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Ion Beam Analysis Techniques for non-Destructive Profiling Studies

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What is IBA?

Ion Beam Analysis (IBA) is based on the interaction, at both the atomic and the nuclear level, between <u>accelerated charged particles</u> and the bombarded material. <u>When a charged particle moving at high speed strikes a material, it</u> <u>interacts with the electrons and nuclei of the material atoms, slows down and</u> <u>possibly deviates from its initial trajectory</u>. This can lead to the emission of particles or radiation whose energy is characteristic of the elements which constitute the sample material.









IBA METHODS:

Method	Acronym	Interaction	
Particle-Induced X-ray Emission	PIXE	Characteristic X-ray emission following ionization by the primary beam	
Rutherford Backscattering Spectrometry	RBS	Elastic scattering at backward angles	
Elastic or Nuclear (non- Rutherford) Backscattering Spectrometry	EBS	Elastic scattering at backward angles	
Elastic Recoil Detection Analysis	ERDA	Elastic recoil at forward angles, not necessarily Rutherford	
Nuclear Reaction Analysis	NRA	Nuclear reaction between incident beam and nuclei in the target, producing a light charged particle	
Particle Induced Gamma – ray Emission	PIGE	Prompt γ-ray emission during ion beam irradiation	







Summary of interactions of accelerated ion with atomic nucleus:

- Elastic scattering (Coulomb and nuclear)
- Inelastic scattering (residual nucleus is excited)
- \bullet Nuclear reactions with emission of particles and $\gamma\text{-rays}$

Positive Common Characteristics of IBA techniques:

> They are generally not destructive (least) and are thus suitable for use with delicate materials.

> They are to a certain extent multielementary and produce high-accuracy quantitative results.

> They require little or no preparation of the sample with the result that a specimen (like an artifact) could be directly analyzed.

> Only very small quantities (mg) of sample are needed.

> They permit the analysis of a very small portion of the sample by reducing the diameter of the ion beam to less than 0.5 mm.







Negative Common Characteristics:

- > Some damage cannot be avoided (thermal, carbon buildup etc.)!
- > A VdG type of accelerator is required.
- > In most of the cases the experiments are carried out in vacuum chambers.
- > Several experimental issues need to be addressed, thus a minimum knowledge of nuclear physics (experimental and theoretical) is mandatory.
- > No <u>direct</u> information about the chemical environment can be produced.
- > Issues like dating and authenticity testing can be addressed only indirectly.
- > Unlike NAA, the IBA analysis concerns only a few microns below the surface of the samples.
- > In most of the cases, a combination of techniques is required to solve a problem, and this implies time consuming experiments!

THUS, DO WE REALLY NEED IBA?

<u>YES</u>, because it can solve problems that cannot be addressed by all the other existing techniques!







IBA PROFILING TECHNIQUES:

• Rutherford Backscattering Spectroscopy / • Charged Particle Activation Nuclear Backscattering Spectroscopy (RBS / EBS) / Channeling

• Elastic Recoil Detection Analysis (ERDA)

• Nuclear Reaction Analysis (particle-particle and particle-gamma reactions)

NON-PROFILING:

Analysis (CPAA), Particle Induced X-Ray Emission (PIXE)

 Neutron Activation Analysis (NAA), Secondary Ion Mass Spectroscopy (SIMS)

SPECIFIC REQUIREMENTS FOR ALL IBA TECHNIQUES:

- Electrostatic accelerator (mainly VdG, single-ended or tandem)
- Scattering chamber (vacuum or in-air)
- Electronics
- Software for acquisition and spectral analysis







AMAKP



IAEA TRAINING COURSE



ION SOURCES (ESPECIALLY DESIGNED TO PRODUCE NEGATIVE IONS):



Duoplasmatron (for gaseous materials)

Duoplasmatron





Cs - sputter (for solid materials)



















Ernest Rutherford (1871 – 1937) >A great talent from New Zealand. Maybe the greatest experimental physicist of his era after Faraday.

>1899 – Discovery and study of α and β radiation. Distinction on the basis of penetrability and measurement of charge. Professor in McGill, Canada till 1907. He named ' γ -rays'. He discovered the law of radioactivity and the existence of nuclear reactions. He measured the age of Earth along with Soddy and Hahn.

>Professor in Manchester till 1919. Famous 'gold foil' experiment with Geiger and Marsden. First nuclear model suggestion. First study of a nuclear reaction: $^{14}N + \alpha -> ^{17}O + p$

➢ Professor at Cavendish, Cambridge in the place of J.J. Thomson till his death. Collaboration with N. Bohr in the new 'atomic model'. Theoretical prediction for the existence of neutron. Brilliant team of students and collaborators: Cockroft, Walton, Chadwick, Wilson, Appleton.

>Nobel prize in Chemistry (!), 1908. In his speech he said: 'It was the quickest transition in my life'!



E. RUTHERFORD: 'It was quite the most incredible event that has ever happened to me in my life. It was almost as incredible as if you fired a 15-inch shell at a piece of tissue paper and it came back and hit you. On consideration, I realized that this scattering backward must be the result of a single collision, and when I made calculations I saw that it was impossible to get anything of that order of magnitude unless you took a system in which the greater part of the mass of the atom was concentrated in a minute nucleus. It was then that I had the idea of an atom with a minute massive centre, carrying a charge.'





$a + X \longrightarrow b + Y$

Cross Sections	Symbol	Technique	Possible Application
Total	σ_t	Attenuation of beam	Shielding
Reaction	σ	Integrate over all angles and all energies of b (all excited states of Y)	Production of radioiso- tope Y in a nuclear reaction
Differential (Angular)	$d\sigma/d\Omega$	Observe b at (θ, ϕ) but inte- grate over all energies	Formation of beam of b particles in a certain direction (or recoil of Y in a certain direc- tion)
Differential (Energy)	dσ/dE	Don't observe b, but observe excitation of Y by subsequent γ emission	Study of decay of excited states of Y
Doubly differential	$d^2\sigma/dE_{ m b}d\Omega$	Observe b at (θ, ϕ) at a specific energy	Information on excited states of Y by angular distribution of b

 $df = nx(2\pi b\,db)$

(df = fraction of the incident particles that pass through the annular ring)

The relative yield differs between high – and low-Z nuclei by almost two orders of magnitude, thus RBS is ideal for heavy elements on light substrates.

Actual cross-sections deviate from Rutherford at low energies for all projectiletarget pairs. This is caused by partial screening of the nuclear charges by the electron shells surrounding both nuclei. This screening is taken into account by a correction factor F: $\sigma = F \sigma_R$

The kinematic factor varies dramatically with the ion beam mass and approaches unity for protons impinging on heavy elements. It depends only on the mass ratio M_1/M_2 and on the scattering angle θ .

To obtain good mass resolution, the coefficient of ΔM_2 has to be as large as possible. To accomplish this one can: \checkmark Increase E_0

- ✓ Increase M₁
- \checkmark Set θ very close to 180°

Fig. 4. Proton backscattering spectrum $E_p = 1.5$ MeV, $\theta = 160^{\circ}$) of sample S1. The solid line represents the RUMP simulation.

RBS/EBS study of the roof at the Municipality of Stockholm, using **1.5 MeV protons – rough samples,** complicated matrix!

Special NRA Characteristics:

POSITIVE

- ✓ High isotopic selectivity
- ✓ The matrix is not so important
- ✓ Clear isolated peaks with practically no background

✓ If the deuteron beam is adopted, one can achieve simultaneous analysis of most of the main light elements (C, O, N, F, B, Li)

NEGATIVE

Not many cross sections available in literature / theoretical evaluation pending

- Usually time-consuming studies
- Not all the elements present low enough MDLs
- Radiation safety is sometimes an issue

Typical sensitivities: 1:10⁴ in atomic proportion

Nuclear Reaction Analysis (particle-gamma) or PIGE

 > Use of a γ-ray detector (preferably HPGe)

Use of standards in identical conditions:

 $\boldsymbol{C}_{i,s} = (\boldsymbol{C}_{st} \; \boldsymbol{S}_{s} \; \boldsymbol{Y}_{i,s}) / (\boldsymbol{S}_{st} \; \boldsymbol{Y}_{i,st})$

> Dealing with the problem of different stopping powers between standard and target

Lack of generally accepted computer simulation code

Lack of accurate/evaluated cross section data for γinducing nuclear reactions in literature

ANN TO

Hydrogen profiling indirectly addresses the dating problem, therefore PIGE information is unique (pre-Colombian quartz artifacts measured at AGLAE, Louvre)!

Extra NRA (particle-gamma) Characteristics:

POSITIVE

- ✓ Excellent for hydrogen profiling, fluorine and aluminum (MDL~1-10 ppm)
- ✓ Quite satisfactory for carbon, nitrogen, oxygen, magnesium, silicon
- ✓ Relatively easy to implement, due to the standards used
- ✓ Excellent for machine calibration (accelerator tuning)!

NEGATIVE

- HPGe detectors are very expensive, fragile, and need liquid nitrogen cooling!
- Usually <u>very</u> time-consuming studies
- Very element-specific technique
- Overlap of resonances might occur if the sample is relatively thick
- C Efficiency calibration of the HPGe detector is a requirement!

Ion Beam Analysis Profiling Techniques: Summary, Present Situation, Future Perspectives

<u>SUMMARY</u>

- 1. Rutherford backscattering (RBS) is ideal for depth-profiling of heavy elements on lighter substrates.
- 2. Elastic recoil detection analysis (ERDA) is excellent for depth-profiling of very light elements in thin films (for very small depths < 1 μ m).
- 3. Nuclear reaction analysis (NRA), is excellent for high resolution depthprofiling of specific isotopes.

PRESENT SITUATION

- 1. A lot of work is being done in channeling, EBS and NRA (cross section measurements and evaluations).
- 2. Micro-beams and measurements in air (Louvre) have enhanced IBA capabilities.

FUTURE PERSPECTIVES

- 1. New techniques are always evolving (e.g. HR-RBS, TOF-ERDA).
- 2. Channeling analytical algorithms?
- 3. CAN WE SOLVE ALL THE PROBLEMS??? NO (BUT MANY YES...)