



UNIVERSITÀ DEGLI STUDI DI TRIESTE

Dipartimento di  
Scienze Chimiche e Farmaceutiche

## Hybrid organic-gold nanoparticles for therapy and imaging

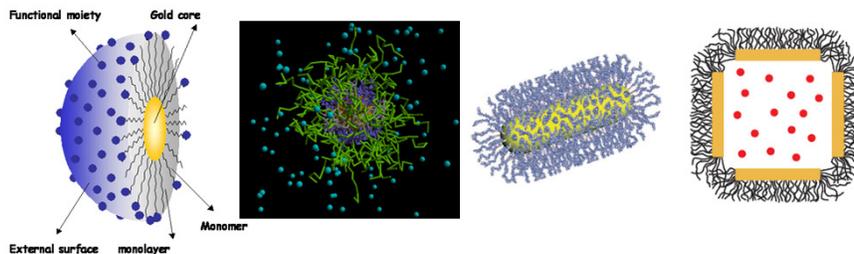
Prof. Lucia Pasquato

Second Conference Nano-Bio-Med-2013

ICTP, Trieste, 14th-18th October 2013

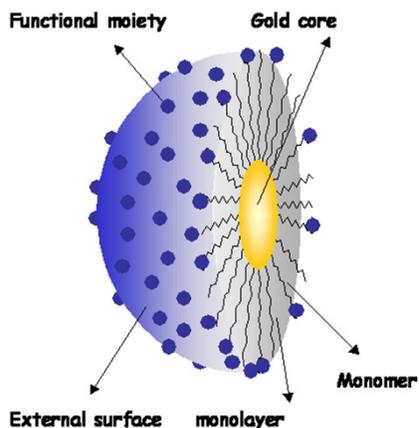
### WHY GOLD?

- Au NPs of many different sizes (1-150 nm) can be fabricated in a controlled fashion with control over size dispersity and shape.
  - the chemistry to protect the gold surface is well know
  - a high density of ligands can be appended for targeting and/or drug-delivery.
  - the monolayer can be modify with a variety of methodologies
  - functional diversity can be readily achieved by creation of a multifunctional monolayer.
  - the physical properties of the gold core may be exploited
- ➔ GNP are ideal nanomaterials to serve as probes for investigating the influence of nanoparticle size, shape, and surface chemistry on their biological interactions.



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## metal nanoparticles protected by a self-assembled organic layer



### the metal core

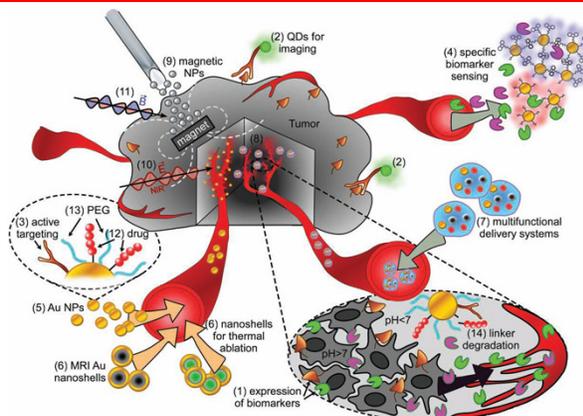
- ✓ the size of the metal core determines the optical and electronic properties
- ✓ the reaction conditions enable to tune the size, the shape and the dispersion

### the self-assembled monolayer

- ✓ it is formed by a self-assembly process
- ✓ it is responsible for the stability
  - the solubility
  - the interaction with the environment
  - the function of the NP

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## GNPs for biomedical applications



C. Minelli, S. B. Lowe, M. M. Stevens *Small* **2010**, *6*, 2336-2357.

Au NPs for **Imaging**  
 Au NPs for **Diagnosis**  
 Au NPs for **Therapy**

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## Gold NPs in Clinical trials

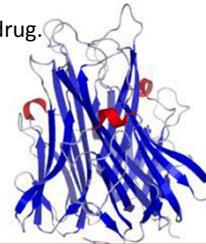
### Phase I and Pharmacokinetic of CYT-6091

L. Tamarkin et al. *Clin. Cancer Res.* 2010, 16, 6139.

CYT-6091 – which is comprised of recombinant human tumor necrosis factor alpha (rhTNF) Bound to the surface of PEGylated 27 nm colloidal gold particles  
85% complete remission rates are routinely observed.

PEGylation of GNPs

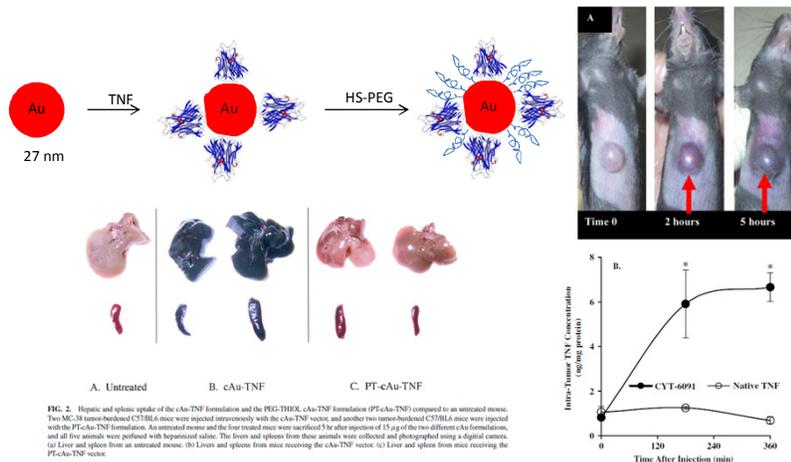
- avoids immediate uptake by the reticuloendothelial reticulum (RES)
  - Reduces the toxicity of rhTNF
  - Allows the nanomedicine to sequester in solid tumors
- maximal antitumor responses were achieved at lower doses of drug.



TNF-alpha is a homotrimer with a subunit molecular mass of 17 kDa

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## Gold NPs in Clinical trials



CytImmune is now collaborating with AstraZeneca to add a chemotherapeutic drug to further improve the killing power of CYT-6091.



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### Delivery

drug delivery via 'active' and 'passive' targeting

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### Delivery

Monolayer coated

Targeted delivery

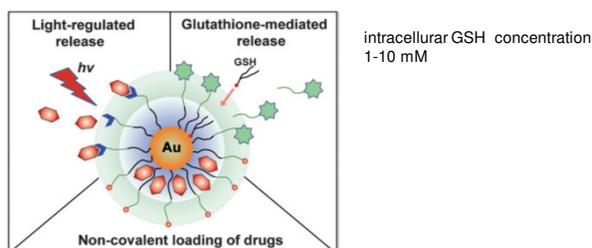
Biomolecule coated

Schematic presentation of the two AuNP surface structures commonly employed in delivery applications.

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## multimodal drug delivery gold nanoparticles

- Prodrugs can be covalently conjugated to AuNPs via cleavable linkers.
- Hydrophobic drugs can be loaded onto AuNPs monolayer exploiting hydrophobic interactions, without structural modification of the drug payload.
- The release can be triggered by either internal (e.g. glutathione) or external (e.g. light) stimuli.



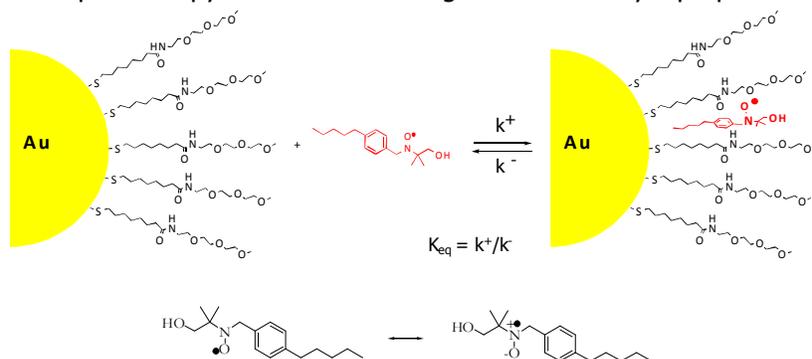
Encapsulation or conjugation of therapeutic agents with NPs has proven to be a successful approach

- to improving therapeutic's water solubility
- circulation time in the body
- uptake, in turn reducing drug payloads and associated toxic effects.

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## non-covalent loading

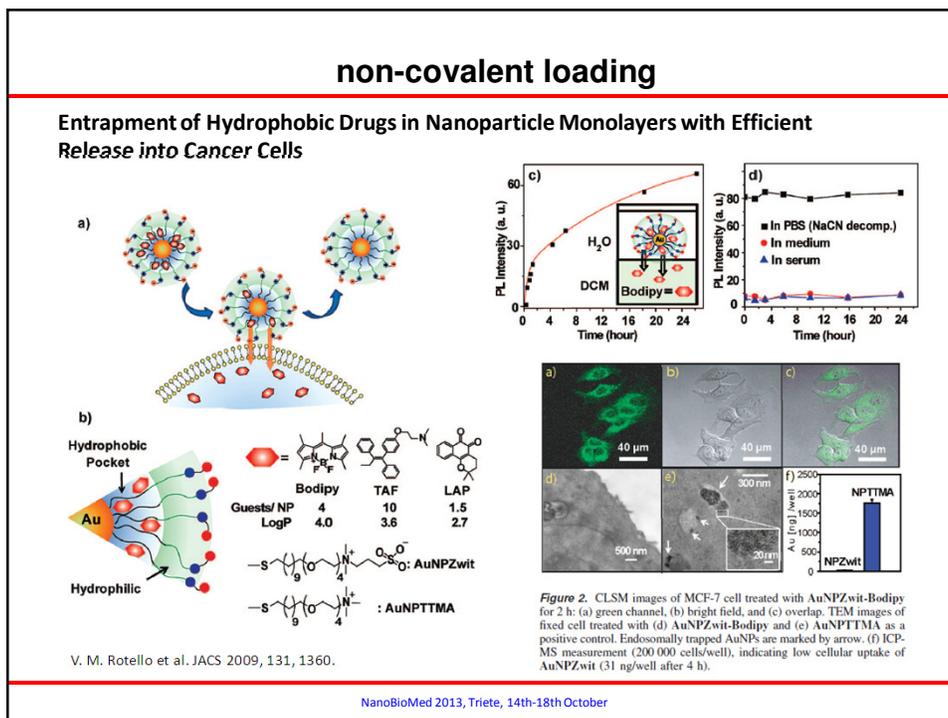
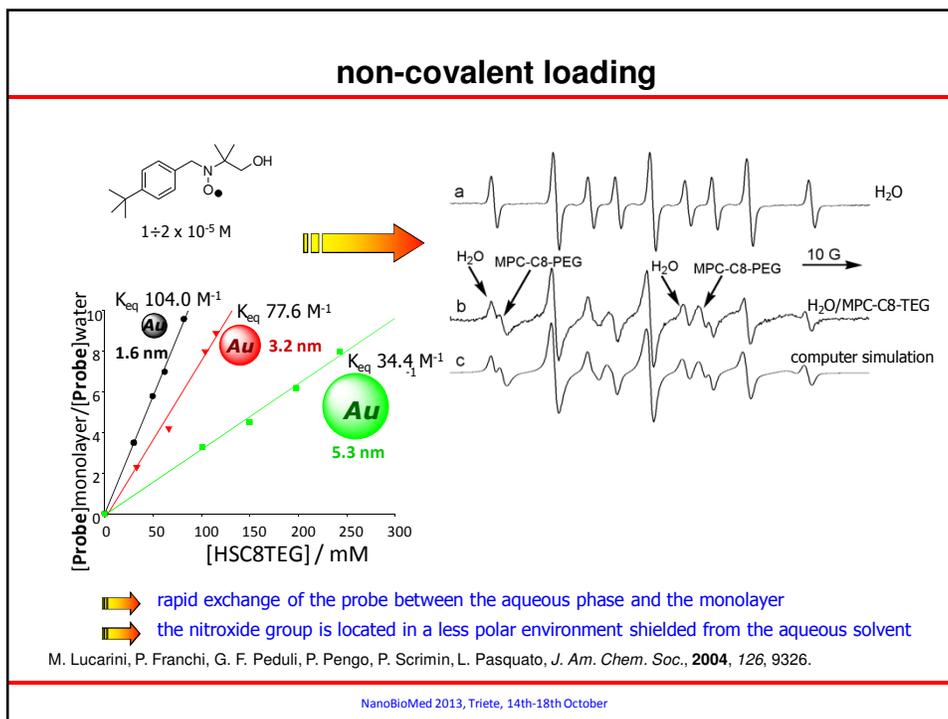
EPR Spectroscopy as a tool to investigate the monolayer properties

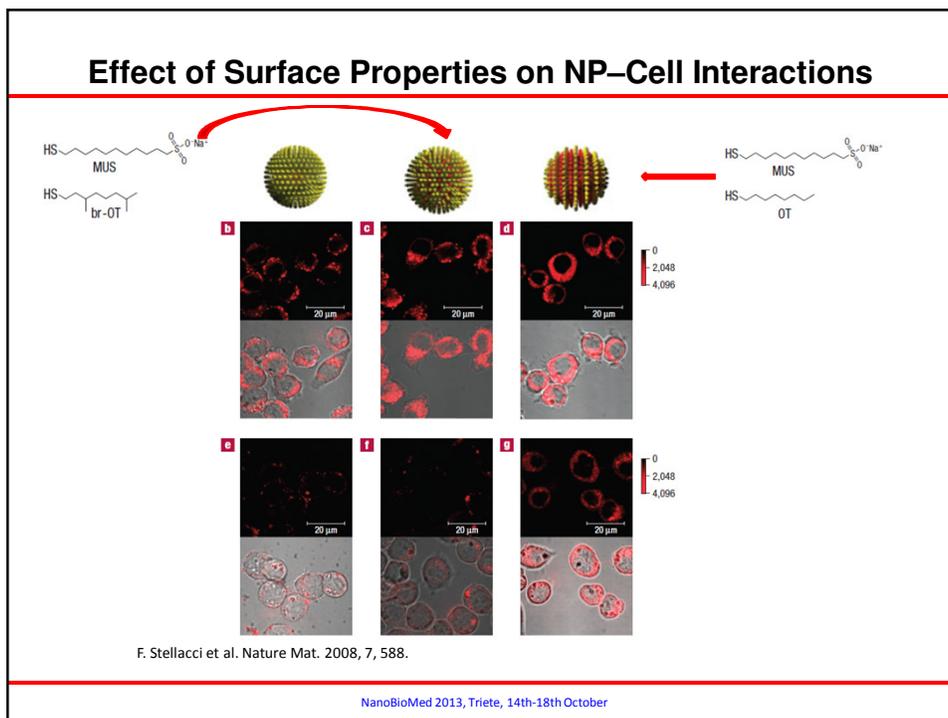
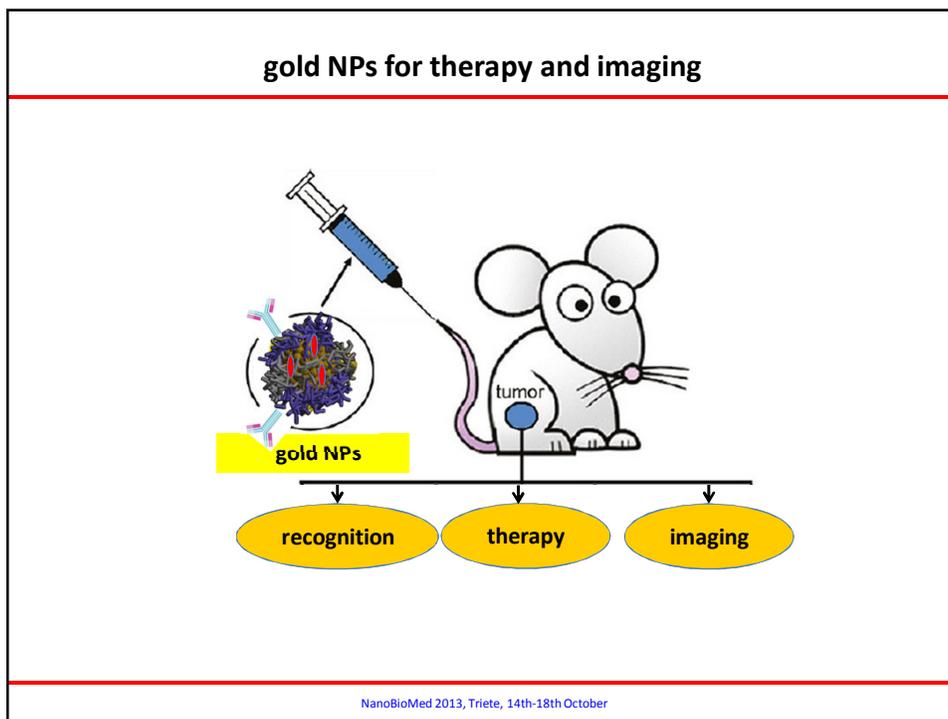


- the hyperfine coupling constants  $a(N)$  and  $a(2H_\beta)$  are larger in polar media

M. Lucarini, P. Franchi, G. F. Peduli, P. Pengo, P. Scrimin, L. Pasquato, *J. Am. Chem. Soc.*, **2004**, *126*, 9326.

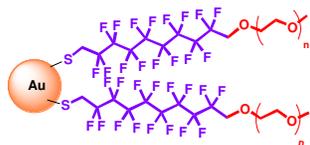
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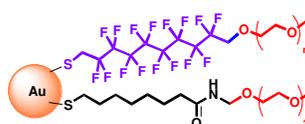


## outline

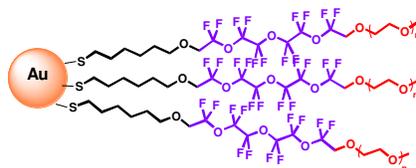
## gold nanoparticles protected by amphiphilic fluorinated ligands



MPC-F8-PEG



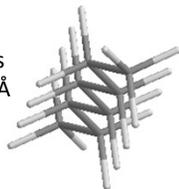
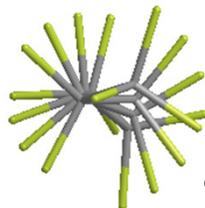
MPC-C8-TEG/F8-PEG

MPC-C6-FEO<sub>3</sub>-PEG

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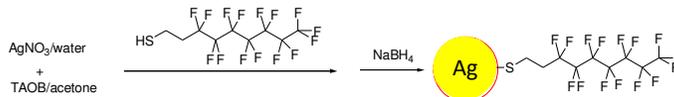
Comparison *F*- vs. *H*-alkyl chains

- (per)fluoroalkyl (*F*) chains are bulkier than alkyl (*H*) chains (cross section of  $\sim 30 \text{ \AA}^2$  and  $20 \text{ \AA}^2$  respectively)
- they have a helical structure rather than a planar zig-zag structure
- monolayers of surfactants containing *F*-chains are more stable than those formed from *H*-chains
- *F*-chains are amphiphobic (*i.e.*, at the same time hydrophobic and lipophobic) this difference promotes self-aggregation
- short perfluoro chains are considered biocompatible and nontoxic
- fluoro-compounds are used for biomedical applications, *e.g.* as contrast agents for  $^{19}\text{F}$ -MRI and ultrasonography, as drug delivery systems or as  $\text{O}_2$  carriers.

van der Waals  
diameter  $4.2 \text{ \AA}$ C<sub>8</sub>H<sub>18</sub>C<sub>8</sub>F<sub>18</sub>van der Waals  
diameter  $5.7 \text{ \AA}$ 

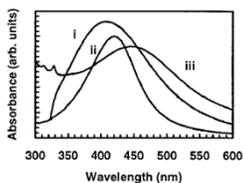
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## metal nanoparticles protected by fluorinated ligands

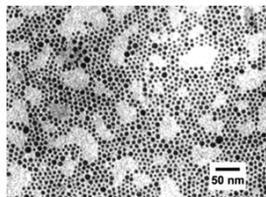


K. P. Johnston, B. A. Korgel et al. *JACS* **2000**, *122*, 4245.

dispersion in acetone and liquid and sc.  $\text{CO}_2$



UV-visible absorbance spectra of AgNPs (i) coated with fluorinated ligands dispersed in acetone; (ii) coated with hydrocarbon ligands dispersed in hexane; (iii) coated with fluorinated ligands dispersed in sc- $\text{CO}_2$ .



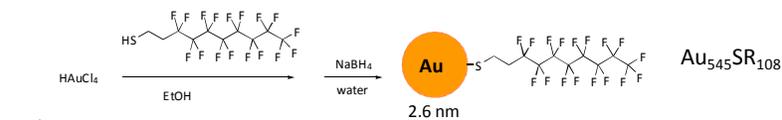
TEM image of silver nanocrystals coated with fluorinated ligands.

average size 5.5 nm

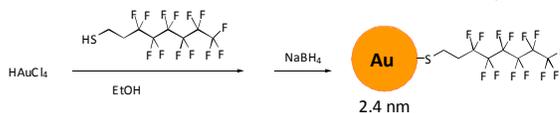
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## gold nanoparticles protected by fluorinated ligands

■ use of commercially available *F*-thiols

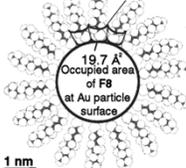


T. Yonezawa, et al. *Langmuir*, **2001**, *17*, 2291-2293.

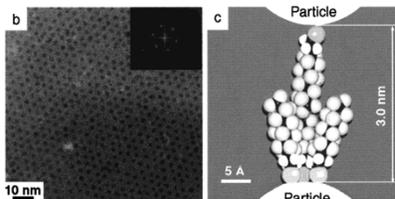


insoluble in common organic solvents:  
ethanol, acetone, chloroform, DMSO

Occupied area of FB  
at  $-\text{CH}_2\text{CF}_3$ - moiety  
35.3 Å<sup>2</sup>



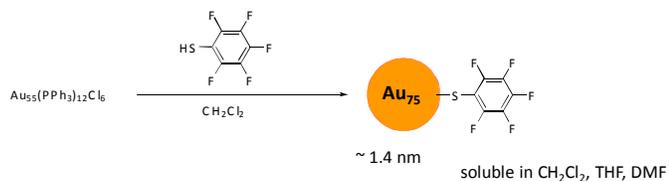
a dispersion in HCFC-225 dropcasted  
on a carbon-coated copper grid displays  
ordered hexagonal assembly



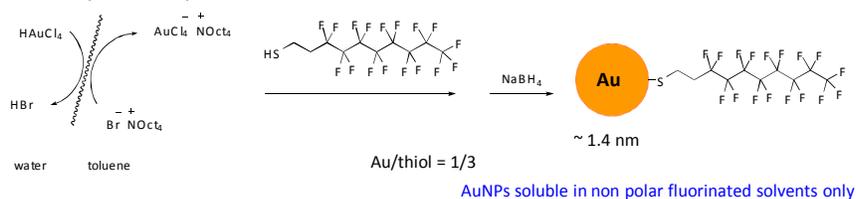
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## gold nanoparticles protected by fluorinated ligands

### ■ synthesis by ligand exchange



### ■ synthesis by Brust reaction

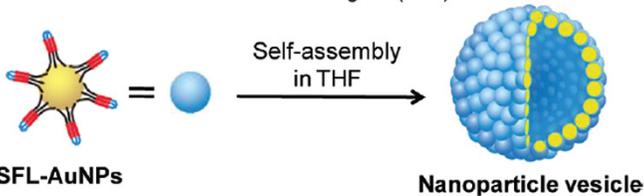
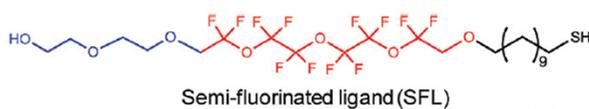


A. Dass, R. Guo, J. B. Tracy, R. Balasubramanian, A. D. Douglas, R. W. Murray *Langmuir*, **2008**, *24*, 310-315.

■ Au/thiol = 3/1 average core diameter 2.5 nm; J. Im, A. Chandekar, J. E. Whitten *Langmuir*, **2009**, *25*, 4288-4292.

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## gold nanoparticles protected by fluorinated ligands



5, 10, 20 nm

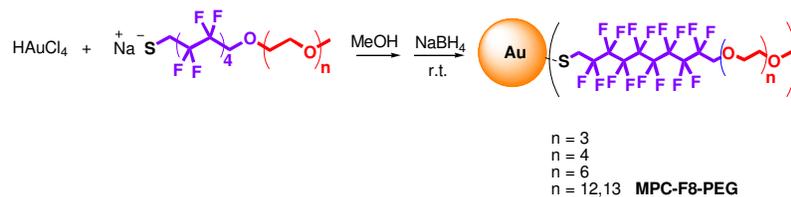
the solvophobic feature of the fluorinated bundles is the driving force for NP assembly

K. Niikura *et al.* *J. Am. Chem. Soc.* **2012**, *134*, 7632.

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## gold nanoparticles protected by amphiphilic fluorinated ligands

### Synthesis

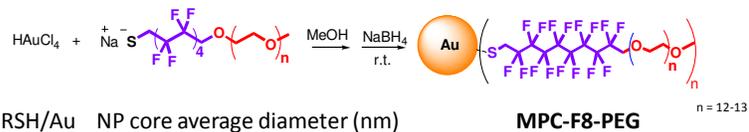


- RSH/Au molar ratio = 2
- reaction at r.t. to favor RSH solubility
- for  $n = 3, 4$  and  $6$  NPs precipitate in MeOH, they are scarcely soluble in DCM and chloroform

C. Gentilini, F. Evangelista, P. Rudolf, P. Franchi, M. Lucarini, L. Pasquato *J. Am. Chem. Soc.* **2008**, *130*, 15678-15682.

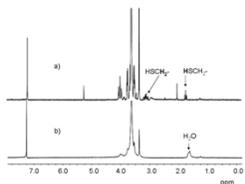
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## gold nanoparticles protected by amphiphilic fluorinated ligands

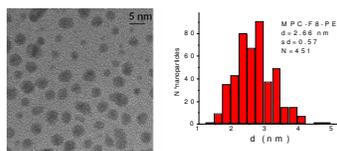


RSH/Au	NP core average diameter (nm)	MPC-F8-PEG
0.7	2.9	average composition: $\text{Au}_{670}(\text{SF8PEG})_{107}$
2	2.7	
2.5	1.6	

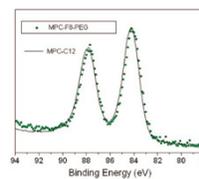
- ✓ NPs are soluble in water, methanol, DCM and chloroform



(a)  $^1\text{H}$  NMR of HS-F8-PEG and (b)  $^1\text{H}$  NMR of MPC-F8-PEG prepared with a thiol/Au ratio of 2:1.



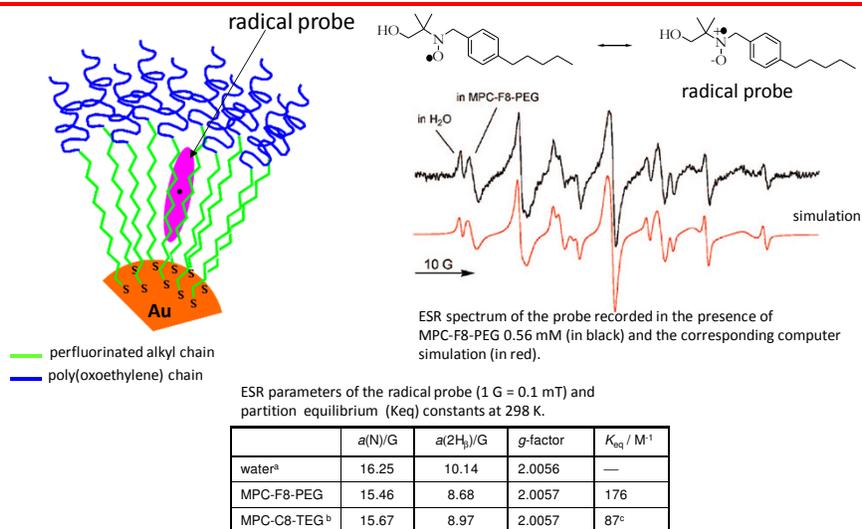
TEM image and histogram of MPC-F8-PEG.



XPS Au 4f core level spectra of MPC-F8-PEG and MPC-C12.

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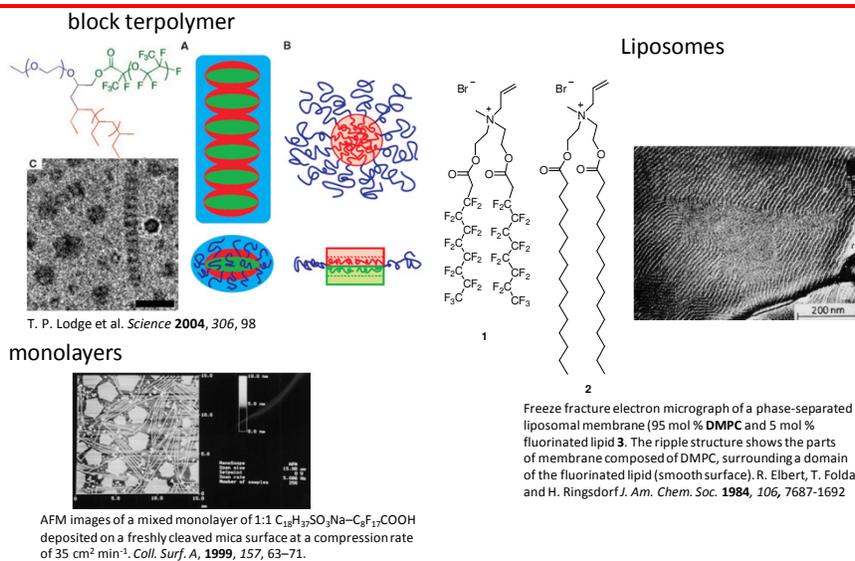
## gold nanoparticles protected by amphiphilic fluorinated ligands



C. Gentilini, F. Evangelista, P. Rudolf, P. Franchi, M. Lucarini, L. Pasquato *J. Am. Chem. Soc.* **2008**, *130*, 15678-15682.

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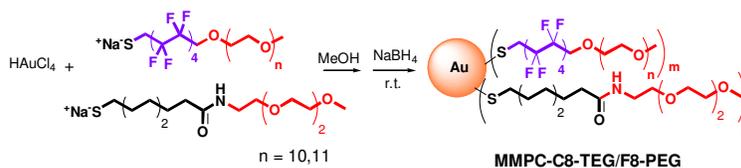
## phase segregation of hydrogenated/fluorinated units



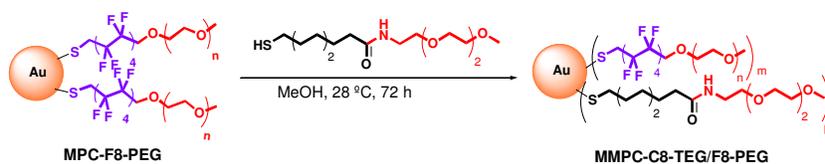
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## gold nanoparticles protected by H-/F- mixed-monolayers

- Homogeneous phase synthesis (methanol/water) using mixtures of thiolates with **immiscible chains**



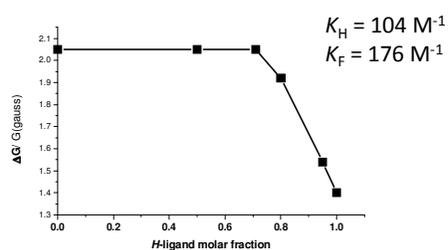
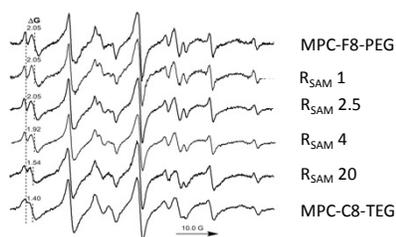
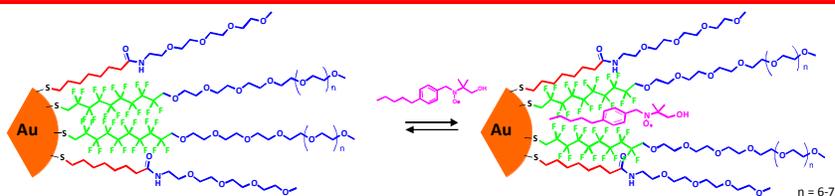
- synthesis of mixed-monolayer by exchange reaction



C. Gentilini, P. Franchi, E. Mileo, S. Polizzi, M. Lucarini, L. Pasquato *Angew. Chem. Int. Ed.* **2009**, *48*, 3060.

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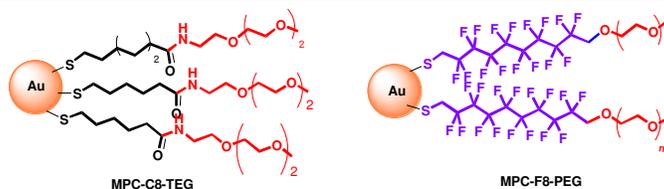
## gold nanoparticles protected by H-/F- mixed-monolayers



C. Gentilini, P. Franchi, E. Mileo, S. Polizzi, M. Lucarini, L. Pasquato *Angew. Chem. Int. Ed.* **2009**, *48*, 3060.

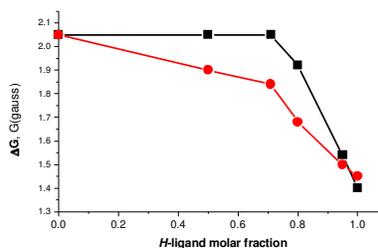
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## gold nanoparticles protected by H-/F- mixed-monolayers



### ESR Parameters in the Presence of Homoligand NP Mixtures

$\chi$	$\Delta G/G$	
0	2.05	2.05
0.5	2.05	1.90
0.71	2.05	1.84
0.80	1.92	1.68
0.95	1.54	1.50
1	1.40	1.45



P. Posocco, C. Gentilini, S. Bidoggia, A. Pace, P. Franchi, M. Lucarini, M. Fermeglia, S. Pricl, L. Pasquato, *ACS Nano* **2012**, *6*, 7243-7253.

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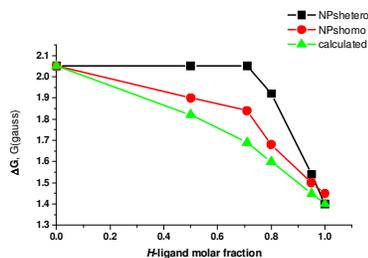
## gold nanoparticles protected by H-/F- mixed-monolayers

$$\Delta G = 1.40X_H + 2.05(1 - X_H)$$

$$X_H = \frac{K_H[H\text{-ligand}]}{K_F[F\text{-ligand}] + K_H[H\text{-ligand}]}$$

$K_H = 104 \text{ M}^{-1}$   
 $K_F = 176 \text{ M}^{-1}$

$\chi$	$\Delta G/G$		
	HeteroL	Homol	calcd.
0	2.05	2.05	2.05
0.5	2.05	1.90	1.82
0.71	2.05	1.84	1.69
0.80	1.92	1.68	1.60
0.95	1.54	1.50	1.45
1	1.40	1.45	1.40



L. Pasquato et al., *ACS Nano* **2012**, *6*, 7243-7253.

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## gold nanoparticles protected by H-/F- mixed-monolayers

### atomistic and mesoscale calculations

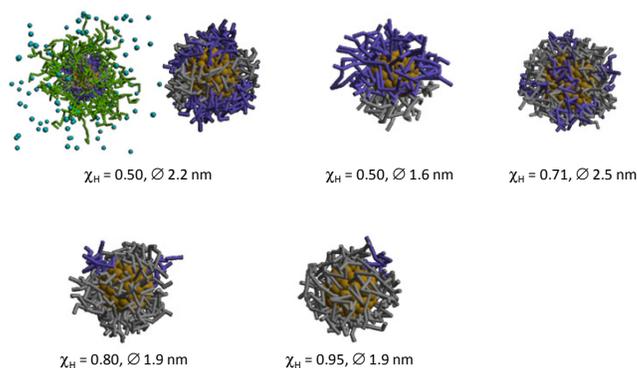
- Models of the NPs were obtained employing an innovative **multiscale molecular simulation** approach
- This approach combines **mesoscale techniques**, developed to describe systems with properties at the nanometer scale, and **atomistic techniques**, which model matter at the level of atoms
- In particular, morphological characterization of the NPs was predicted using mesoscale models (*Dissipative Particle Dynamics*), whose chemical potentials were derived from simulations at a lower level of detail (*Molecular Dynamics*)
- Simulation have been carried out in the presence of water as solvent, at 298 K, and considering the mobility of the thiolates on the gold surface

L. Pasquato et al., *ACS Nano* **2012**, 6, 7243-7253.

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## gold nanoparticles protected by H-/F- mixed-monolayers

Ligand organization on the surface of gold NPs at different molar fraction of the two ligands



P. Posocco, C. Gentilini, S. Bidoggia, A. Pace, P. Franchi, M. Lucarini, M. Fermeglia, S. Pricl, L. Pasquato, *ACS Nano* **2012**, 6, 7243-7253.

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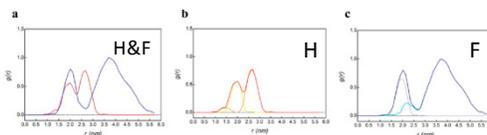
## gold nanoparticles protected by *H*-/*F*- mixed-monolayers

Equilibrium constants (T = 298 K) in the presence of Heteroligand mixed-monolayers

$\chi$	$K_{\text{mix}}/M^{-1}$	$K_H/K_F^c$	$K_F/M^{-1}$	$K_H/M^{-1}$
0	176 <sup>d</sup>		176 <sup>d</sup>	
0.50	100 ± 7.7 <sup>b</sup>	<0.04	200	<10
0.71	100 ± 8.1	<0.04	350	<10
0.80	189 ± 20	0.06	762 <sup>f</sup>	45 <sup>f</sup>
0.95	120 ± 11	0.16	600 <sup>f</sup>	96 <sup>f</sup>
1	104 <sup>e</sup>			104 <sup>e</sup>

**shell thickness** F8PEG 2.82 nm in homoligand NPs  
C8TEG 1.40 nm

F8PEG 2.60 nm in heteroligand NPs  
C8TEG 1.59 nm



Radial distribution functions (RDFs) for the SAM components of MPC-C8-TEG/F8-PEG, 1:1

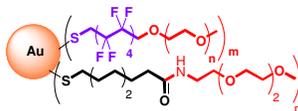
it is apparent that the small ligand chains somewhat stretch out toward the longer ones and these in turn bend over the smaller counterpart.

L. Pasquato et al., *ACS Nano* **2012**, *6*, 7243-7253.

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## gold nanoparticles protected by *H*-/*F*- mixed-monolayers

### conclusions



**Phase segregation** occurs for all molar fractions considered, even when about three **F**-chains (4%) are present in the monolayer.

The shape of the domains is triggered by:

- the **relative length of the ligands** as the consequence of the optimum balance between enthalpic and entropic force.
- the **size of the NP core** : below 2 nm Janus domains are formed, whereas stripes-like domains spontaneously form for larger NPs.
- the composition of the mixed-monolayer

mixtures of *H*-ligands and *F*-ligands of different length are ideal to control the morphology of mixed self-assembled monolayers in three dimensions.

mixed-monolayers present features that are different from those of the parent homoligand monolayers.

L. Pasquato et al., *ACS Nano* **2012**, *6*, 7243-7253.

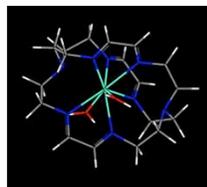
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## MAGNETIC RESONANCE IMAGING

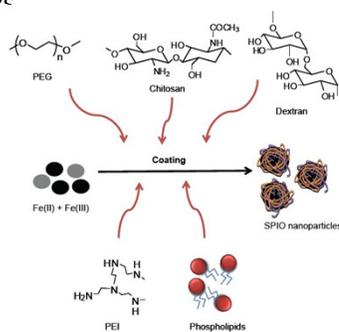
- MRI is a particular attractive imaging modality for clinical trials.
  - ✿ it provides high-resolution,
  - ✿ detailed structural images,
  - ✿ three-dimensional spatial reconstruction,
  - ✿ no ionizing radiation
- It can be used for:
  - ✿ disease detection
  - ✿ therapeutic monitoring
  - ✿ treatment efficacy
  - ✿ cell tracking

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- $^1\text{H}$  MRI
- Gadolinium-based contrast agents (GBCAs)



- SuperParamagnetic Iron Oxide (SPIO) NP<sup>c</sup>
- 50 – 180 nm
- USPIO, 10-50 nm
- VSPIO, < 10 nm
- 



Feridex, Resovist

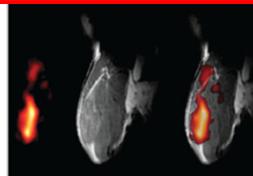
Clinically approved  
Rapidly removed

- $^{19}\text{F}$  MRI

NanoBioMed 2013, Trieste, 14th-18th October

**$^{19}\text{F}$  MRI**

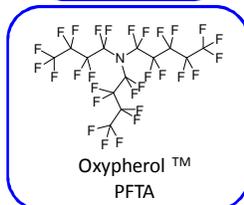
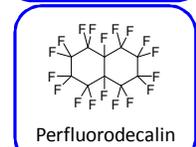
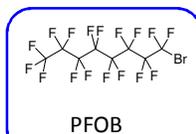
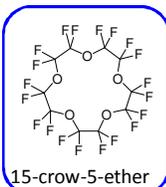
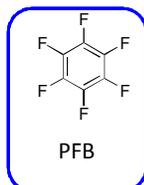
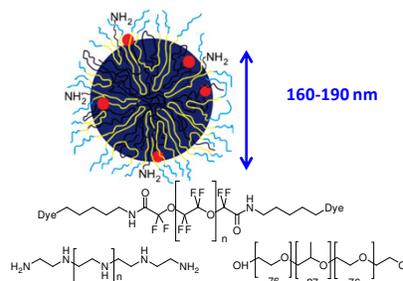
- high relative sensitivity (83% of  $^1\text{H}$ )
- 100% natural abundance of  $^{19}\text{F}$
- its resonance differs by only 6% from that of  $^1\text{H}$
- an extremely broad range (> 200 ppm) which permits unambiguous identification of distinctive  $^{19}\text{F}$ -containing compounds, but sensitive to the environment
- tissues have a negligible background  $^{19}\text{F}$  signal, quantitative analysis are allowed
- perfluorocarbons (PFCs) have been extensively studied for biomedical applications.
- a very high  $F$  concentration is needed



*Nat. Biotechnol.* **2005** 23,983.

Ruiz-Cabello, J.; Barnetta, B. P.; Bottomley, P. A.; Bulte, J. W. M. *NMR Biomed.* **2011**, *24*, 114–129.  
 Srinivas, M.; Heerschap, A.; Ahrens, E.T.; Figdor, C. G.; de Vries, I. J. M. *Trends Biotech.* **2010**, *28*, 363-370.

NanoBioMed 2013, Trieste, 14th-18th October

 **$^{19}\text{F}$  MRI****PFCs****Perfluoroether nanoemulsion****Dual  $^{19}\text{F}$  MRI-fluorescence detection**

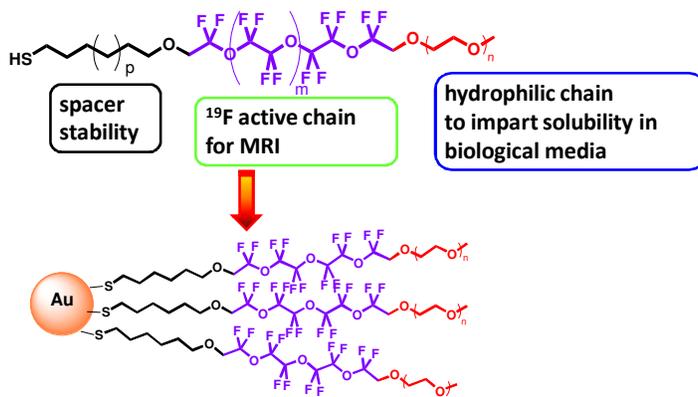
Ahrens, E. T.; Janjic, J. M.; Srinivas, M.; Kadayakkara, D. K. K. *J. Am. Chem. Soc.* **2008**, *130*, 2831-2841.

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## GNPs for $^{19}\text{F}$ MRI

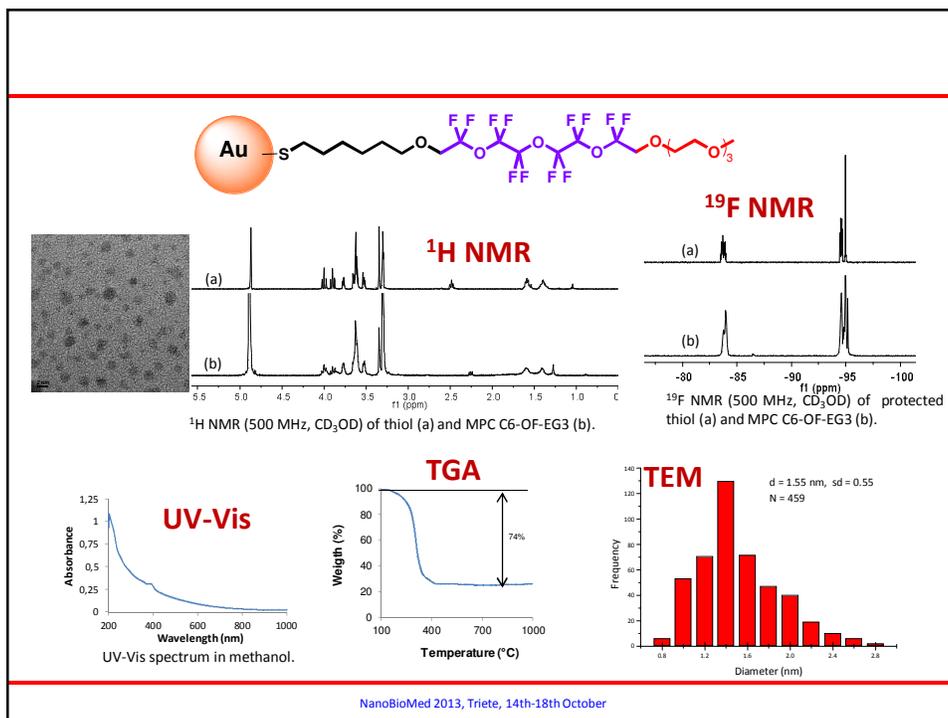
design, synthesis and use of gold NPs protected by fluorinated ligands as nanomaterial for imaging and therapy

**modular system**



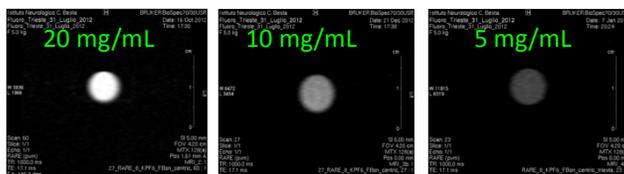
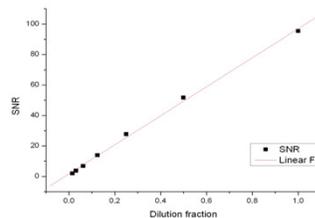
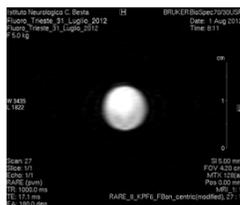
L. Pasquato et al. *Chem. Commun.* **2013**, 49, 8794.

NanoBioMed 2013, Trieste, 14th-18th October



## MPC-C6-FEO-EG3, <sup>19</sup>F MRI

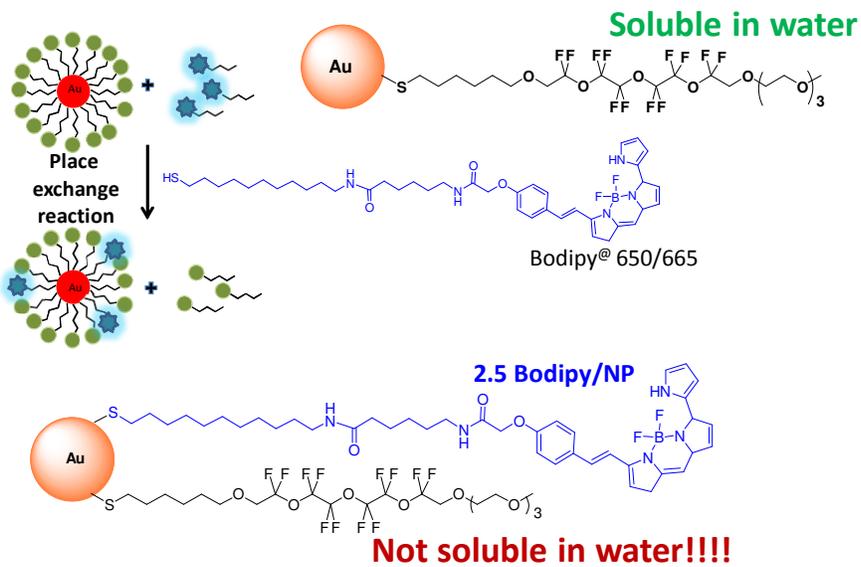
40 mg/mL of NP (1.75x10<sup>20</sup> <sup>19</sup>F/mL)



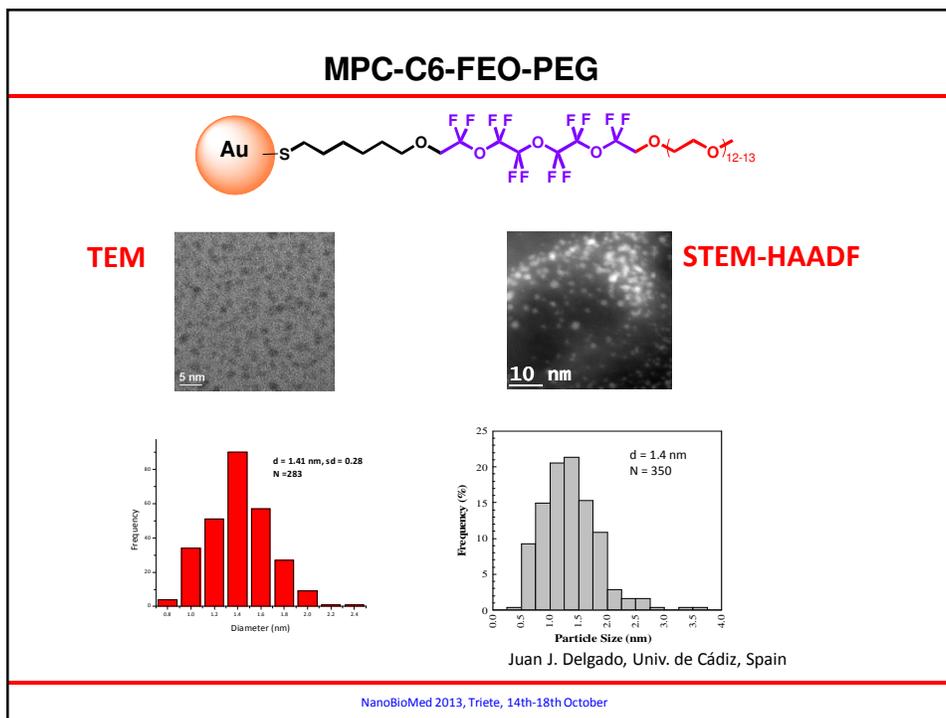
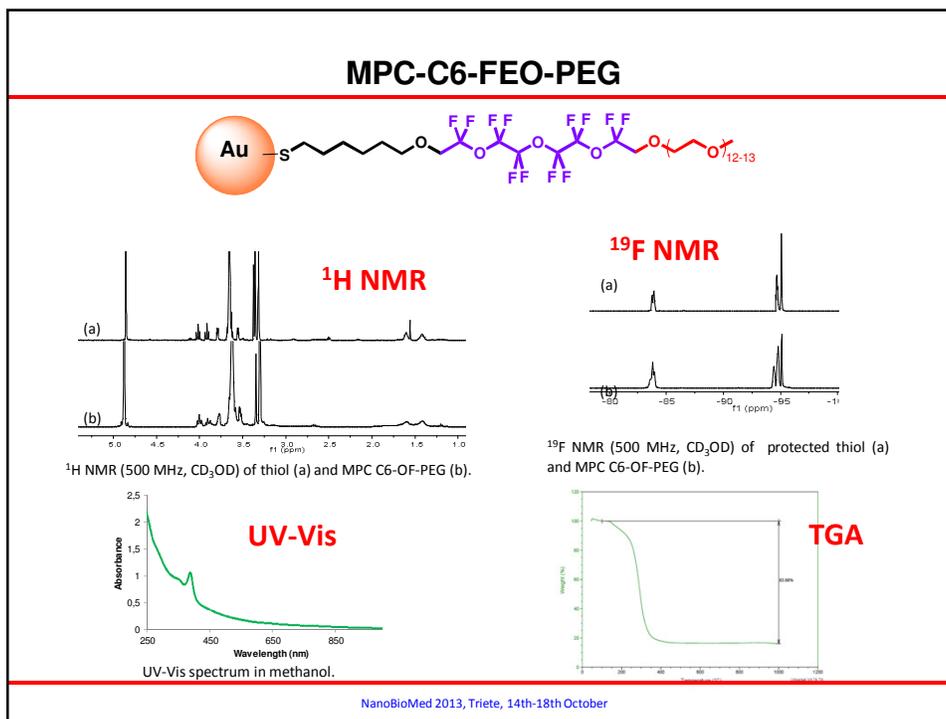
I. Zucca, Institute C. Besta, Milan

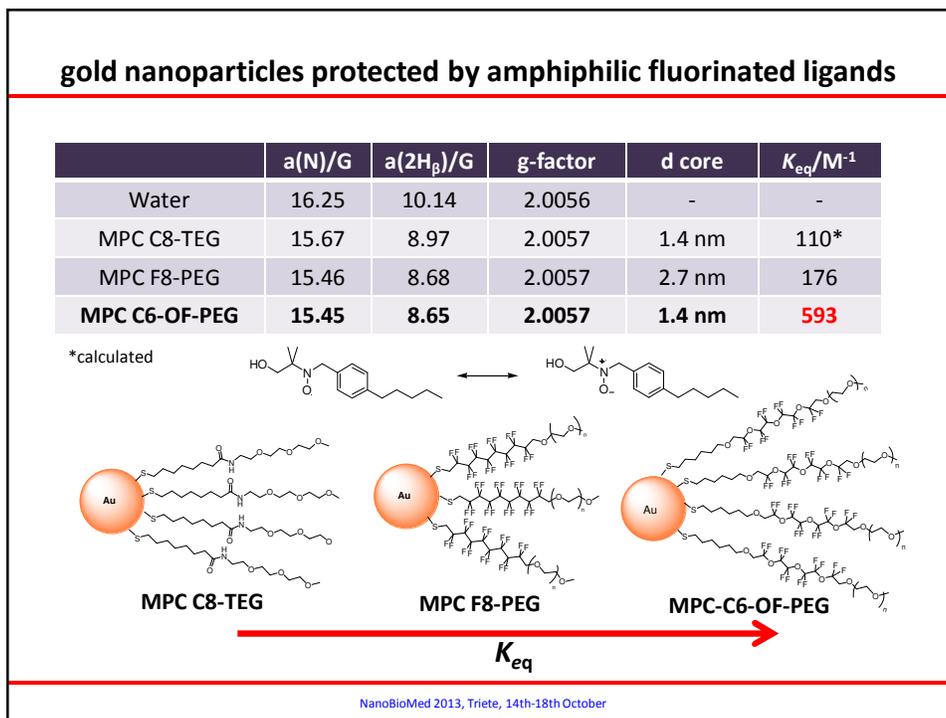
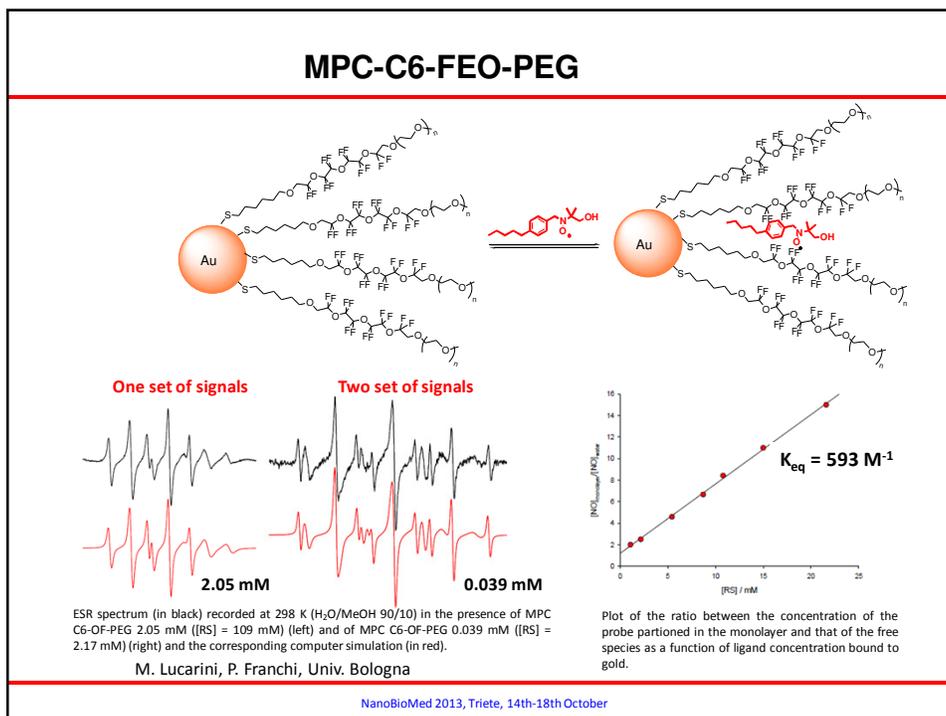
NanoBioMed 2013, Trieste, 14th-18th October

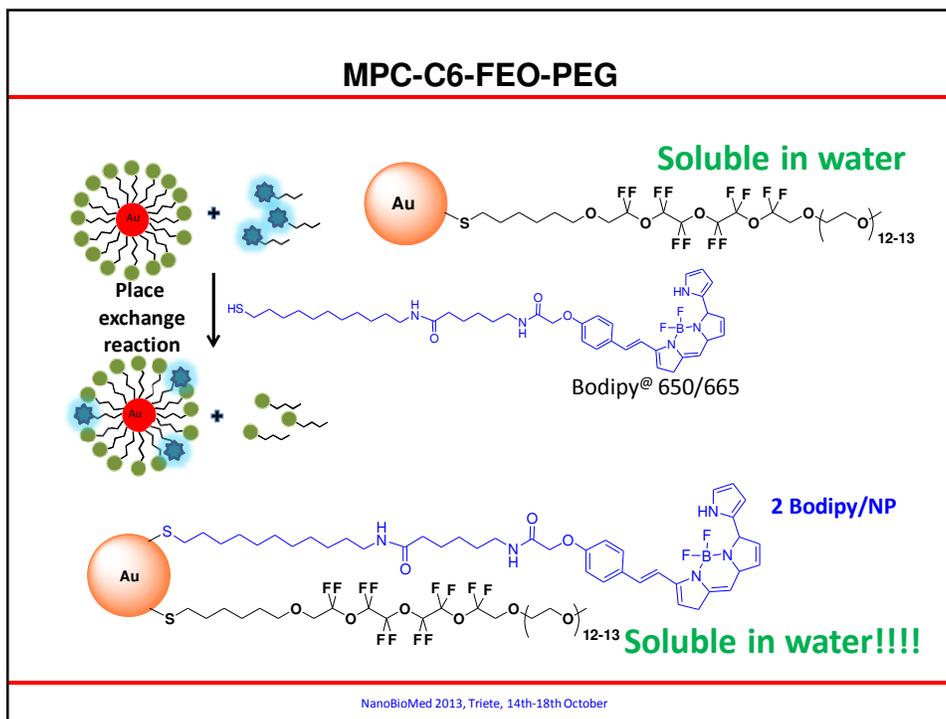
## MPC-C6-FEO-EG3



NanoBioMed 2013, Trieste, 14th-18th October







### MPC-C6-FEO-PEG, cellular uptake

**4 h incubation with HeLa cells at 37 °C, and 30 min RBC**

Confocal laser microscopy images of HeLa cells (nucleus stained in blue, Hoechst dye) loaded with F-MPCs **4b** (red fluorescent signal) for 4 h at 37 °C.

**No unbound Bodipy** was detected by RBC test.

**4 h incubation with HeLa cells at 4 °C** (endocytic/ pinocytic mechanisms are arrested)

**No visible red signal**, only very little is possible visualized with the enhanced signal.

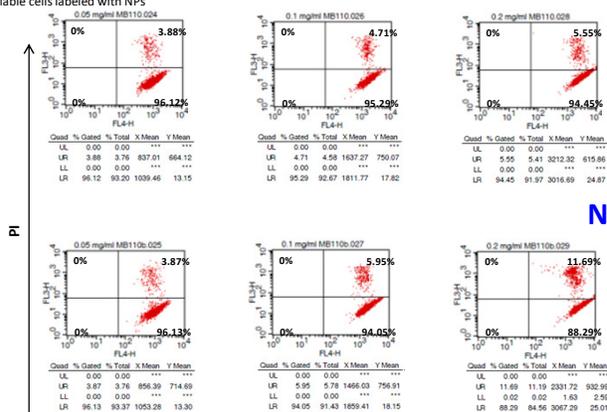
F. Sousa, IEO, Milan

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## MPC-C6-FEO-PEG, cellular viability

UL – dead cells not labeled with NPs  
 UR - dead cells labeled with NPs  
 LL – viable cells not labeled with NPs  
 LR- viable cells labeled with NPs

FACS for PI of HeLa incubated 4 hrs with MPC C6-FEO-PEG



No cytotoxicity

The percentage of viable cells is above 95% after taken up NPs. The percentage of dead cells labeled with NPs are very similar to all concentrations tested.

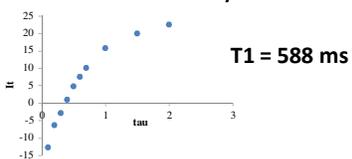
NanoBioMed 2013, Trieste, 14th-18th October

## GNPs for <sup>19</sup>F MRI

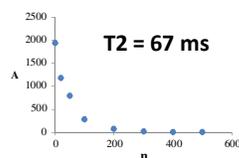
NMR/MRI

Relaxation times (<sup>19</sup>F-NMR)

T1 (spin-lattice relaxation time)  
 Inversion Recovery

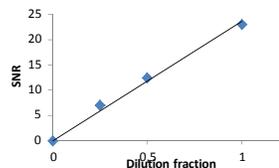
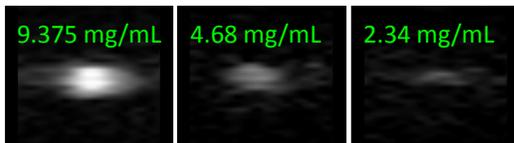


T2 (spin-spin relaxation time)  
 CPMG sequence  
 (Carr-Purcell Meiboom-Gill)



<sup>19</sup>F-MR images

18.75mg/mL of NPs (6.06x10<sup>19</sup> <sup>19</sup>F/mL)



L. Pasquato et al. *Chem. Commun.* 2013, 49, 8794.

NanoBioMed 2013, Trieste, 14th-18th October

## Concluding remarks

- ✓ Amphiphilic *F*-ligands have been used successfully in the passivation of gold NPs which become very stable and soluble in organic solvents and in water
- ✓ *F*-ligands when in the monolayer become interesting tools to introduce new features and the properties of *F* nucleous
- ✓ mixtures of *F*- and *H*-ligands enable to control the morphology of mixed-monolayers grafted on spherical gold surfaces. Moreover, they show phase segregation independently on the monolayer composition. This behaviour may favor multivalent recognition and cooperative catalysis processes
- ✓ The properties of mixed-monolayers are different from those of the parent homoligand monolayers
- ✓ Gold NPs coated with fluorinated ligands present features suitable for  $^{19}\text{F}$  MRI
- ✓ These monolayers may be used on NP of different materials

NanoBioMed 2013, Trieste, 14th-18th October

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NanoBioMed 2013, Trieste, 14th-18th October