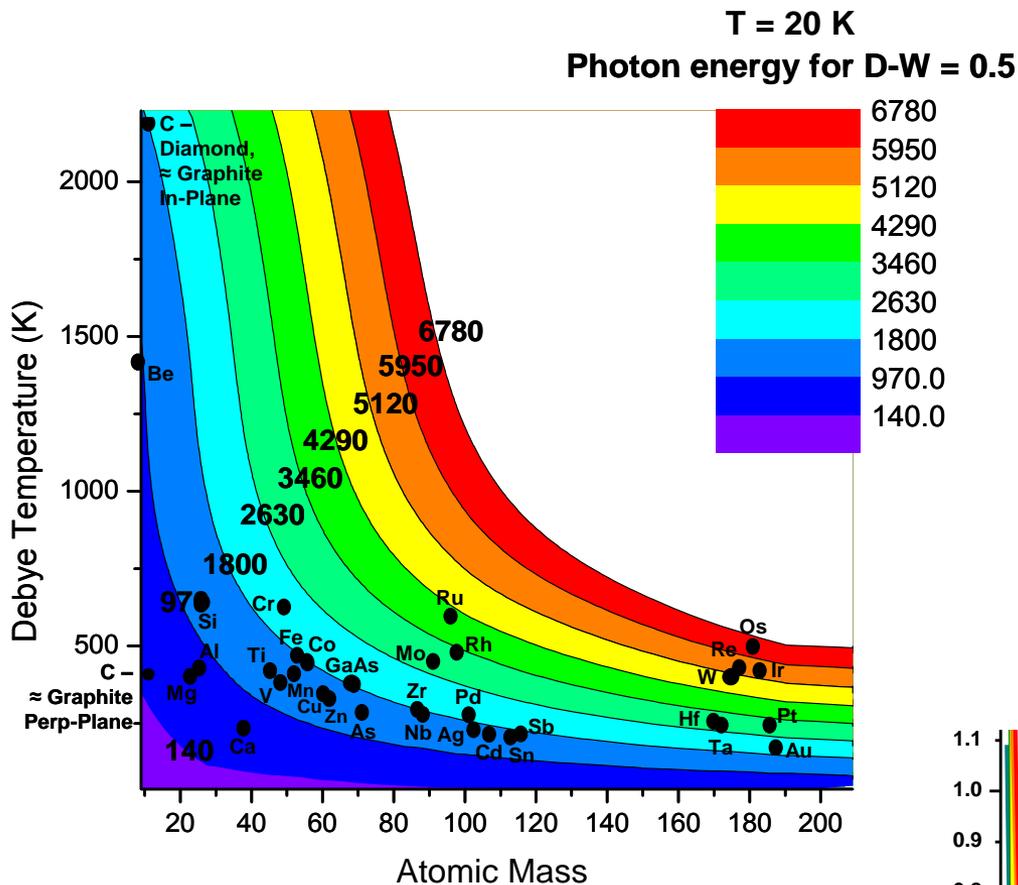


Gray, Minar, Dubon (sample) et al., TBP

Hard x-ray ARPES--GaAs and the dilute magnetic semiconductor $\text{Ga}_{0.97}\text{Mn}_{0.03}\text{As}$

How does Mn alter the electronic structure so as to produce ferromagnetic coupling? Two differing views:

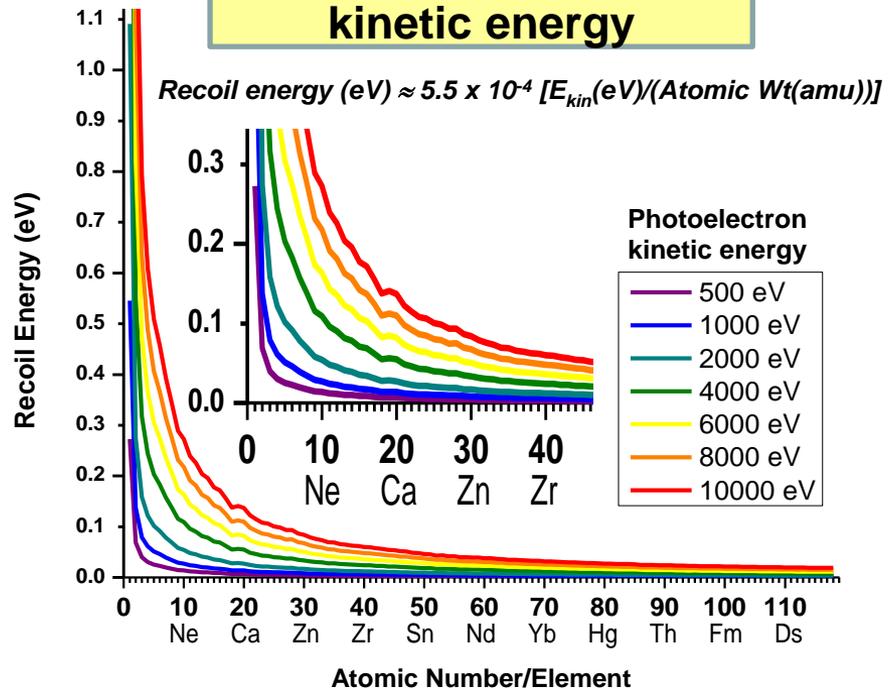




**Photon energy for ca.
50% k-conserving
transitions @ 20K**

**HARPES-How
high can we go?
Photoemission
Debye-Waller
Factors and
Recoil Energies**

**Recoil energy as a fn.
of atomic number and
kinetic energy**



**C. Papp, L. Plucinski, et al.,
Phys. Rev. B 84, 045433 (2011)**

Basic Concepts and Experiments

Core-Level Photoemission

- Intensities and Quantitative Analysis, the 3-Step Model
- Varying Surface and Bulk Sensitivity
- Chemical shifts
- Multiplet Splittings
- Electron Screening and Satellite Structure
- Magnetic and Non-Magnetic Dichroism
- Resonant Photoemission
- Photoelectron Diffraction and Holography

Valence-Level Photoemission

- Band-Mapping in the Ultraviolet Photoemission Limit
- Densities of States in the X-Ray Photoemission Limit

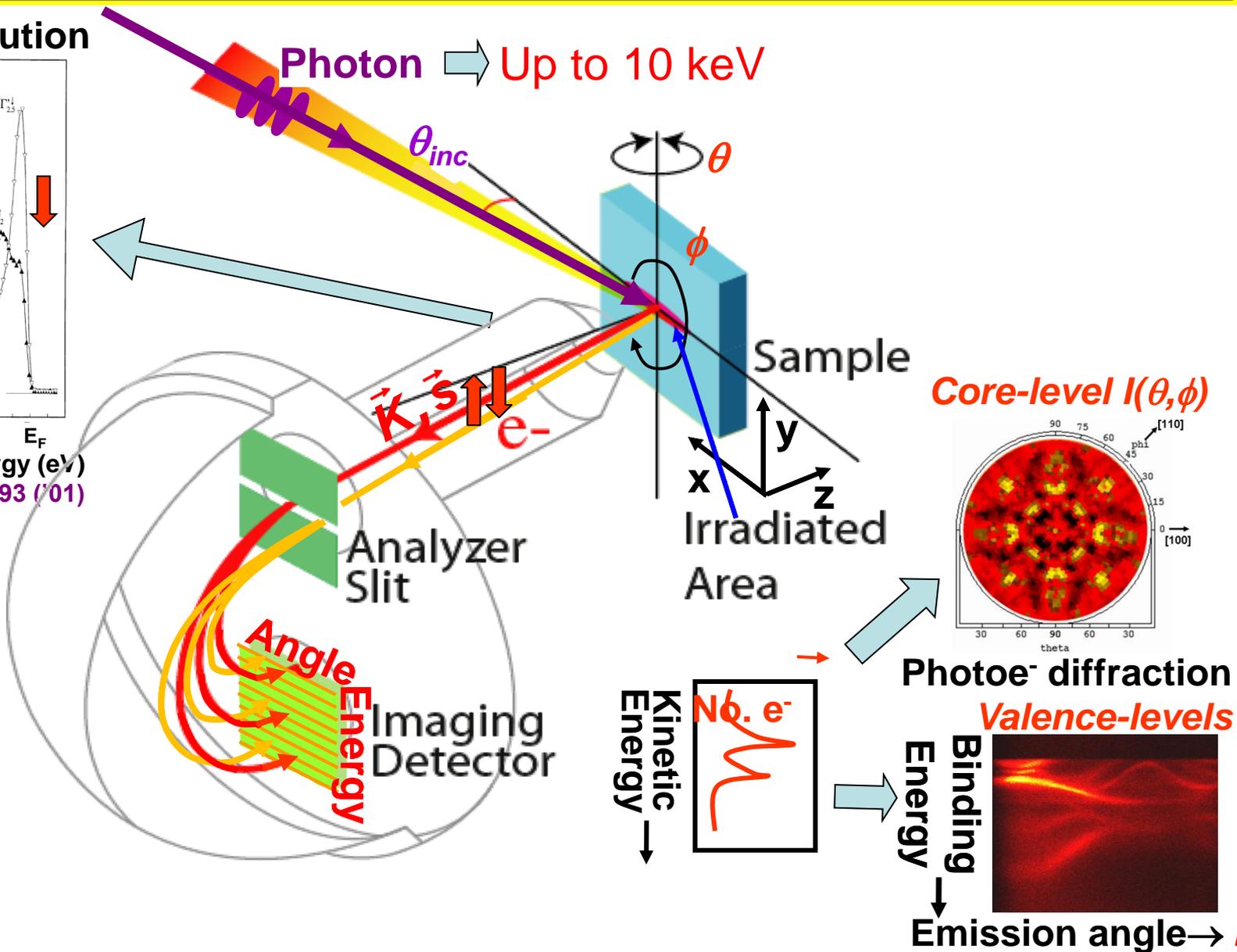
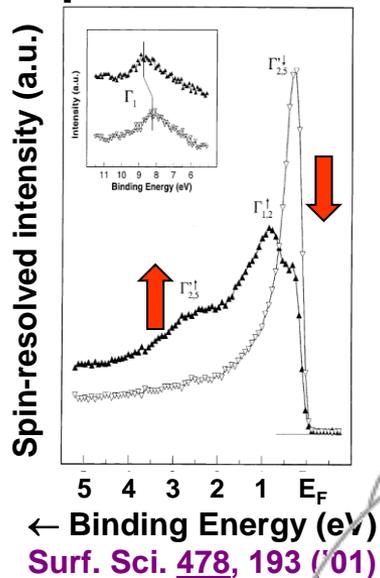
Some New Directions



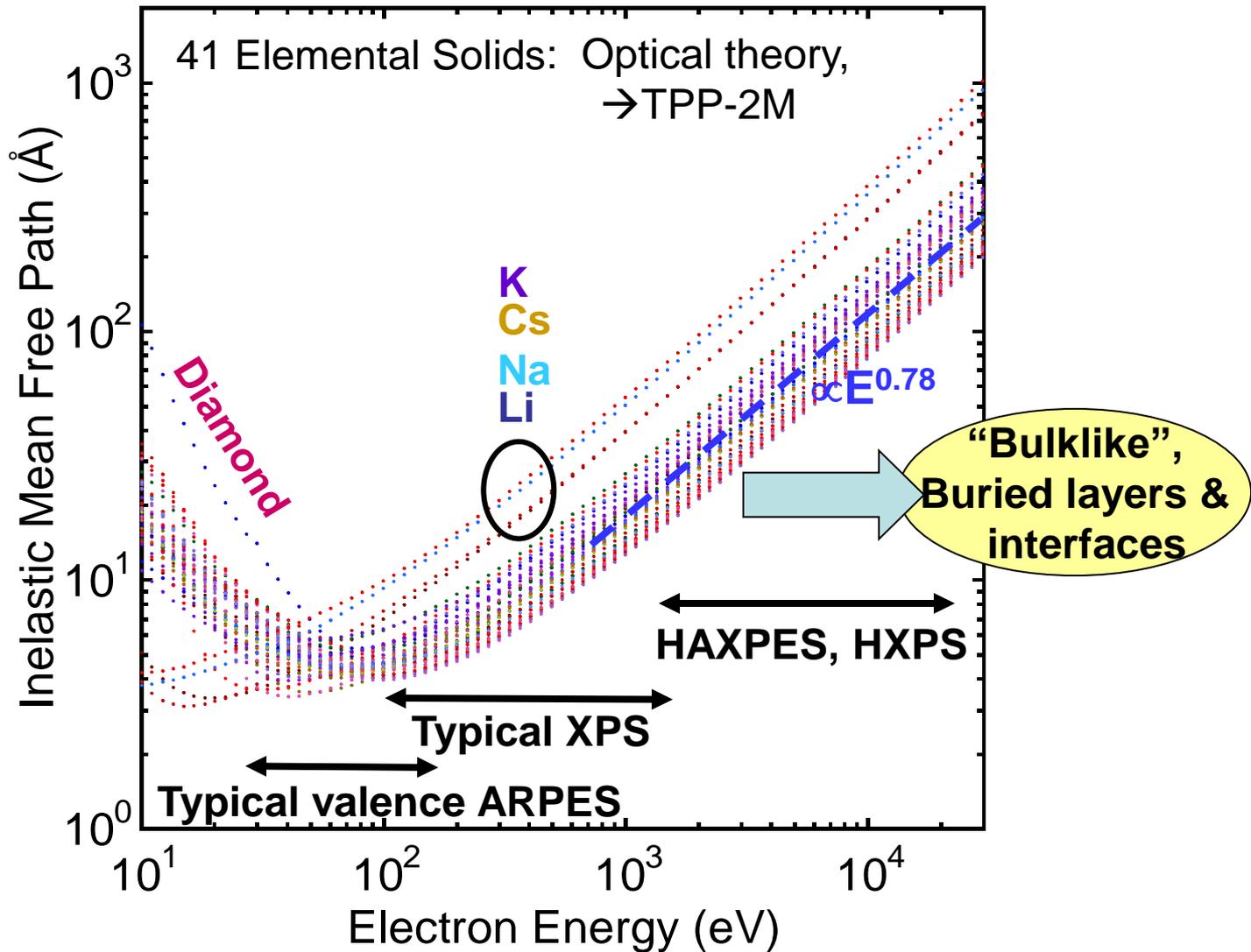
- Photoemission with Hard X-Rays (throughout lectures)
- Photoemission with Standing Wave Excitation
- Photoemission with Spatial Resolution/Photoelectron Microscopy
- Temporal Resolution
- @ Higher Pressures

Typical experimental geometry for energy- and angle-resolved photoemission measurements

Spin-resolution



Surface sensitivity and why we may want to go to 5-10 keV in XPS



Varying surface sensitivity for lower electron takeoff angles

Simplest interpretation:

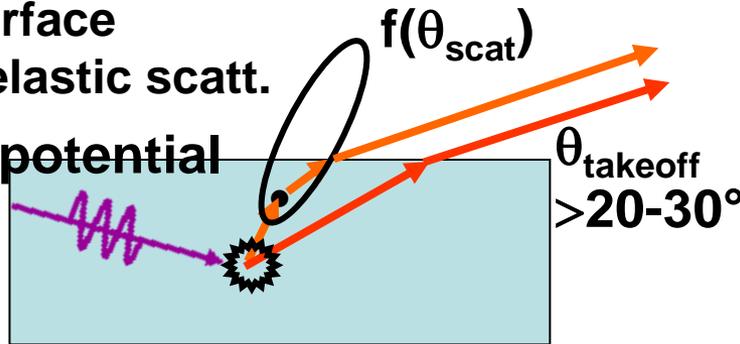
Average emission depth = $\Lambda_{\text{inelastic}} \sin \theta_{\text{takeoff}}$

How valid?

$E_{\text{kin}} \approx 500-1000 \text{ eV}$

Surface inelastic scatt.

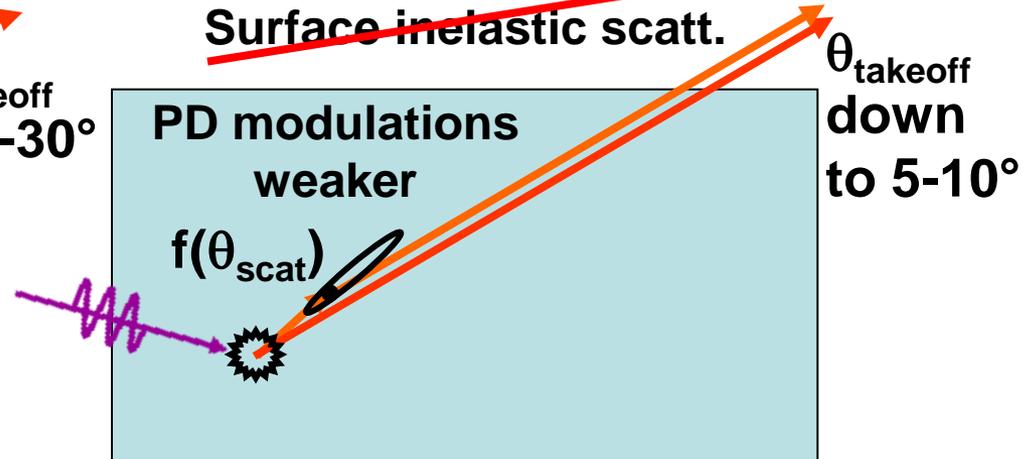
Inner potential



E.g.: A. Jablonski and C. J. Powell,
J. Vac. Sci. Tech. A 21, 274 (2003):
→ Mean Emission Depth (MED)
more relevant than $\Lambda_{\text{inelastic}}$

$E_{\text{kin}} \approx 10,000 \text{ eV}$

Surface inelastic scatt.



Approx. constant analyzer transmission

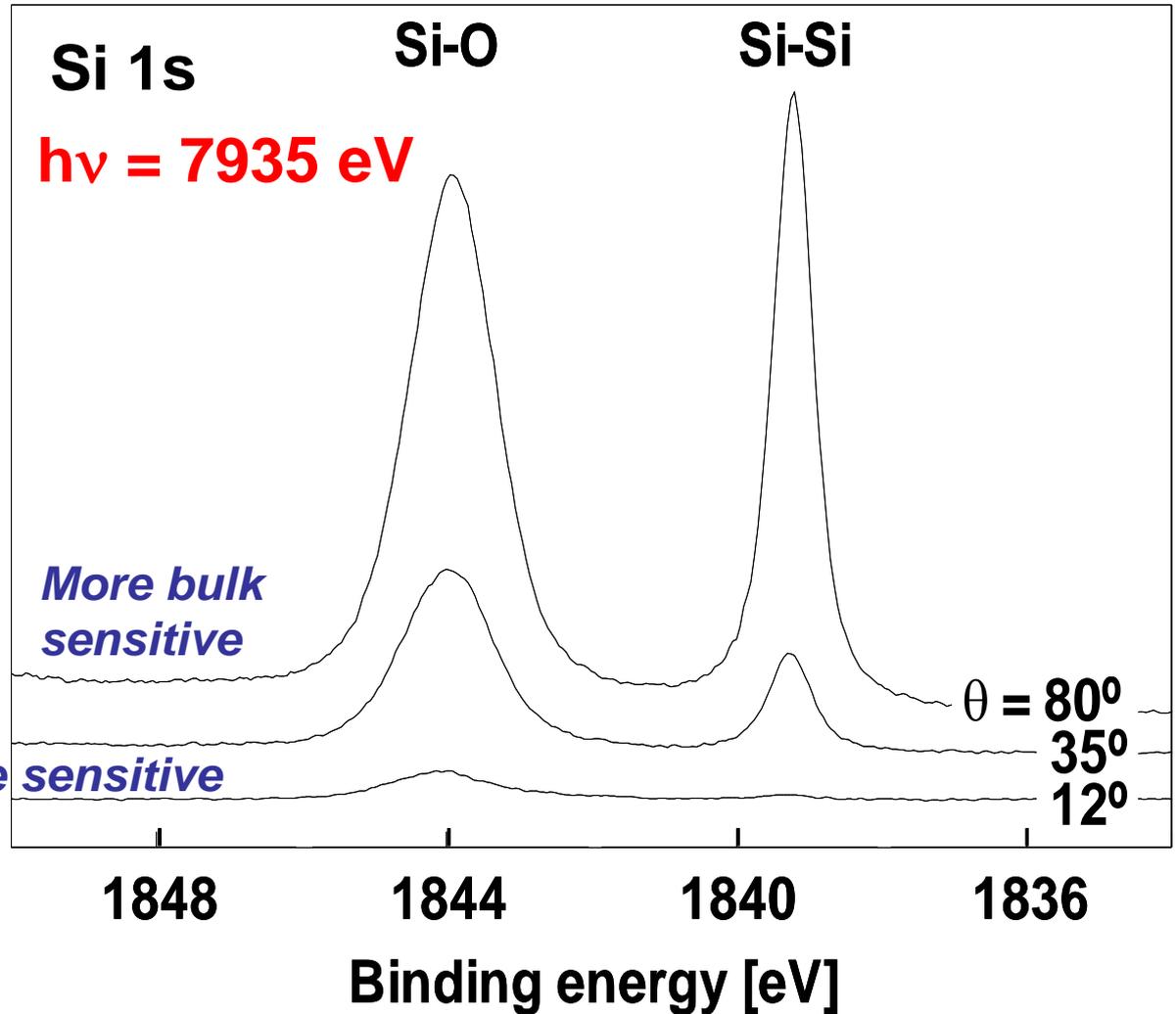
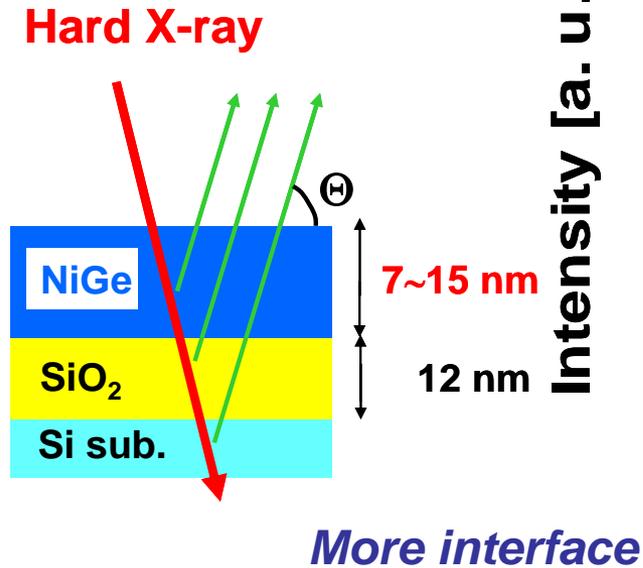
Simpler analysis

Cleaner bulk & surface distinction

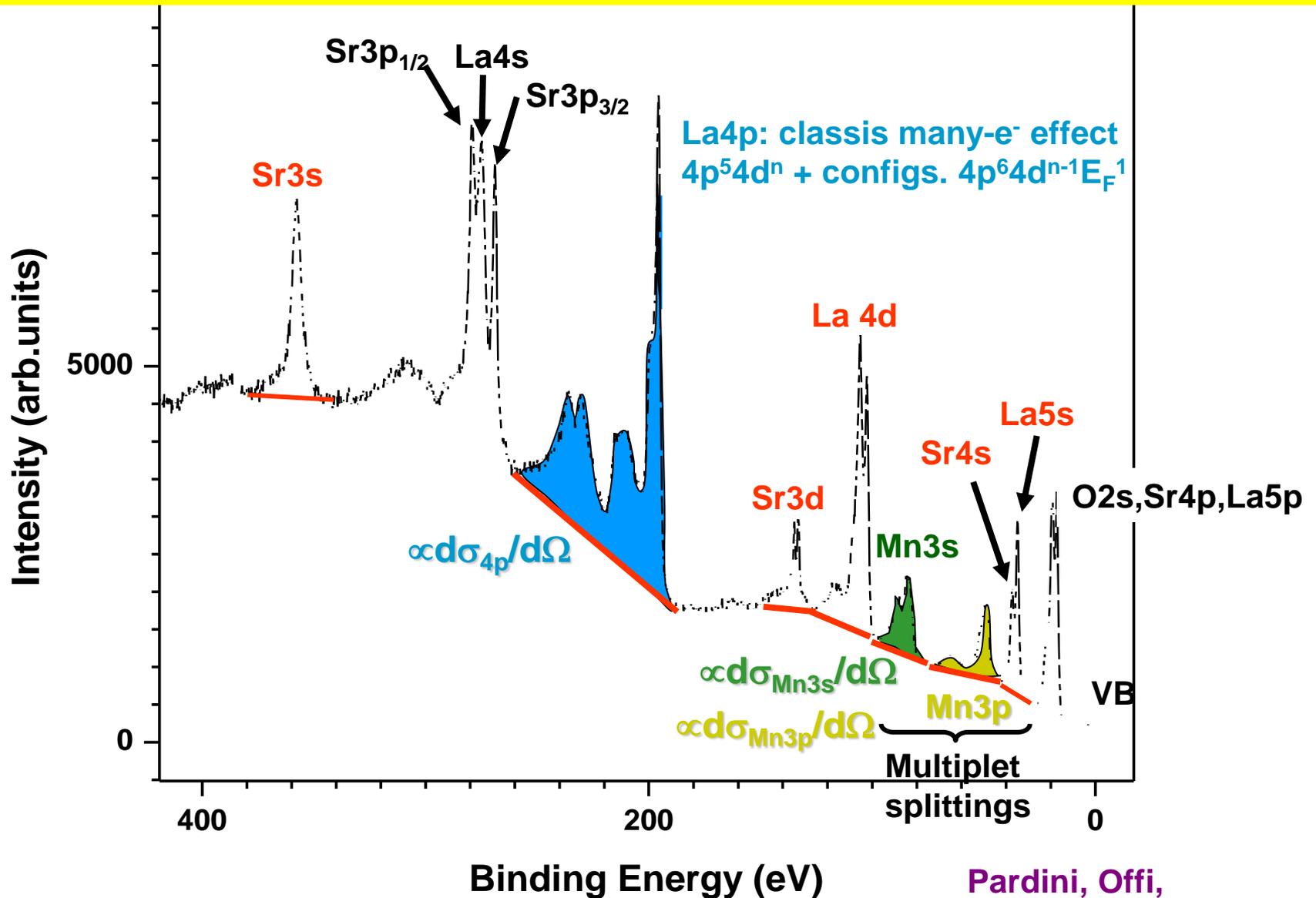
C.S.F., Nucl. Inst. & Meth. A 547, 24 (2005)

Kover, Werner, Drube, et al., Surf. & Int. Anal. 38, 569 (2006)

Looking into nanoscale devices--Variable takeoff-angle Si 1s photoelectron spectra from NiGe(12-nm)/SiO₂(12-nm)/Si(100)

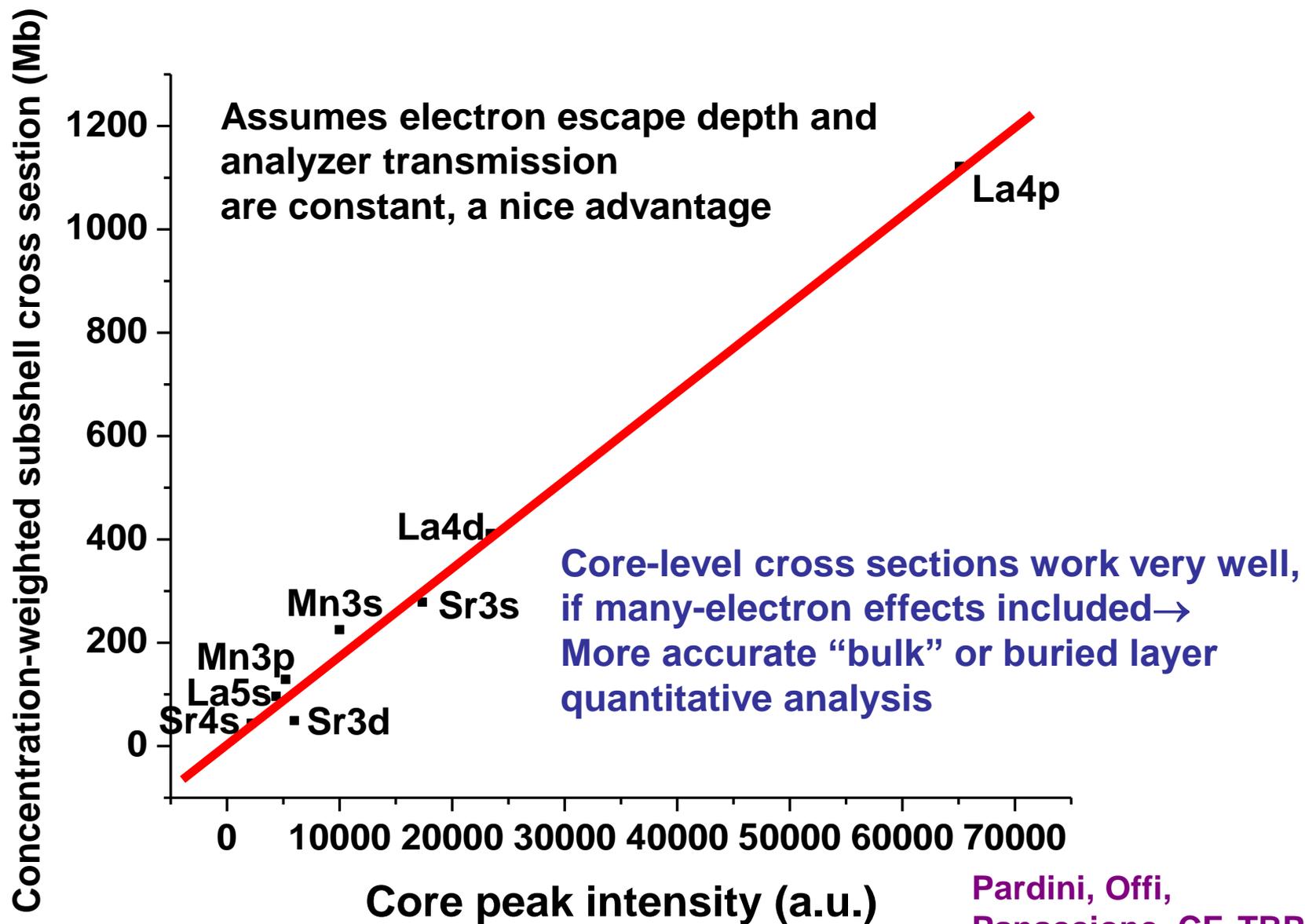


Hard x-ray photoemission--Quantitative analysis of peak intensities using theoretical cross sections: $\text{La}_{0.7}\text{Sr}_{0.3}\text{MnO}_3$, $h\nu = 7700 \text{ eV}$

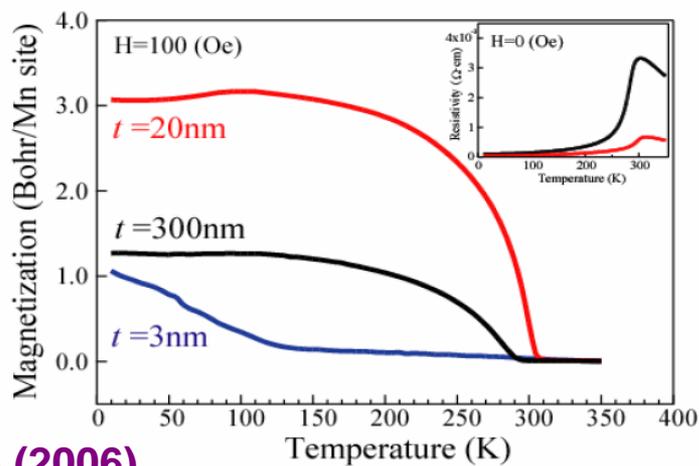
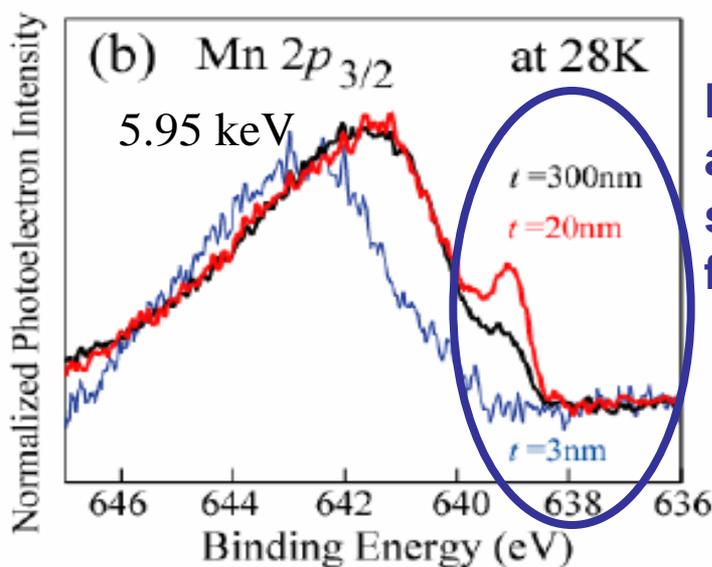
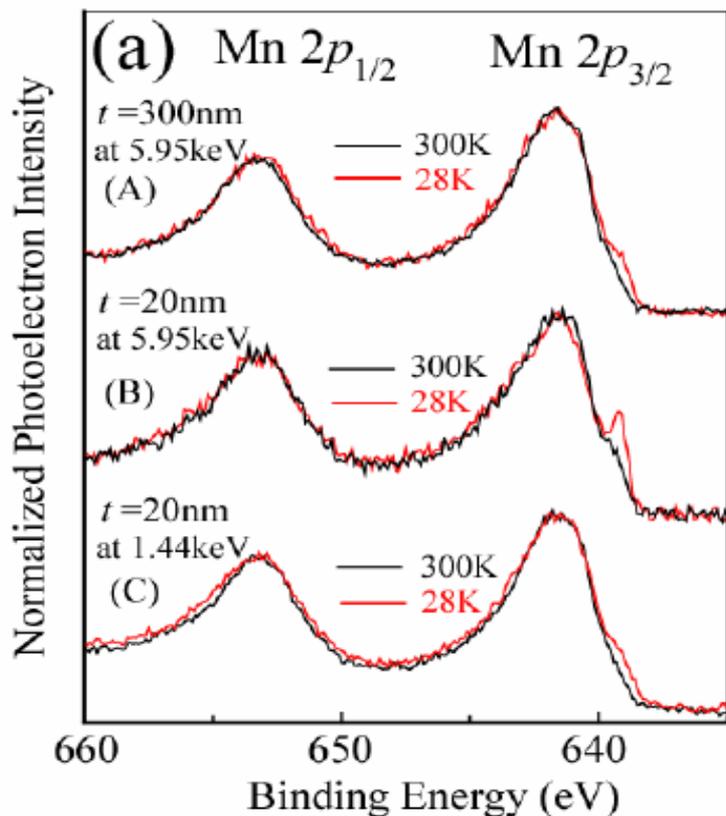


Pardini, Offi,
Panaccione, CF, TBP

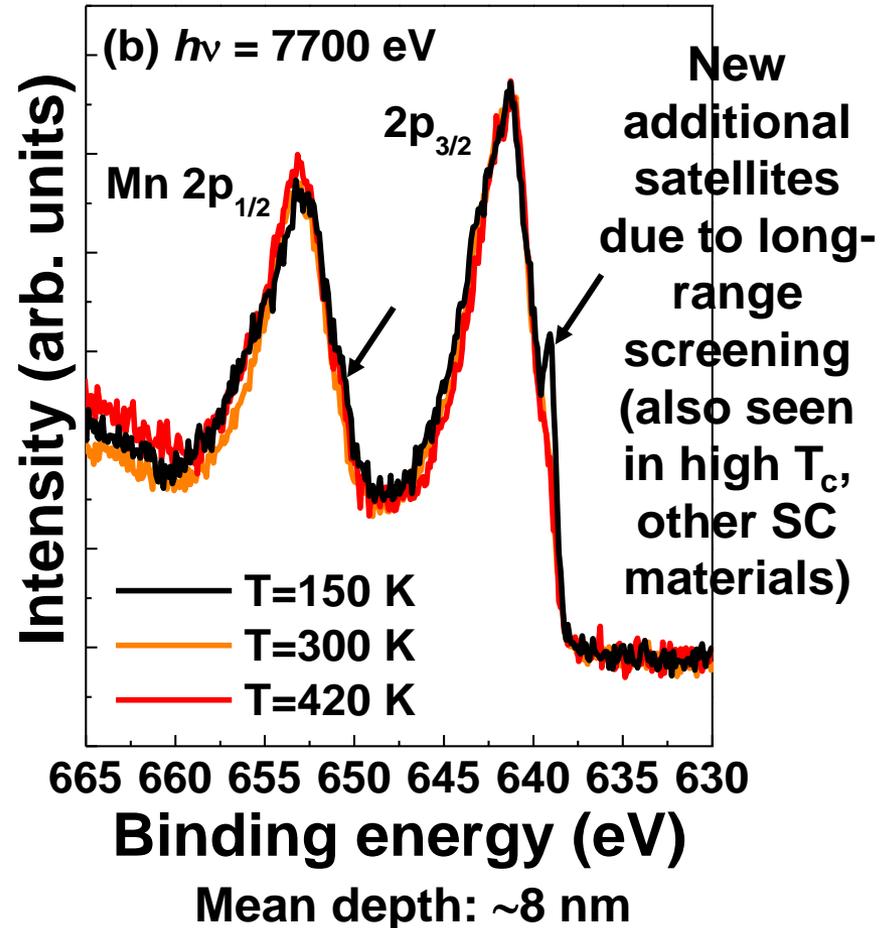
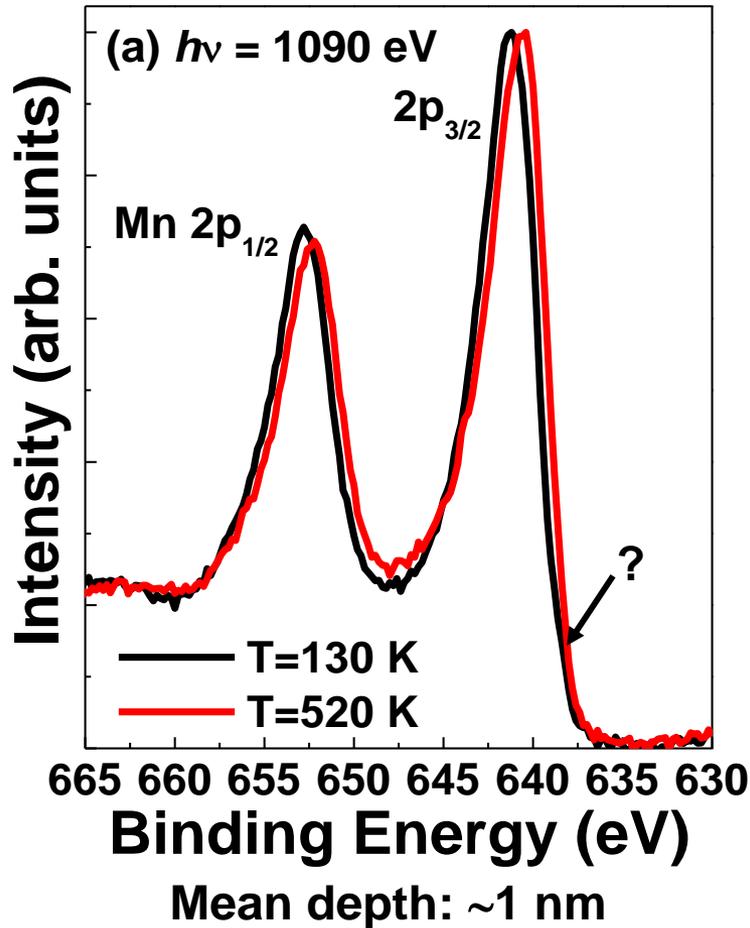
Quantitative analysis of peak intensities using theoretical cross sections (Scofield): $\text{La}_{0.7}\text{Sr}_{0.3}\text{MnO}_3$, $h\nu = 7700 \text{ eV}$



Electronic Structure of Strained Manganite Thin Films with Room Temperature Ferromagnetism Investigated by Hard X-ray Photoemission Spectroscopy:



Temperature dependence of Mn2p spectra: $\text{La}_{0.7}\text{Sr}_{0.3}\text{MnO}_3$ New satellite structures in core spectra



→ Suggests bulk electronic structure not reached until ca. 8 nm depth

Basic Concepts and Experiments

Core-Level Photoemission

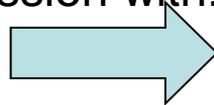
Intensities and Quantitative Analysis, the 3-Step Model
Varying Surface and Bulk Sensitivity
Chemical shifts
Multiplet Splittings
Electron Screening and Satellite Structure
Magnetic and Non-Magnetic Dichroism
Resonant Photoemission
Photoelectron Diffraction and Holography

Valence-Level Photoemission

Band-Mapping in the Ultraviolet Photoemission Limit
Densities of States in the X-Ray Photoemission Limit

Some New Directions

Photoemission with: Hard X-Rays (throughout lectures)



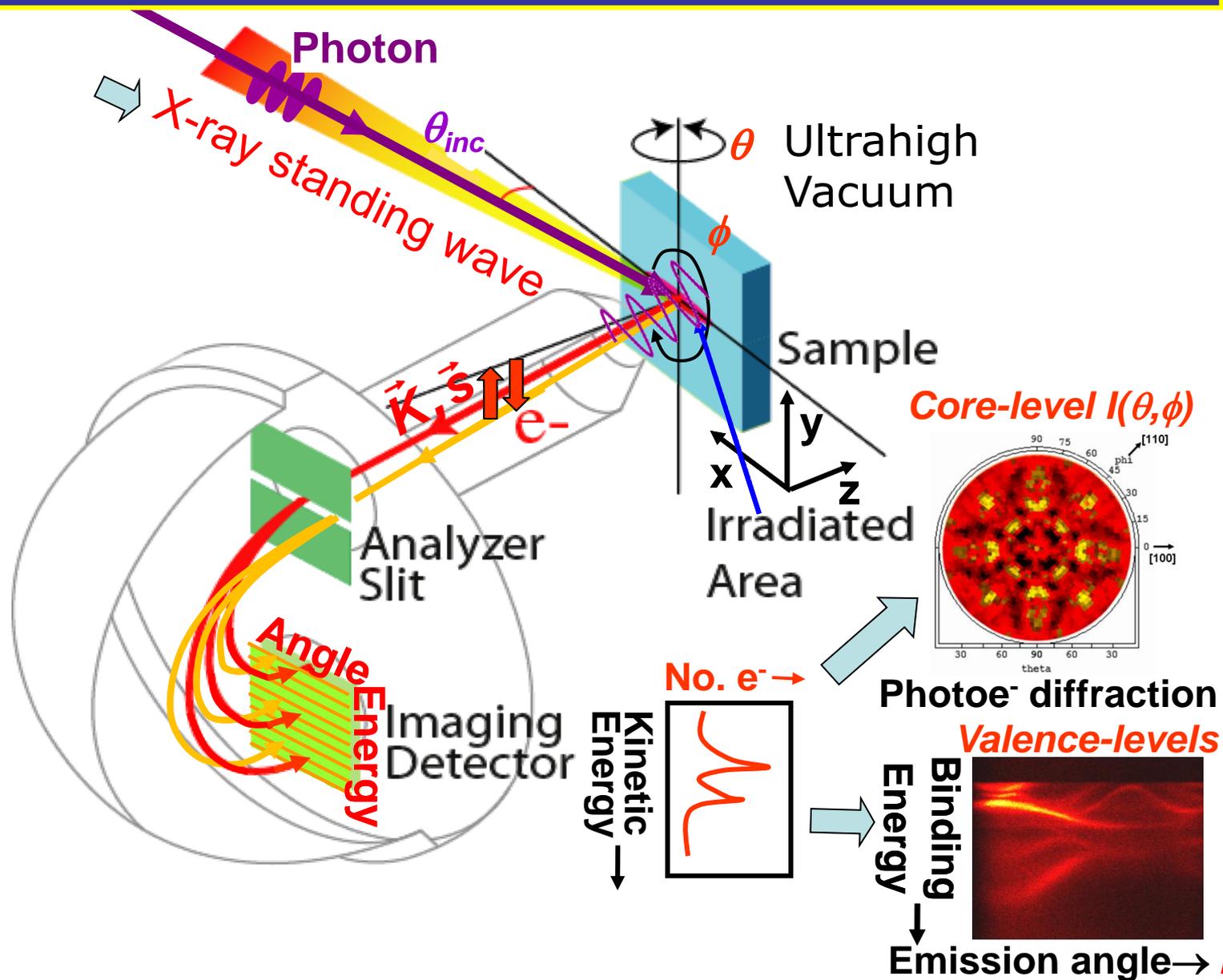
Standing Wave Excitation

Spatial Resolution/Photoelectron Microscopy

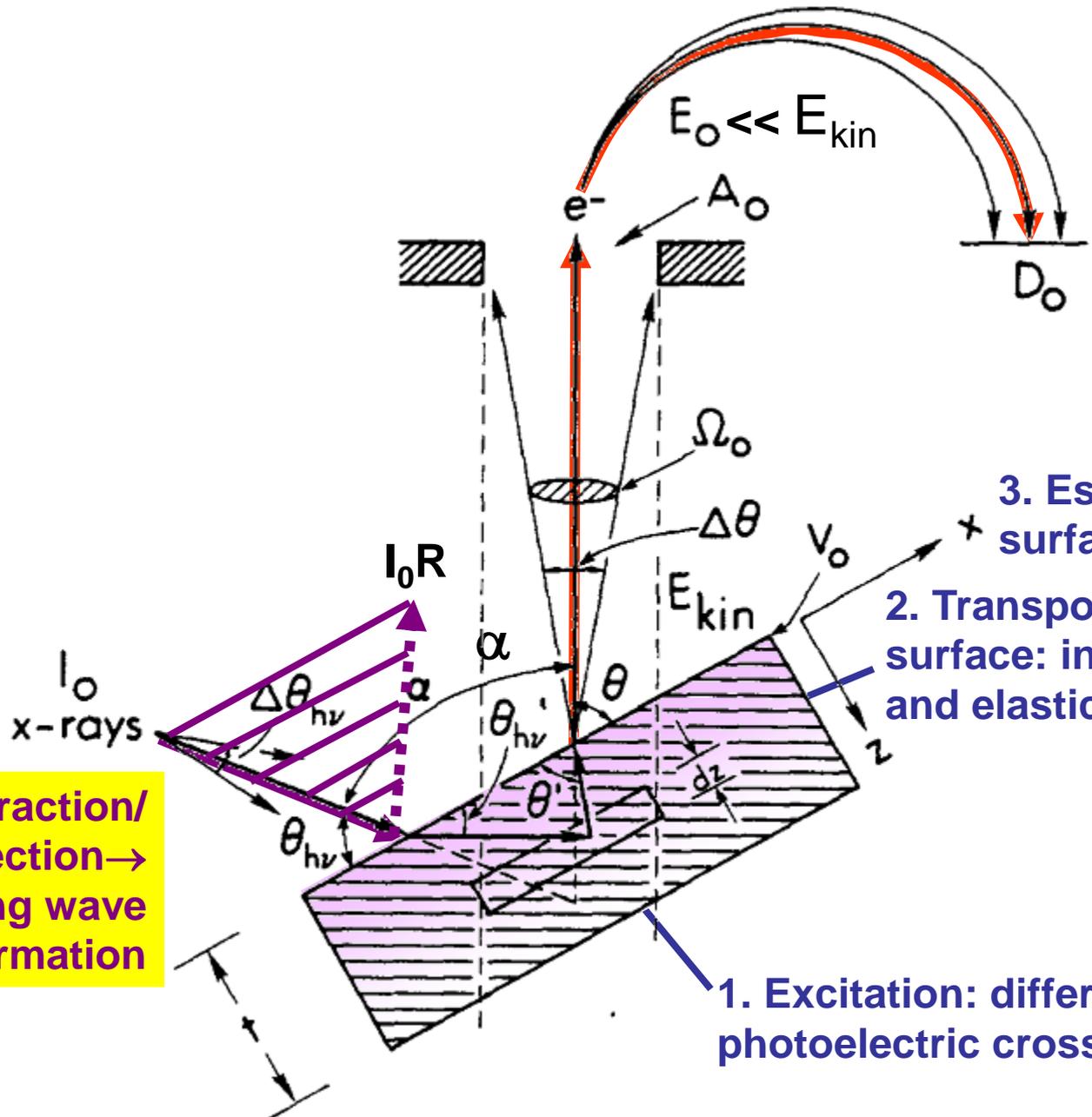
Temporal Resolution

@ Higher Pressures

Typical experimental geometry for energy- and angle-resolved photoemission measurements



PHOTOELECTRON INTENSITIES—THE 3-STEP MODEL



+X-ray refraction/
reflection →
standing wave
formation

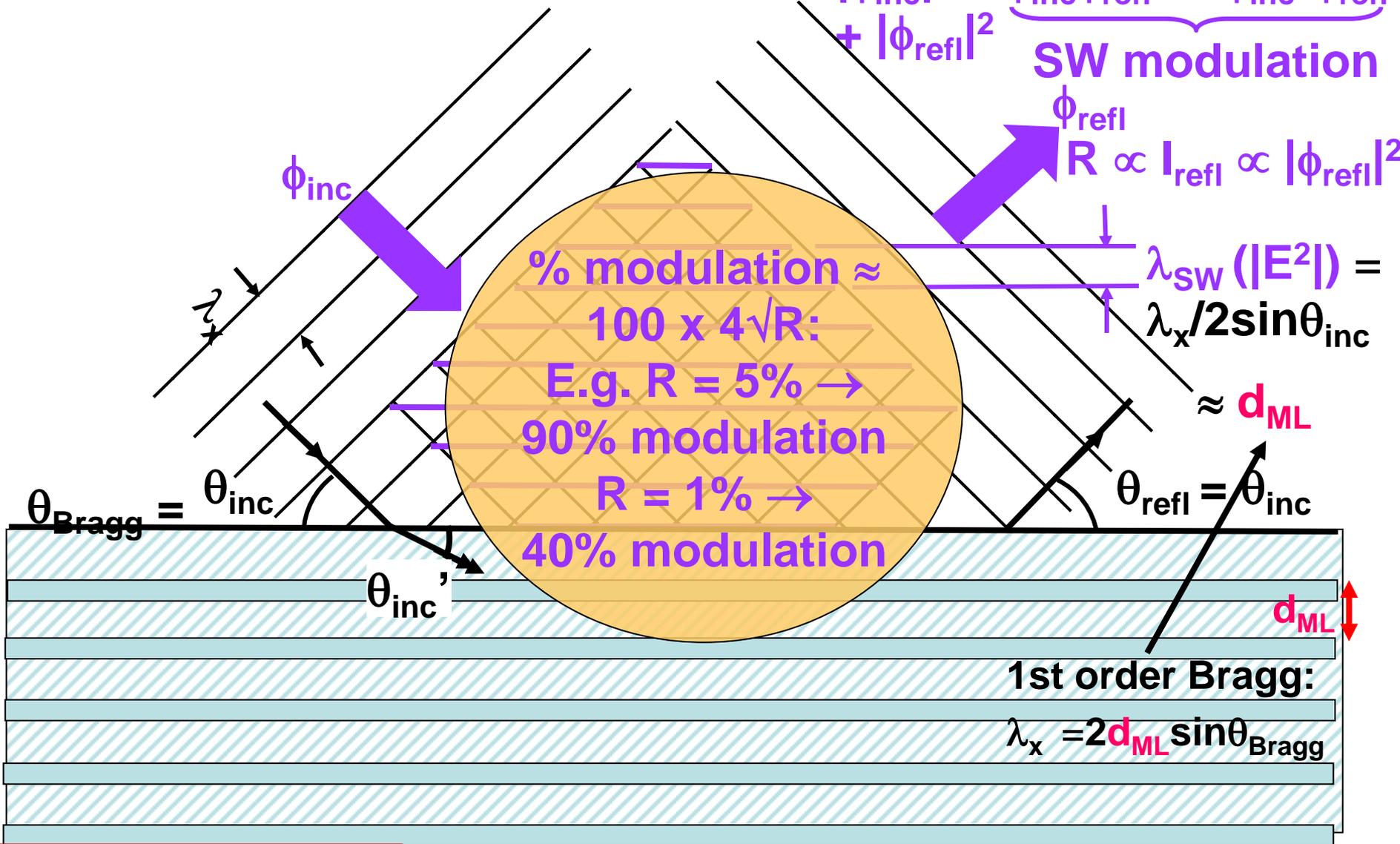
3. Escape across the surface barrier (V_0)
2. Transport to the surface: inelastic (Λ_e) and elastic ($f(\theta)$) scattering

1. Excitation: differential photoelectric cross section ($d\sigma/d\Omega$)

Standing wave formation:

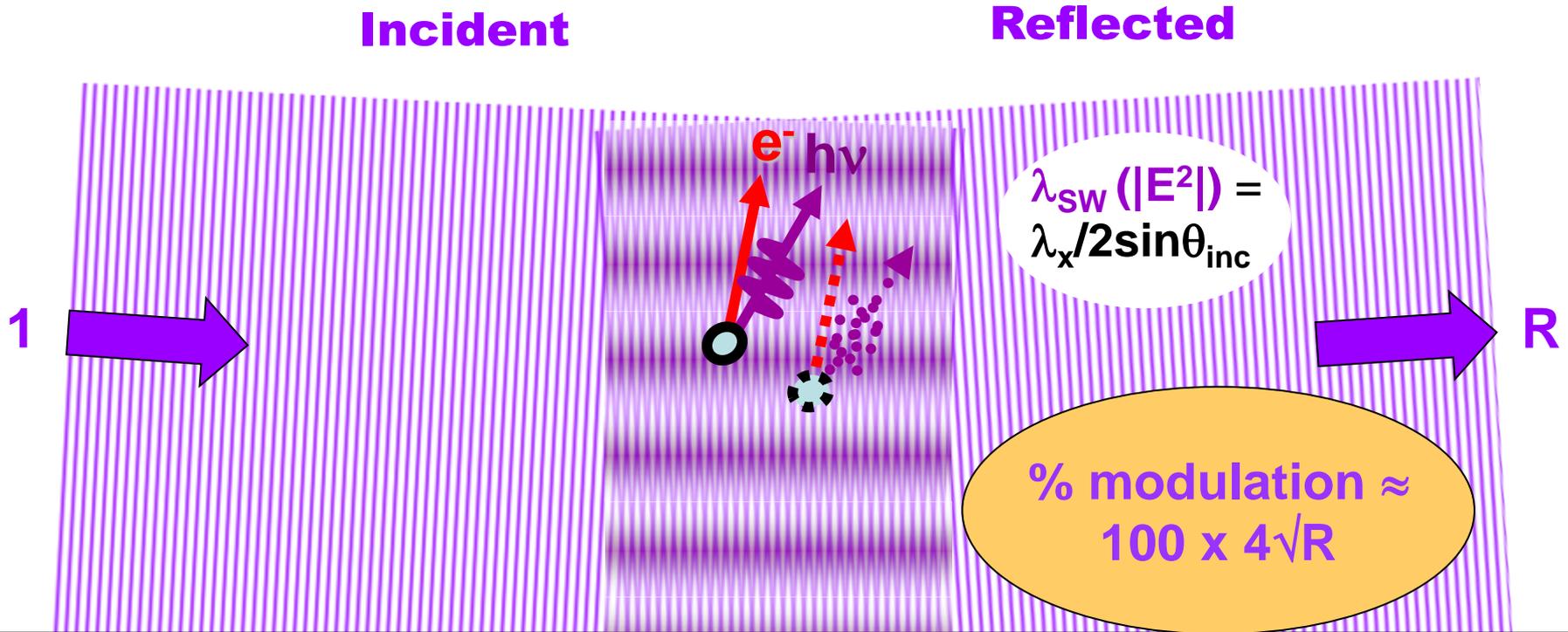
$$I_{sw} (|E^2|) \propto |\phi_{inc} + \phi_{refl}|^2$$

$$= |\phi_{inc}|^2 + \underbrace{\phi_{inc}\phi_{refl}^* + \phi_{inc}^*\phi_{refl}}_{\text{SW modulation}} + |\phi_{refl}|^2$$



Periodic atomic planes: $d_{ML} \leq \sim 5 \text{ \AA}$ -- Si(111) -- $\leq 3.74 \text{ \AA}$, $\text{LaB}_{66}(001)$ -- $\leq 23.52 \text{ \AA} \rightarrow 5.88 \text{ \AA}$, mica -- 10.0 \AA ; or **Synthetic multilayers:** $d_{ML} \approx \sim 20-40 \text{ \AA}$

Standing wave formation in reflection from a surface, or single-crystal Bragg planes⁺, or a multilayer mirror



- **Rocking curve:**

$$I(\theta_{inc}) \propto 1 + R(\theta_{inc}) + 2\sqrt{R(\theta_{inc})} f \cos[\varphi(\theta_{inc}) - 2\pi P]$$

- **Photon energy scan:**

$$I(h\nu) \propto 1 + R(h\nu) + 2\sqrt{R(h\nu)} f \cos[\varphi(h\nu) - 2\pi P]$$

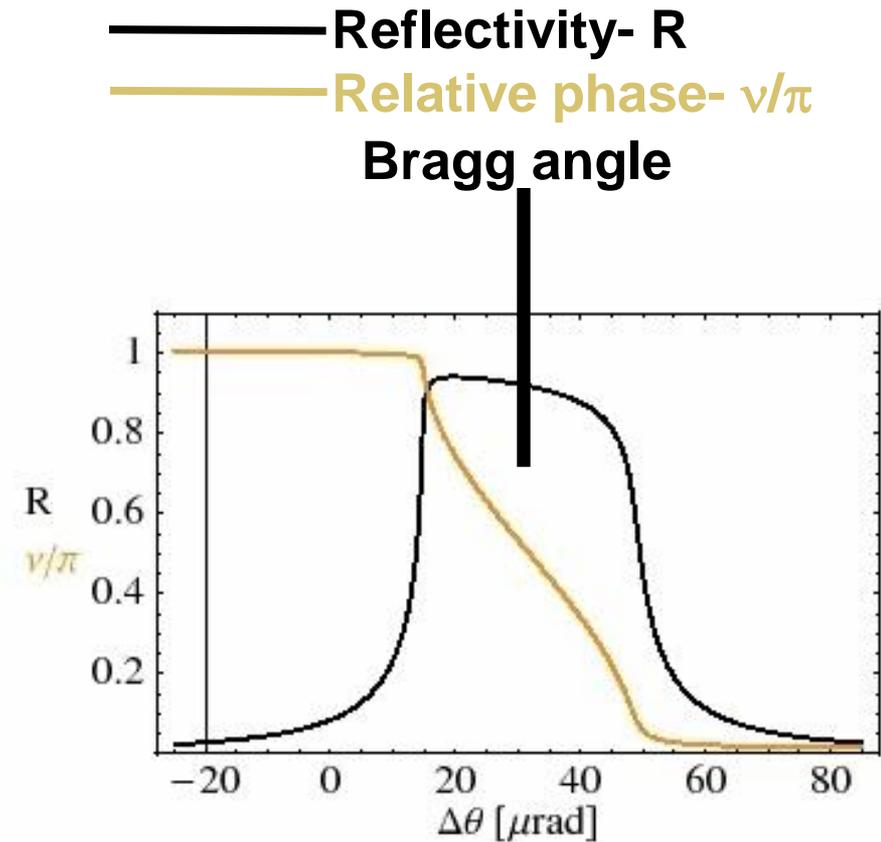
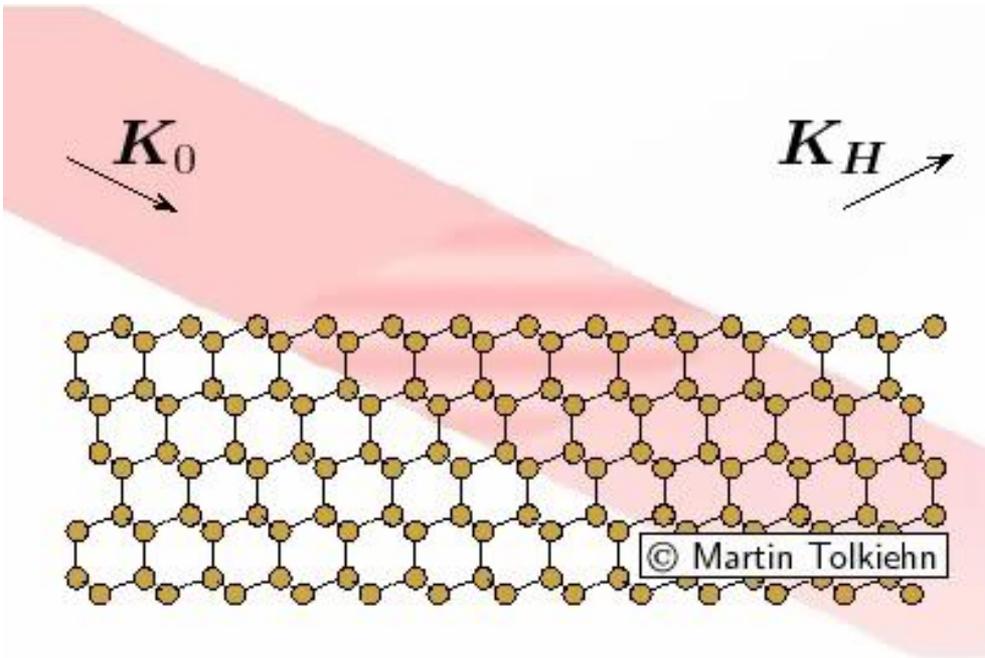
with: f = coherent fraction of atoms, P = phase of coherent-atom position

- **Phase scan with wedge-shaped sample (“Swedge” method):**

XMCD—Kim, Kortright, PRL 86, 1347 (2001)

⁺Standing waves via Bragg reflection of hard x-rays: Batterman, Phys. Rev A 133, 759 (1964)

Standing Wave Behavior During a Rocking Curve Scan

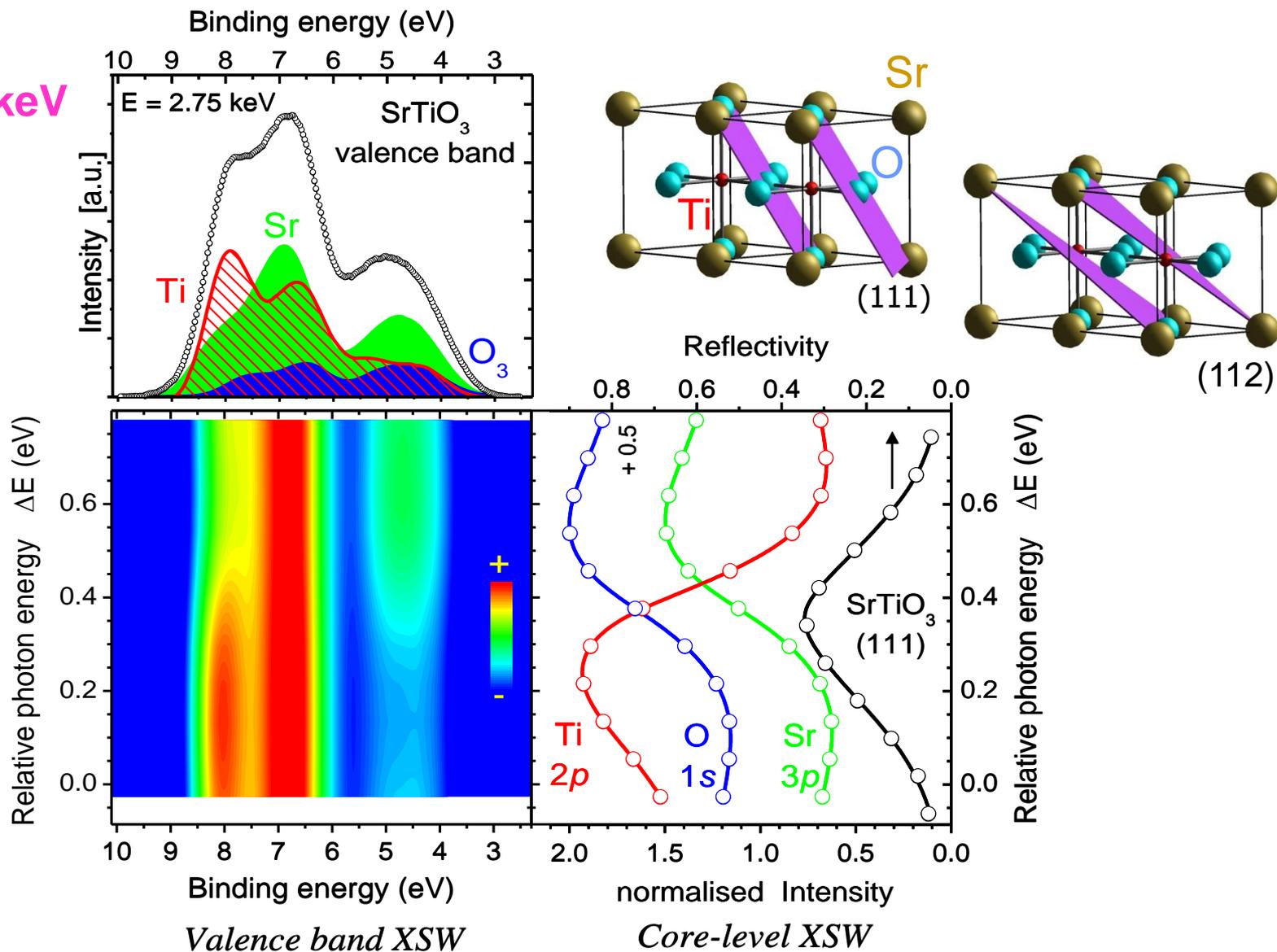


+Same general forms if **photon energy** is scanned

With thanks to Martin Tolkiehn, Dimitri Novikov, DESY

Site-specific valence electronic structure of SrTiO₃

$h\nu = 2.75 \text{ keV}$



Valence band XSW

Core-level XSW

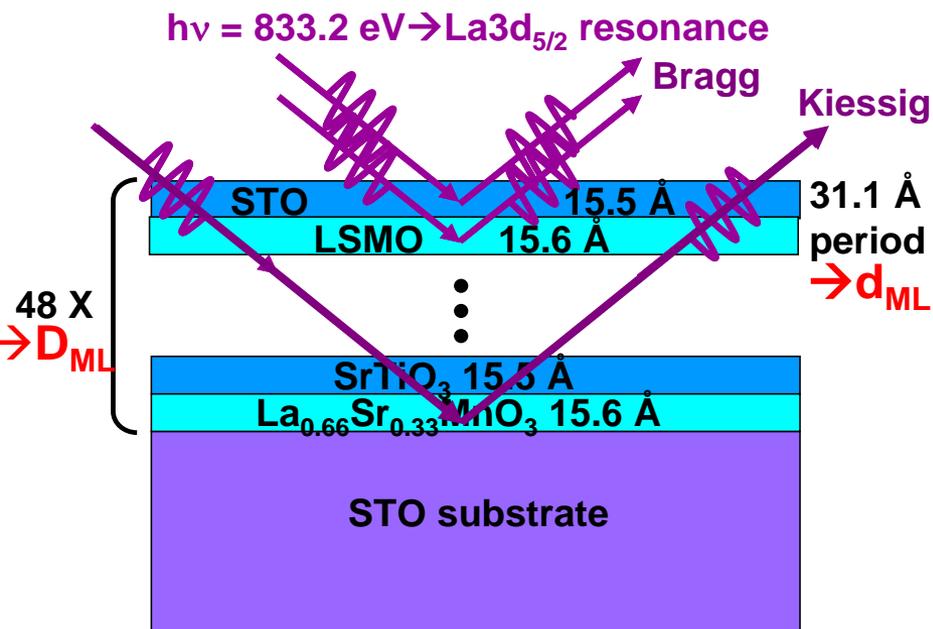
Lee et al., Solid State Communications 150 (2010) 553,

Talk by Materlik

Case study: Standing wave/rocking curve analysis of an epitaxial SrTiO₃/La_{0.67}Sr_{0.33}MnO₃ interface: Resonant soft x-ray excitation

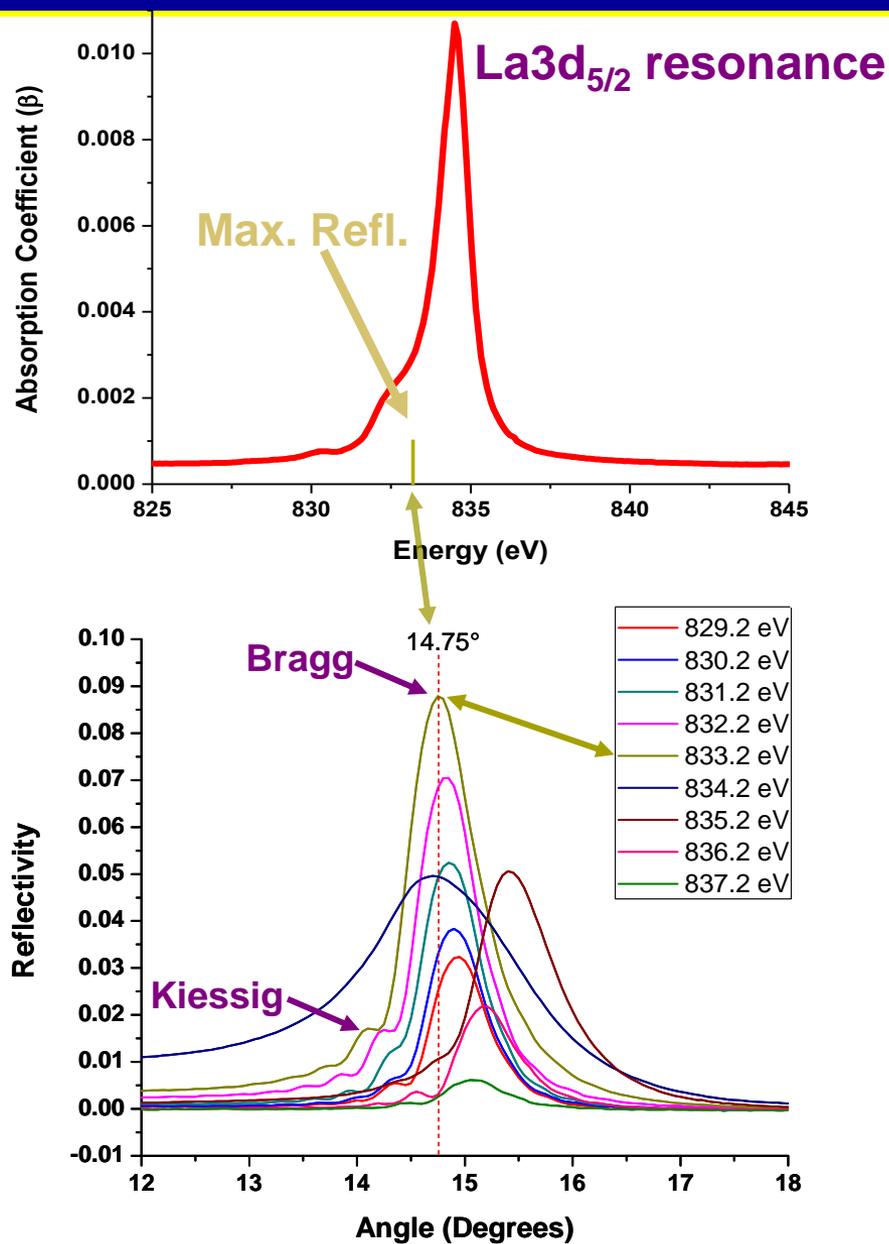


The Advanced Light Source

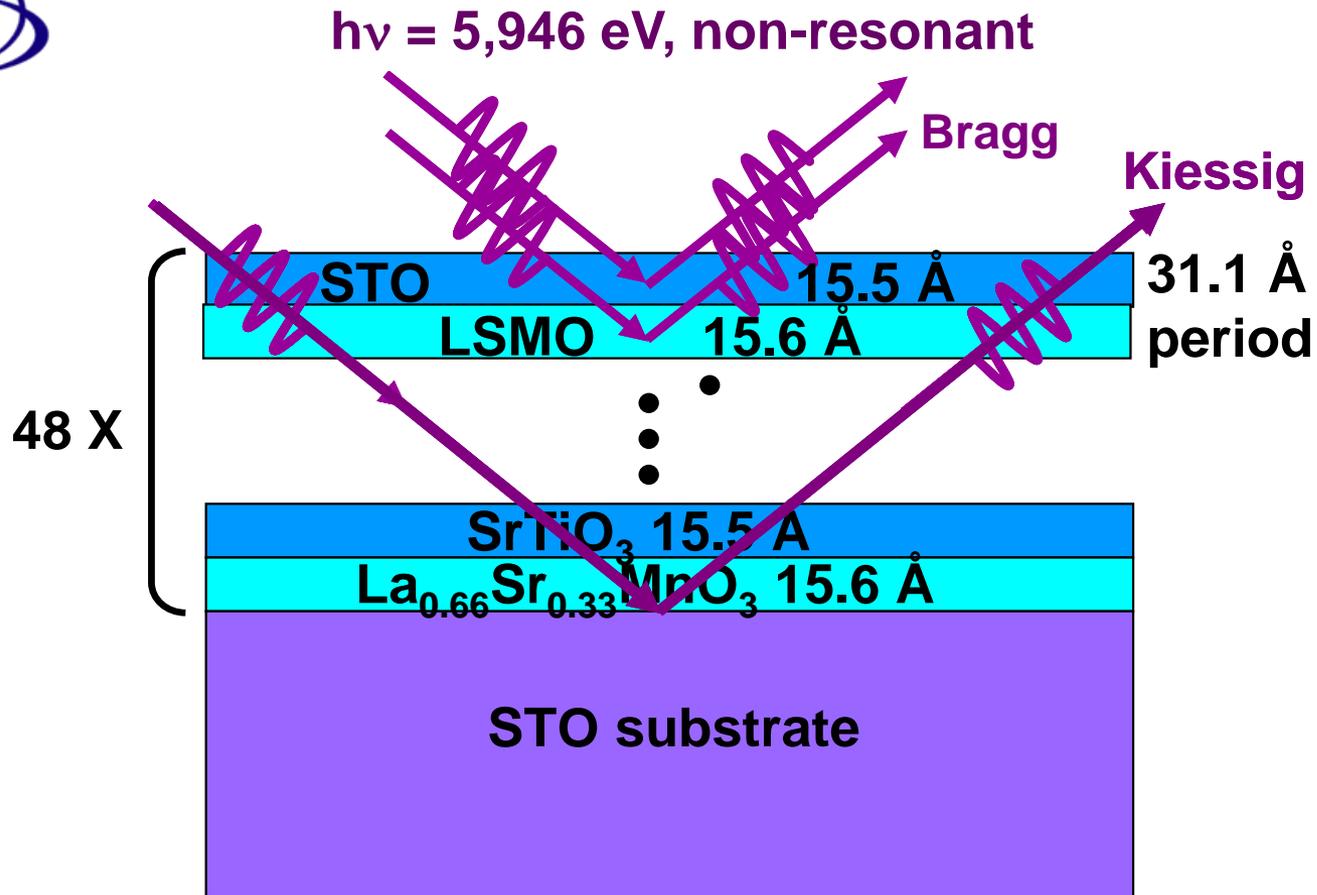


$$\lambda_x = 2d_{ML} \sin \theta_{\text{Bragg}}$$

Gray et al., Phys. Rev. B 82, 205116 (2011)
Samples: Ramesh, Huijben



Standing wave/rocking curve analysis of an epitaxial $\text{SrTiO}_3/\text{La}_{0.67}\text{Sr}_{0.33}\text{MnO}_3$ interface: hard x-ray excitation



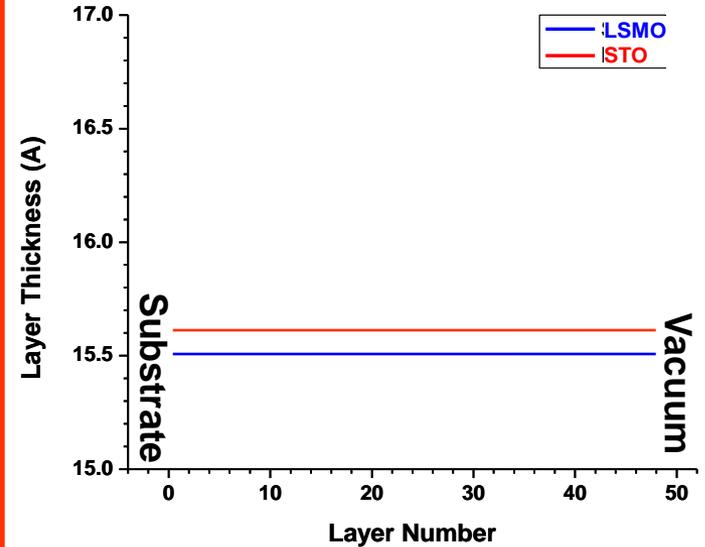
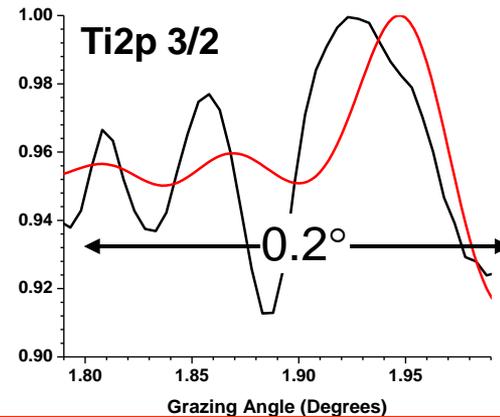
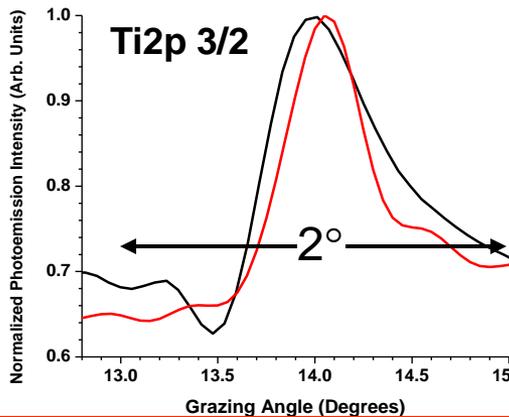
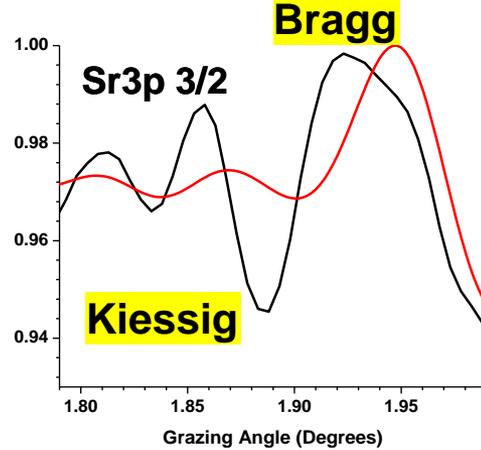
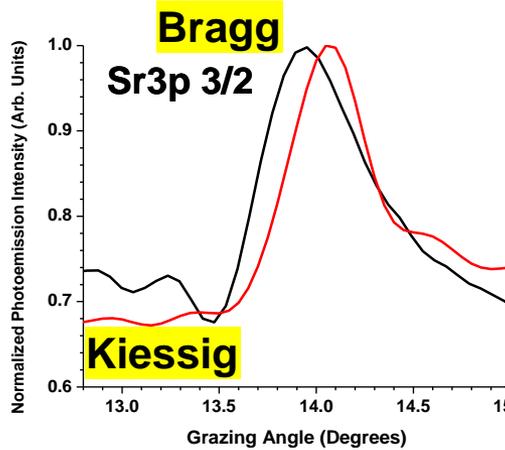
SrTiO₃/La_{0.67}Sr_{0.33}MnO₃ Multilayer Analysis of Rocking Curves

Expt.
Calc.

Ideal Bilayer
Thickness
Gradient Profile

hν = 833.2 eV

hν = 5956.4 eV



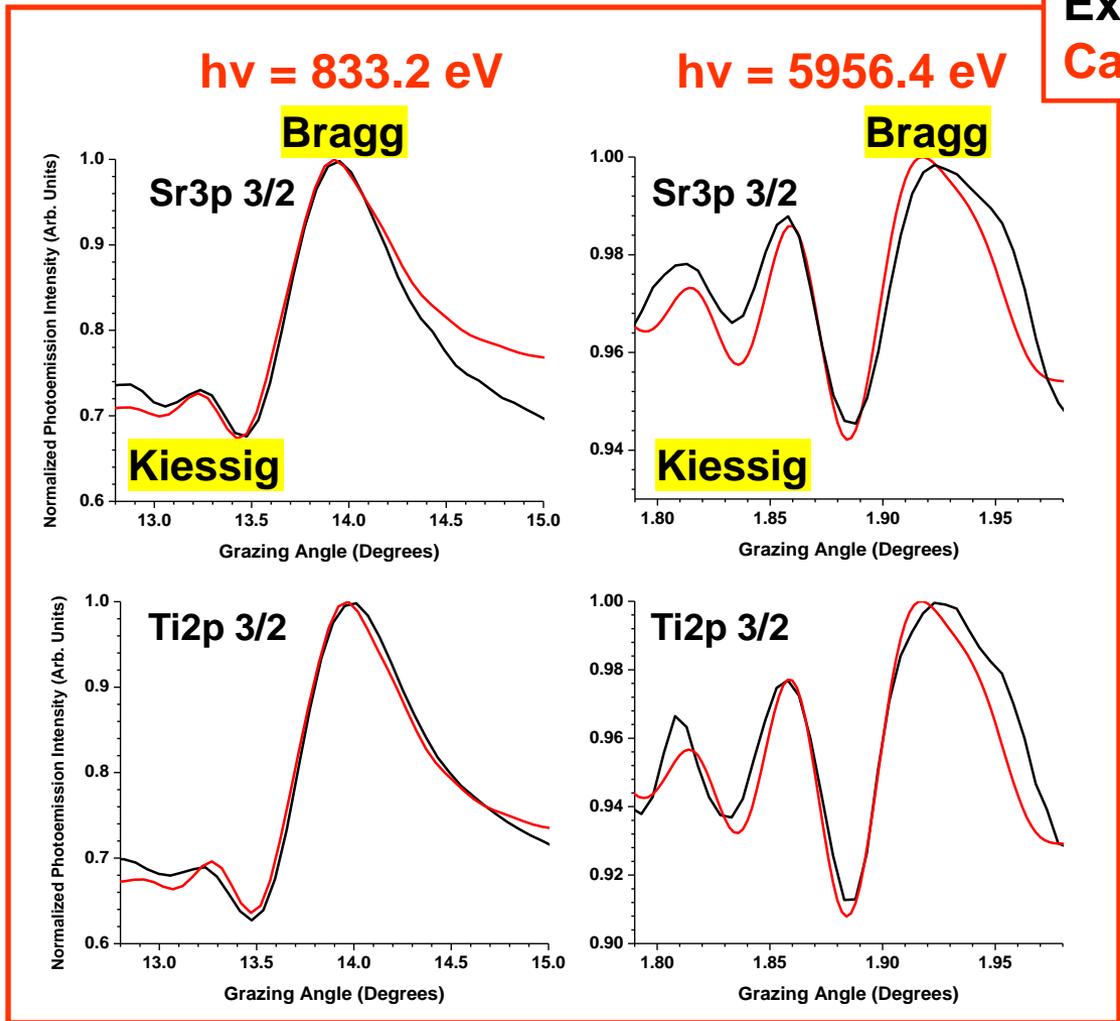
- Relative amplitude of the predicted Kiessig fringes does not agree with experiment
- Strong Kiessig fringes predicted on both sides of the rocking curves, esp. @ 5.9 keV



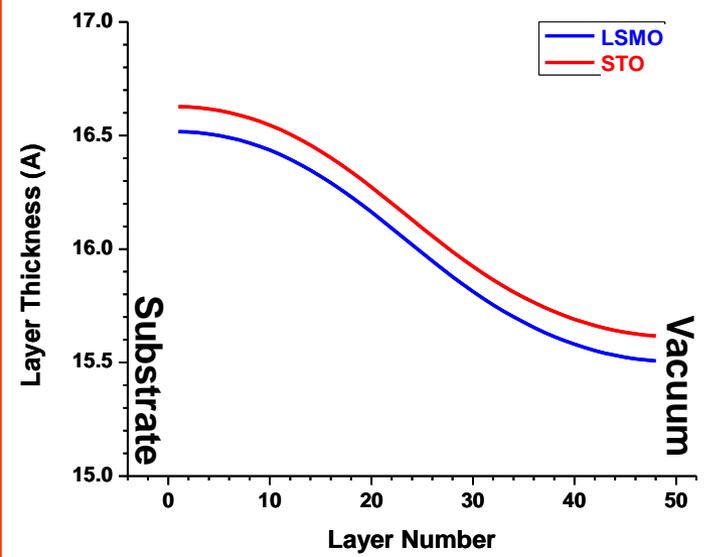
SrTiO₃/La_{0.67}Sr_{0.33}MnO₃ Multilayer

Analysis of Rocking Curves

Exp.
Calc.



Bilayer Thickness Gradient Profile



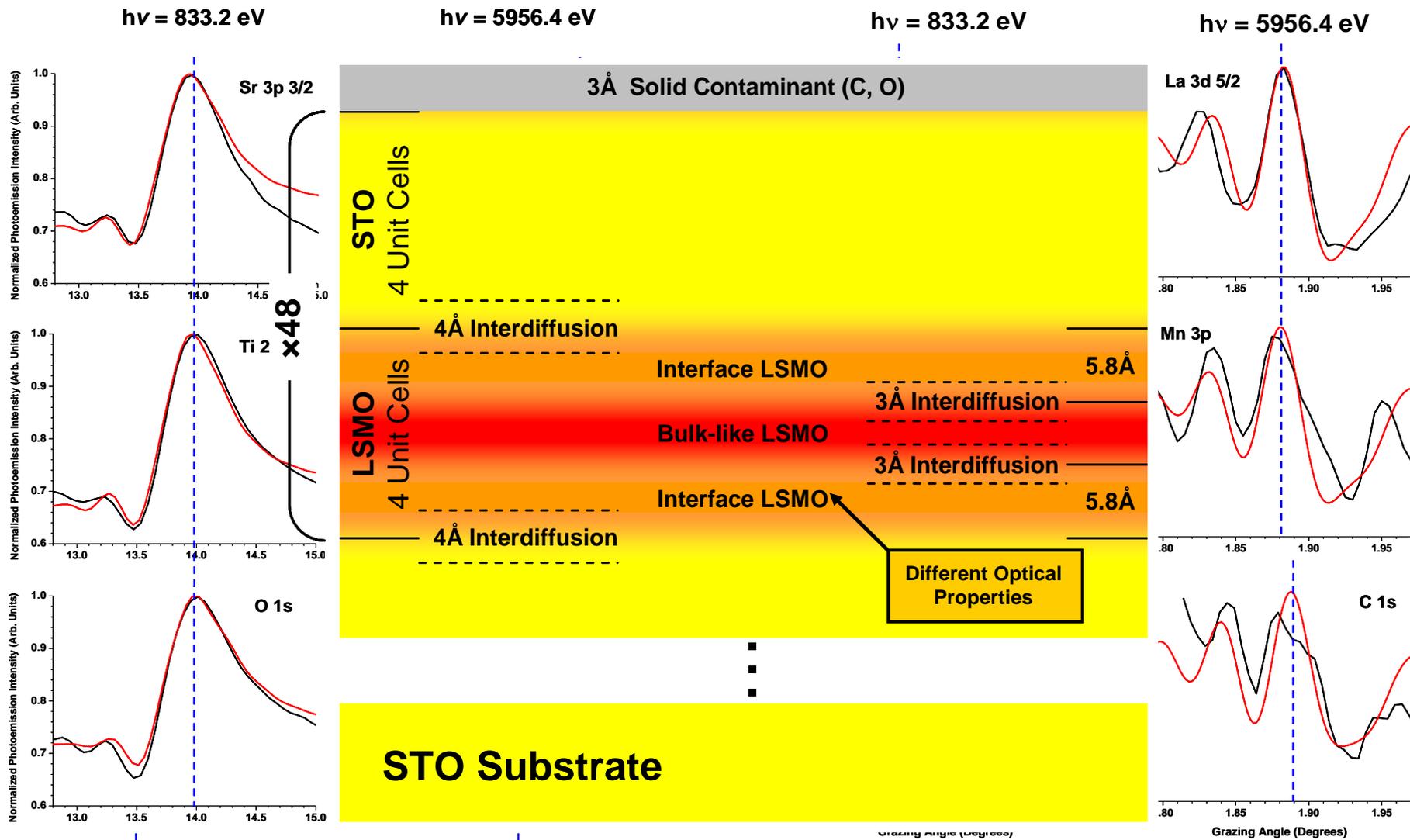
→ Average multilayer d_{ML} changes by about $-2 \text{ \AA} \approx -6\%$ or half a unit cell from top to bottom



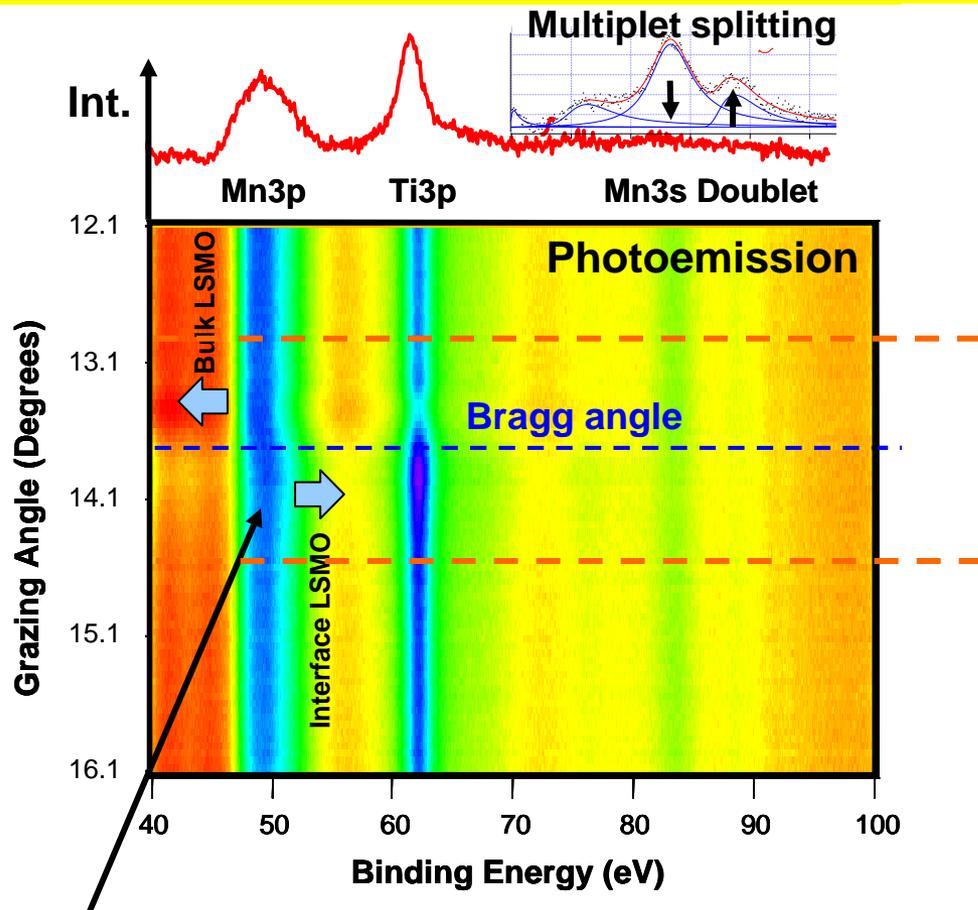
BEST FIT



Fitting of Rocking Curves—All Elements Present, Soft and Hard X-rays

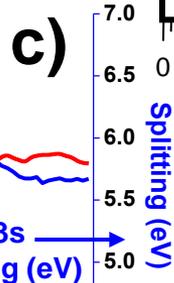
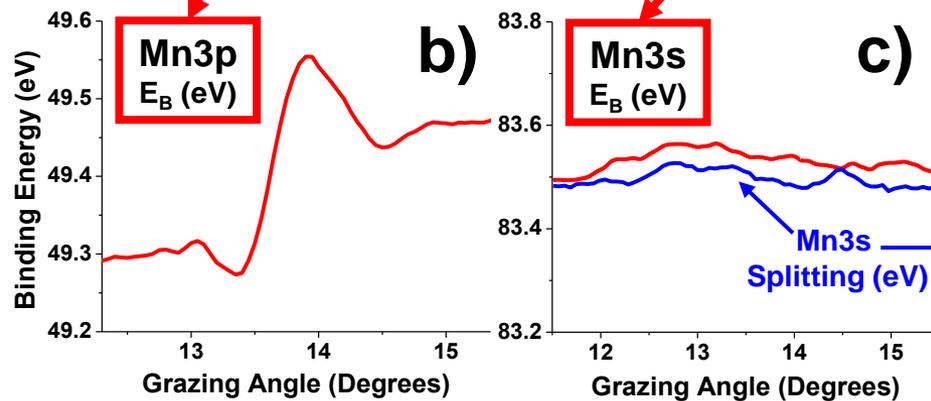
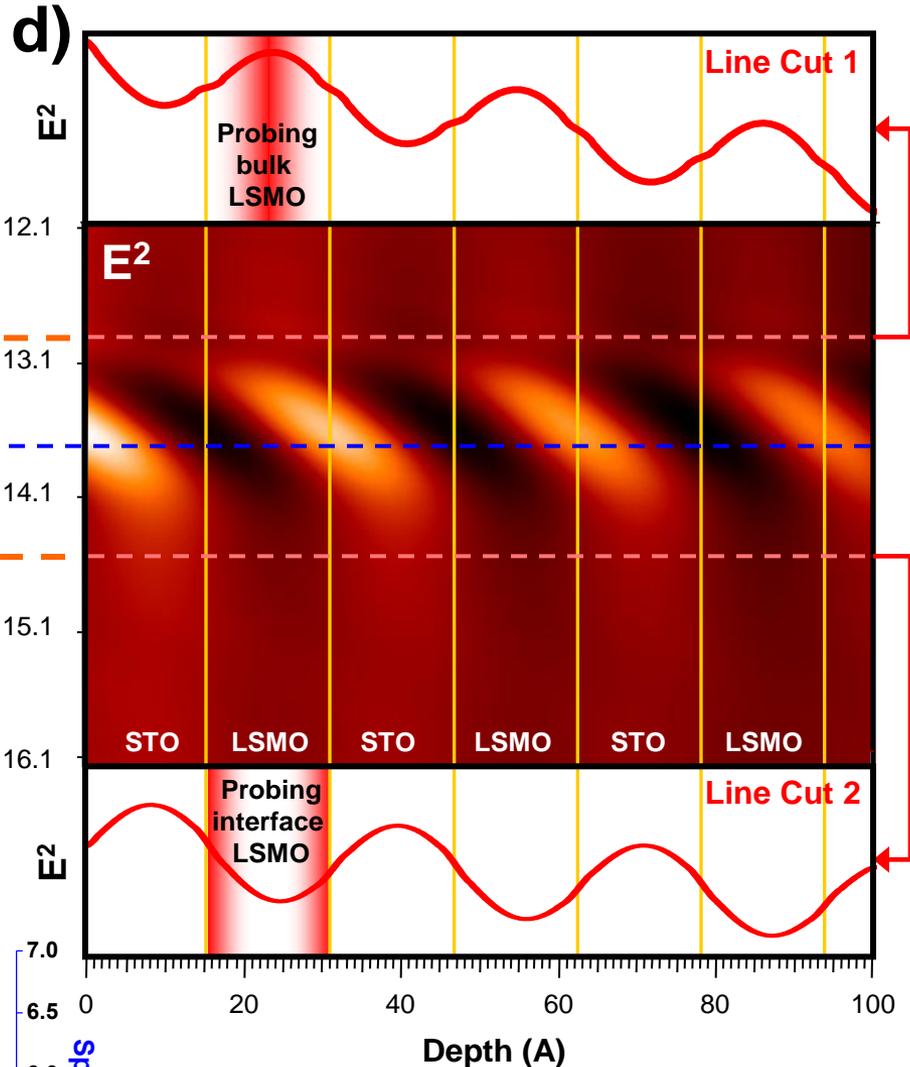
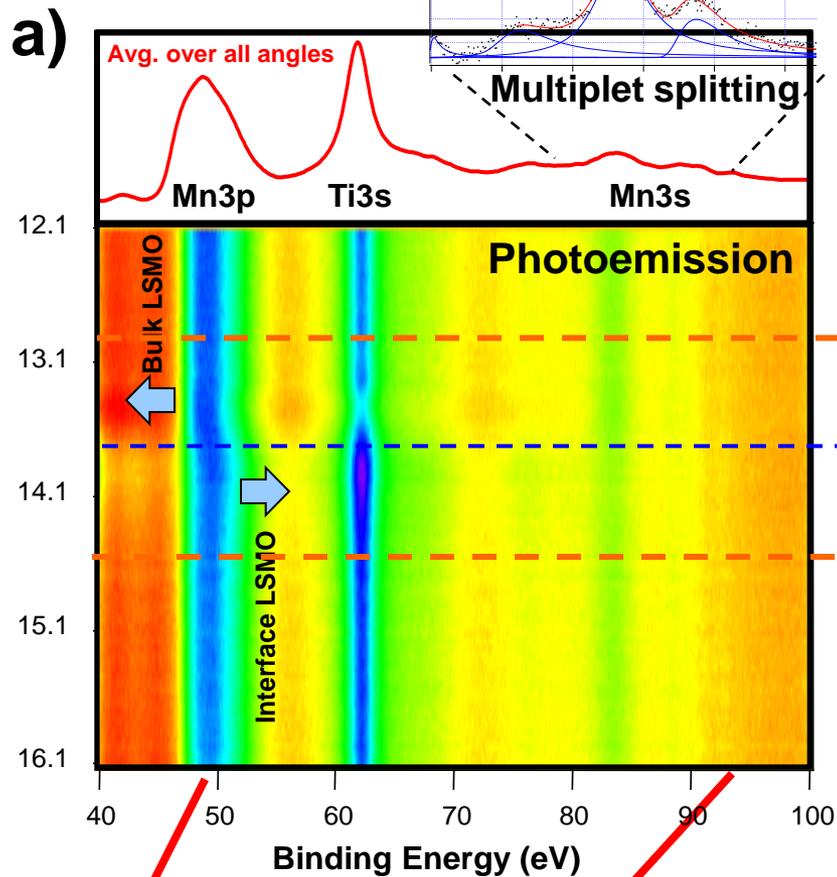


STO/LSMO-Resonant soft x-ray standing wave/rocking curves at 833 eV: core photoelectron peaks compared to calculated standing-wave field



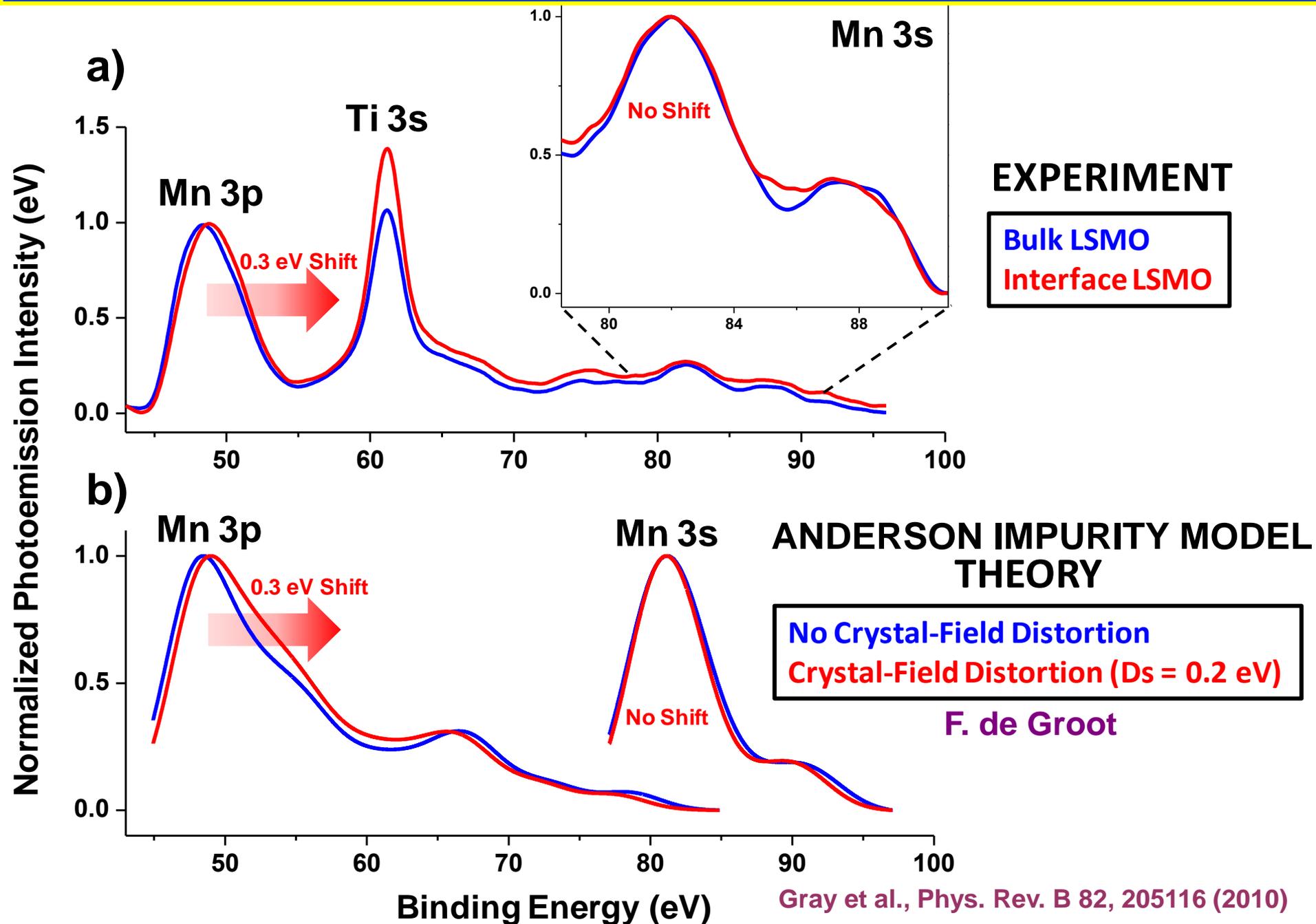
- Clear chemical/final-state shift at interface seen in Mn 3p
- No change in Mn 3s
- No change in Ti 3p—near surface



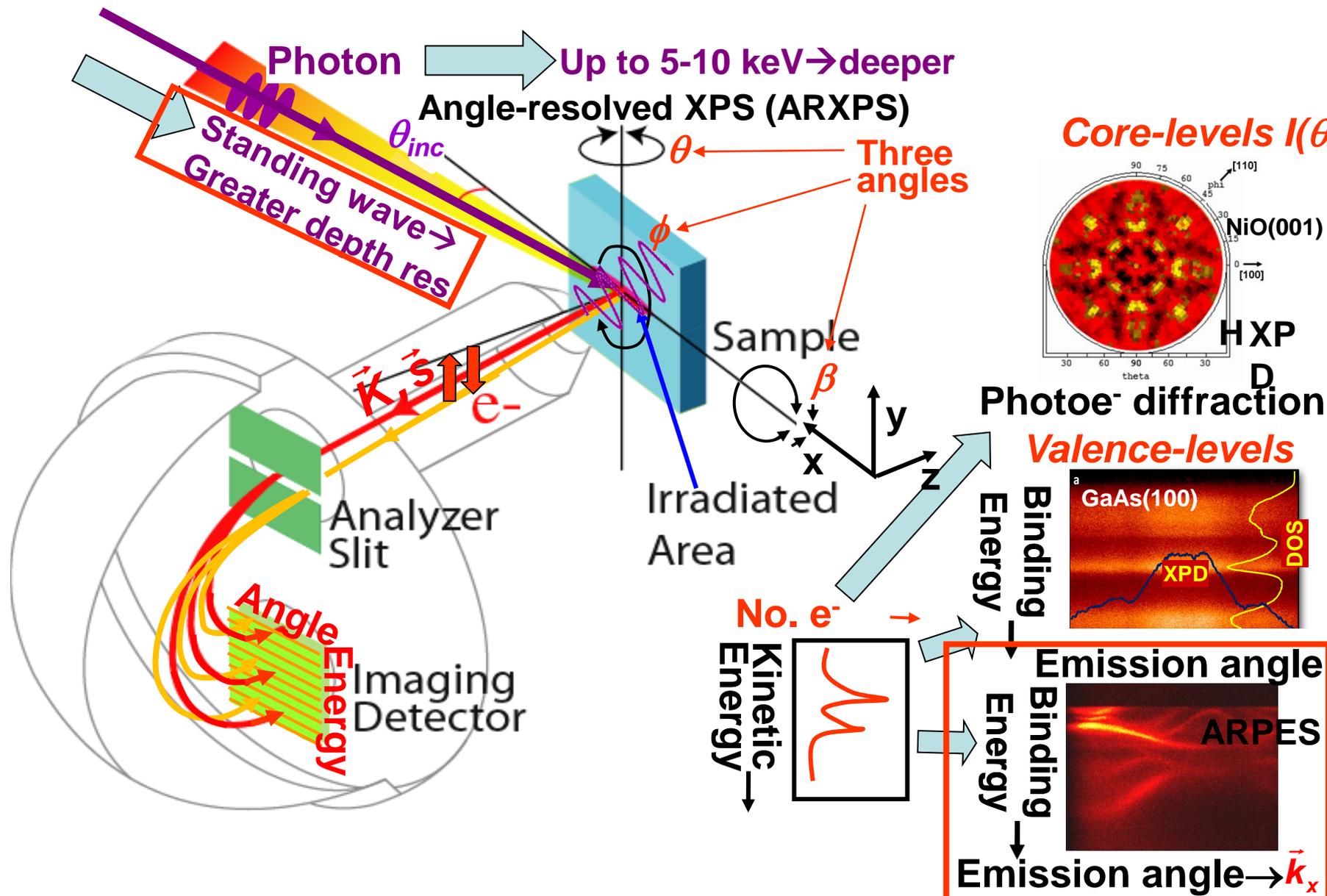


- **Clear shift at interface in Mn 3p**
- **No change in Mn 3s binding energy or multiplet splitting**

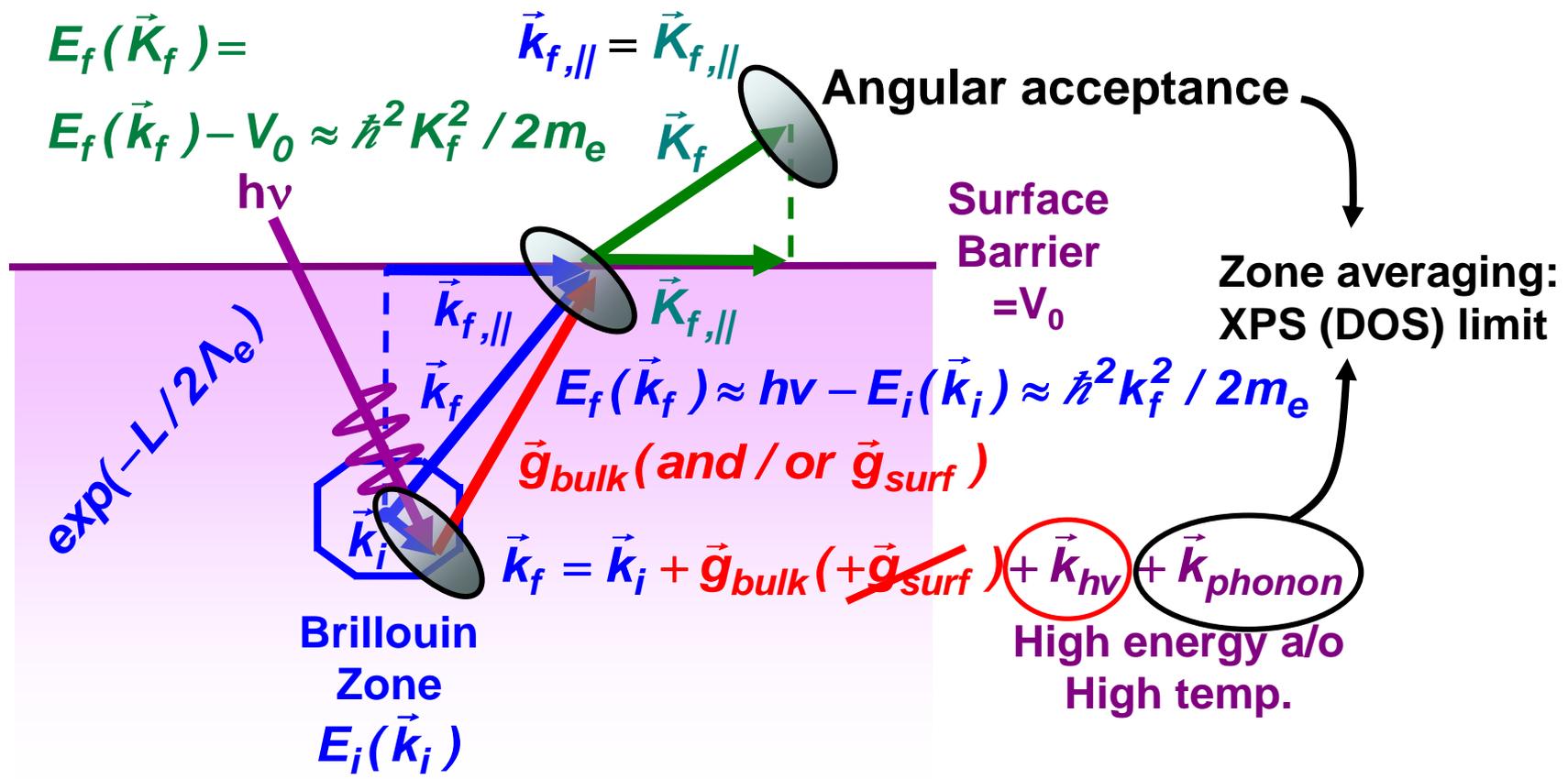
STO/LSMO-Explaining the Difference Between Mn 3p and Mn 3s behavior



X-ray photoemission: some key elements



ARPES—How high can we go in energy and temperature?



Fraction DTs \approx Debye-Waller factor = $W(T) \approx \exp[-(k^f)^2 \langle u^2(T) \rangle]$
 $\approx \exp[-C_1 (k^f)^2 T / (m\Theta_D^2)] \approx \exp(-C_2 E_{kin} T)$

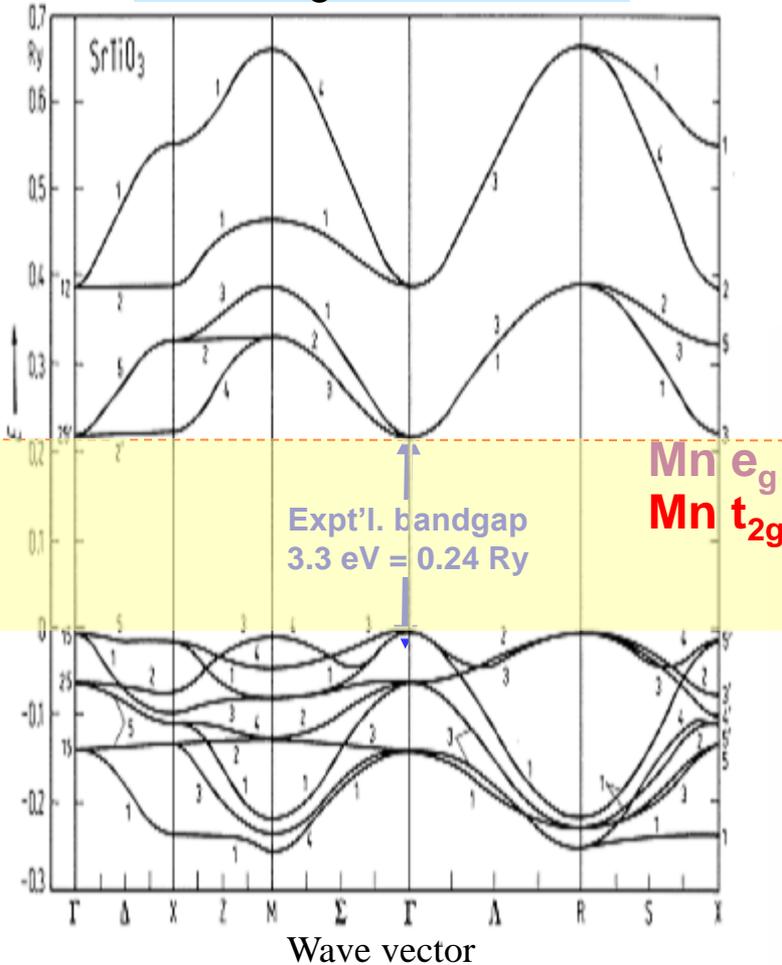
ARPES \rightarrow bands, quasiparticles
 (Low $h\nu$, Low T , High angul. Res.)

XPS \rightarrow DOS
 (High $h\nu$, High T , Low angul. Res.)

Shevchik, Phys. Rev. B 16, 3428 (1977)
 Hussain....CF, Phys. Rev. B 34 (1986) 5226

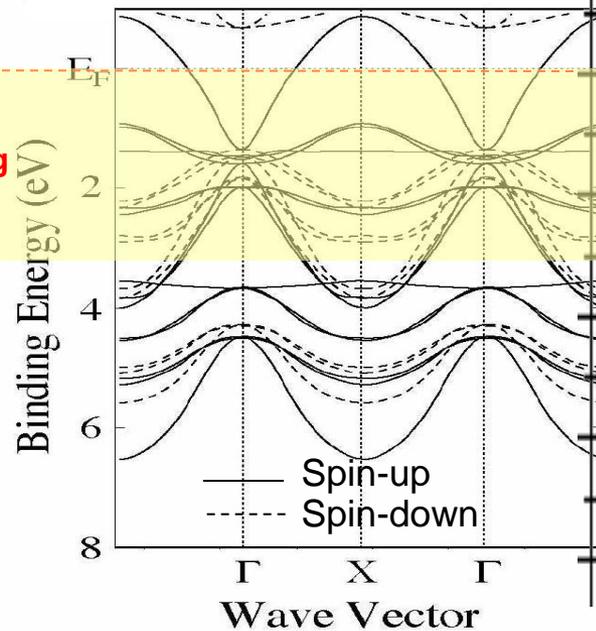
SrTiO₃ and La_{0.67}Sr_{0.33}MnO₃ band structures and DOS

SrTiO₃-insulator



Mattheiss, PRB 6, 4718 (1972)

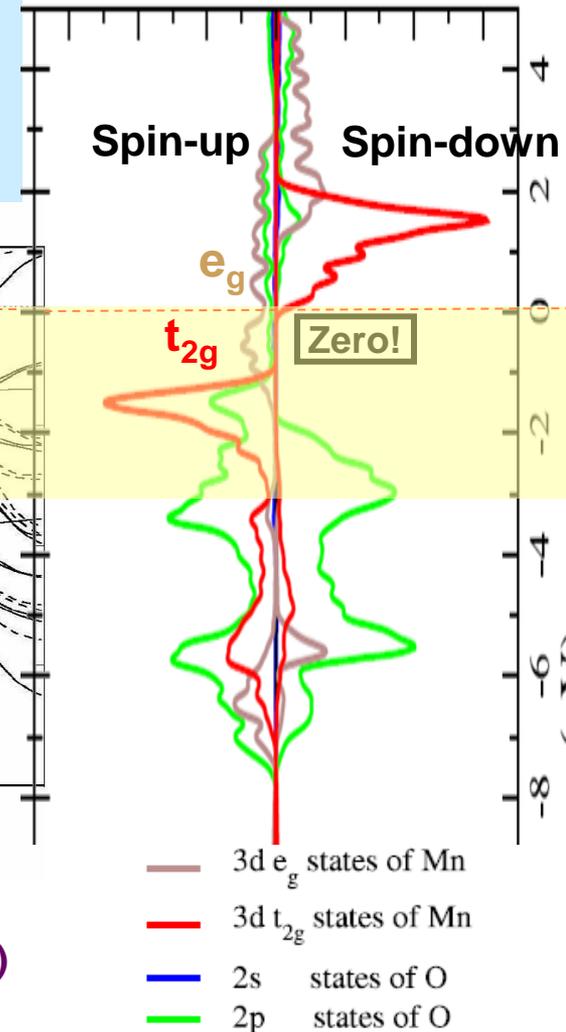
La_{0.67}Sr_{0.33}MnO₃- Half-Metallic Ferromagnetic metal



Chikamatsu et al.,
PRB 73, 195105 (2006)

Zheng, Binggeli, J. Phys.
Cond. Matt. 21, 115602 (2009)

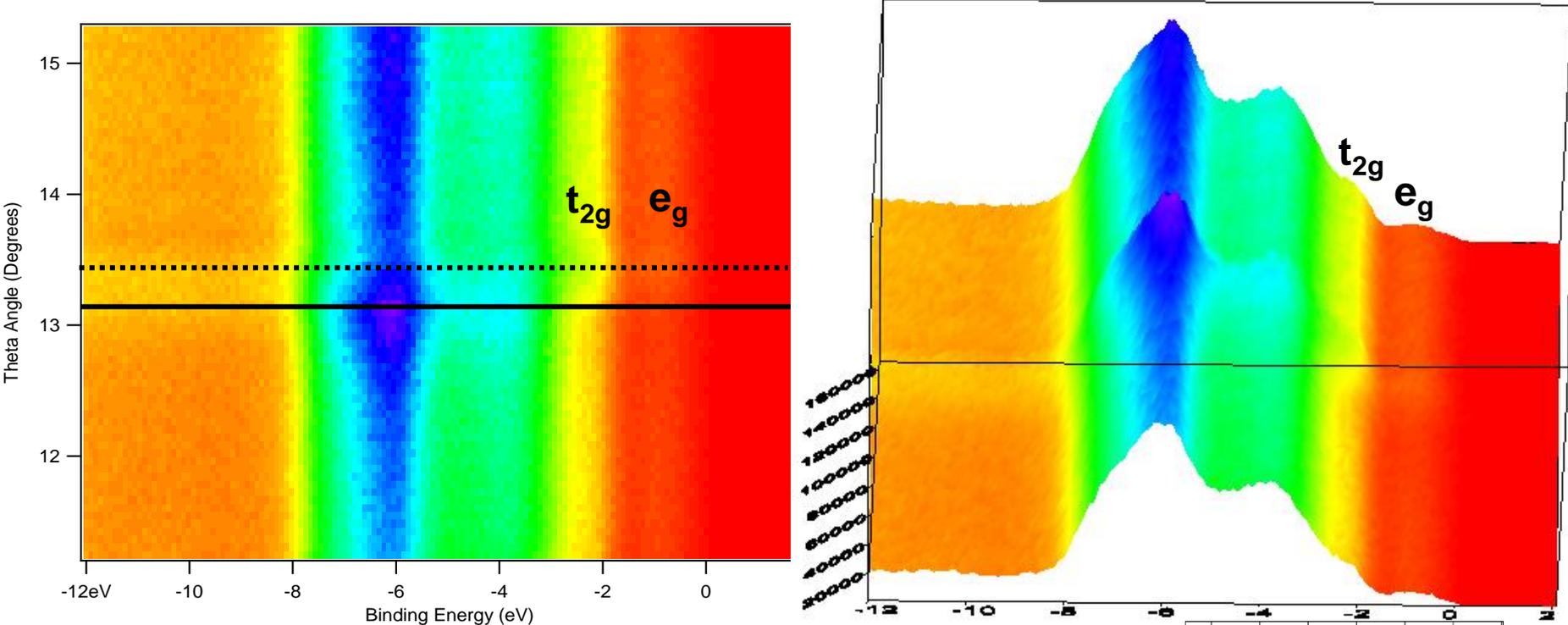
Projected DOSs



- 3d e_g states of Mn
- 3d t_{2g} states of Mn
- 2s states of O
- 2p states of O

STO/LSMO-Standing wave/rocking curves of valence region: 833 eV, 300K

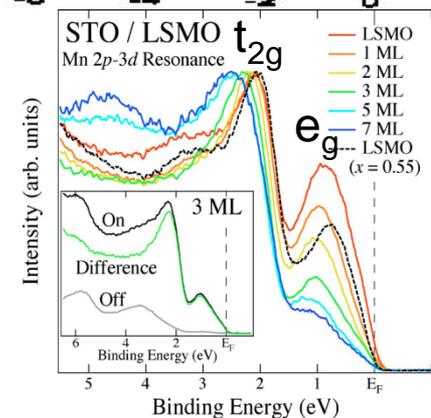
Debye-Waller $\approx 0.013 \rightarrow$ DOS limit



A. Gray et al., Phys. Rev. B 82, 205116 (2010)

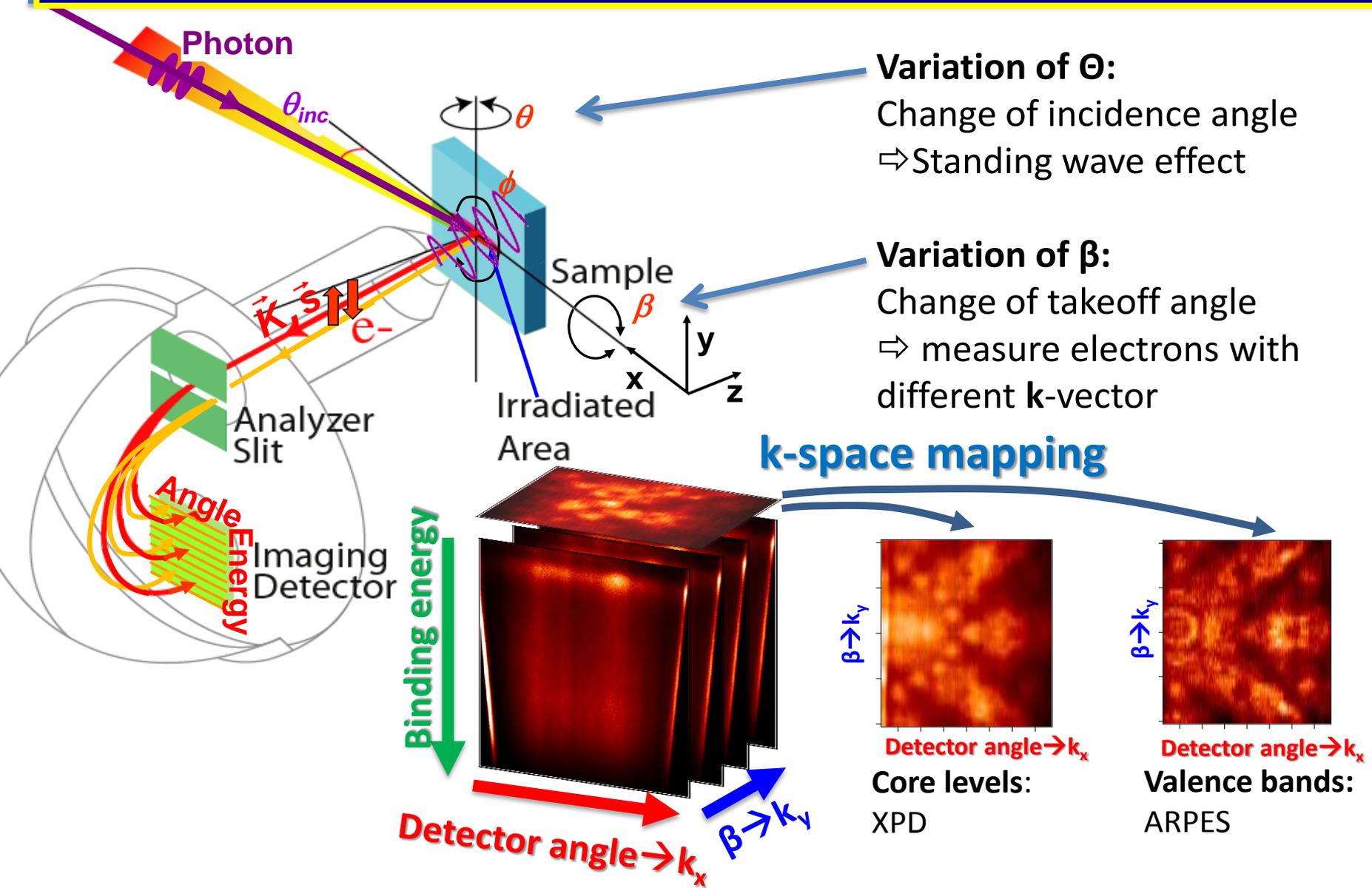


The Advanced
Light Source



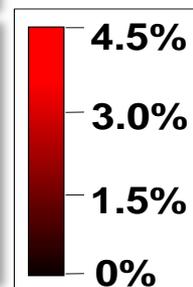
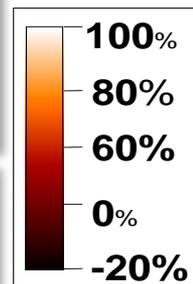
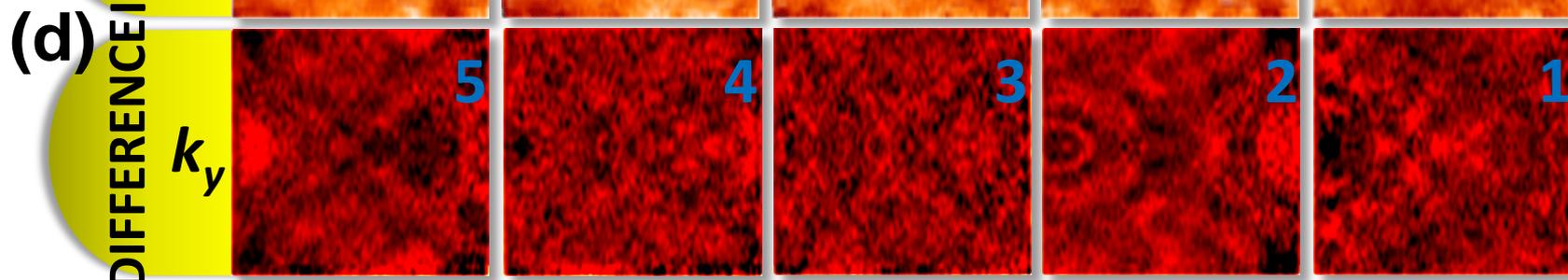
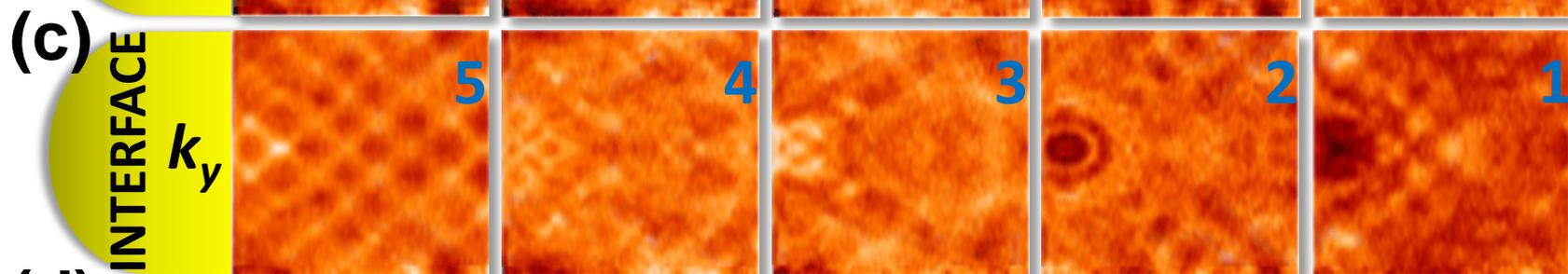
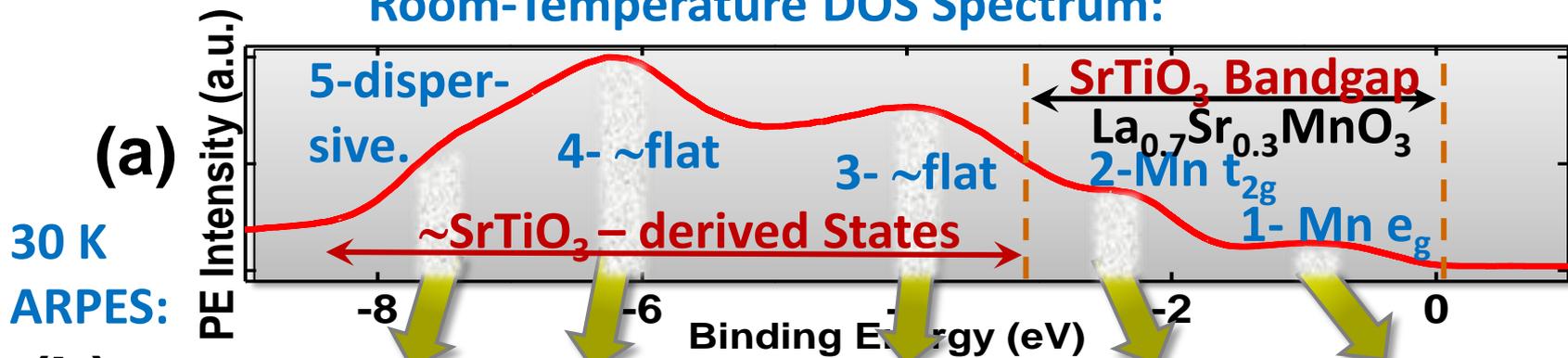
Prior resonant PS: Fujimori et al., J.A.P 99, 08S903 (2006)

Standing-wave angle-resolved photoemission



STO/LSMO Depth-resolved ARPES: $h\nu=833$ eV, 20K

Room-Temperature DOS Spectrum:



k_x

k_x

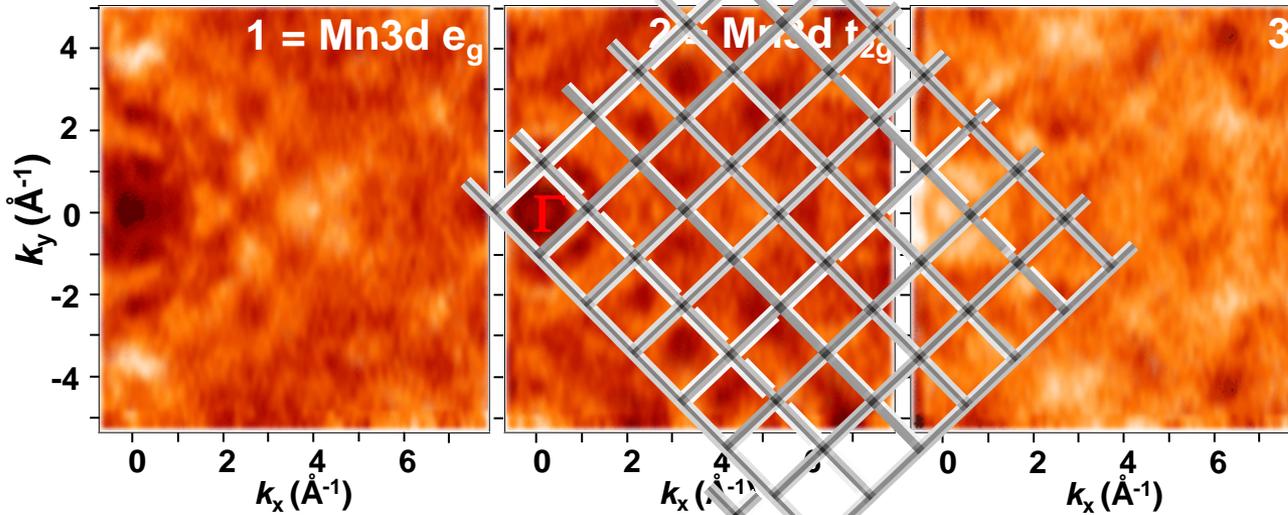
k_x

k_x

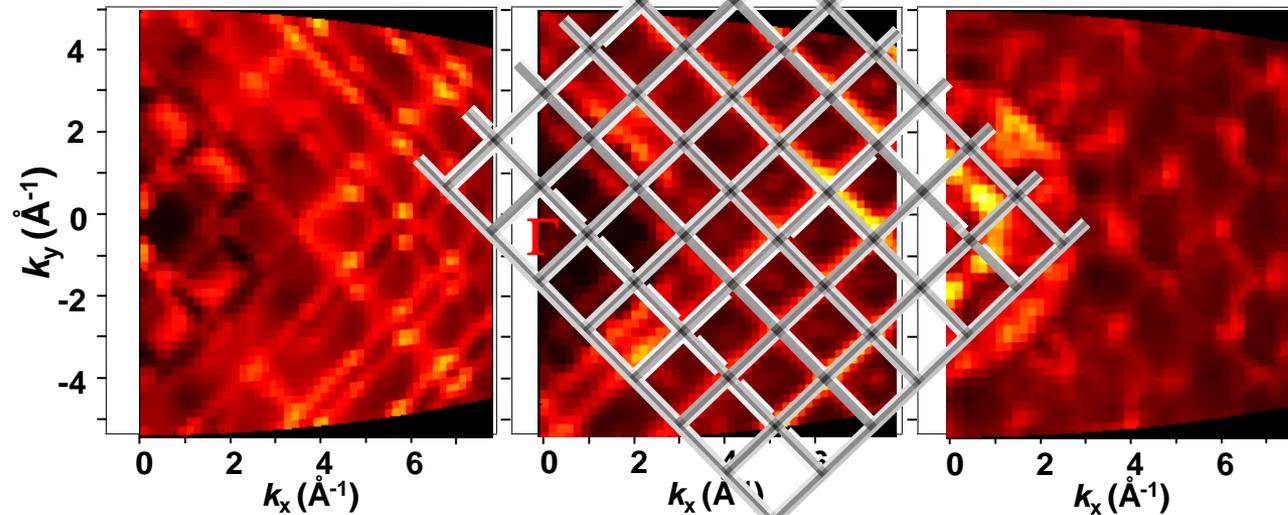
k_x

First comparison to one-step photoemission theory for LSMO

Experiment: Bulk-Sensitive Scans



One-Step Theory: Bulk LSMO



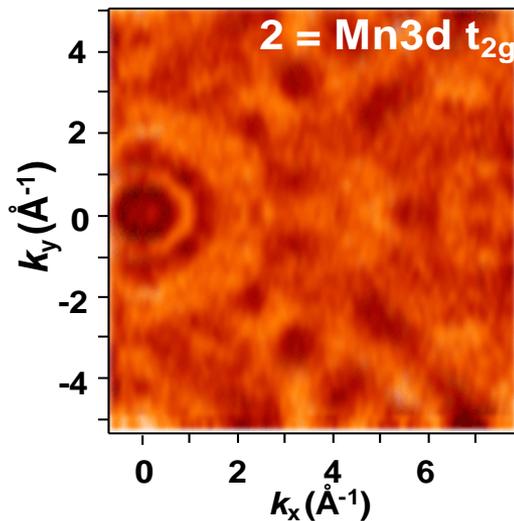
Theory from:
H. Ebert, J. Minar et al.,
Munich: one-step time-
reversed LEED, with
matrix elements

Approx. BZ map

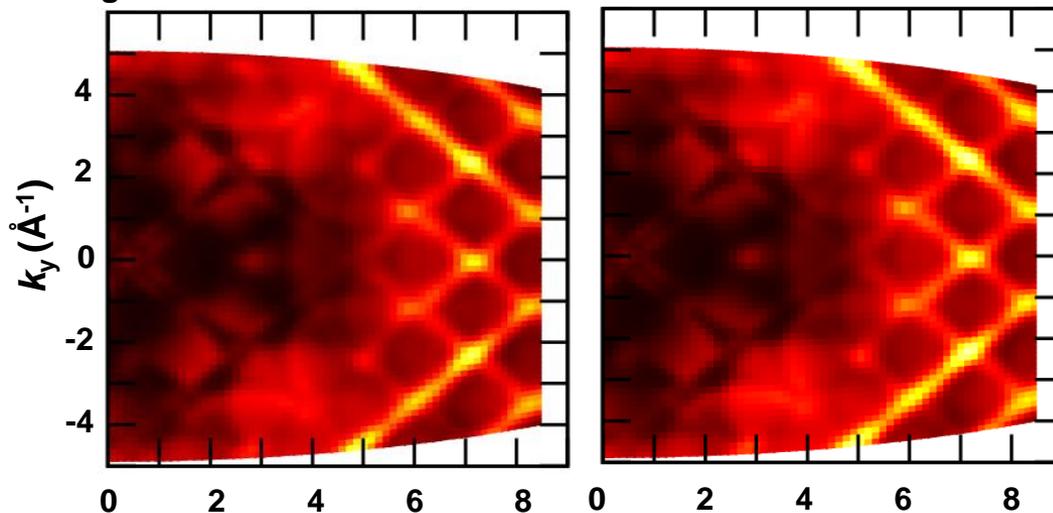
A. Gray, J. Minar et al., TBP

First comparison to one-step photoemission theory for STO/LSMO

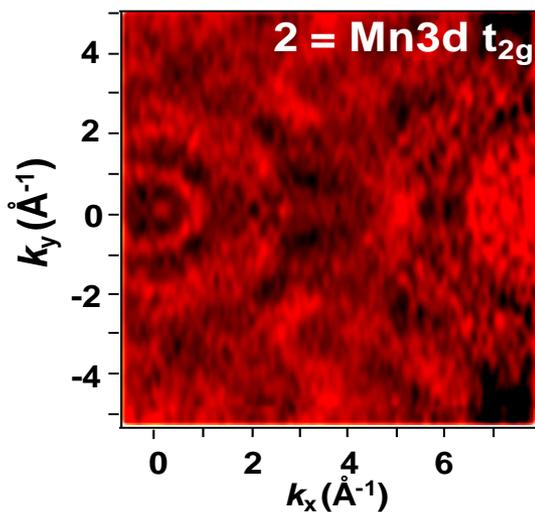
Experiment:
 t_{2g} : Bulk LSMO



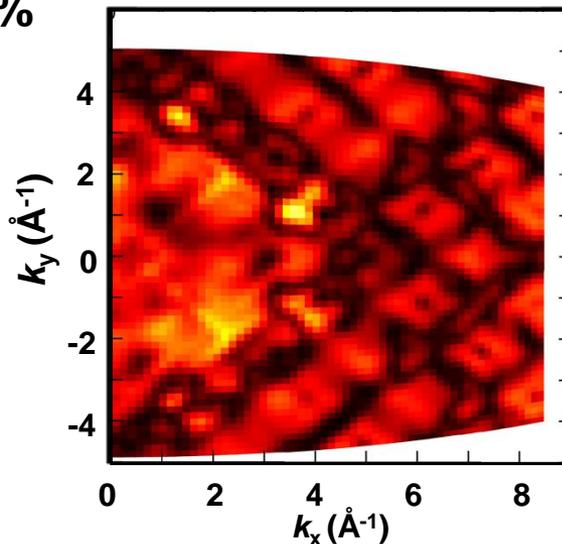
One-Step Theory, LDA+U, with SW E-field profile:
 t_{2g} : Bulk LSMO Interface LSMO



Bulk – Interface- ~5%

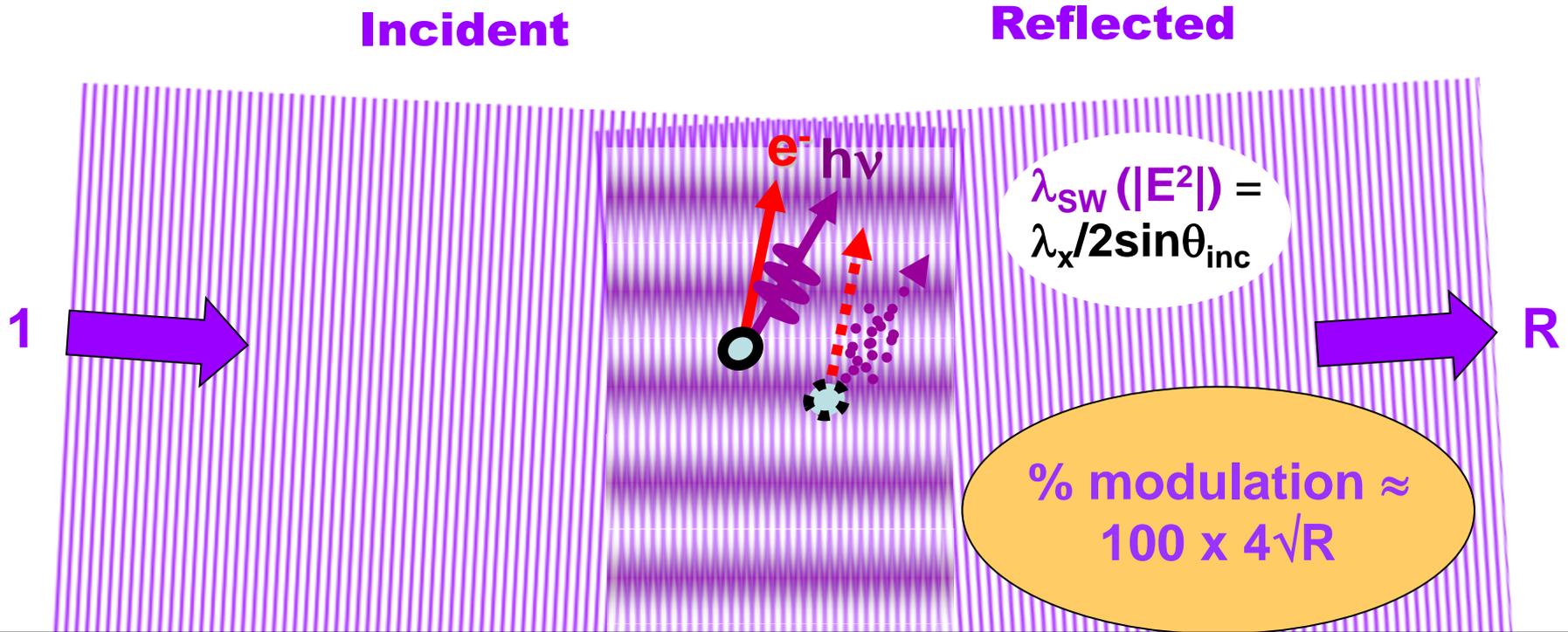


Bulk – Interface- ~8%



Calculations with
more complex
Interface
relaxations in
progress-
Minar, Braun,
Ebert

Standing wave formation in reflection from a surface, or single-crystal Bragg planes⁺, or a multilayer mirror



1. Rocking curve:

$$I(\theta_{inc}) \propto 1 + R(\theta_{inc}) + 2\sqrt{R(\theta_{inc})} f \cos[\varphi(\theta_{inc}) - 2\pi P]$$

2. Photon energy scan:

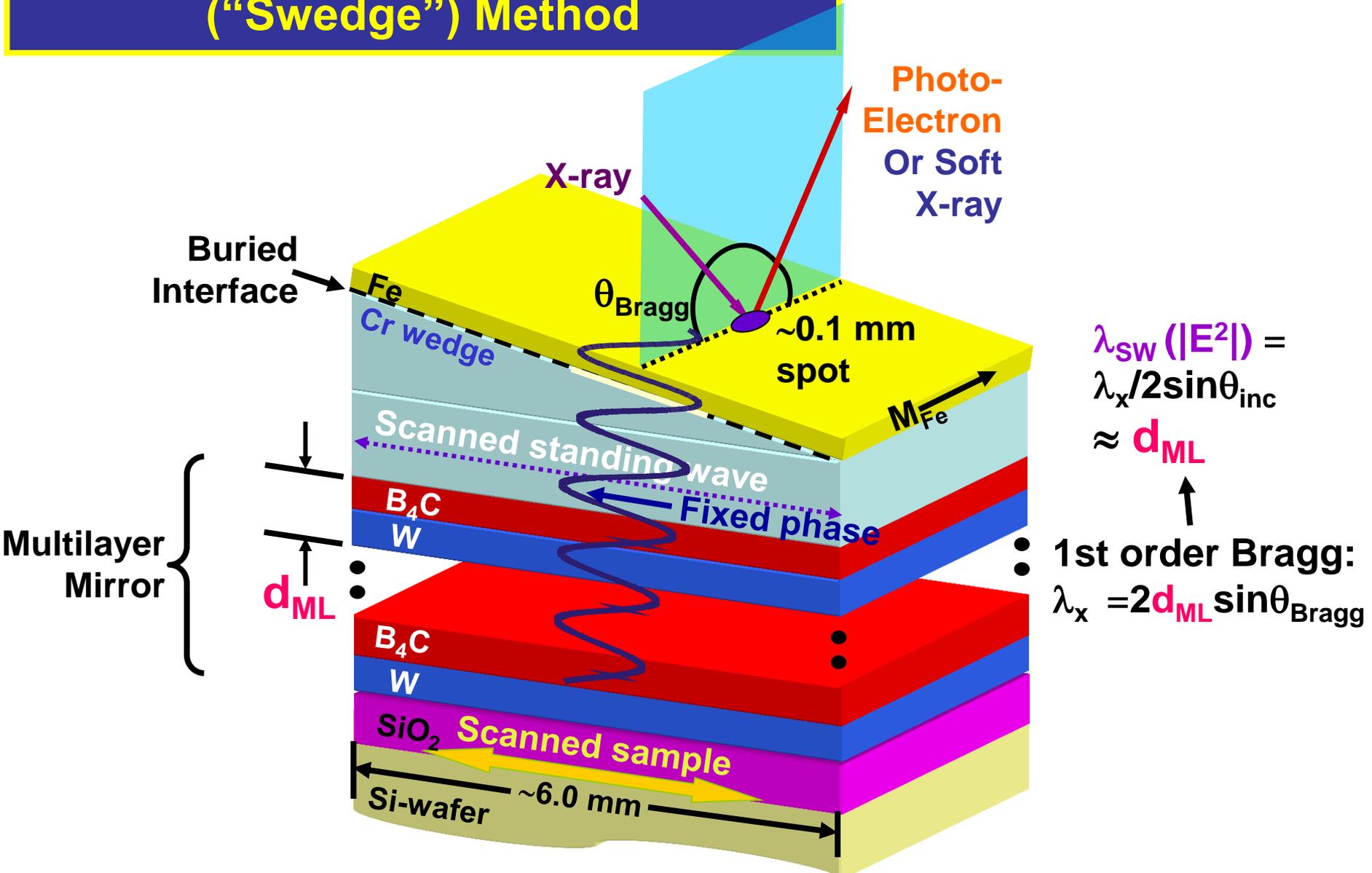
$$I(h\nu) \propto 1 + R(h\nu) + 2\sqrt{R(h\nu)} f \cos[\varphi(h\nu) - 2\pi P]$$

with: f = coherent fraction of atoms, P = phase of coherent-atom position

3. Phase scan with wedge-shaped sample ("Swedge" method):

Fe/Cr GMR—Yang, J. Phys. Cond. Matt. 14, L406 (2002); ; Co/CoFeB/Al₂O₃ TMR—Yang, J. Phys.: Cond. Matt. 18, L259 (2006); Fe/MgO TMR—Döring, Yang, Balke—PRB B 83, 165444 (2011) & 84, 184410 (2011)

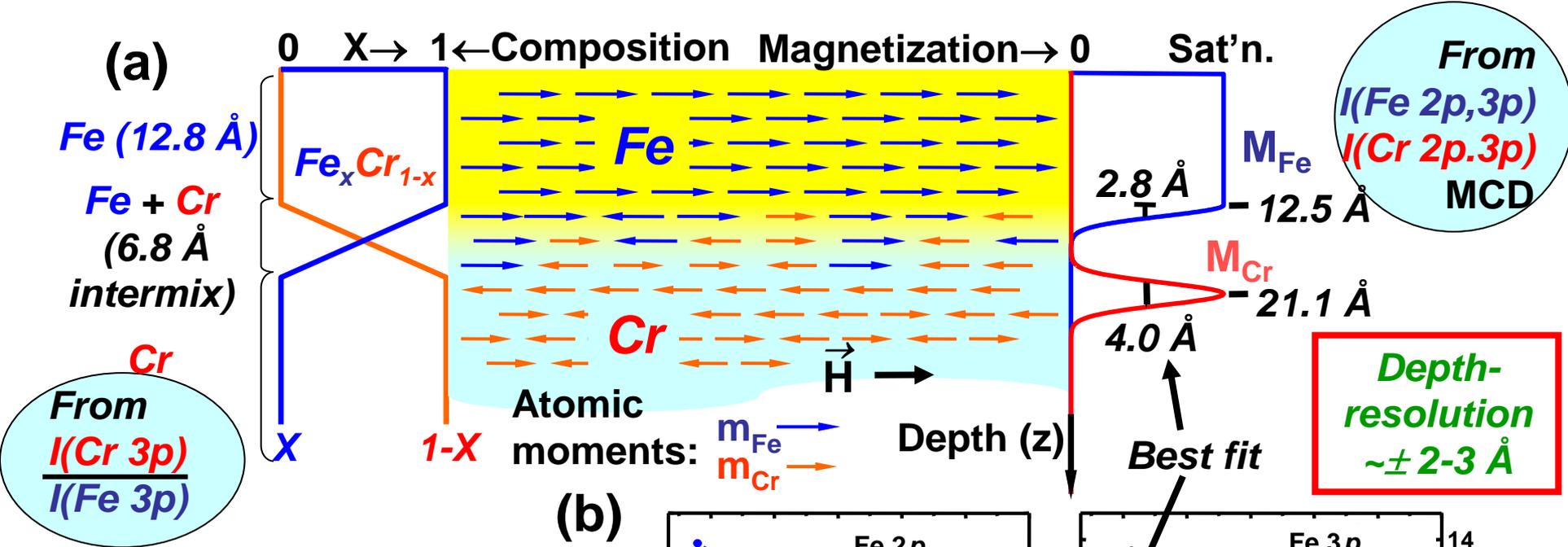
Probing Buried Interfaces: The Standing Wave-Wedge ("Swedge") Method



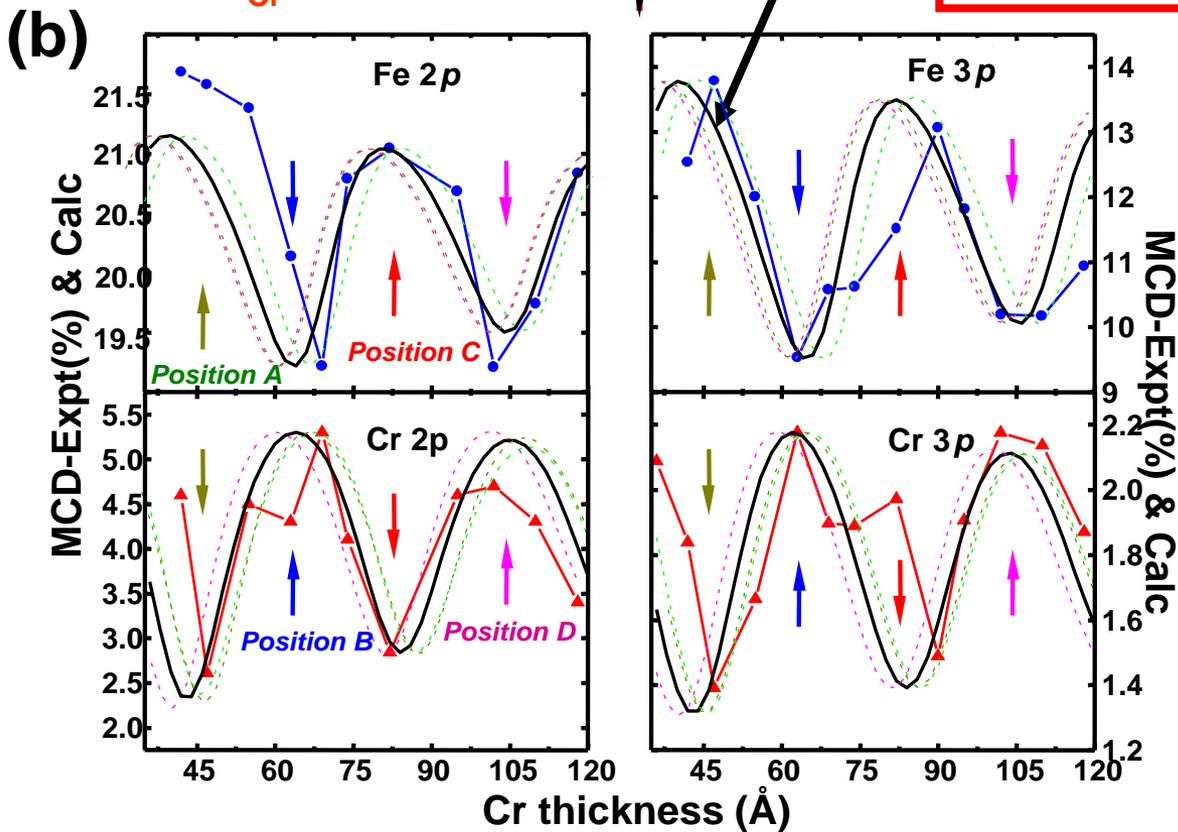
$$\lambda_{sw} (|E^2|) = \lambda_x / 2 \sin \theta_{inc} \approx d_{ML}$$

↑

• 1st order Bragg: $\lambda_x = 2 d_{ML} \sin \theta_{Bragg}$

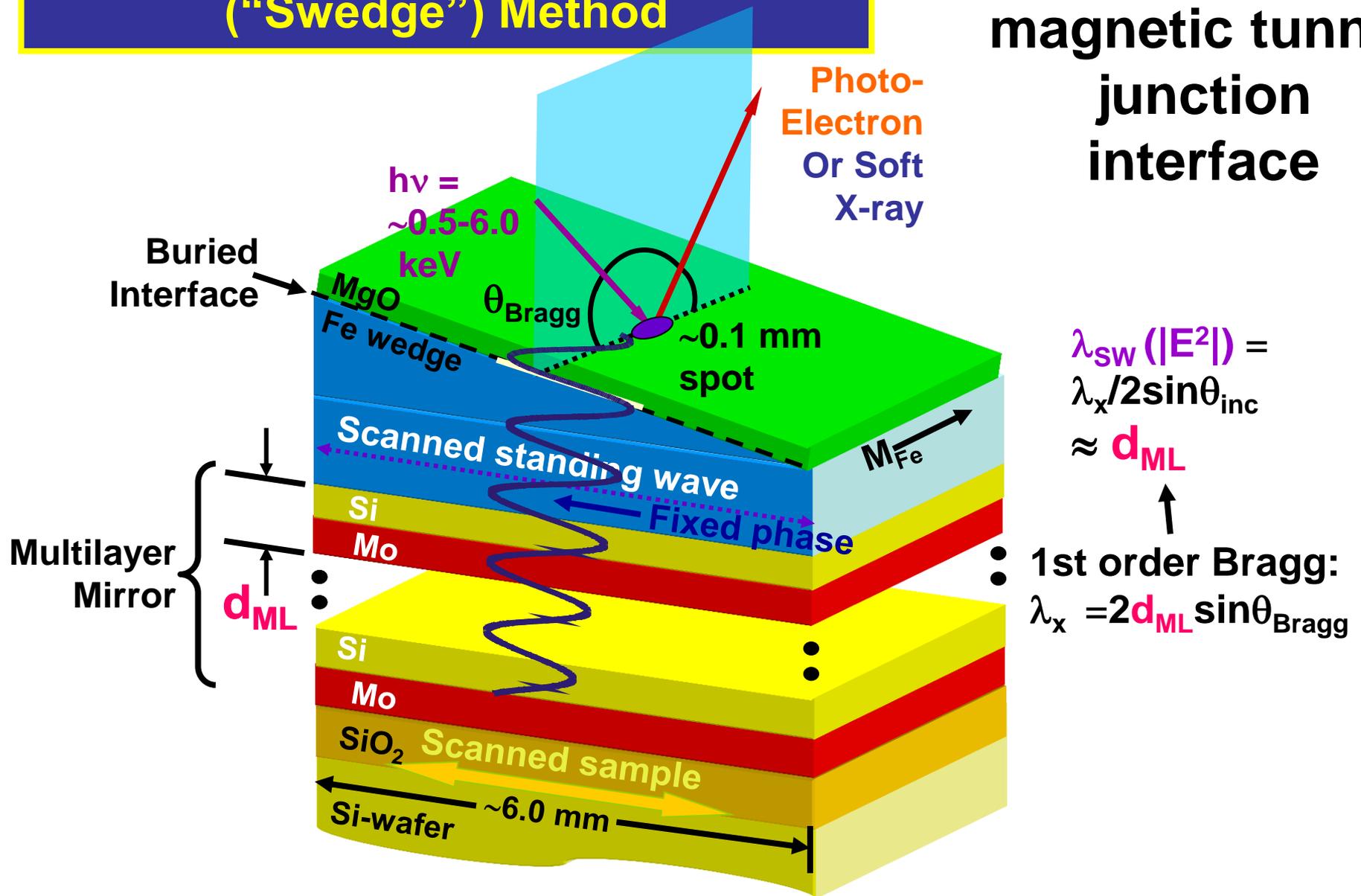


Fitting core PS experiment to XRO theory: non-destructive, depth-resolved determination of composition and magnetization profiles

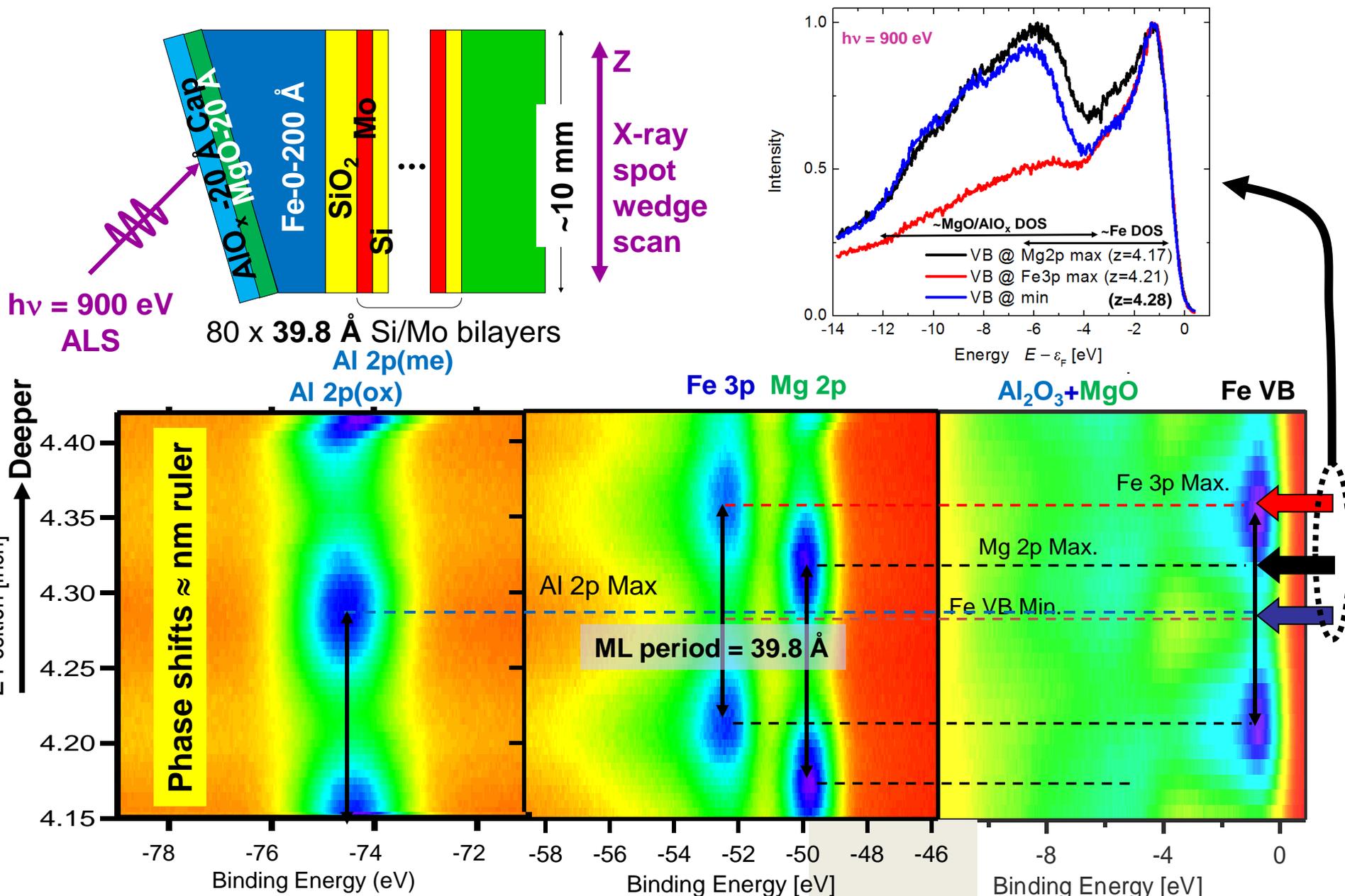


Probing Buried Interfaces: The Standing Wave-Wedge ("Swedge") Method

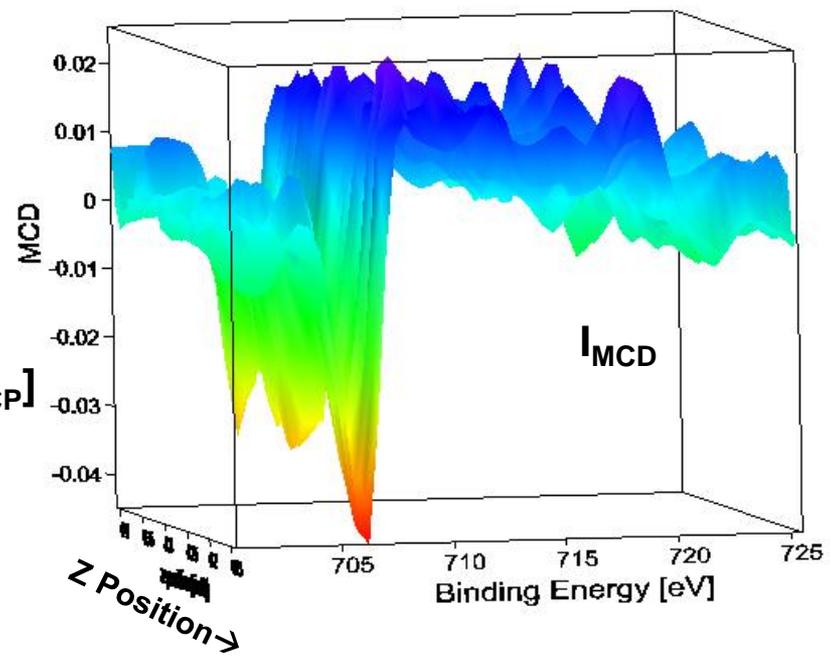
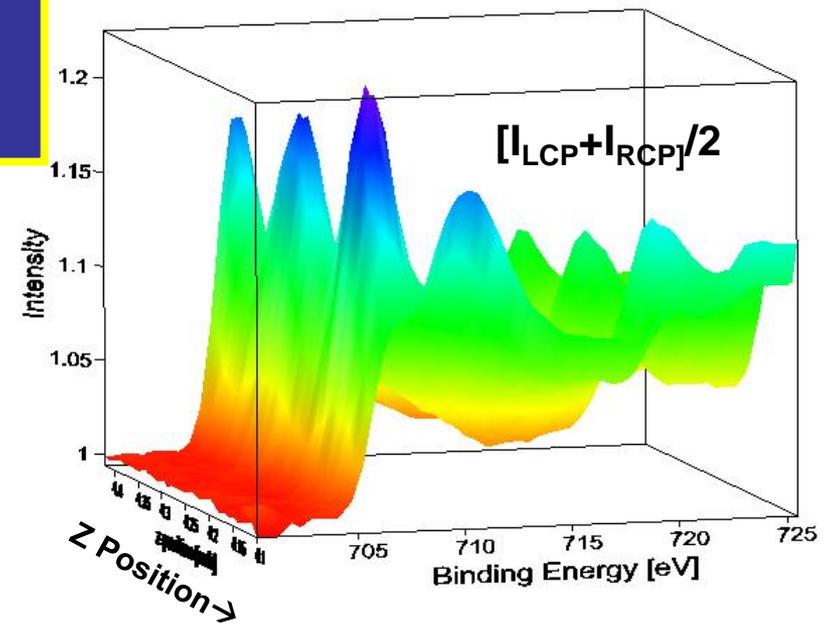
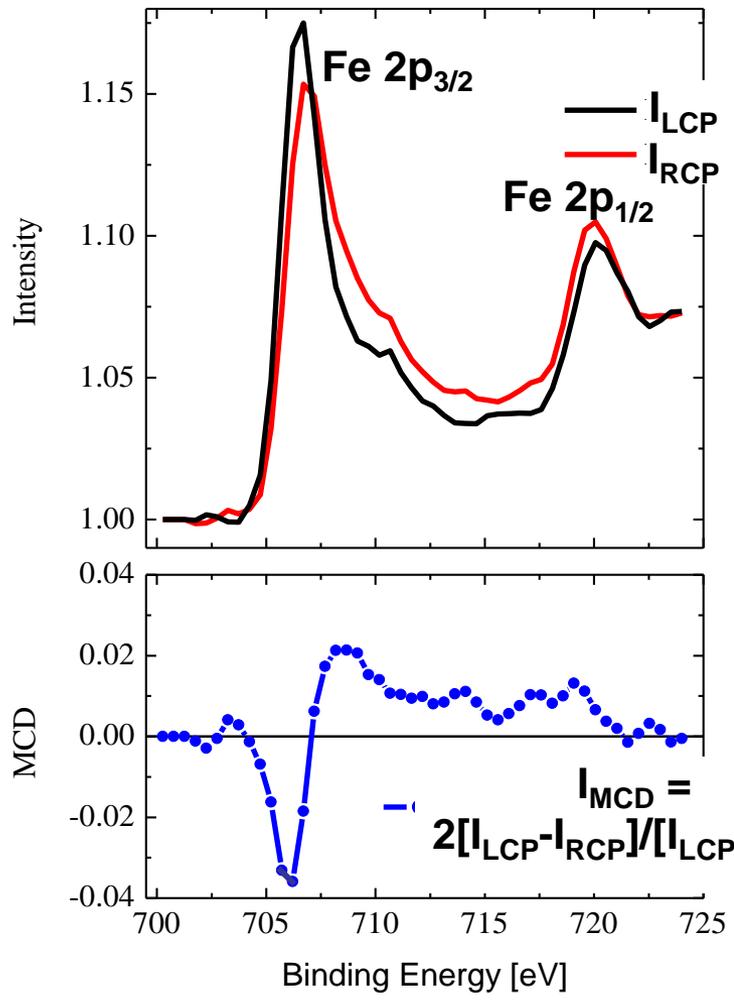
Example: The
MgO/Fe
magnetic tunnel
junction
interface

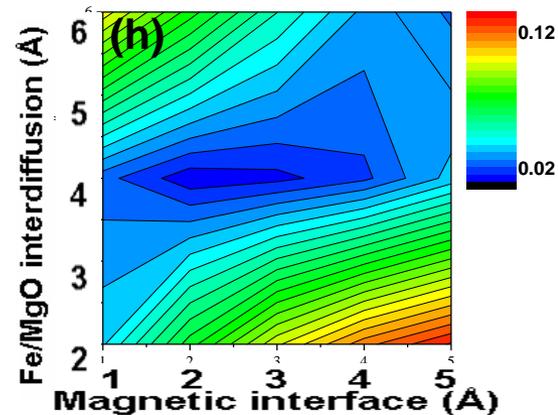
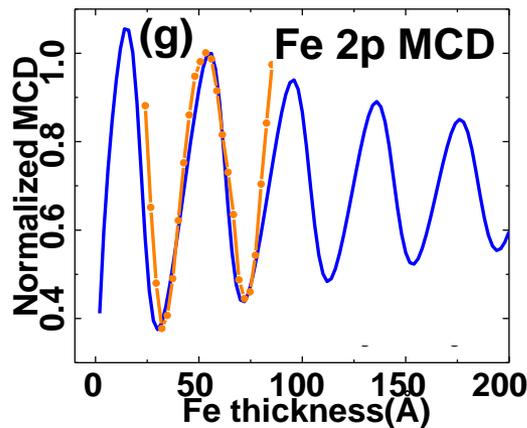
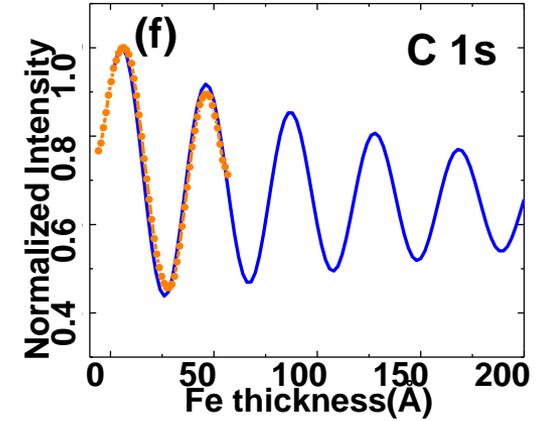
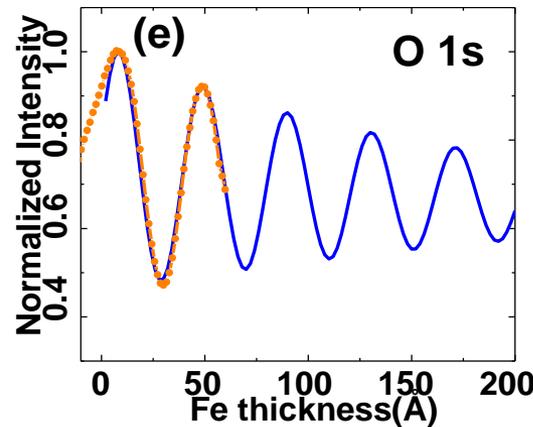
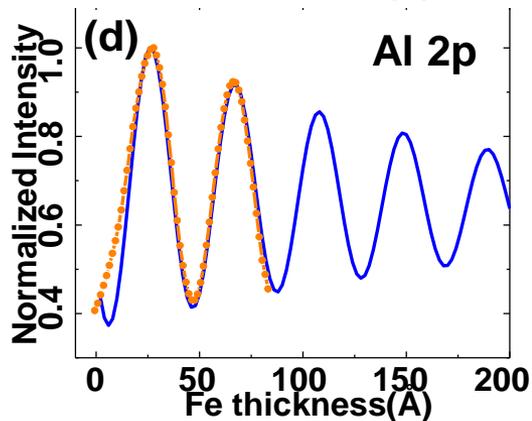
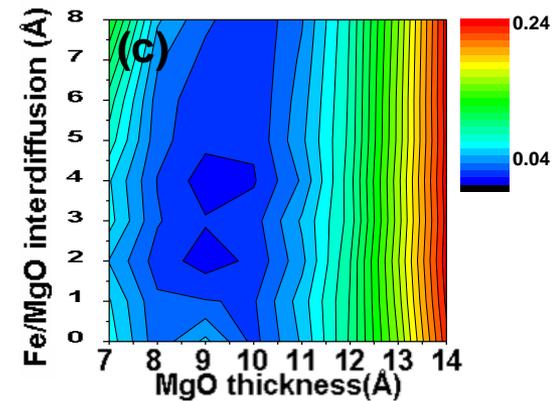
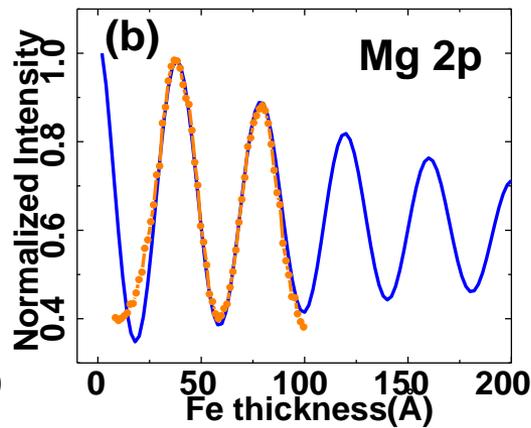
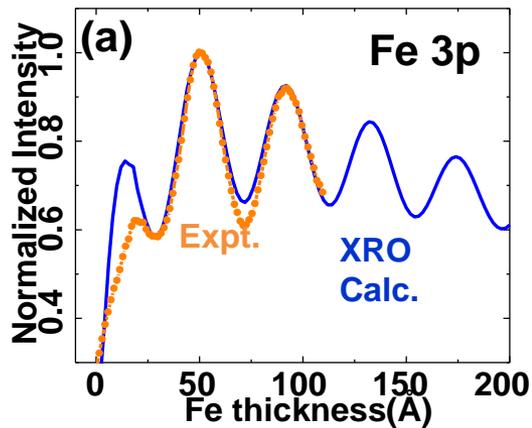


Soft x-ray standing-wave wedge scans through a magnetic tunnel junction



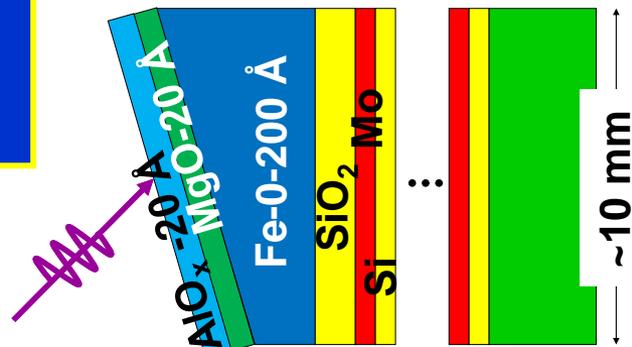
(D) Magnetic Circular Dichroism with Standing Wave Excitation- MgO/Fe, $h\nu = 900$ eV



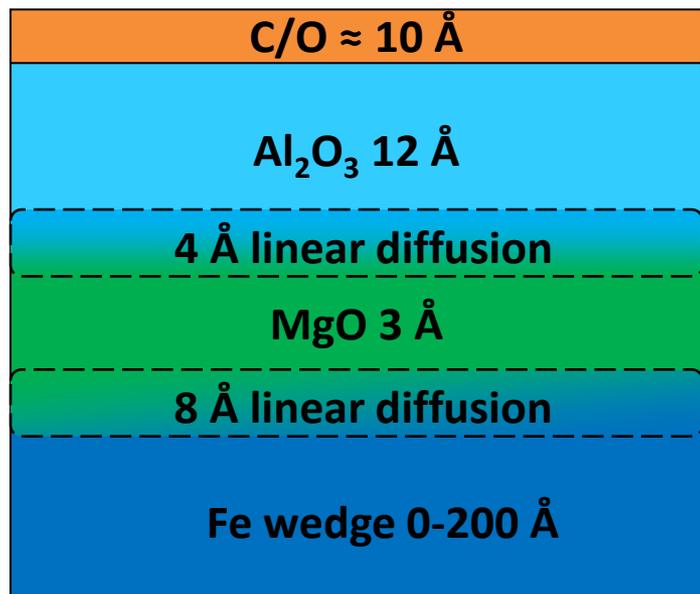


Standing wave/wedge analysis of an Fe/MgO tunnel junction multilayer: final fits of expt. to x-ray optical calcs.

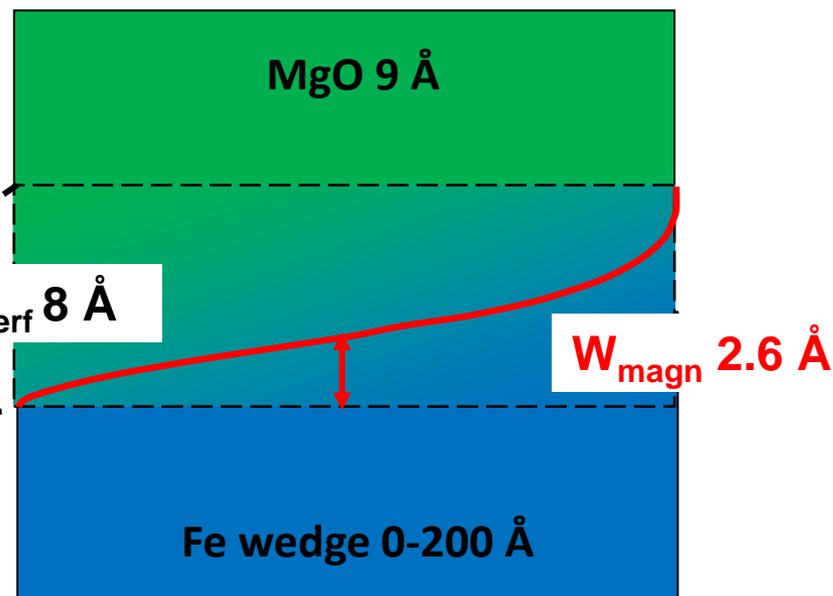
Final profiles of concentration and magnetization



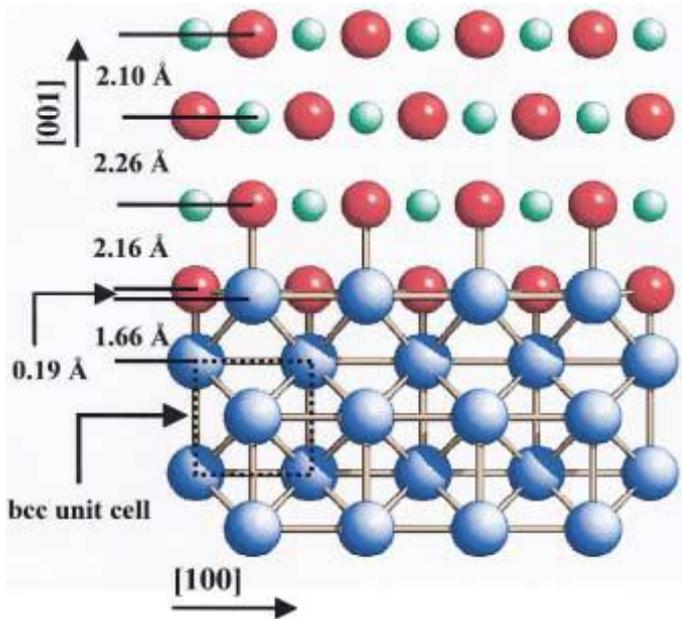
Concentration



Magnetization

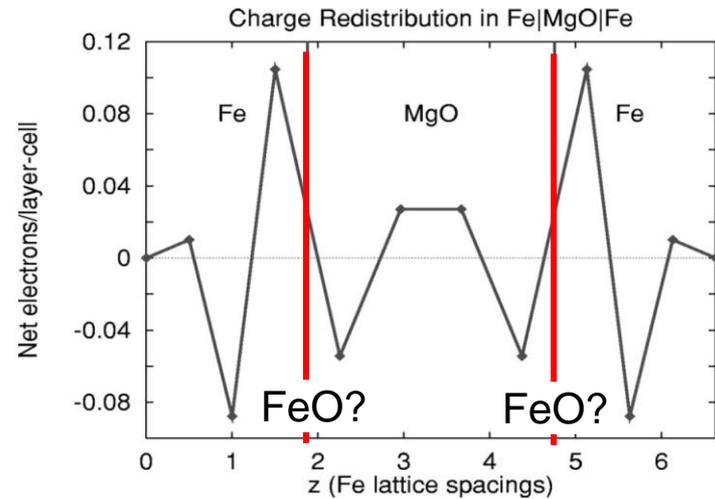


MgO/Fe tunnel junction- the real interface



Meyerheim PRL 87, 076102 (2001).

- *Is there FeO at the interface?*
- *What is the density of states at the interface?*
- Δ_1 *controls tunneling?*
- Can we see bands at epitaxial interfaces?*
- (Future project)*



Fe Minority DOS near Interface with MgO

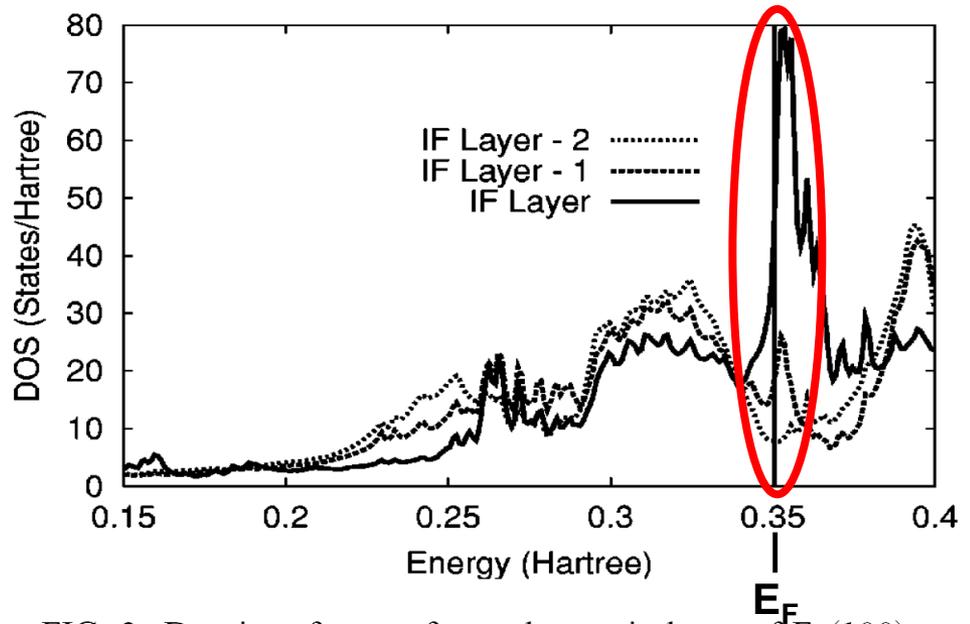
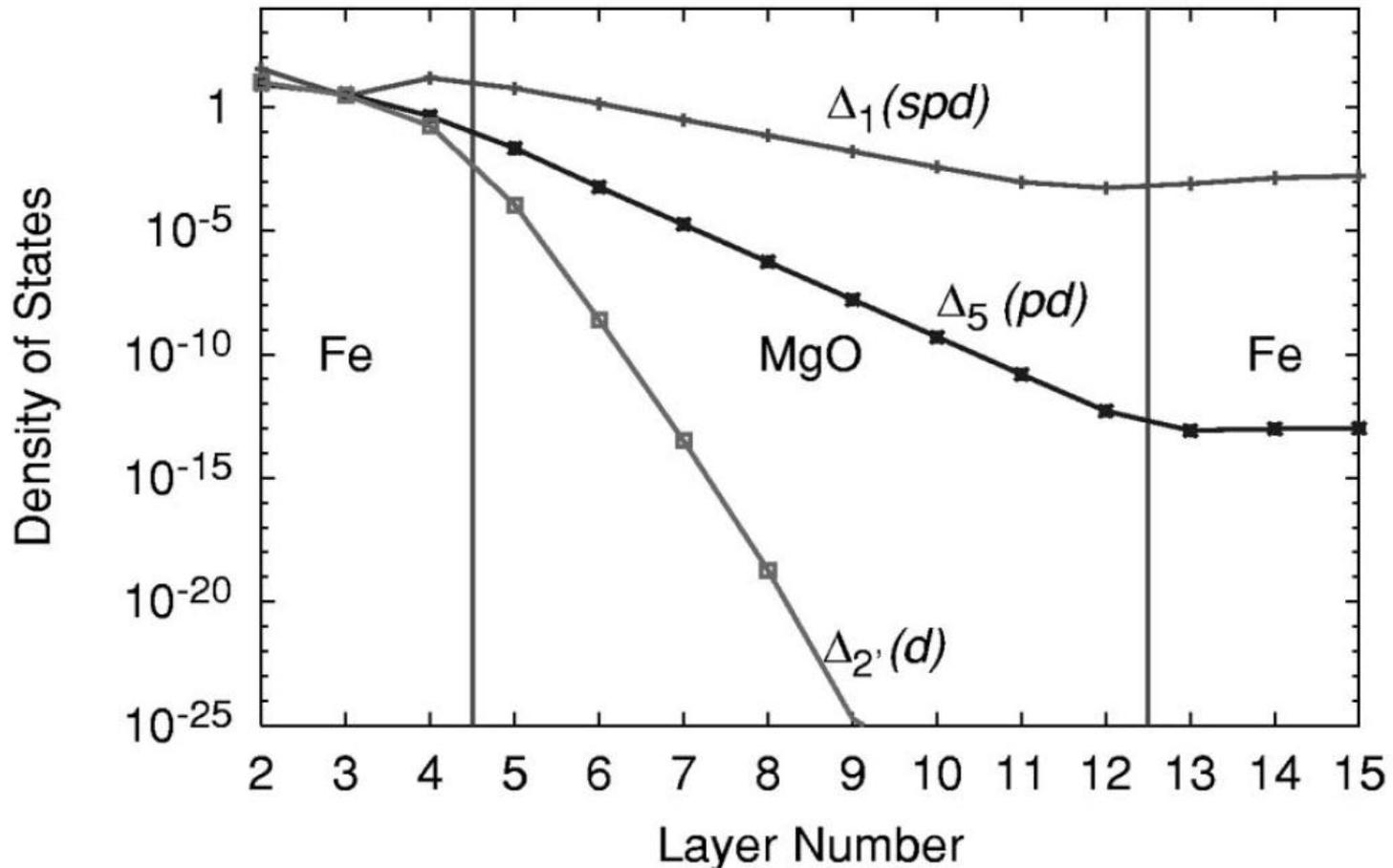


FIG. 3. Density of states for each atomic layer of Fe(100) near an interface with MgO. One hartree equals 27.2 eV.

Butler et al., PRB 63, 054416 (2001);
 Mathon & Umerski, PRB 63, 220403 (2001);
 Mertig et al., PRB 73, 214441 (2006)

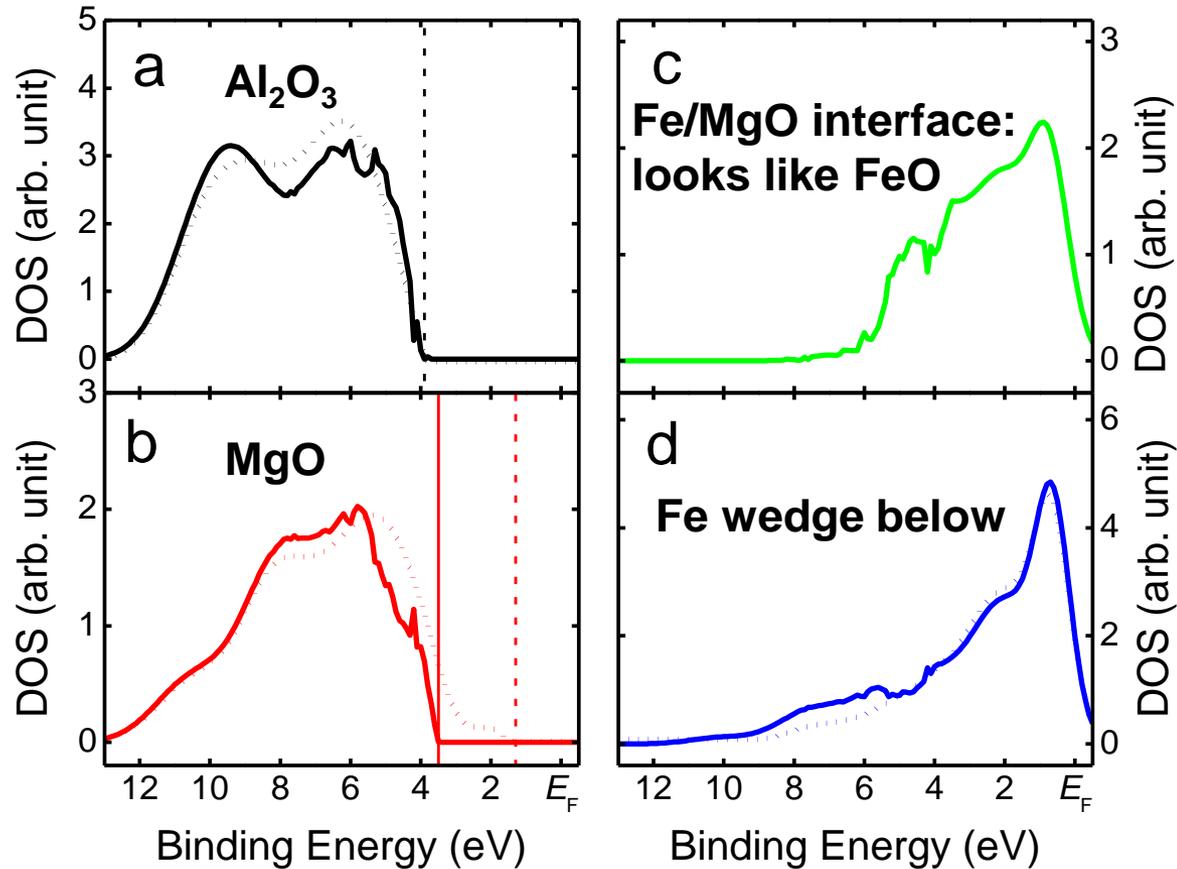
MgO/Fe tunnel junction- Δ_1 states dominant in tunneling for ideal interface

Majority Density of States for Fe|MgO|Fe

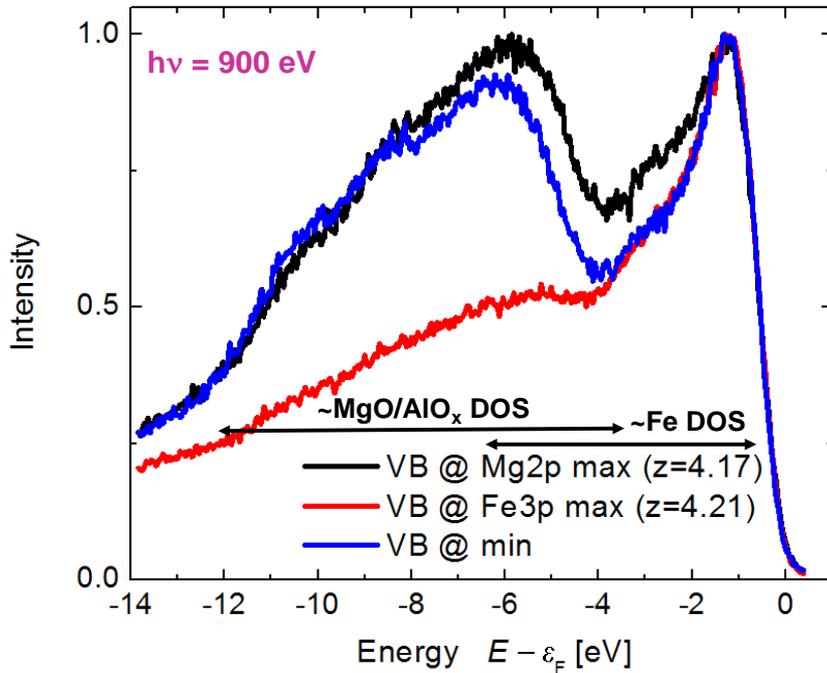


Standing wave/wedge derivation of depth-dependent densities of states of states: Fe/MgO tunnel junction

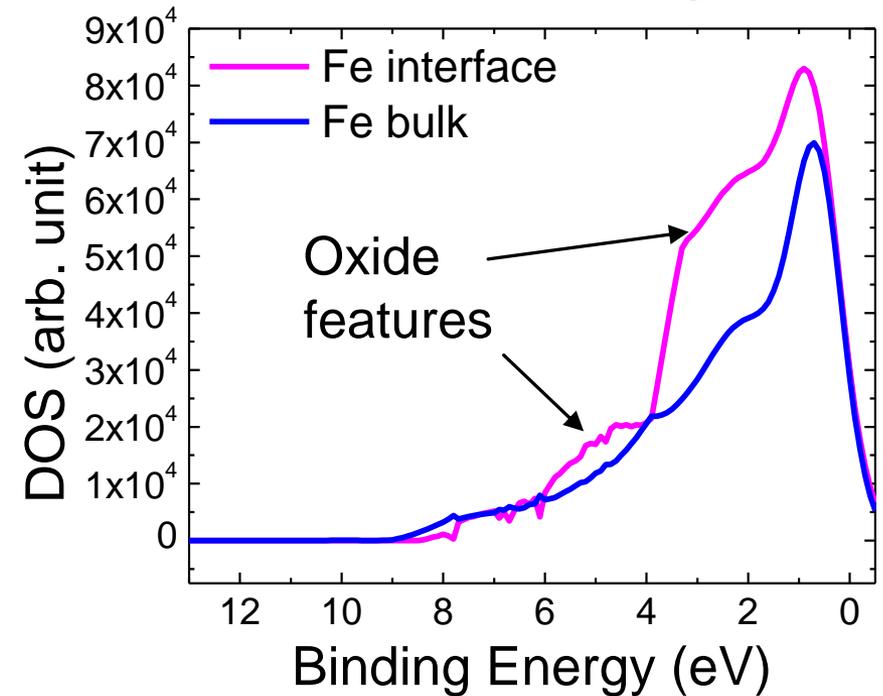
$\tilde{D}_L(E_{\text{kin},j})$ = matrix-element weighted density of states in layer L



Standing wave/wedge derivation of depth-dependent densities of states: Fe/MgO tunnel junction



→ Oxidation at the Fe/MgO interface



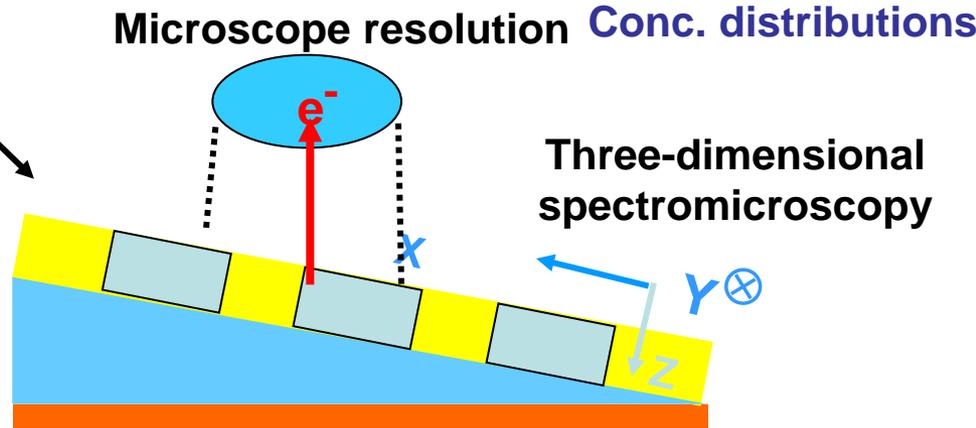
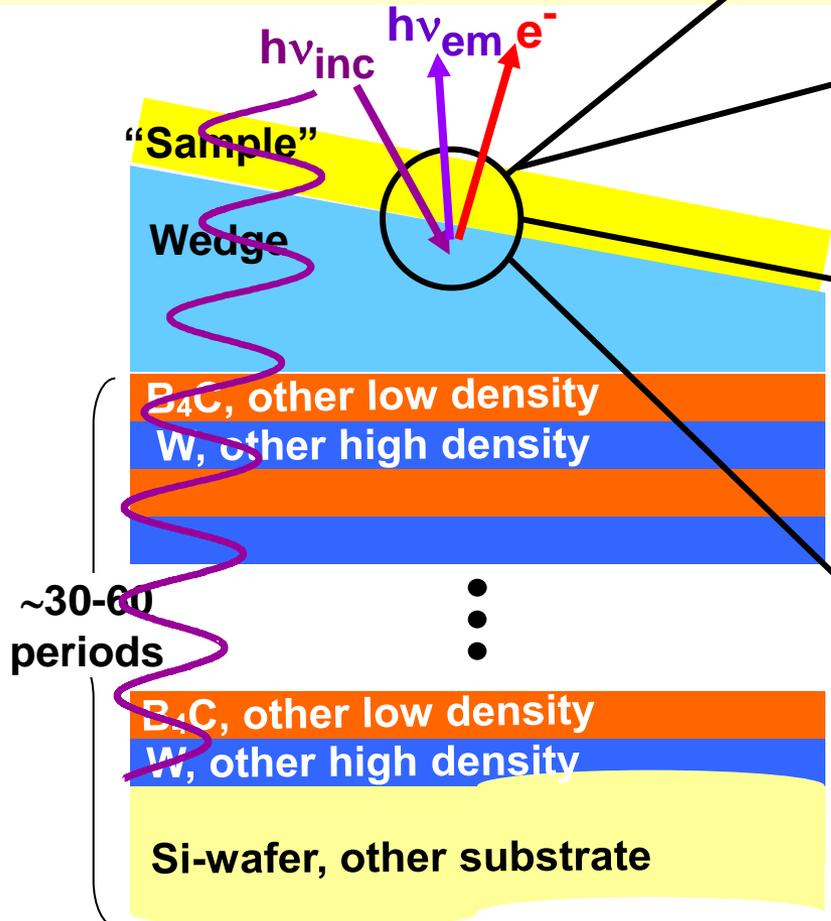
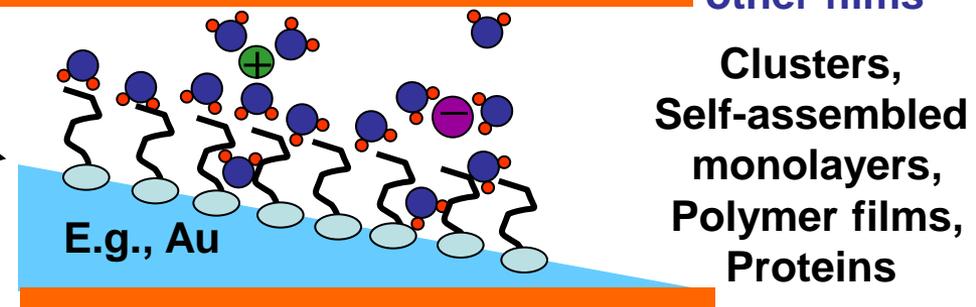
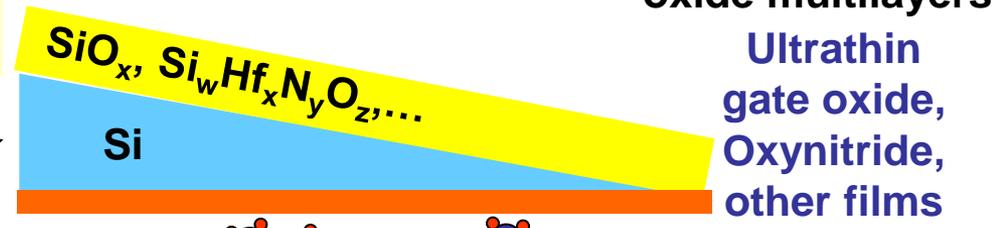
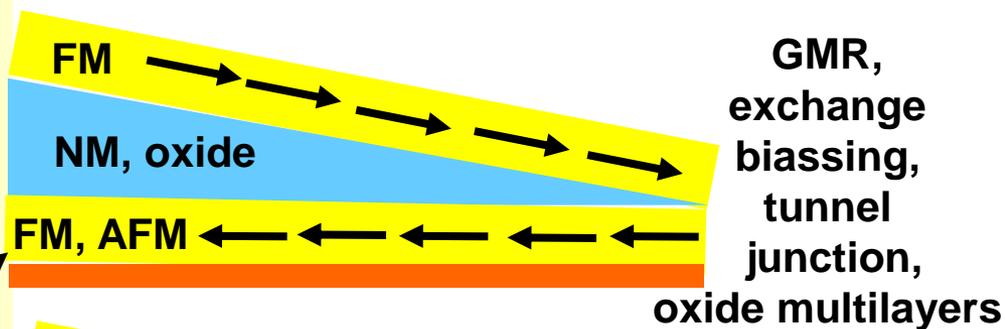
Self-consistent
X-ray optical
modeling of layer-
resolved densities
of states

- Other material pairs in multilayer ($B_4C/W, Si/Mo, \dots$) + epitaxial multilayers ($AlAs/GaAs, STO/LSMO$) → epitaxial samples

- Smaller periods (to $\sim 25-30 \text{ \AA}$) → smaller SW period, better resolution

- Lower $h\nu_{inc}$ → higher Bragg angles → perpend. component of M

- X-ray emission → deeper layers, more sensitivity to SW position



Basic Concepts and Experiments

Core-Level Photoemission

Intensities and Quantitative Analysis, the 3-Step Model
Varying Surface and Bulk Sensitivity
Chemical shifts
Multiplet Splittings
Electron Screening and Satellite Structure
Magnetic and Non-Magnetic Dichroism
Resonant Photoemission
Photoelectron Diffraction and Holography

Valence-Level Photoemission

Band-Mapping in the Ultraviolet Photoemission Limit
Densities of States in the X-Ray Photoemission Limit

Some New Directions

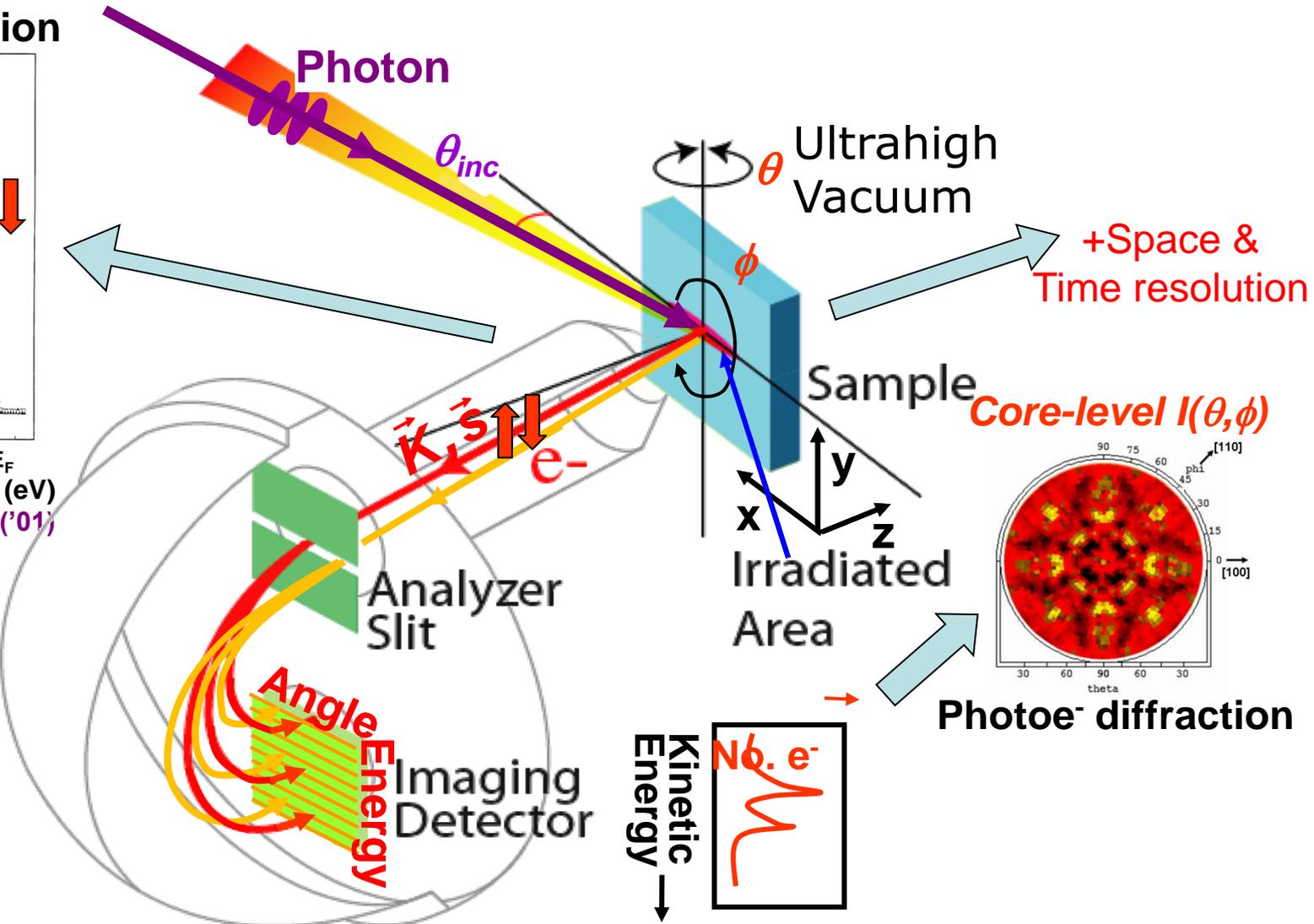
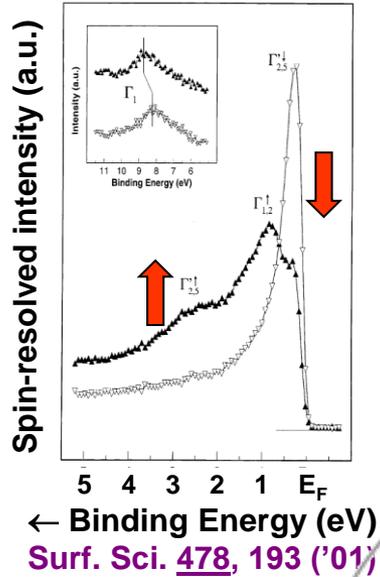
Photoemission with Hard X-Rays (throughout lectures)
Photoemission with Standing Wave Excitation
Photoemission with Spatial Resolution/Photoelectron Microscopy
Temporal Resolution
@ Higher Pressures



Kiskinova
Locatelli

Typical experimental geometry for energy- and angle-resolved photoemission measurements

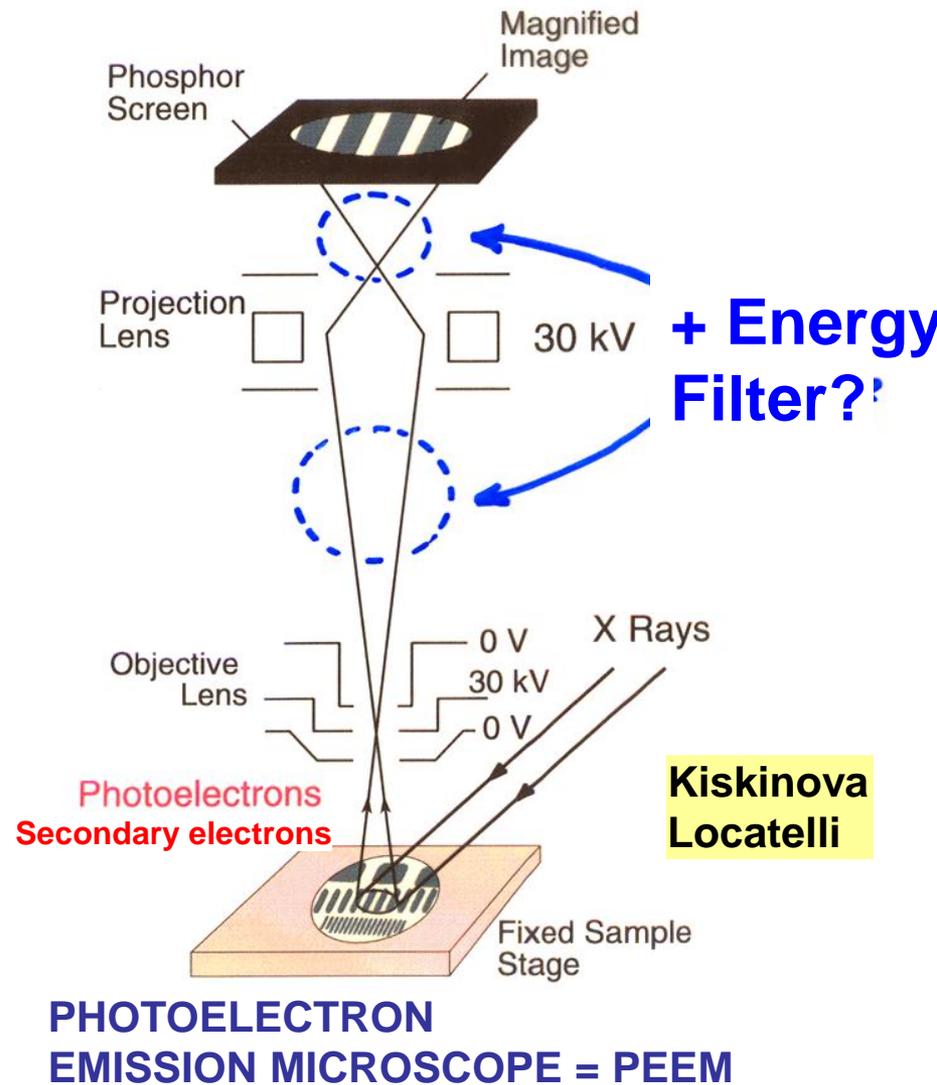
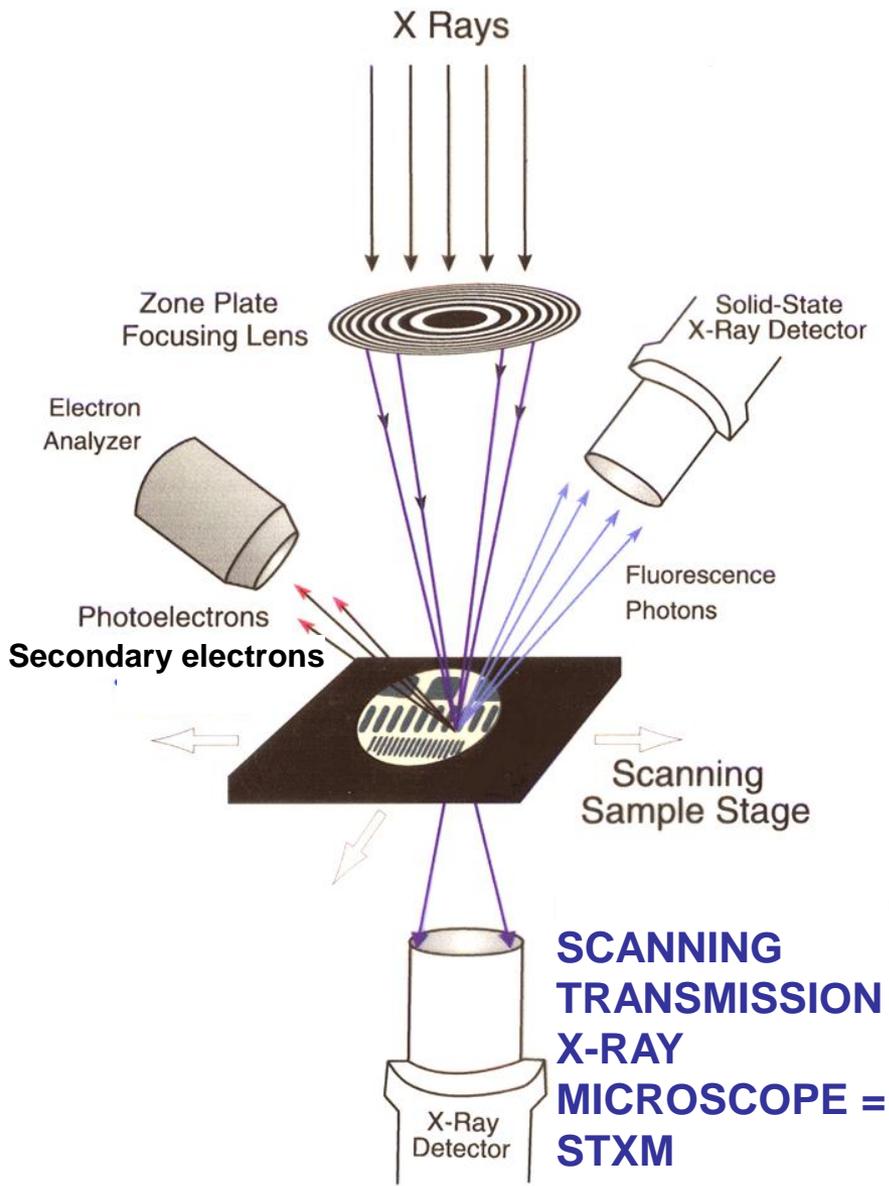
Spin-resolution



Imaging with soft x-ray photoemission microscopes—two types

Scanning X-ray Microscopy

X-Ray Photoelectron Microscopy



Basic Concepts and Experiments

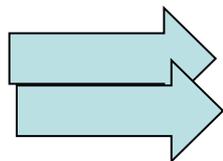
Core-Level Photoemission

Intensities and Quantitative Analysis, the 3-Step Model
Varying Surface and Bulk Sensitivity
Chemical shifts
Multiplet Splittings
Electron Screening and Satellite Structure
Magnetic and Non-Magnetic Dichroism
Resonant Photoemission
Photoelectron Diffraction and Holography

Valence-Level Photoemission

Band-Mapping in the Ultraviolet Photoemission Limit
Densities of States in the X-Ray Photoemission Limit

Some New Directions

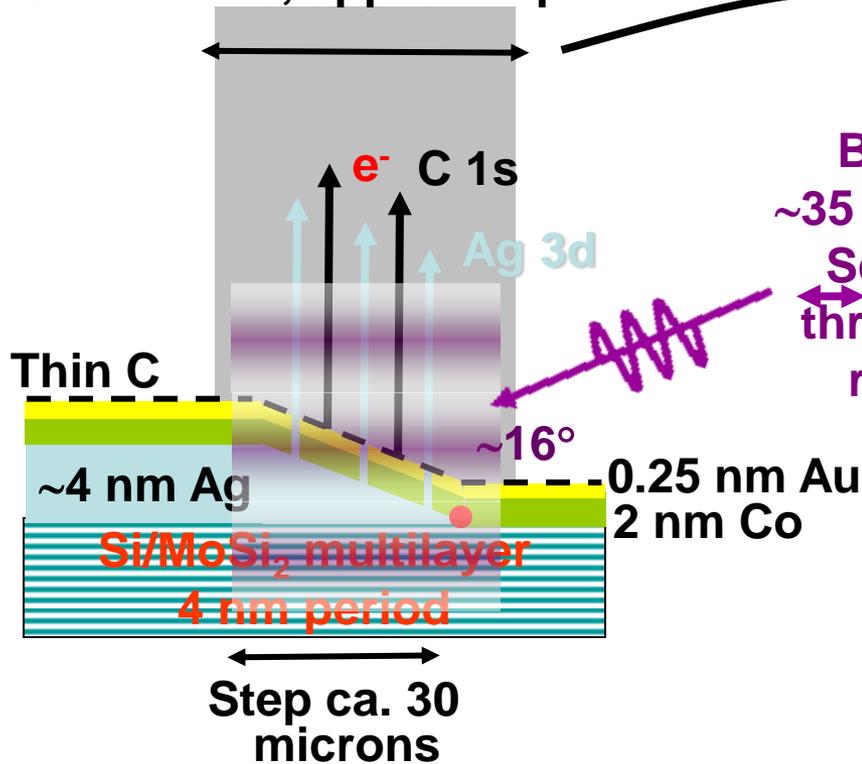


Photoemission with Hard X-Rays (throughout lectures)
Photoemission with Standing Wave Excitation
Photoemission with Spatial Resolution/Photoelectron Microscopy
Temporal Resolution
@ Higher Pressures

Adding depth resolution to the photoelectron microscope via standing wave excitation

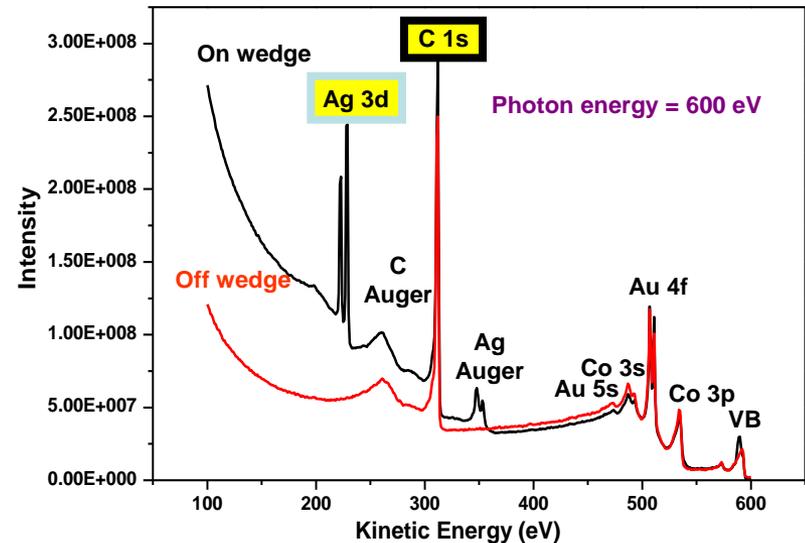
**Spin-Polarized Photoelectron
Microscope—BESSY, Berlin**

Microscope field of view
~50 microns, approx. 1 period

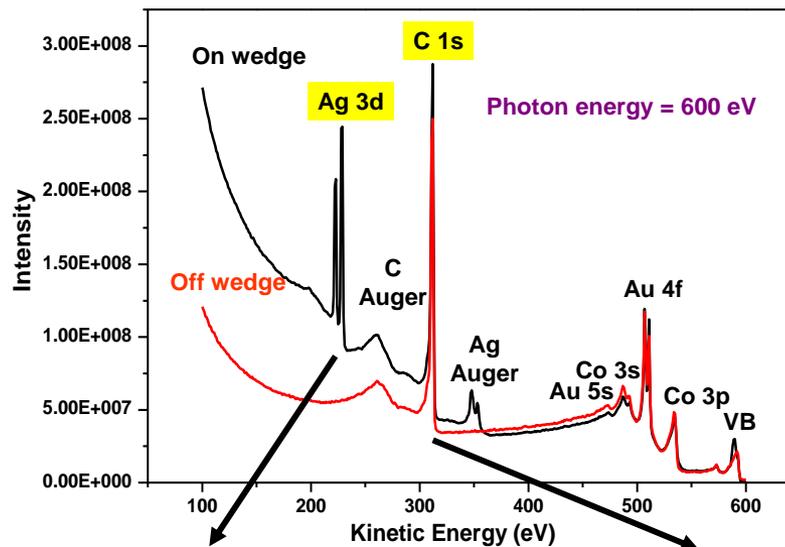


Imaging with
element-specific
photoelectron peaks

F. Kronast et al., Appl. Phys. Lett. 93, 243116 (2008)

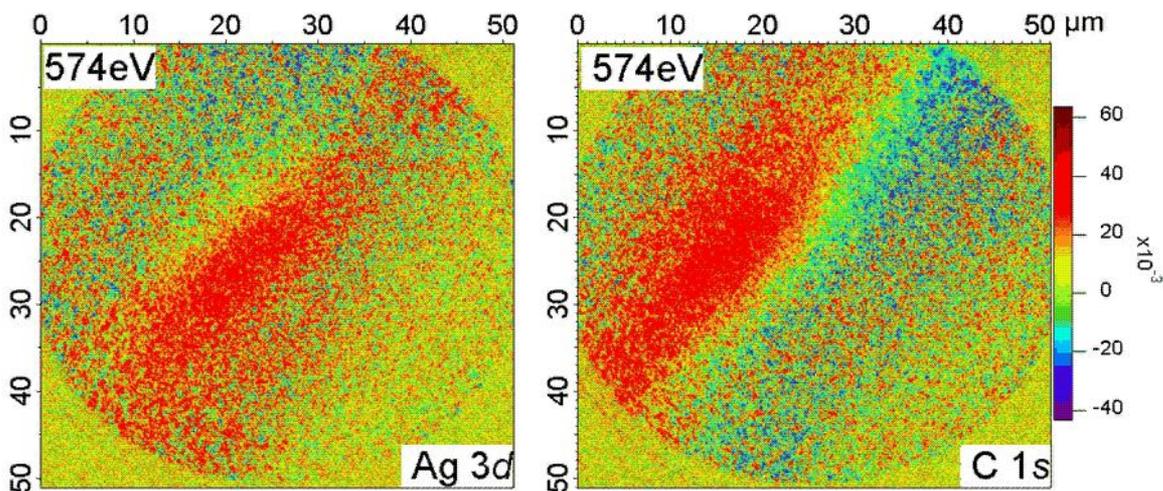


Standing wave effects in a photoelectron microscope



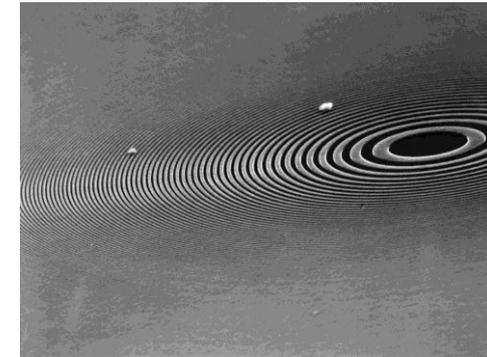
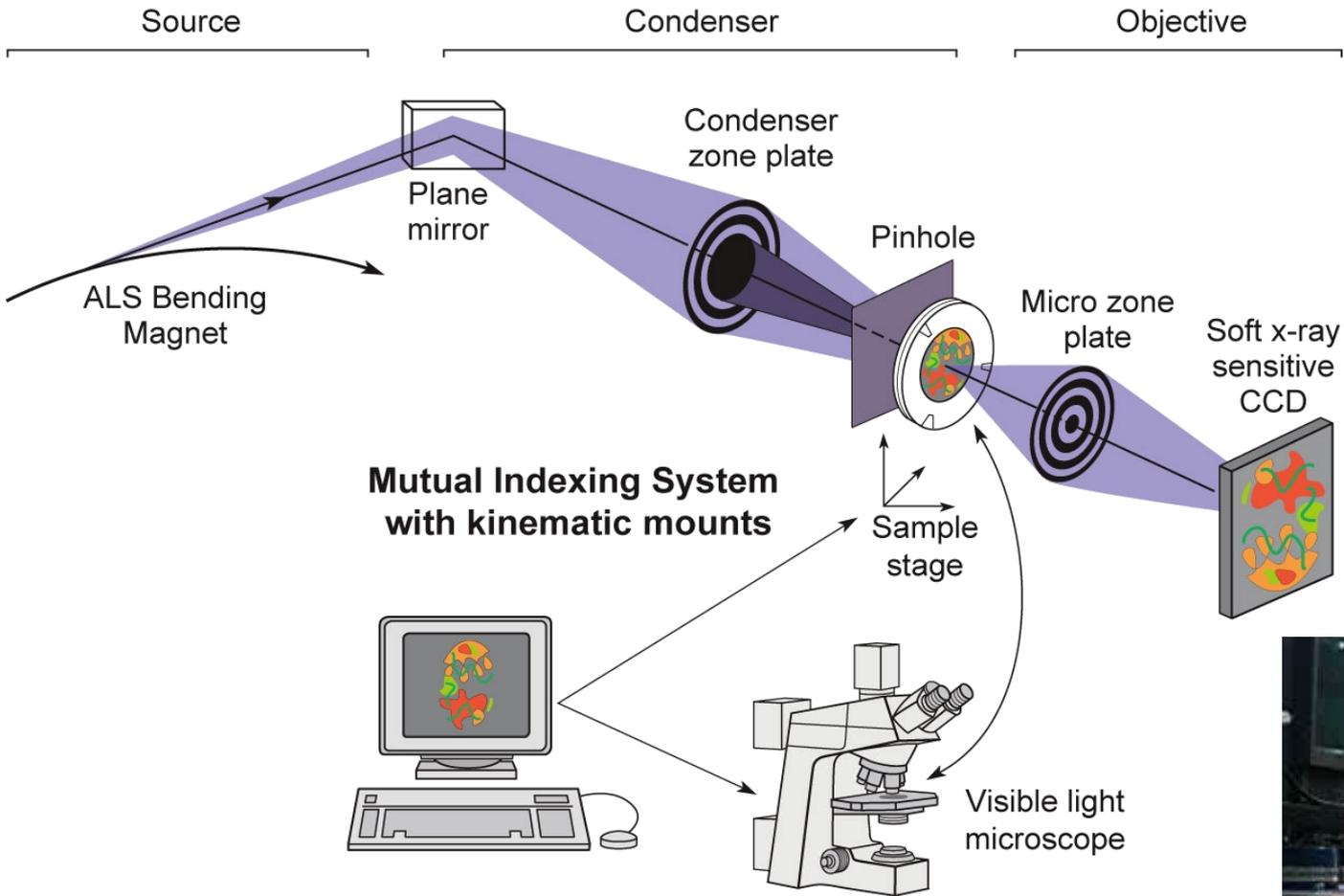
Ag 3d

C 1s



F. Kronast, H. Dürr,
 BESSY
 D. Buegler, R.
 Scheiber, C.
 Schneider, Jülich
 Yang, IBM, CF,
 Appl. Phys. Lett.
 93, 243116 (2008)

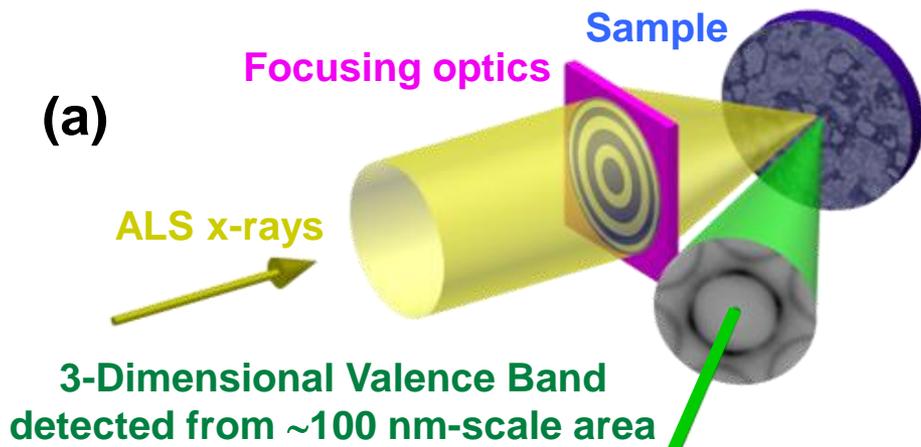
Third type: Imaging Zone-Plate X-ray microscope XM-1 @ ALS



X-ray lens = zone plate
Outer rings 100 atoms apart

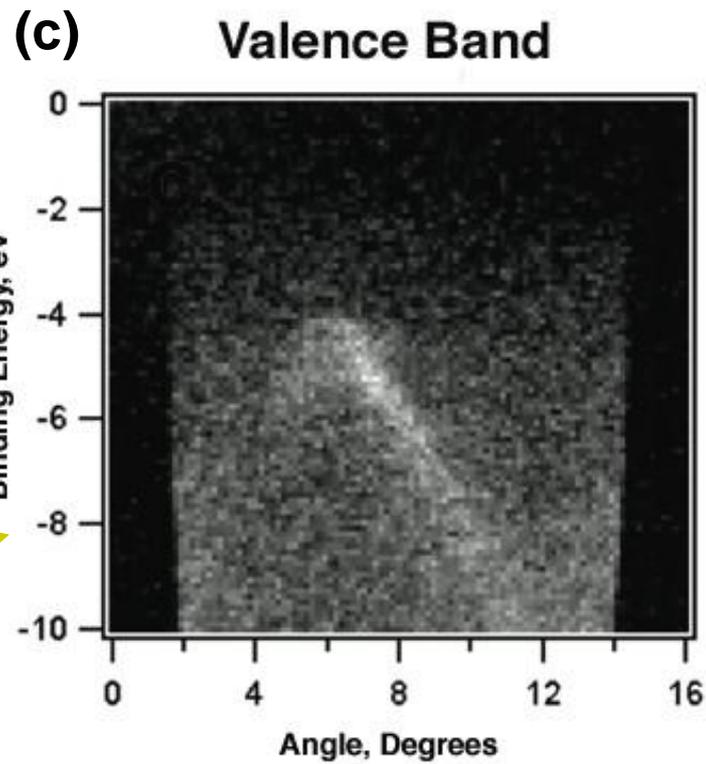
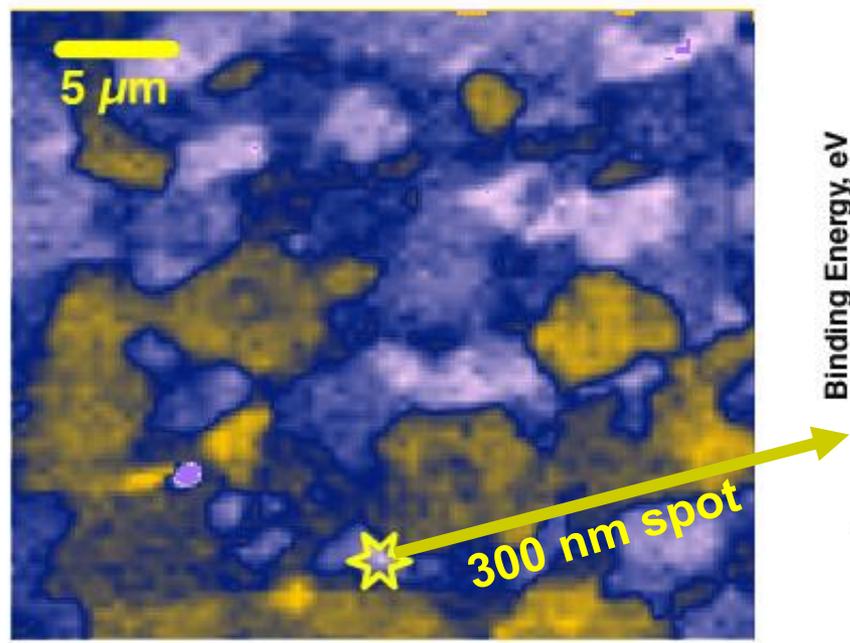


Fischer et al.



**Nanometer-Scale
Angle-Resolved
Photoemission**

(b) **HOPG Graphite**
Imaged with Valence Band Contrast



Basic Concepts and Experiments

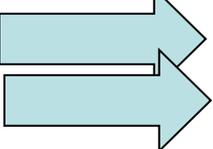
Core-Level Photoemission

Intensities and Quantitative Analysis, the 3-Step Model
Varying Surface and Bulk Sensitivity
Chemical shifts
Multiplet Splittings
Electron Screening and Satellite Structure
Magnetic and Non-Magnetic Dichroism
Resonant Photoemission
Photoelectron Diffraction and Holography

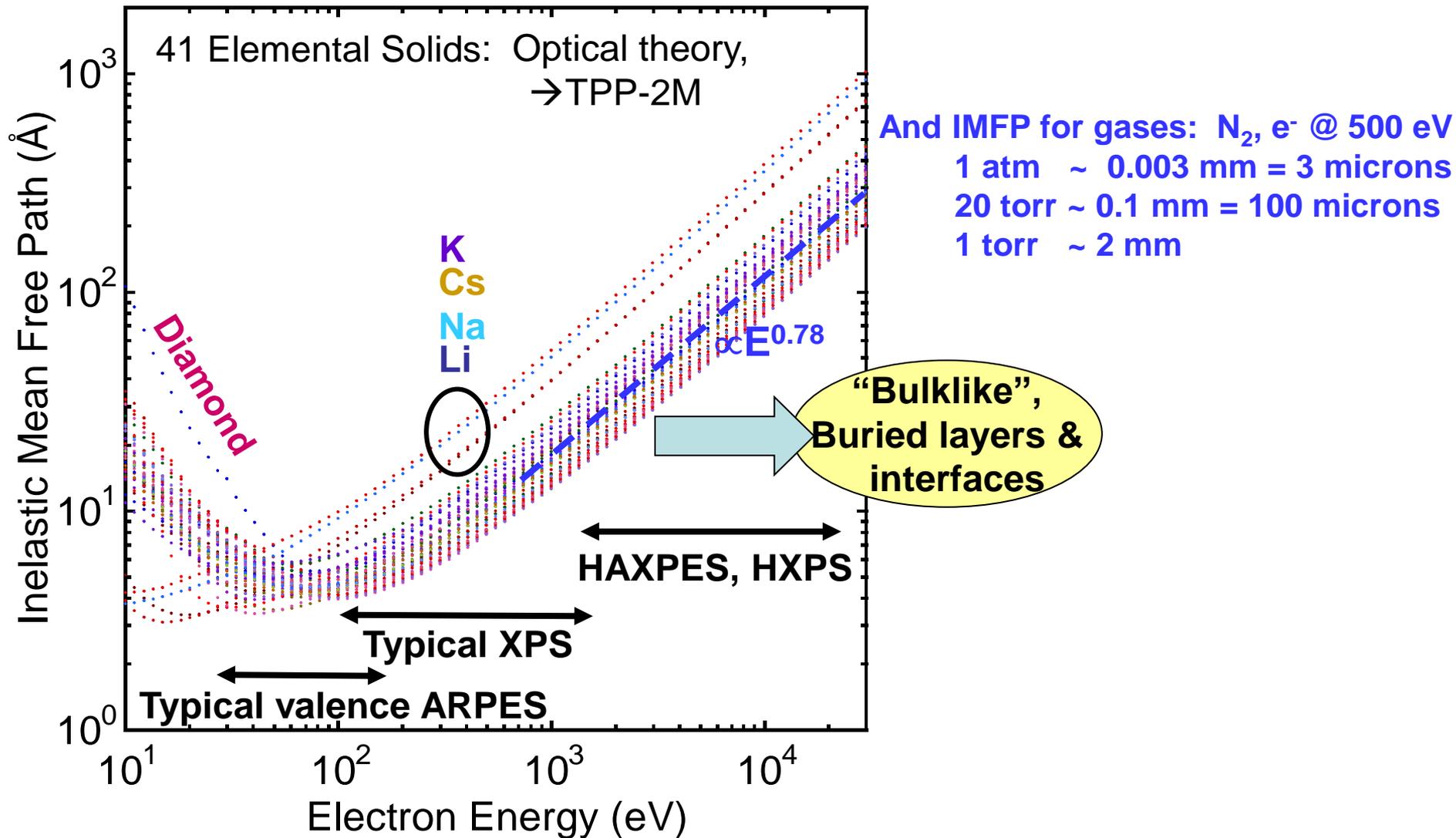
Valence-Level Photoemission

Band-Mapping in the Ultraviolet Photoemission Limit
Densities of States in the X-Ray Photoemission Limit

Some New Directions

Photoemission with Hard X-Rays (throughout lectures)
Photoemission with Standing Wave Excitation
Photoemission with Spatial Resolution/Photoelectron Microscopy
 Temporal Resolution
@ Higher Pressures

Surface sensitivity and why we may want to go to 5-10 keV in XPS



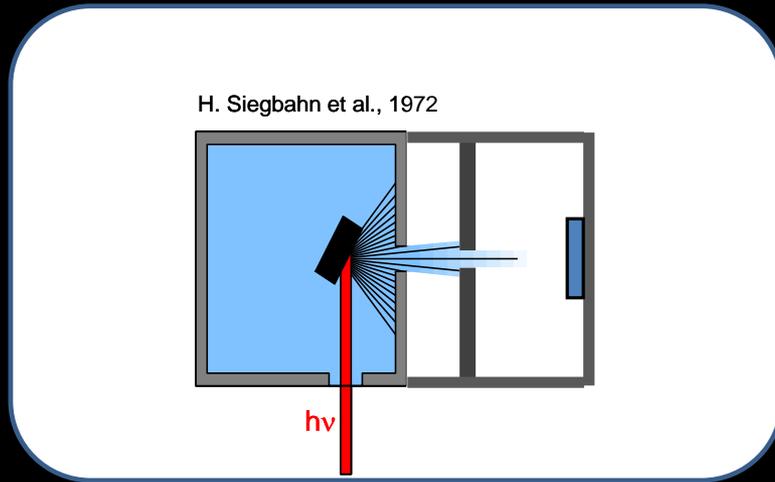
Challenges for high-pressure photoemission: analyzer Pressure and short electron mean free path

IMFP: N_2 @ 500 eV

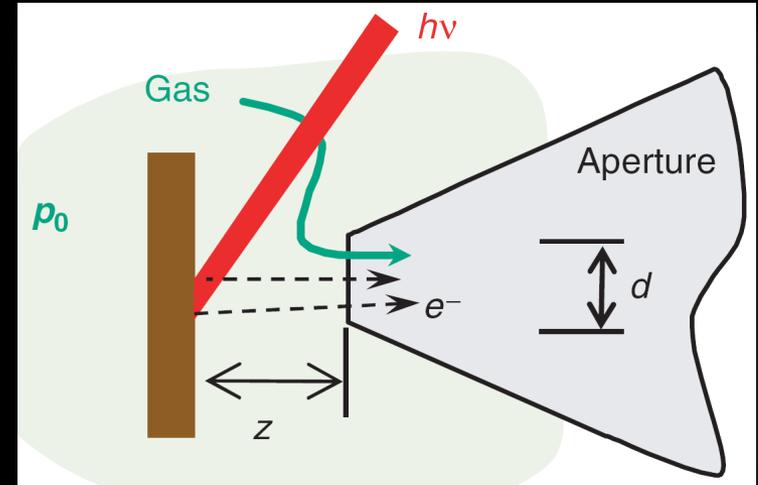
1 atm ~ 0.003 mm = 3 microns

20 torr ~ 0.1 mm = 100 microns

1 torr ~ 2 mm

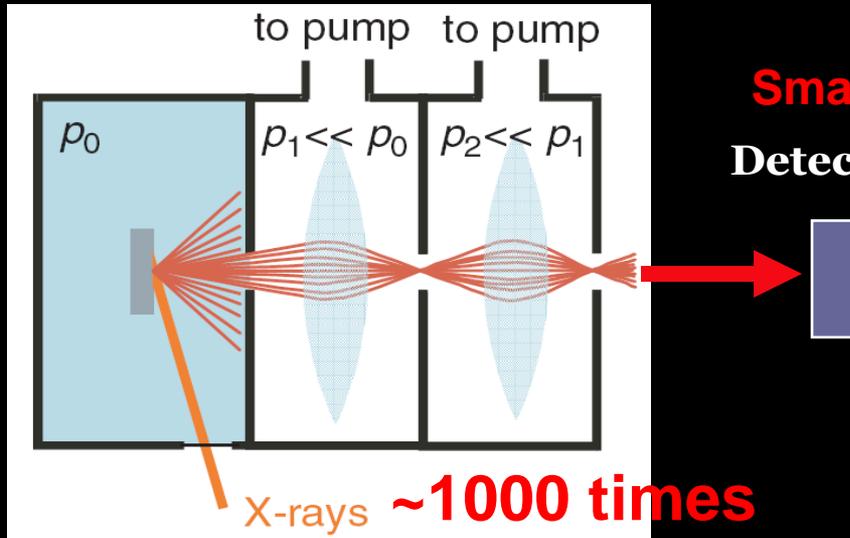


Very low efficiency



Smaller x-ray spot & $z \rightarrow$ Higher Pressure...

Detector



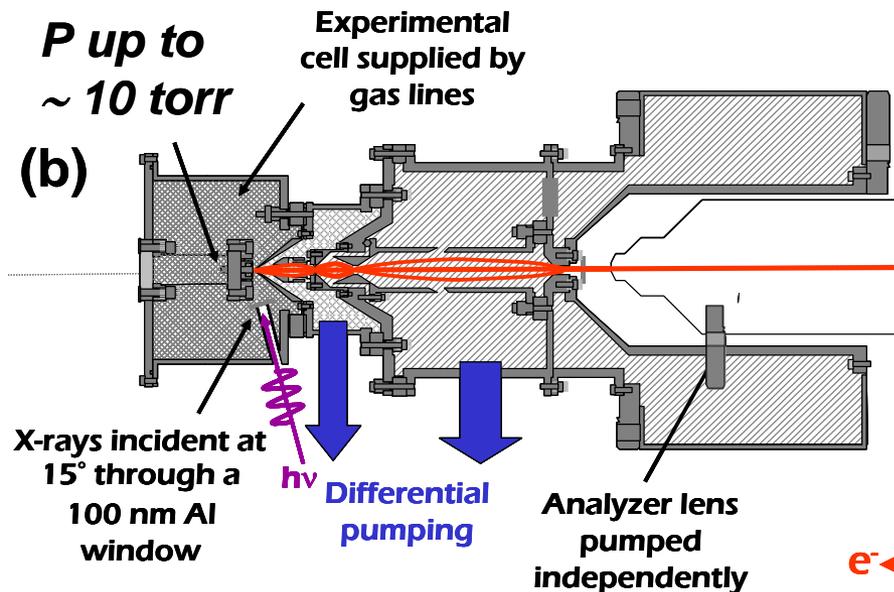
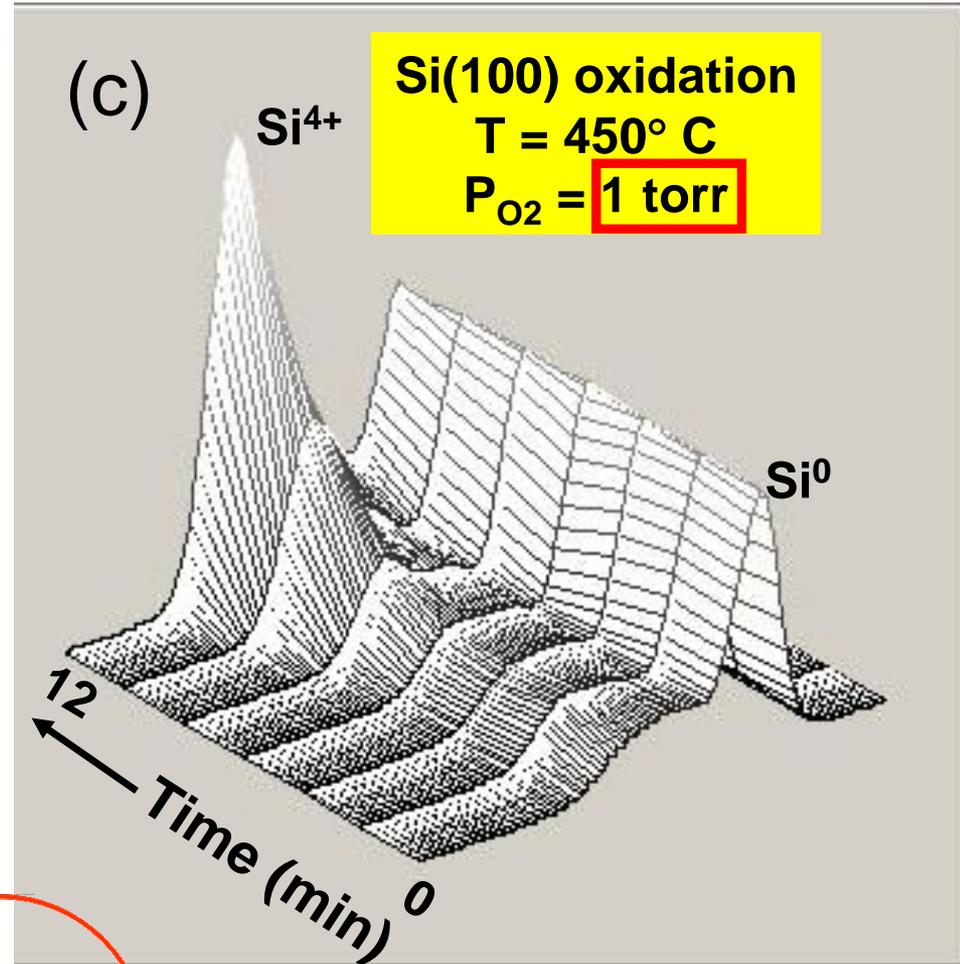
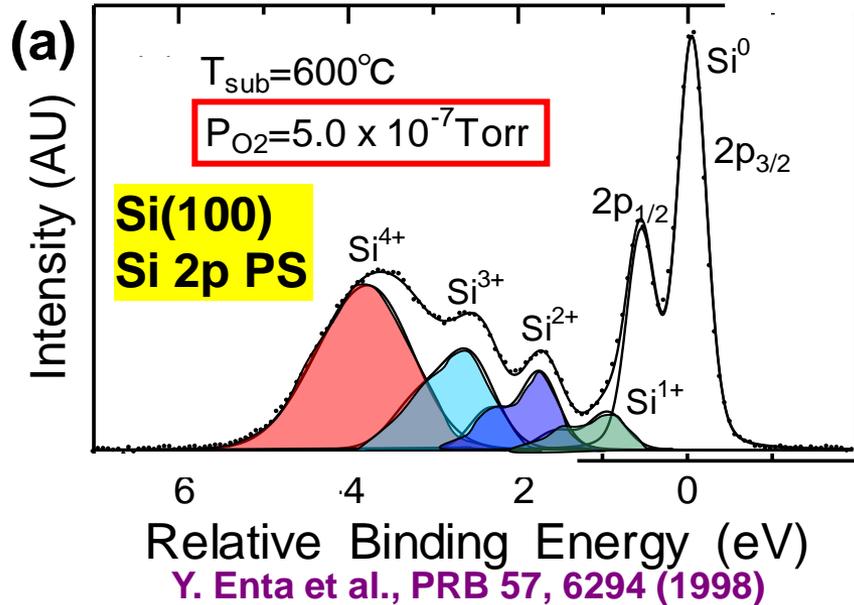
The first endstation at a SR facility (ALS, BL9.3.2):

D.F. Ogletree, H. Bluhm, G. Lebedev, C.S. Fadley, Z. Hussain, M. Salmeron, Rev. Sci. Instrum. 73 (2002) 3872.

A good review paper:

M. Salmeron and R. Schlögl, Surf. Sci. Rep. 63, 169-199 (2008).

Bridging the Pressure Gap: Chemical-State- and Time- Resolved Oxidation of Si at Multi-Torr Pressures

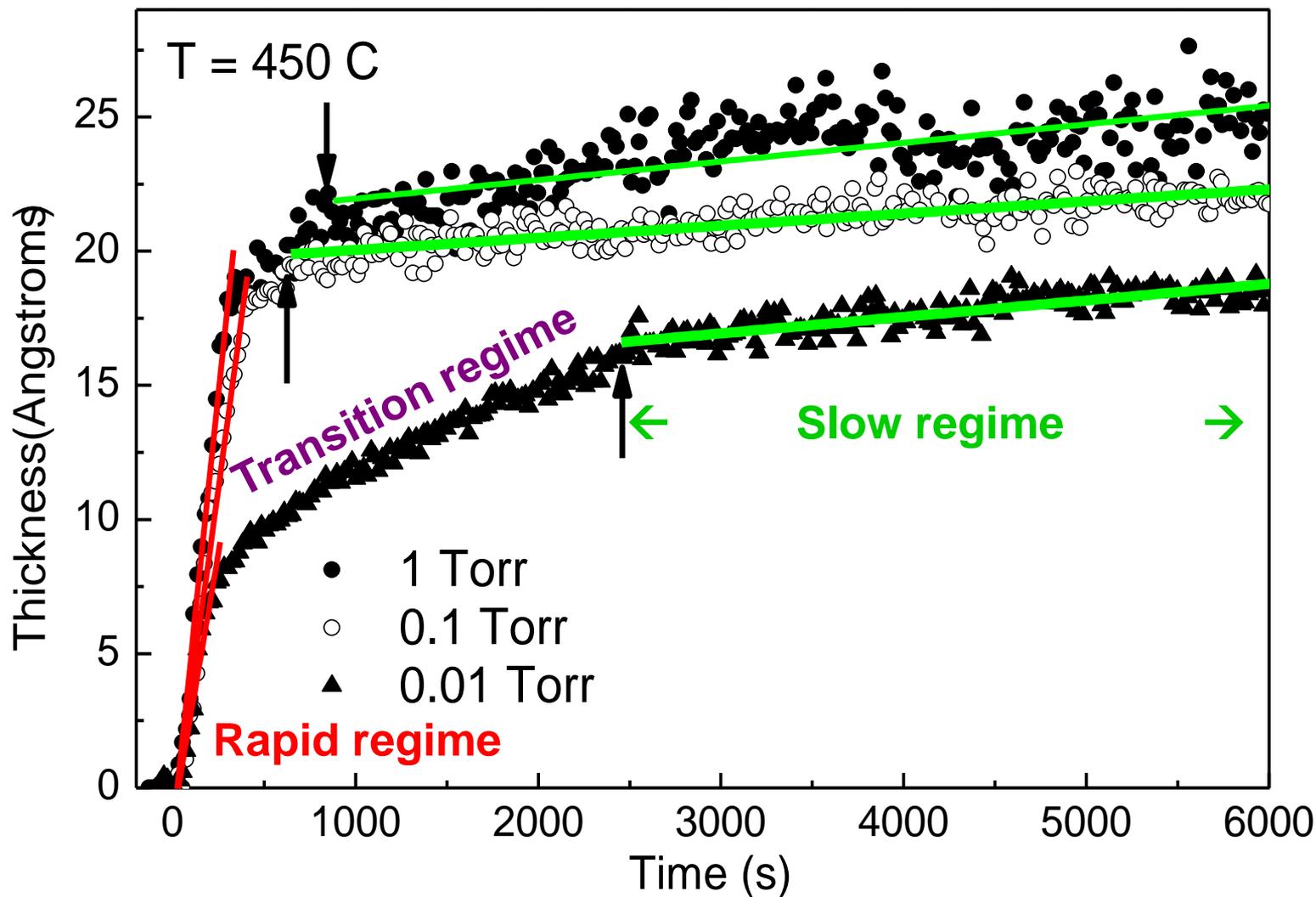


$P \approx 10^{-7} \text{ torr}$
or better

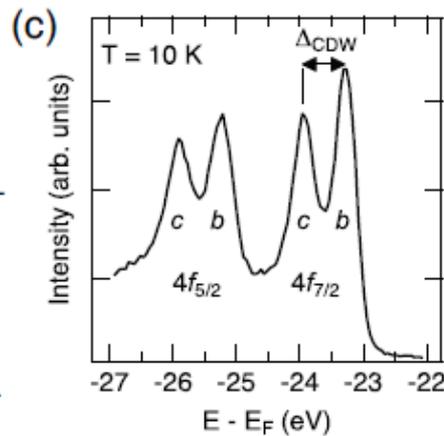
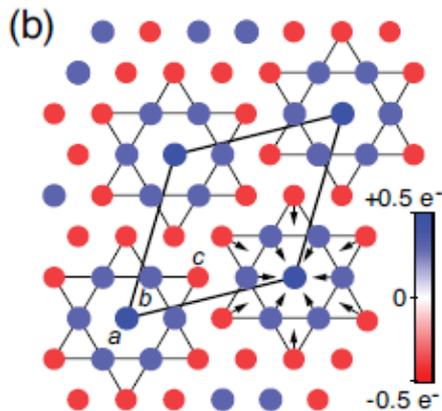
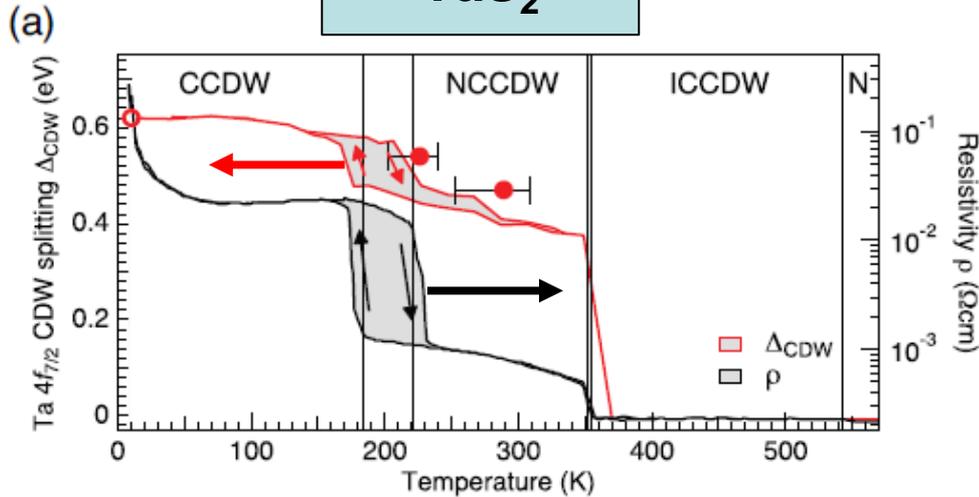
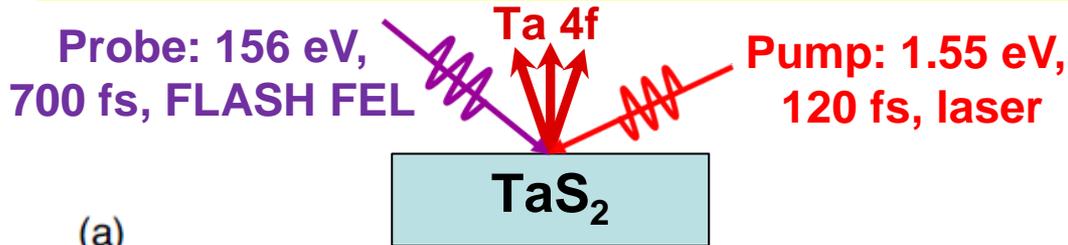
Ogletree et al., Rev. Sci. Instr. 73, 3872 (2002)—SR, ALS
 Bluhm, Salmeron, Schlögl—ALS, BESSY

Enta, Mun et al., Appl. Phys. Lett. 92, 012110 (2008); J. Appl. Phys. 103, 044104(2008)

Watching the oxide grow in real time: constant P, variable T

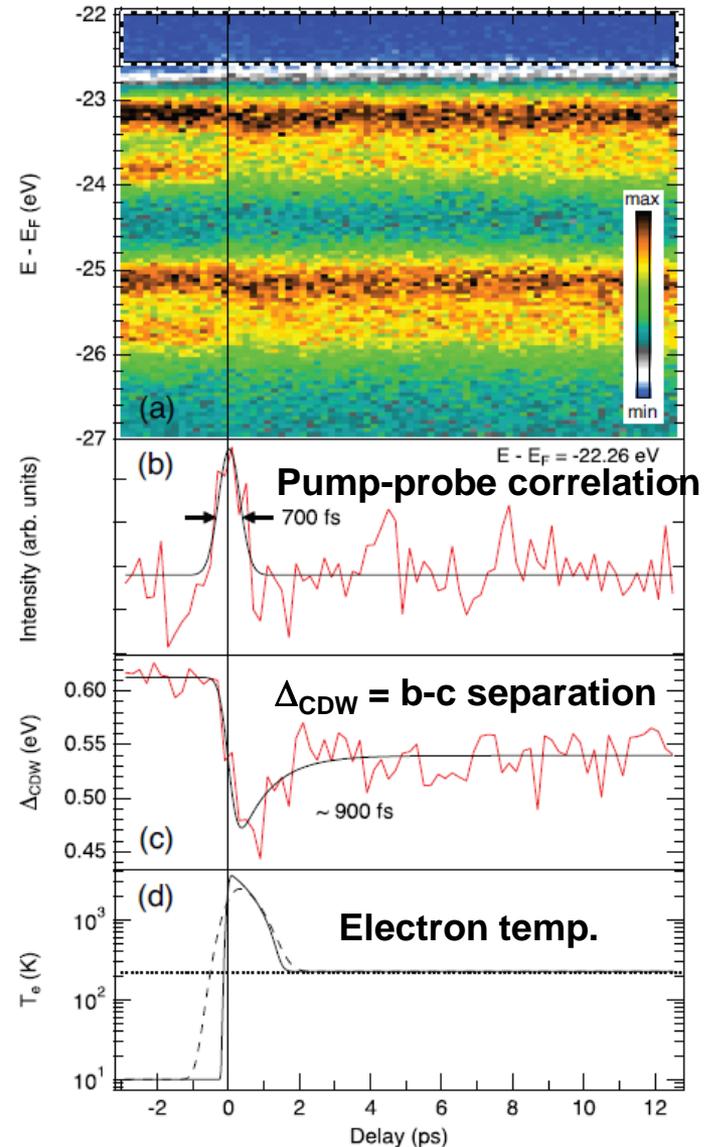


Time-resolved core-level XPS(\rightarrow HXPS) as a monitor of a charge density wave in Mott insulator $1T\text{-TaS}_2$

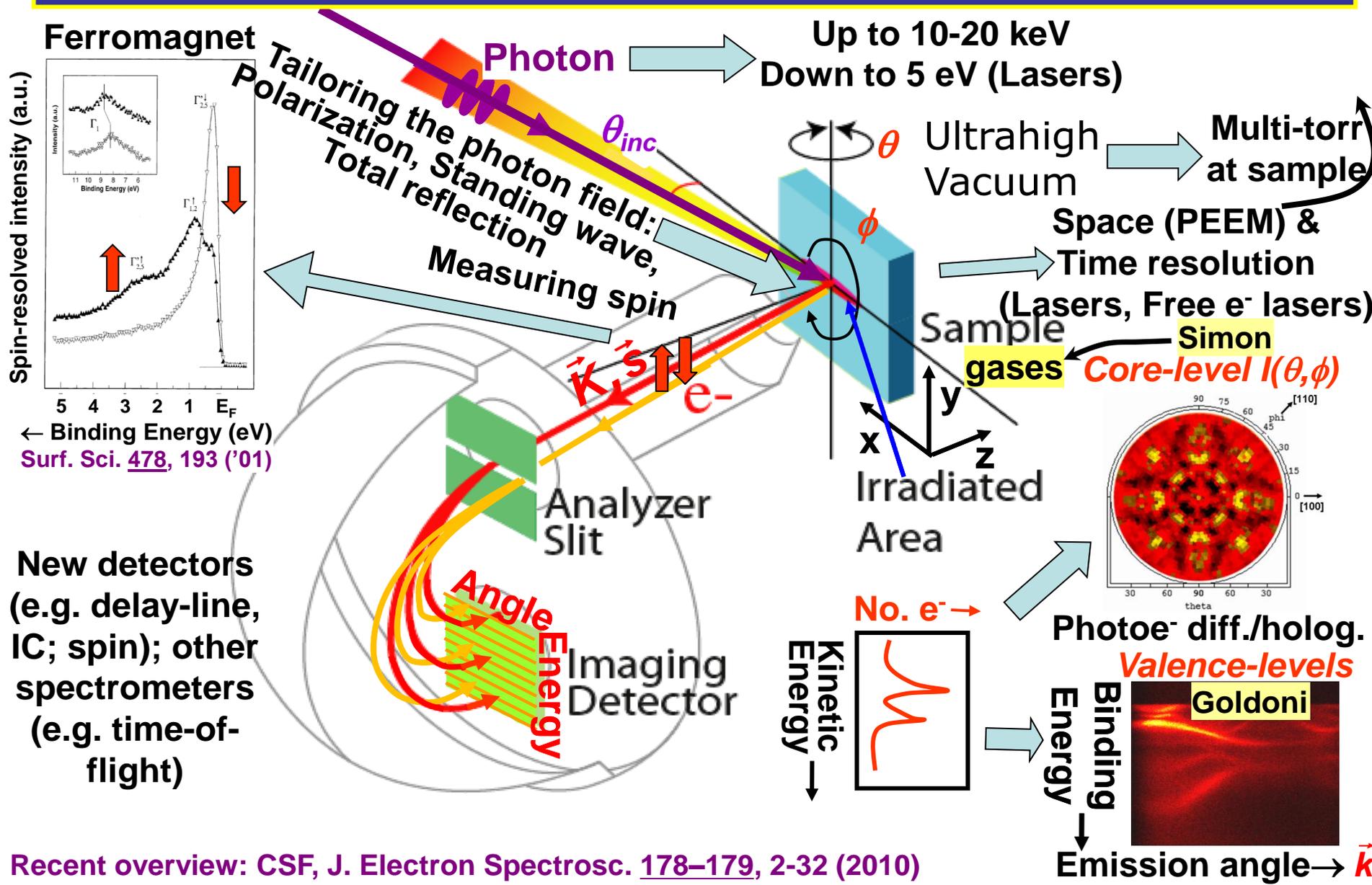


~~$a:b:c = 1:6:6$~~

Hellman et al., PRL 105, 187401 (2010)



Typical experimental geometry for energy- and angle-resolved photoemission measurements



Basic Concepts and Experiments

The photoelectric effect: 100+ years old! Much enhanced by synchrotron radiation, now moving to the harder x-ray regime and short-pulse FELs

Core-Level Photoemission— an element specific probe

Intensities and the Three-Step Model— quantitative near-surface analysis

Varying Surface and Bulk Sensitivity— vary energy or takeoff angle
standing waves

Chemical shifts— chemical state specific spectroscopy and stoichiometry

Multiplet Splittings— spin state, valence configuration(s)

Electron Screening and Satellite Structure— valence configuration(s)

Magnetic Circular Dichroism— long-range magnetic order

Resonant photoemission— enhancing a given atom's contribution

Photoelectron Diffraction and Holography— element-specific atomic structure

Valence-Level Photoemission

Band-Mapping in the Ultraviolet Photoemission Limit— valence electronic states
Fermi surfaces, for many materials and nanostructures,

Densities of States in the X-Ray Photoemission Limit— densities of states with
greater bulk sensitivity, low temp. measurements → band mapping

Some New Directions

Photoemission with Hard X-Rays— deeper probing of nanostructures,
densities of states, true bulk electronic structure

Photoemission with Standing Wave (SW) Excitation— selective probing of buried
interfaces and layers

Photoemission with: Spatial Resolution— nanoscale band mapping and
(several spectromicroscopies)

Temporal Resolution— lasers, FELs

@ Higher Pressures— several torr, Realtime kinetics

Miniussi et al., Ru T dependent lineshapes