

Introduction to SR for Environmental Sciences

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Take home message:

XRF + XAS + E-CT + XRD + ...

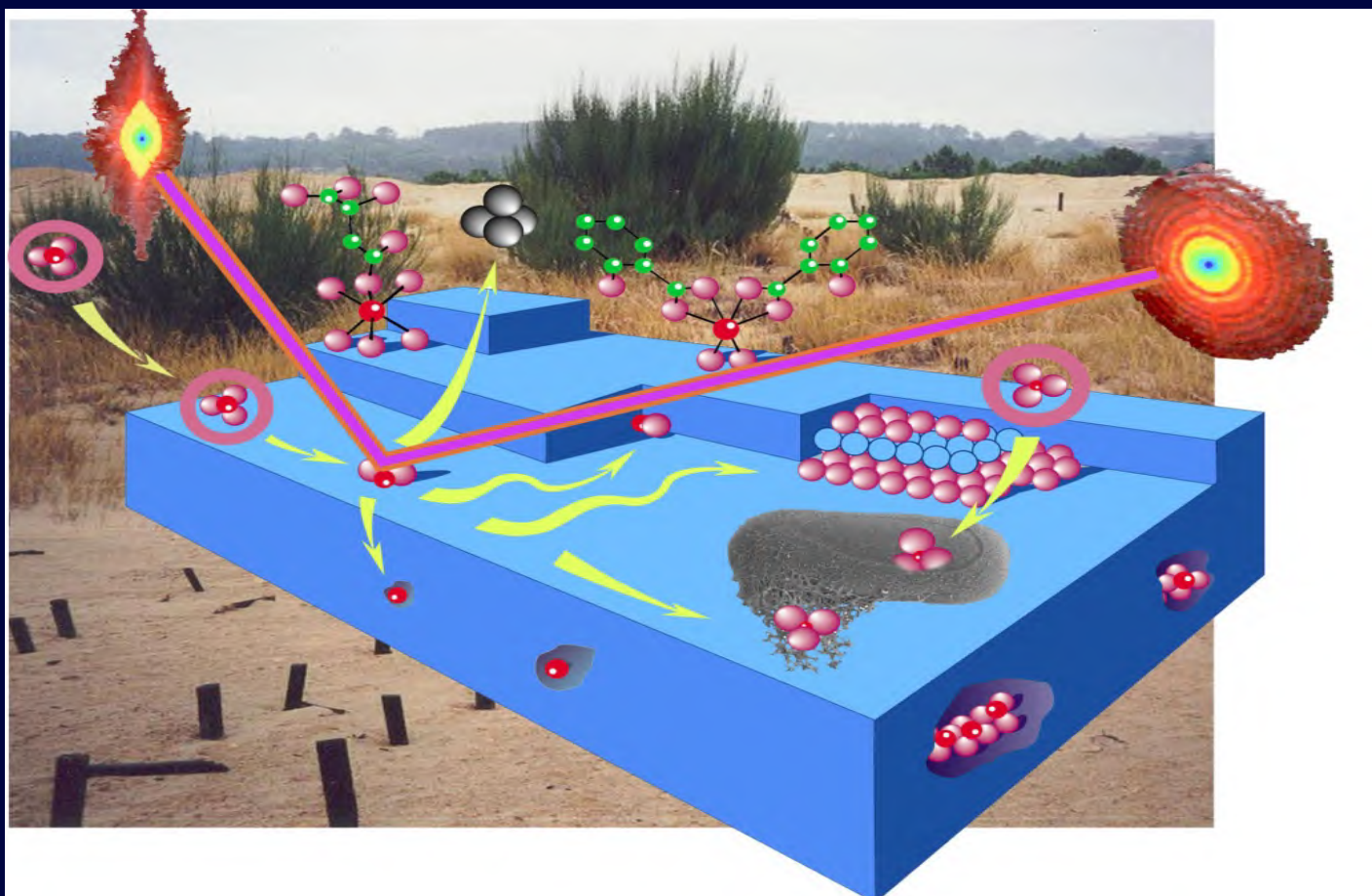
+

IMAGING

☀ Paradigm:

Speciation $\leftrightarrow$ Solubility $\leftrightarrow$ Bio-availability $\leftrightarrow$ TOXICITY

Toxicity of TE is linked to their bio-availability, itself linked to the chemical form at atomic scales $\rightarrow$ speciation



Bio-geochemistry goals:

- **Disentangle** concentration & speciation in the evaluation of ecotoxicity of trace elements
- **Identify** mobile/imobile forms and detoxification forms for the biosphere

WHY ?

... to evaluate/attenuate the chemical risk, to « **heal** » the environment

Difficulties :

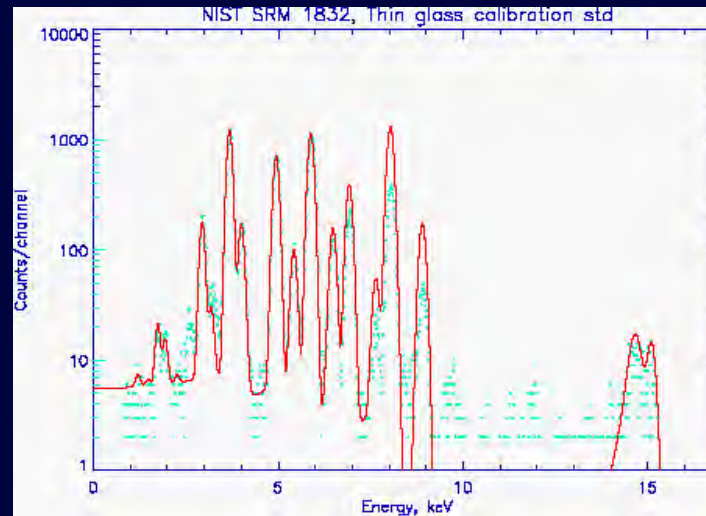
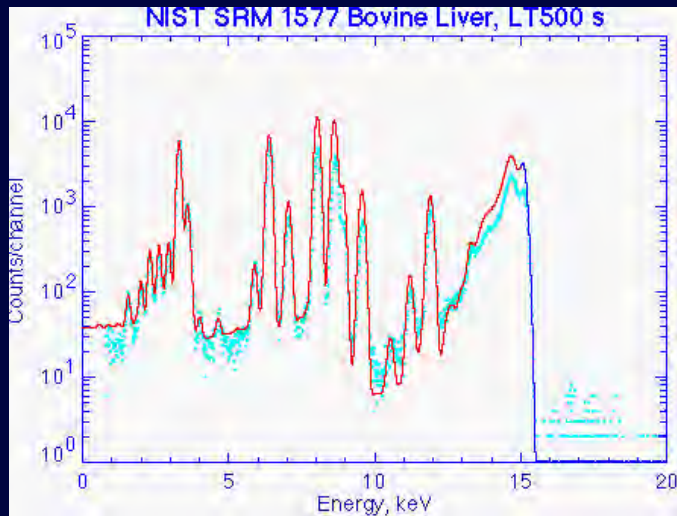
- scientific challenge because of the **low abundance** of most toxic elements (e.g. As, Hg, Cd), of the problems of the *in-situ* characterization of their chemistry and of the **high heterogeneity** of natural systems

Techniques :

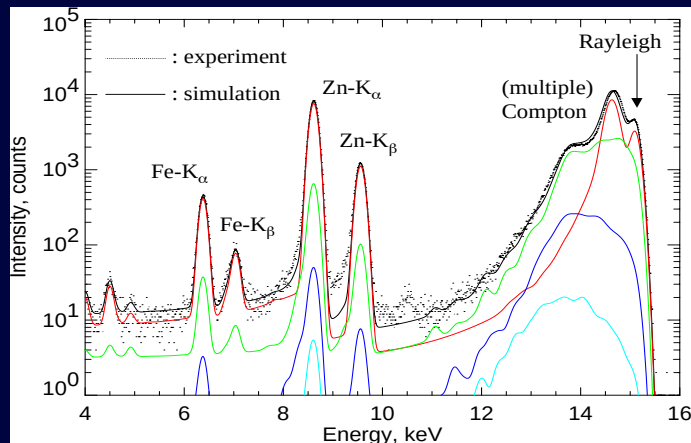
- X-Ray fluorescence (**XRF**) => chemical correlations
- X-Ray diffraction (**XRD**) => identification of host phases
- X-Ray absorption (**XAS**) => chemical forms at atomic scales
- X-ray tomography (**XCT**) => morphology & anatomy

μ Fluorescence : M-C Simulation / Experiment

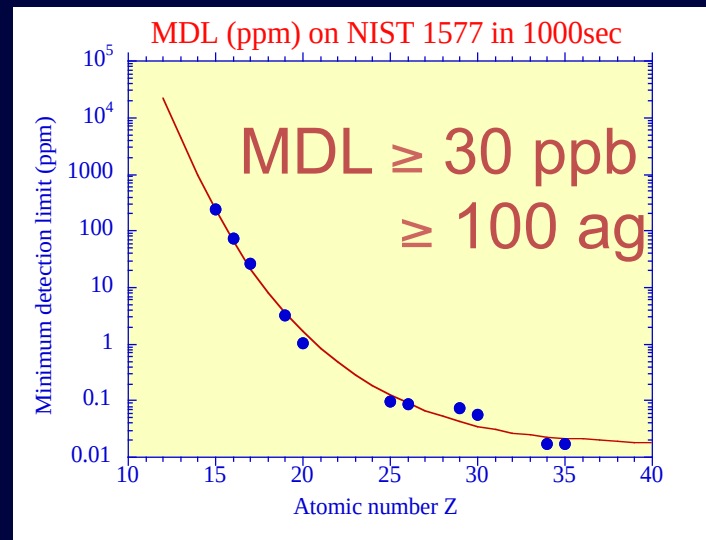
collab. : L. Vincze, T. Schoonjans, Mitac, Univ. of Antwerp



Currently:
1E13 ph/s
47 nm,
MDL=30 zg



Polypropylene 2 μ m



MDL, E=15 keV, 1 x 2 μ m²

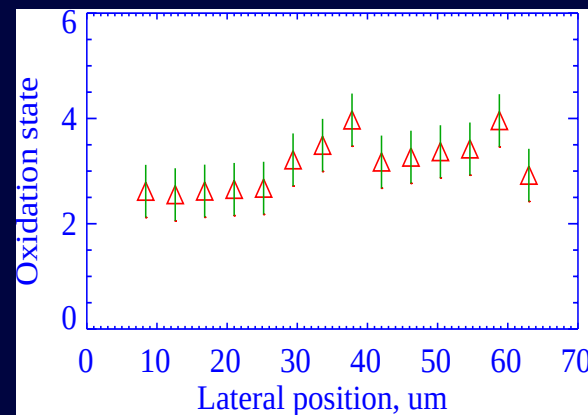
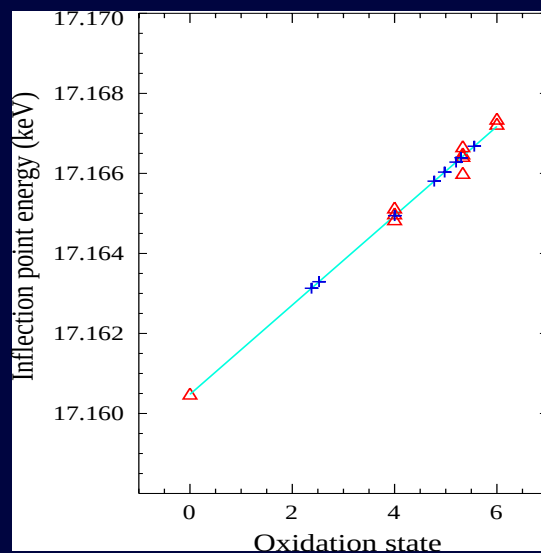
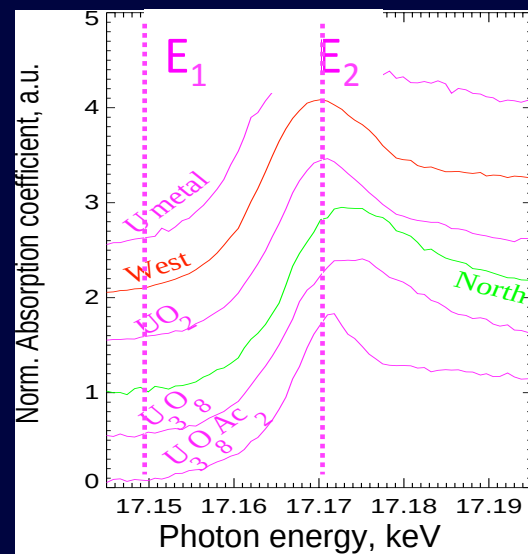
μ - speciation: post-Chernobyl aerosols

XANES on Uranium L_{III}

Univ. of Ås, Norway: B. Salbu, T. Krekling, O.C. Lind
Univ. of Antwerp, Belgium, K. Janssens

- ★ calibration: U metal, UO_2 and U_3O_8 ($0^+ - 6^+$)
- ★ $FWHM \geq 0.15$ eV, $\delta E \approx 0.8$ eV/q
- ★ **2 x 5 μm** , flux 10^9 ph/s., **2 - 5** sec./point

Salbu *et al*,
Nucl. Instr.& Meth. A **467-468**,
1249-1252, 2001

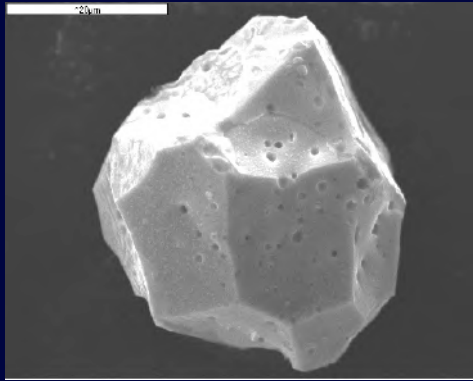


Xanes line scan
across a particle

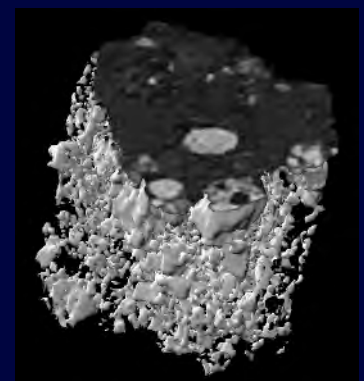
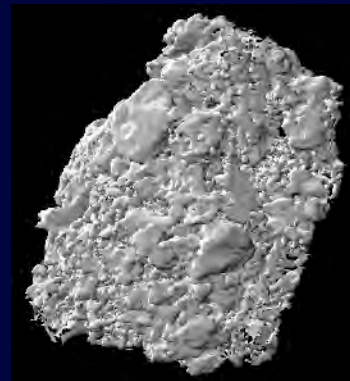
Spectroscopy = starting point of combined analyses:

- Xanes calibration spectra
- multi-elemental analysis

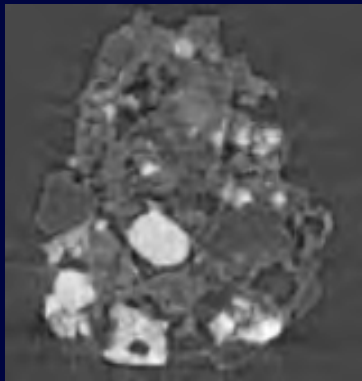
SEM Image



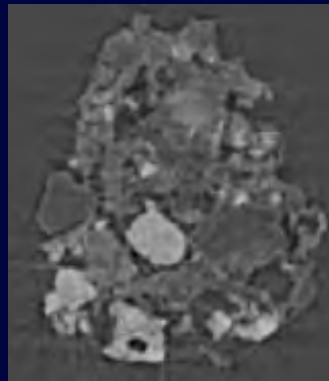
μ -tomo, 1 μ m res., E = 20 keV



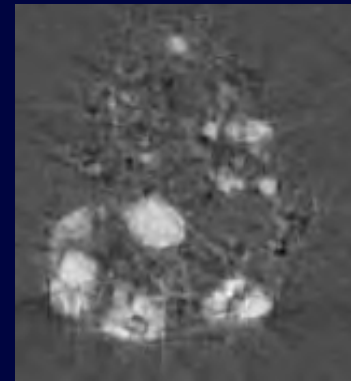
Xanes-Tomography = XANET / Spectro-tomography



E $\approx$ 17.2 keV



E $\approx$ 17.1 keV



Difference

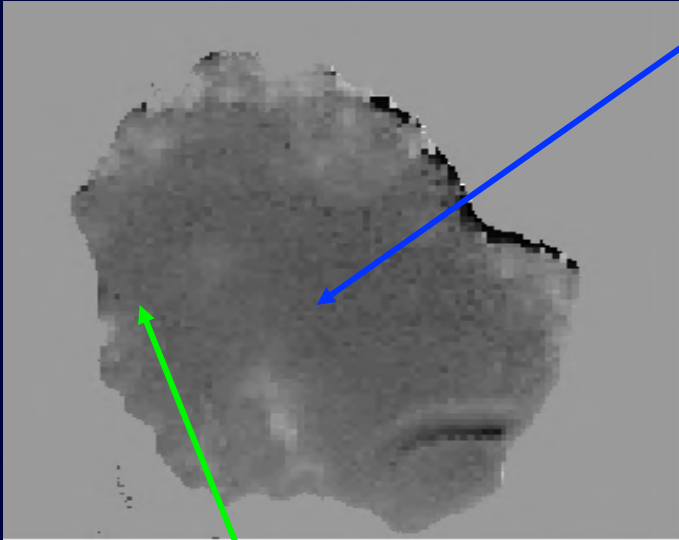
Dual energy tomography: (U. Bonse, 1986)

Particle weathering rate vs. U oxidation state

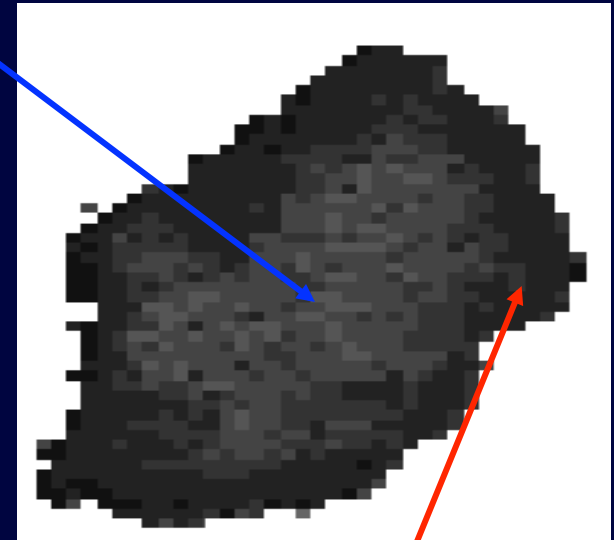
West: initial release
(explosion)

Oxidation
state: $+4 \pm 0.5$

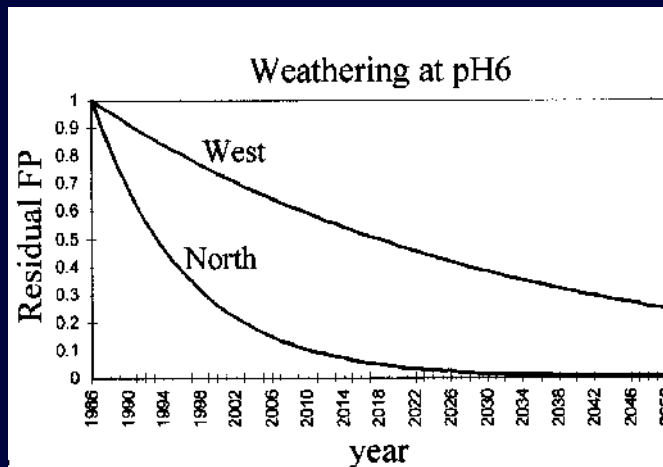
North: subsequent release
(fire)



Oxidation
state: $+2.5 \pm 0.5$



Oxidation
state: $+5 \pm 0.5$

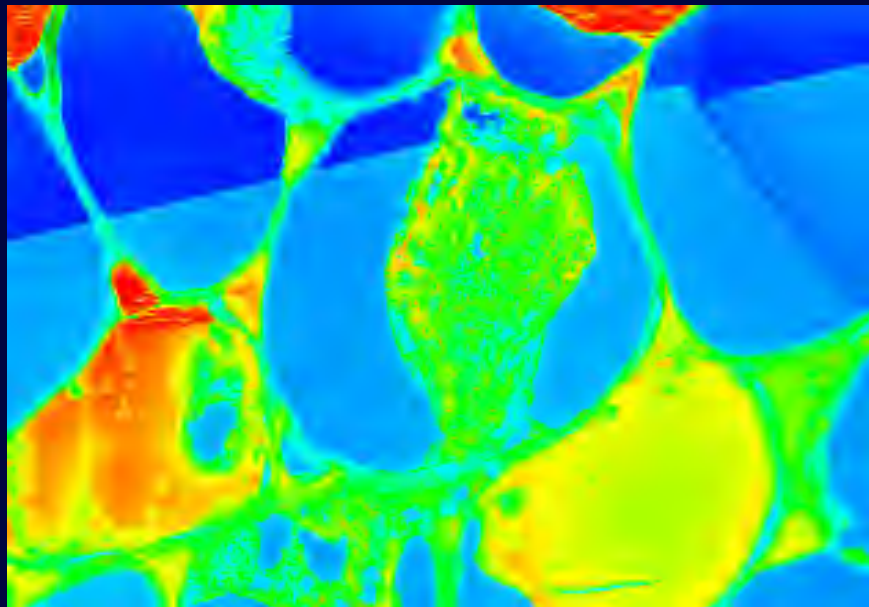
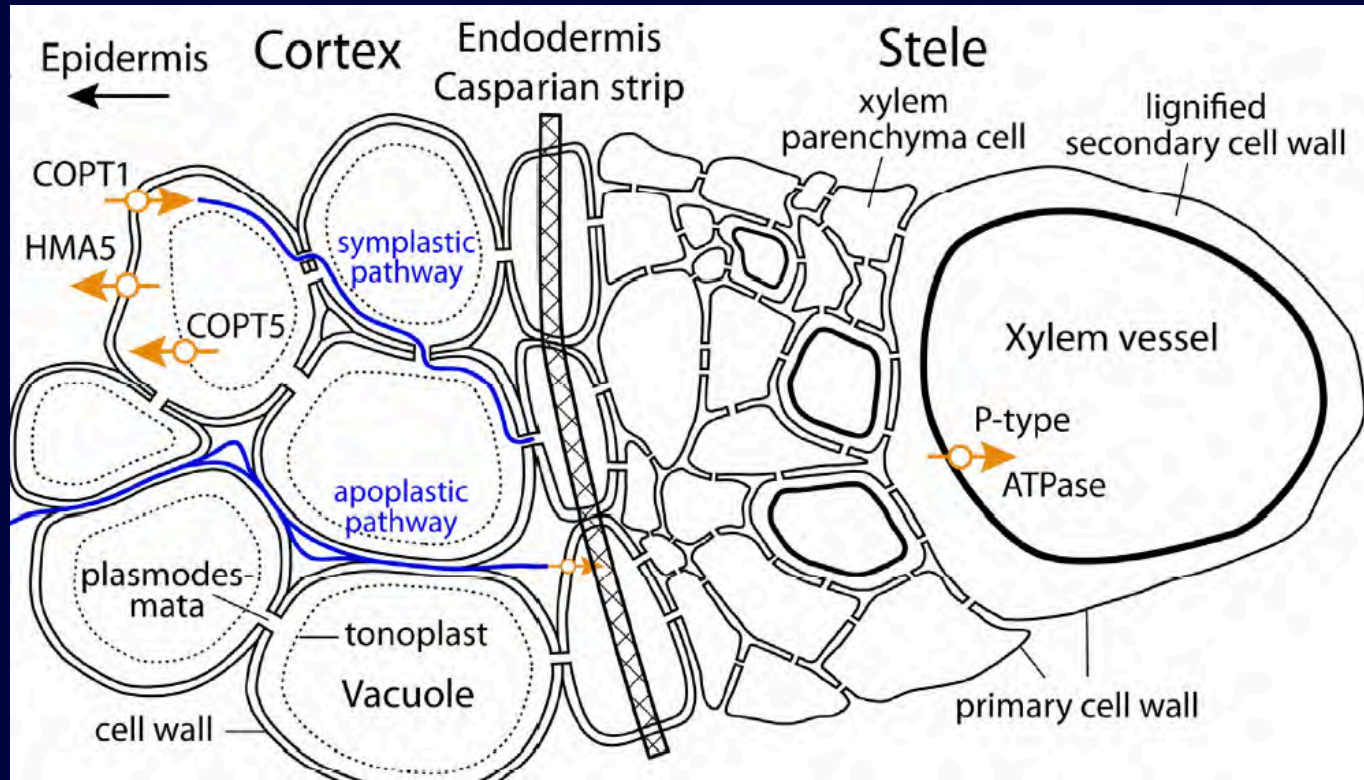


Cu storage in roots and export from roots to shoots

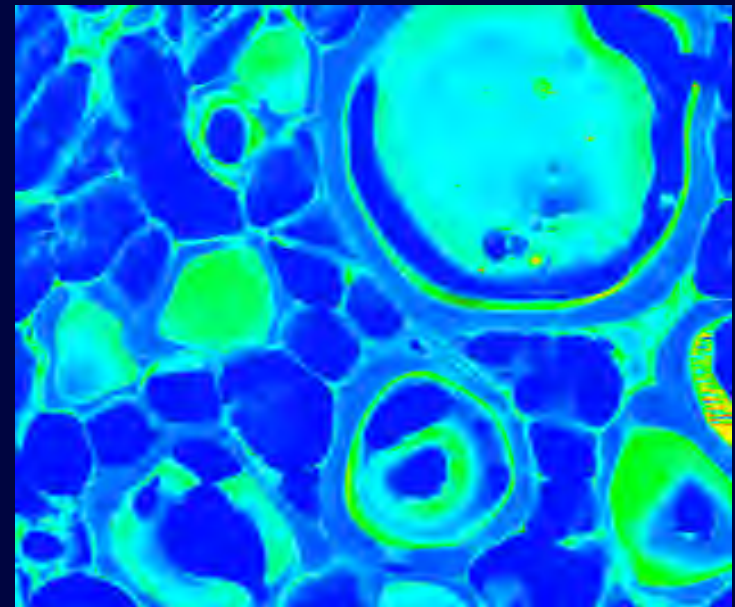
(A. Manceau, ISTerre, CNRS&UJF)

Copper is essential for:

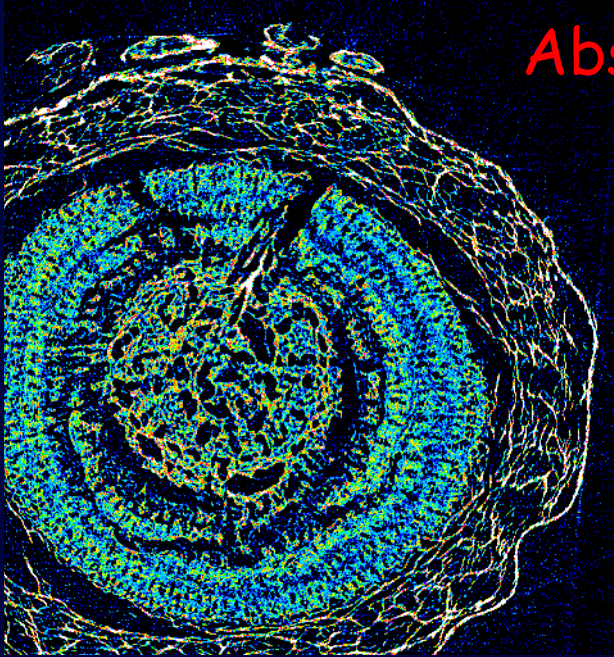
- photosynthesis
- mitochondrial respiration
- lignin synthesis
- root growth
- ethylene sensing
- reactive oxygen metabolism



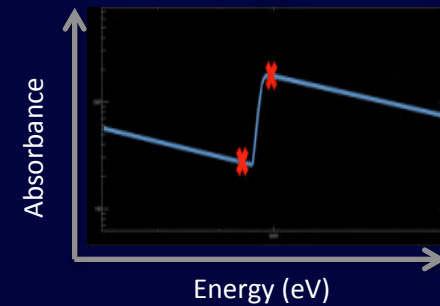
Res. $\approx$
70 nm



Absorption tomography (from 8.3.2)

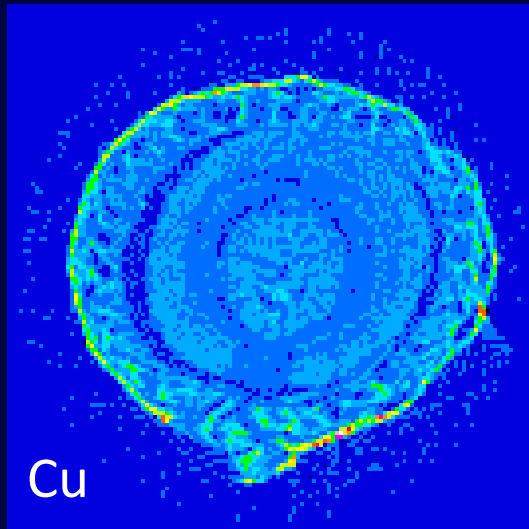


Cu, Zn K edges



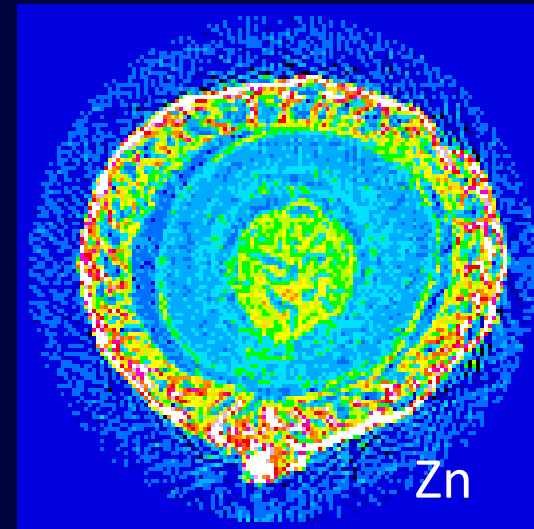
Absorption contrast (Zn)
(from ALS-8.3.2)

Fluorescence tomography (from ALS-10.3.2)

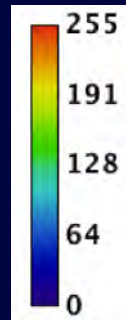


Cu

5.4 mm

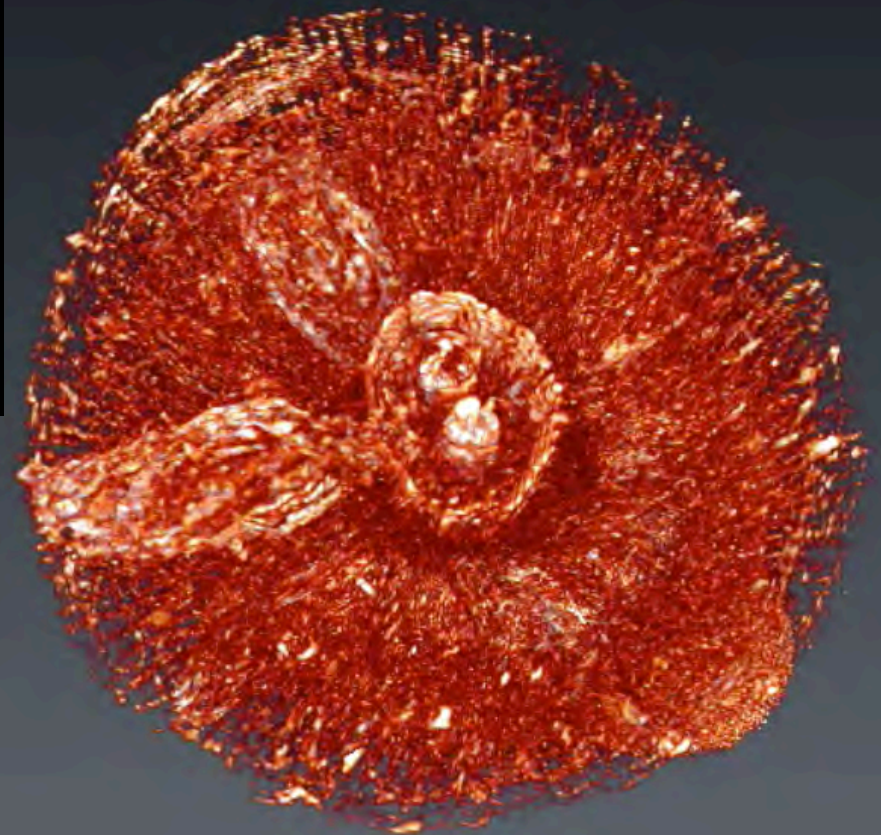
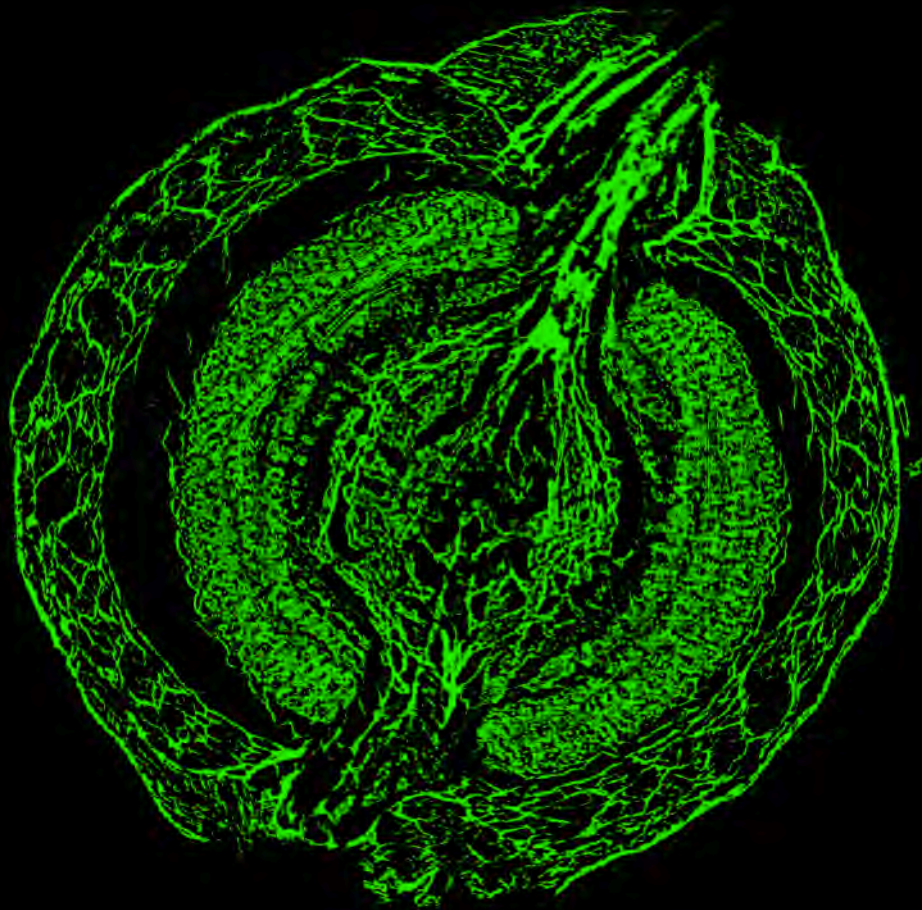


Zn



Res. 5 μ m

Distribution of Cu and Zn by AT/FT



3D XANES tomography

XANES
absorption spectroscopy
oxidation state

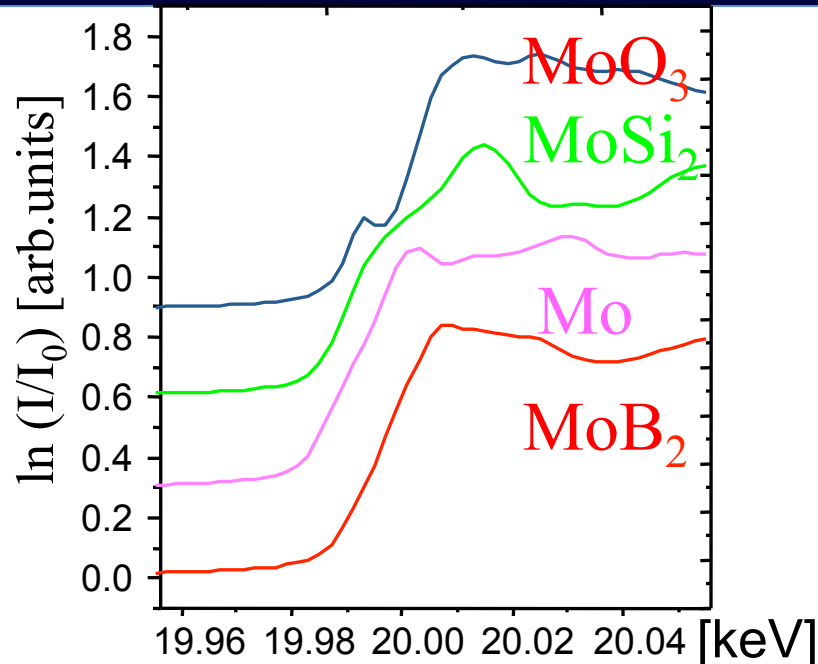
X-ray tomography
absorption radiography
morphology

Tomography with chemical speciation

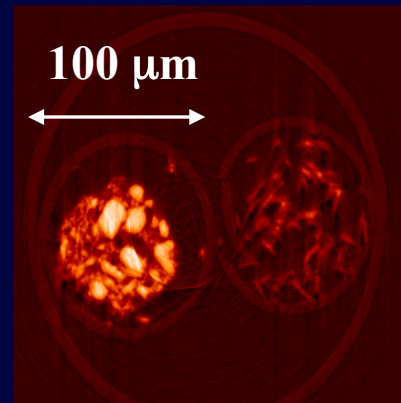
Sample: Mo
compounds in cap.
($\varnothing 100 \mu\text{m}$)

$t_{\text{exp.}}/\text{image}$: 0.2 s

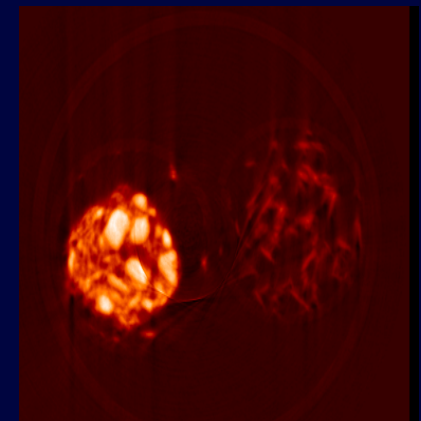
resolution: $\sim 1 \mu\text{m}$



$E_{\text{inc}} = 19.956 \text{ keV}$



$E_{\text{fin}} = 20.056 \text{ keV}$



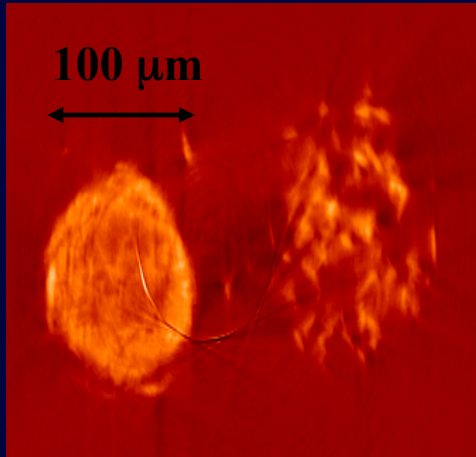
Rau et al, Nucl. Instr.&
Meth. A, 200C, 445-450,
2003

Voxel resolved XANES spectra (2eV)

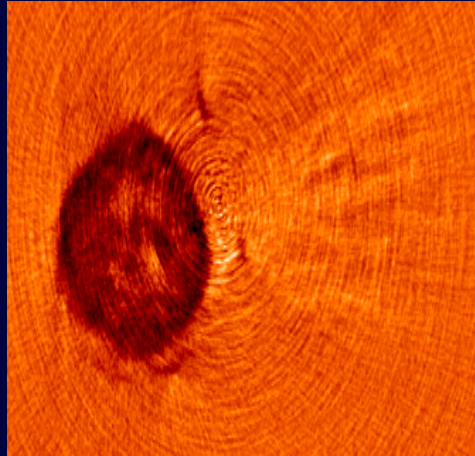
-Scientific case: phase separations in metals

3D imaging: Difference of same reconstructed slice at characteristic energies

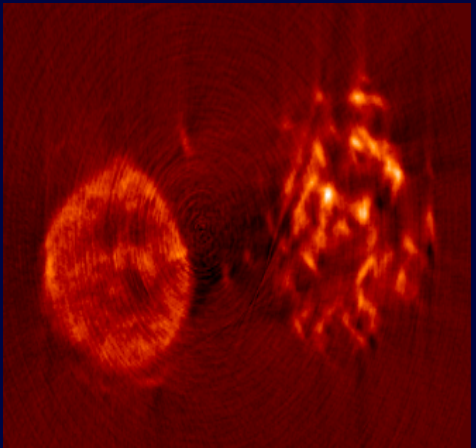
energy 4-energy 2



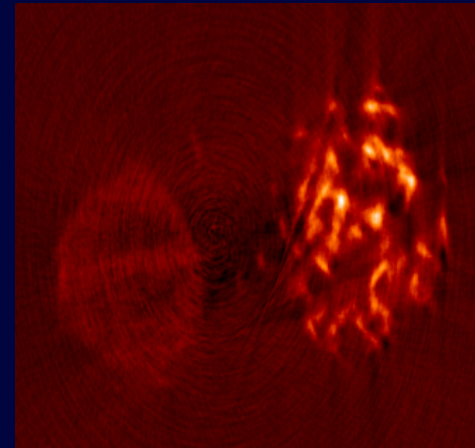
energy 5-energy 6



energy 5-energy 3

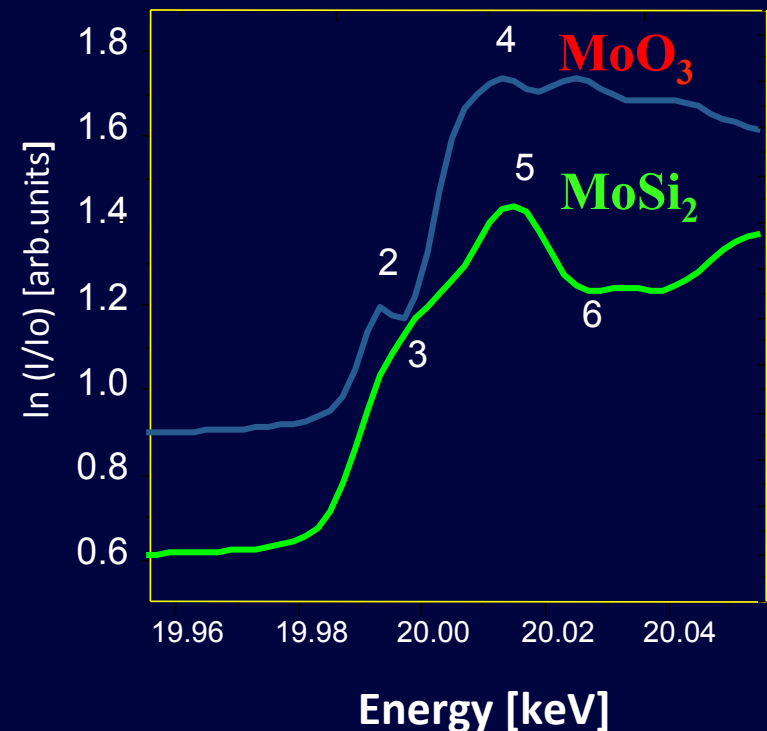


energy 6-energy 3



3D XANES full field imaging
with a CCD HR camera
(0.7 μm resolution)

PIXEL resolved XANES spectra



MoSi₂

MoO₃



EnviroSynch

Environment – Synchrotron

Erasmus for All Master
in **EcoX** Equipex (A. Manceau)



- 5th year students; program of **18 months**
- 2 semesters of Courses + practicals/lab = 2 x 30 ECTS
- 1 semester of internship in **member/partner** labs
- 30% redundancy: **backbone** + **options** on 3 campuses



New member



Members

FRANCE – Université J. Fourier, Grenoble

ASimionovici, BLanson, PDonnadieu, LSpadini, MMunoz

Royaume Uni – Manchester University

MDenecke +

Allemagne – Karlsruhe Institute of Technology

JGöttlicher +

Synchrotron partners

ESRF, Grenoble

PGlatzel, PCloetens, ASolé,
JSusini, MCotte

Diamond, Oxford

FMosselmans

ANKA, Karlsruhe

JGöttlicher

CLS, Saskatoon

IPickering, GGeorge

Invited lecturers

PFenter, KNagy - Univ. of Illinois, RDähn- PSI Villigen, AManceau - ISTerre,
TReich - Univ. of Mainz, LMichot - LEM + + +

Courses in Grenoble

Backbone

X-ray Absorption/Emission Spectroscopy – PGlatzel	24 h – 3 ECTS
Crystal structure & organization of finely divided solids – BLanson	56h – 6 ECTS
Methods and tools of modern mineralogy - ASimionovici + MMunoz	24 h – 3 ECTS
Imaging with electrons: SEM and TEM – PDonnadieu	24 h – 3 ECTS
Mineralogy and Environment – MMunoz + ASimionovici	24 h – 3 ECTS
Environmental Chemistry – LSpadini	24 h – 3 ECTS
Soil chemistry and pollution – ASimionovici	24 h – 3 ECTS
XCT, XAS imag., Hyperspectral imag. – PCloetens, ASolé, ASimionovici	24 h – 3 ECTS
Lecturers – PFenter, KNagy, RDähn, AManceau, TReich, FMosselmans, LMichot	24 h – 3 ECTS
TOTAL	248h - 30 ECTS

Starting $\geq$ Fall 2014